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Li et al.(10) **Pub. No.: US 2018/0226592 A1**(43) **Pub. Date: Aug. 9, 2018**(54) **METAL COMPOUNDS, METHODS, AND
USES THEREOF****Publication Classification**(71) Applicant: **Arizona Board of Regents on behalf
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Hangzhou Zhejiang (CN)(21) Appl. No.: **15/882,358**(22) Filed: **Jan. 29, 2018****Related U.S. Application Data**(63) Continuation of application No. 14/430,454, filed on
Mar. 23, 2015, now Pat. No. 9,882,150, filed as
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2013.(60) Provisional application No. 61/704,880, filed on Sep.
24, 2012.(51) **Int. Cl.****H01L 51/00** (2006.01)**C09K 11/06** (2006.01)**C07F 15/00** (2006.01)**H01L 51/50** (2006.01)**H01L 51/42** (2006.01)(52) **U.S. Cl.**CPC .. **H01L 51/0087** (2013.01); **C09K 2211/1059**
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(2013.01); **C09K 2211/1029** (2013.01); **C09K**
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(57)

ABSTRACTDisclosed herein are metal compounds useful in devices,
such as, for example, OLEDs.

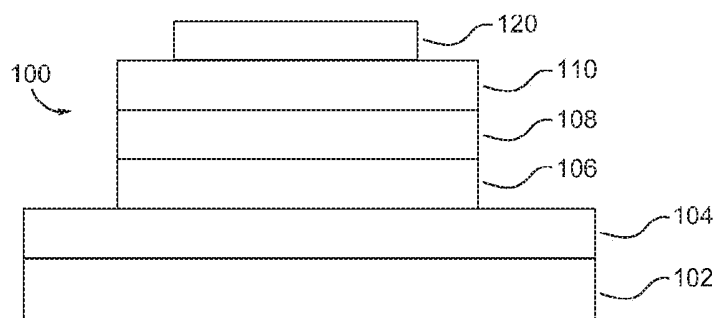


FIG. 1

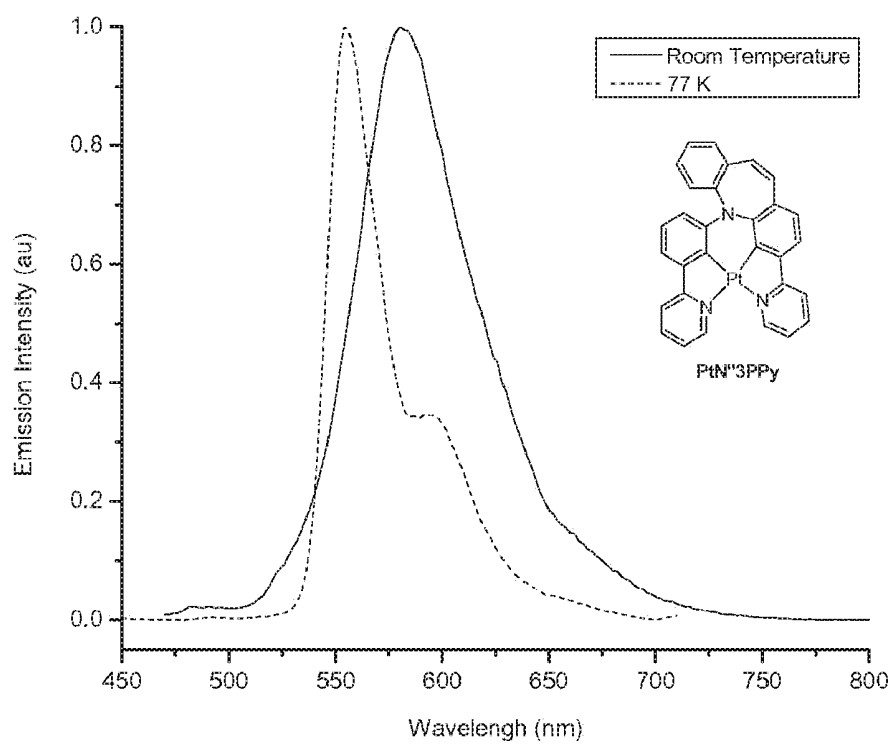


FIG. 2

METAL COMPOUNDS, METHODS, AND USES THEREOF

BACKGROUND

Technical Field

[0001] The present disclosure relates to metal compounds that are useful in devices, such as, for example, organic light emitting diodes (OLEDs).

Technical Background

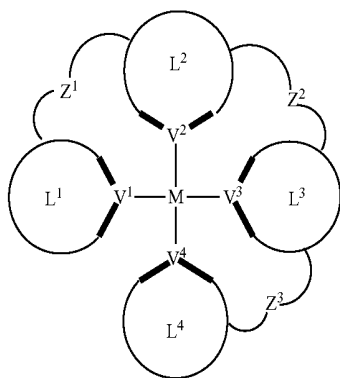
[0002] Compounds capable of absorbing and/or emitting light can be ideally suited for use in a wide variety of optical and electro-optical devices, including, for example, photo-absorbing devices such as solar- and photo-sensitive devices, photo-emitting devices, OLEDs, or devices capable of both photo-absorption and emission. Much research has been devoted to the discovery and optimization of organic and organometallic materials for using in optical and electro-optical devices. Generally, research in this area aims to accomplish a number of goals, including improvements in absorption and emission efficiency, as well as improvements in processing ability.

[0003] Despite significant advances in research devoted to optical and electro-optical materials, many currently available materials exhibit a number of disadvantages, including poor processing ability, inefficient emission or absorption, and less than ideal stability, among others. Thus, a need exists for new materials which exhibit improved performance in optical and electro-optical devices. This need and other needs are satisfied by the present invention.

SUMMARY

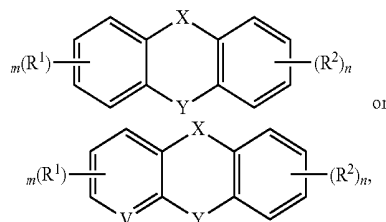
[0004] Disclosed herein are metal compounds that are useful in devices, such as, for example, organic light emitting diodes (OLEDs).

[0005] Disclosed herein is a compound having the formula

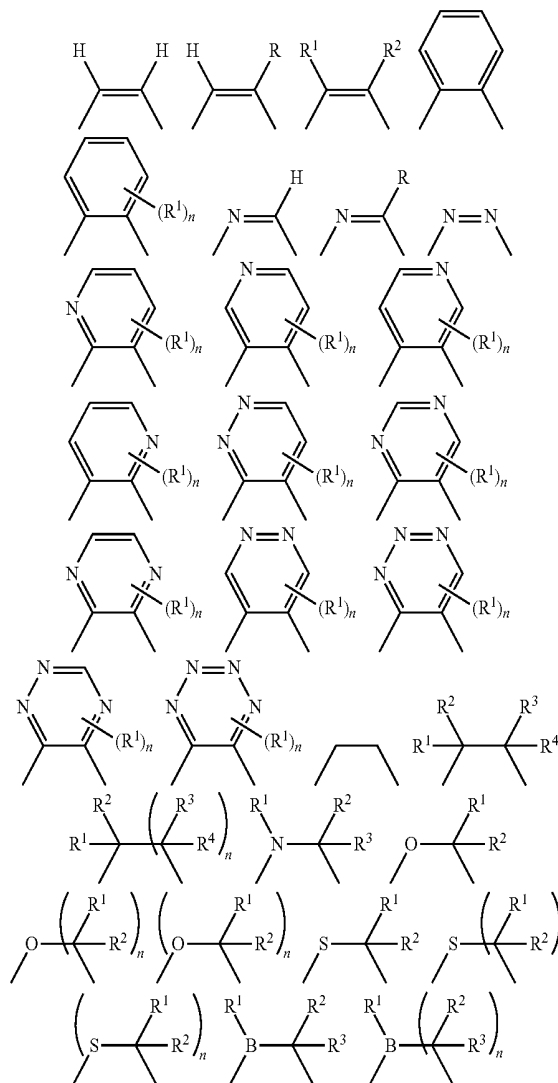


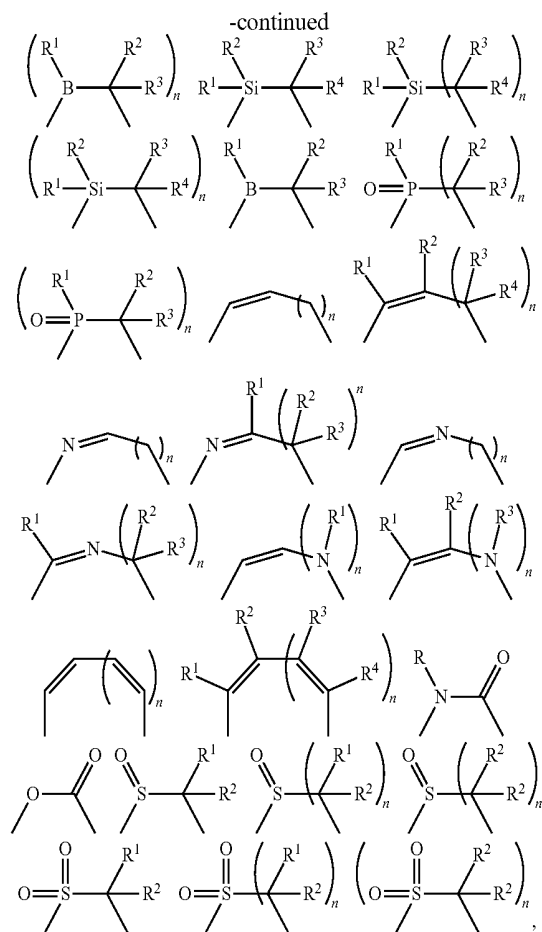
wherein each of L^1 , L^2 , L^3 , and L^4 independently is a substituted or unsubstituted aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, carbene, or N-heterocyclic carbene, wherein each of Z^1 , Z^2 , and Z^3 independently is present or absent, and if present each of Z^1 , Z^2 , and Z^3 independently is A, A^1 , or A^2 ,

wherein each of A, A^1 , or A^2 independently is O, S, S=O, SO₂, Se, NR³, PR³, RP=O, CR¹R², C=O, SiR¹R², GeR¹R², BH, P(O)H, PH, NH, CR¹H, CH₂, SiH₂, SiHR¹, BH, or BR³, wherein M is Pt, Pd, Au, Ir, Rh, Ru, Fe, Co, Ni, Cu, Zn, Ag, Hg, Cd, or Zr, wherein at least one of L^1 , L^2 , L^3 , and L^4 comprises



wherein m and n independently is an integer from 0 to 4, wherein each of V, X, and Y independently is V¹, V², V³, V⁴, O, S, S=O, SO₂, Se, NR³, PR³, R¹P=O, CR¹R², C=O, SiR¹R², GeR¹R², BH, P(O)H, PH, NH, CR¹H, CH₂, SiH₂, SiHR¹, BH, or BR³, or any one of





wherein at least one of V, X, and Y is V¹, V², V³, or V⁴, wherein m and n independently is an integer from 0 to 4, wherein each of V¹, V², V³ and V⁴ independently is coordinated to M, and wherein each of V¹, V², V³ and V⁴ independently is N, C, CH, P, B, SiH, or Si,

wherein each of R, R¹, R², R³, and R⁴, independently is hydrogen, aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, alkyl, alkenyl, alkynyl, deuterium, halogen, hydroxyl, thiol, nitro, cyano, amino, a mono- or di-alkylamino, a mono- or diaryl amino, alkoxy, aryloxy, haloalkyl, aralkyl, ester, nitrile, isonitrile, alkoxy carbonyl, acylamino, alkoxy carbonylamino, aryloxy carbonylamino, sulfonylamino, sulfamoyl, carbamoyl, alkylthio, sulfinyl, ureido, phosphoramidate, mercapto, sulfo, carboxyl, hydrazino, substituted silyl, or polymerizable, or any conjugate or combination thereof,

wherein each of V¹, V², V³ and V⁴ independently optionally is substituted for any one of V¹, V², V³ and V⁴.

[0006] Also disclosed herein are devices, such as, for example, OLEDs, comprising one or more of the disclosed compounds.

BRIEF DESCRIPTION OF THE FIGURES

[0007] The accompanying figures, which are incorporated in and constitute a part of this specification, illustrate several aspects and together with the description serve to explain the principles of the invention.

[0008] FIG. 1 shows a drawing of a cross-section of an exemplary organic light-emitting diode (OLED).

[0009] FIG. 2 is a emission spectra of PtN³PPy in CH₂Cl₂ at room temperature and in 2-MeTHF at 77K.

[0010] Additional aspects of the invention will be set forth in part in the description which follows, and in part will be obvious from the description, or can be learned by practice of the invention. The advantages of the invention will be realized and attained by means of the elements and combinations particularly pointed out in the appended claims. It is to be understood that both the foregoing general description and the following detailed description are exemplary and explanatory only and are not restrictive of the invention, as claimed.

DESCRIPTION

[0011] The present invention can be understood more readily by reference to the following detailed description of the invention and the Examples included therein.

[0012] Before the present compounds, devices, and/or methods are disclosed and described, it is to be understood that they are not limited to specific synthetic methods unless otherwise specified, or to particular reagents unless otherwise specified, as such can, of course, vary. It is also to be understood that the terminology used herein is for the purpose of describing particular aspects only and is not intended to be limiting. Although any methods and materials similar or equivalent to those described herein can be used in the practice or testing of the present invention, example methods and materials are now described.

1. Definitions

[0013] As used in the specification and the appended claims, the singular forms “a,” “an” and “the” include plural referents unless the context clearly dictates otherwise. Thus, for example, reference to “a component” includes mixtures of two or more components.

[0014] Ranges can be expressed herein as from “about” one particular value, and/or to “about” another particular value. When such a range is expressed, another aspect includes from the one particular value and/or to the other particular value. Similarly, when values are expressed as approximations, by use of the antecedent “about,” it will be understood that the particular value forms another aspect. It will be further understood that the endpoints of each of the ranges are significant both in relation to the other endpoint, and independently of the other endpoint. It is also understood that there are a number of values disclosed herein, and that each value is also herein disclosed as “about” that particular value in addition to the value itself. For example, if the value “10” is disclosed, then “about 10” is also disclosed. It is also understood that each unit between two particular units are also disclosed. For example, if 10 and 15 are disclosed, then 11, 12, 13, and 14 are also disclosed.

[0015] As used herein, the terms “optional” or “optionally” means that the subsequently described event or circumstance can or cannot occur, and that the description includes instances where said event or circumstance occurs and instances where it does not.

[0016] Disclosed are the components to be used to prepare the compositions of the invention as well as the compositions themselves to be used within the methods disclosed herein. These and other materials are disclosed herein, and it is understood that when combinations, subsets, interactions, groups, etc. of these materials are disclosed that while

specific reference of each various individual and collective combinations and permutation of these compounds cannot be explicitly disclosed, each is specifically contemplated and described herein. For example, if a particular compound is disclosed and discussed and a number of modifications that can be made to a number of molecules including the compounds are discussed, specifically contemplated is each and every combination and permutation of the compound and the modifications that are possible unless specifically indicated to the contrary. Thus, if a class of molecules A, B, and C are disclosed as well as a class of molecules D, E, and F and an example of a combination molecule, A-D is disclosed, then even if each is not individually recited each is individually and collectively contemplated meaning combinations, A-E, A-F, B-D, B-E, B-F, C-D, C-E, and C-F are considered disclosed. Likewise, any subset or combination of these is also disclosed. Thus, for example, the sub-group of A-E, B-F, and C-E would be considered disclosed. This concept applies to all aspects of this application including, but not limited to, steps in methods of making and using the compositions of the invention. Thus, if there are a variety of additional steps that can be performed it is understood that each of these additional steps can be performed with any specific embodiment or combination of embodiments of the methods of the invention.

[0017] As referred to herein, a linking atom connect two groups such as, for example, a N and C group. The linking atom can, if valency permits, have other chemical moieties attached. For example, an oxygen would not have any other chemical groups attached as the valency is satisfied once it is bonded to the two groups (N and/or C groups). In another example, when carbon is the linking atom, two additional chemical moieties would be attached to the carbon as valency would require such. Suitable chemical moieties includes, but are not limited to, hydrogen, hydroxyl, alkyl, alkoxy, =O, halogen, nitro, amine, amide, thiol, aryl, heteroaryl, cycloalkyl, and heterocyclyl.

[0018] The term “alkyl” as used herein is a branched or unbranched saturated hydrocarbon group of 1 to 24 carbon atoms, such as methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, s-butyl, t-butyl, n-pentyl, isopentyl, s-pentyl, neopentyl, hexyl, heptyl, octyl, nonyl, decyl, dodecyl, tetradecyl, hexadecyl, eicosyl, tetracosyl, and the like. The alkyl group can be cyclic or acyclic. The alkyl group can be branched or unbranched. The alkyl group can also be substituted or unsubstituted. For example, the alkyl group can be substituted with one or more groups including, but not limited to, optionally substituted alkyl, cycloalkyl, alkoxy, amino, ether, halide, hydroxy, nitro, silyl, sulfo-oxo, or thiol, as described herein. A “lower alkyl” group is an alkyl group containing from one to six (e.g., from one to four) carbon atoms.

[0019] The terms “amine” or “amino” as used herein are represented by the formula $NA^1A^2A^3$, where A^1 , A^2 , and A^3 can be, independently, hydrogen or optionally substituted alkyl, cycloalkyl, alkenyl, cycloalkenyl, alkynyl, cycloalkynyl, aryl, or heteroaryl group as described herein.

[0020] The term “halide” as used herein refers to the halogens fluorine, chlorine, bromine, and iodine.

[0021] The term “hydroxyl” as used herein is represented by the formula —OH.

[0022] The term “nitro” as used herein is represented by the formula —NO₂.

[0023] The term “nitrile” as used herein is represented by the formula —CN.

[0024] The term “thiol” as used herein is represented by the formula —SH.

[0025] The term “heterocyclyl” or the like terms refer to cyclic structures including a heteroatom. Thus, “heterocyclyl” includes both aromatic and non-aromatic ring structures with one or more heteroatoms. Non-limiting examples of heterocyclic includes, pyridine, isoquinoline, methylpyrrole and thiophene etc. “Heteroaryl” specifically denotes an aromatic cyclic structure including a heteroatom.

[0026] As briefly described above, the present invention is directed to multidentate, for example, tetradentate cyclometalated metal complexes. In one aspect, such complexes can be incorporated with heptacyclic or higher order materials. In another aspect, such complexes can be useful, for example, in displays and lighting applications.

[0027] In various aspects, the compounds of the present disclosure can comprise one or more of platinum (Pt) complexes, palladium (Pd) complexes, gold (Au) complexes, iridium (Ir) complexes, rhodium (Rh) complexes, ruthenium (Ru) complexes, iron (Fe) complexes, cobalt (Co) complexes, nickel (Ni) complexes, copper (Cu) complexes, zinc (Zn) complexes, silver (Ag) complexes, mercury (Hg) complexes, cadmium (Cd) complexes, zirconium (Zr) complexes, or other metal complexes not specifically recited herein which are capable of emitting light and are thus useful as an emissive materials in devices.

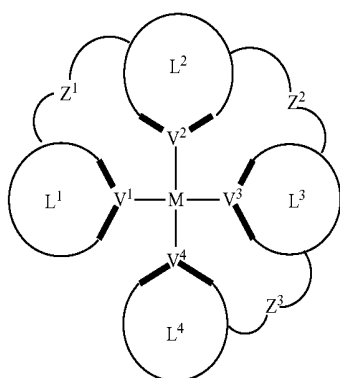
[0028] The term “each of V¹, V², V³ and V⁴ independently optionally is substituted for any one of V¹, V², V³ and V⁴” as used herein means that any one of V¹, V², V³, and V⁴ can be substituted for another of V¹, V², V³, and V⁴. For example, V¹ can be substituted for V², V³, or V⁴. In another example, V³ can be substituted for V¹, V², or V⁴. In one example, V² can be substituted for V¹. In another example, V³ can be substituted for V¹. In another example, V⁴ can be substituted for V². In another example, V³ can be substituted for V². In one aspect, V, as described herein, can be V¹, V², V³, or V⁴ as described herein. For example, V can be V¹. In another example, V can be V². In another example, V can be V³. In another example, V can be V⁴.

[0029] As used herein, the terms “compound” and “complex” are used interchangeably.

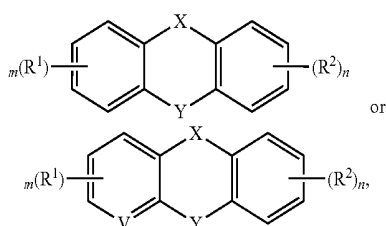
2. Compounds

[0030] In one aspect, disclosed herein are compounds comprising a metal, such as platinum (Pt), palladium (Pd), gold (Au), iridium (Ir), rhodium (Rh), ruthenium (Ru), iron (Fe), cobalt (Co), nickel (Ni), copper (Cu), zinc (Zn), silver (Ag), mercury (Hg), cadmium (Cd), or zirconium (Zr). The disclosed compounds can emit electromagnetic radiation. In another aspect, the emission of the disclosed compounds can be tuned, for example, from the ultraviolet to near-infrared, by, for example, modifying the ligand structure. In another aspect, the disclosed compounds can provide emission over a majority of the visible spectrum. In a specific example, the disclosed compounds can emit light over a range of from about 400 nm to about 700 nm. In another aspect, the disclosed compounds have improved stability and efficiency over traditional emission complexes. In yet another aspect, the disclosed compounds can be useful as luminescent labels in, for example, bio-applications, anti-cancer agents, emitters in organic light emitting diodes (OLED), or a combination thereof. In another aspect, the disclosed compounds

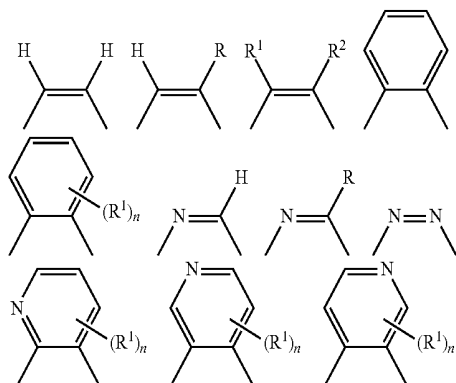
[0032] Disclosed herein are compounds having the formula



wherein each of A, A¹, or A² independently is O, S, S=O, SO₂, Se, NR³, PR³, RP=O, CR¹R², C=O, SiR¹R², GeR¹R², BH, P(O)H, PH, NH, CR¹H, CH₂, SiH₂, SiHR¹, BH, or BR³, wherein M is Pt, Pd, Au, Ir, Rh, Ru, Fe, Co, Ni, Cu, Zn, Ag, Hg, Cd, or Zr.

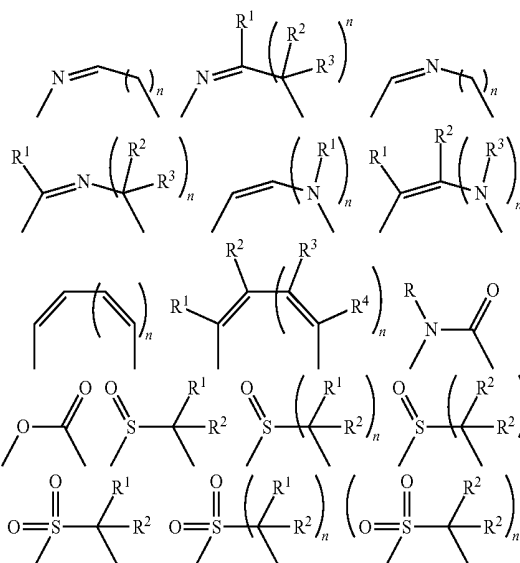


wherein m and n independently is an integer from 0 to 4, wherein each of V, X, and Y independently is V¹, V², V³, V⁴, O, S, S=O, SO₂, Se, NR³, PR², R¹P=O, CR¹R², C=O, SiR¹R², GeR¹R², BH, P(O)H, PH, NH, CR¹H, CH₂, SiH₂, SiHR¹, BH, or BR², or any one of



Continued

The continuation of Figure 1 shows a grid of chemical structures. The first three rows contain various heterocyclic structures (pyridine, pyrimidine, triazine, selenazine, oxadiazole, isoxazole, and 1,2,4-triazole) with substituents $(R^1)_n$. The subsequent rows show a variety of functional groups and linkages, including carbon-carbon bonds, amines, ethers, sulfides, boranes, silanes, and phosphines, all with substituents R^1, R^2, R^3, R^4 and a subscript n indicating multiple units. The structures are arranged in a grid-like fashion, with some structures having multiple substituents or being part of a polymer chain.



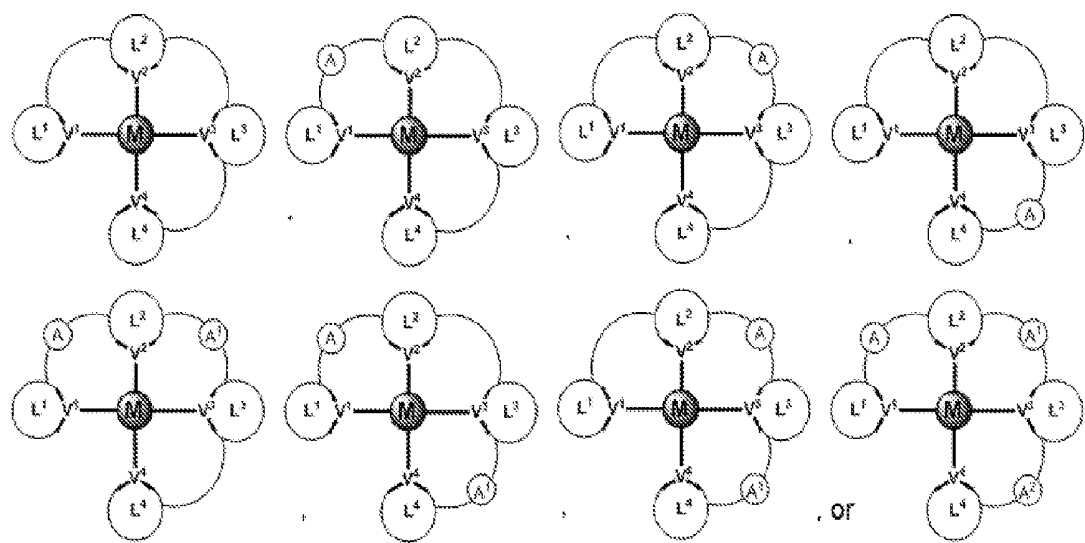
wherein at least one of V, X, and Y is V¹, V², V³, or V⁴, wherein m and n independently is an integer from 0 to 4, wherein each of V¹, V², V³ and V⁴ independently is coordinated to M, and wherein each of V¹, V², V³ and V⁴ independently is N, C, CH, P, B, SiH, or Si,

wherein each of R, R¹, R², R³, and R⁴, independently is hydrogen, aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, alkyl, alkenyl, alkynyl, deuterium, halogen, hydroxyl, thiol, nitro, cyano, amino, a mono- or di-alkylamino, a mono- or diaryl amino, alkoxy, aryloxy, haloalkyl, aralkyl, ester, nitrile, isonitrile, alkoxycarbonyl, acylamino, alkoxycarbonylamino, aryloxy carbonylamino, sulfonylamino, sulfamoyl, carbamoyl, alkylthio, sulfinyl, ureido, phosphoramidate, mercapto, sulfo, carboxyl, hydrazino, substituted silyl, or polymerizable, or any conjugate or combination thereof,

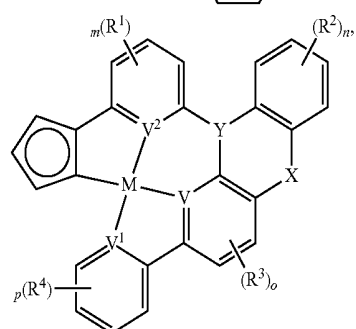
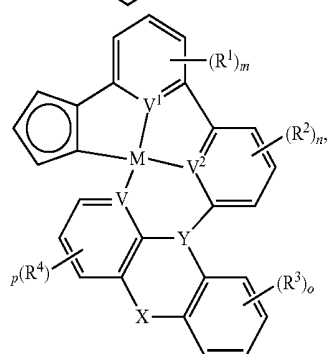
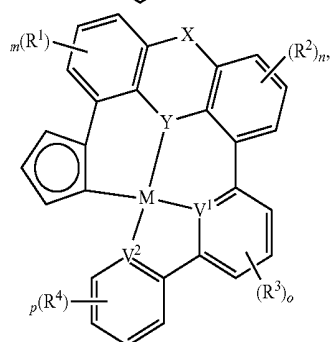
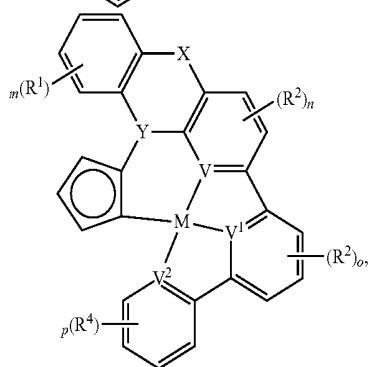
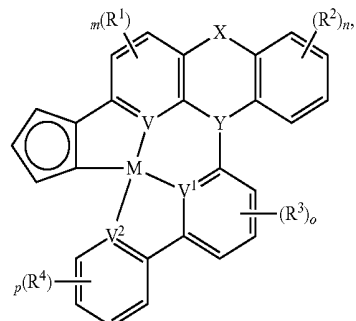
wherein each of V¹, V², V³ and V⁴ independently optionally is substituted for any one of V¹, V², V³ and V⁴.

[0033] In one aspect each of Z¹, Z², and Z³ independently is present. In another aspect, Z¹ and Z² are present and Z³ is absent. In yet another aspect, Z² and Z³ are present and Z¹ is absent. In yet another aspect, Z¹ and Z³ are present and Z² is absent. In yet another aspect, each of Z¹, Z², and Z³ independently is absent. In another aspect, Z¹ and Z² are absent and Z³ is present. In yet another aspect, Z² and Z³ are absent and Z¹ is present. In yet another aspect, Z¹ and Z³ are absent and Z² is present.

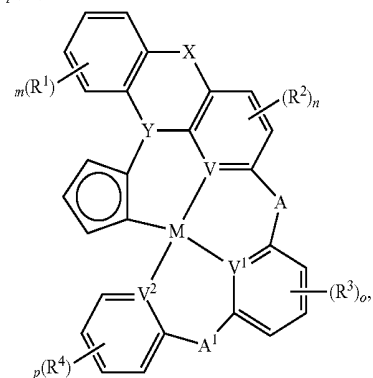
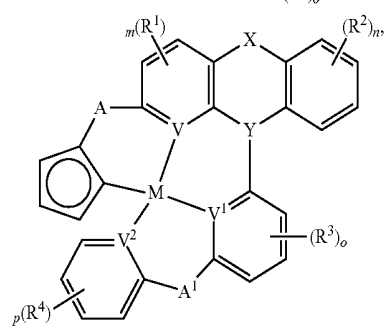
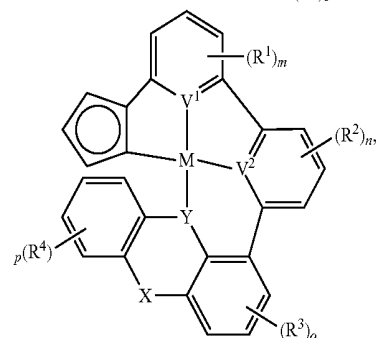
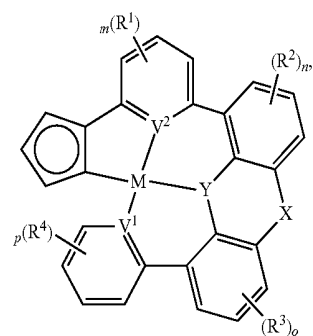
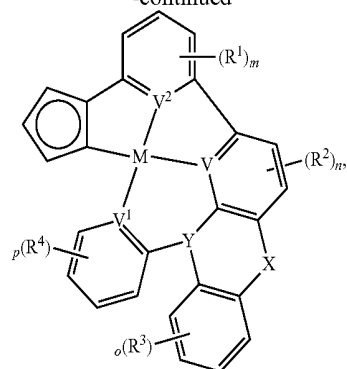
[0034] In one aspect, the compound can have the structure

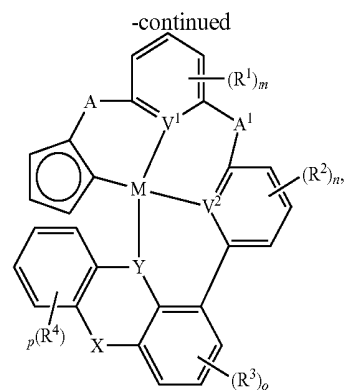
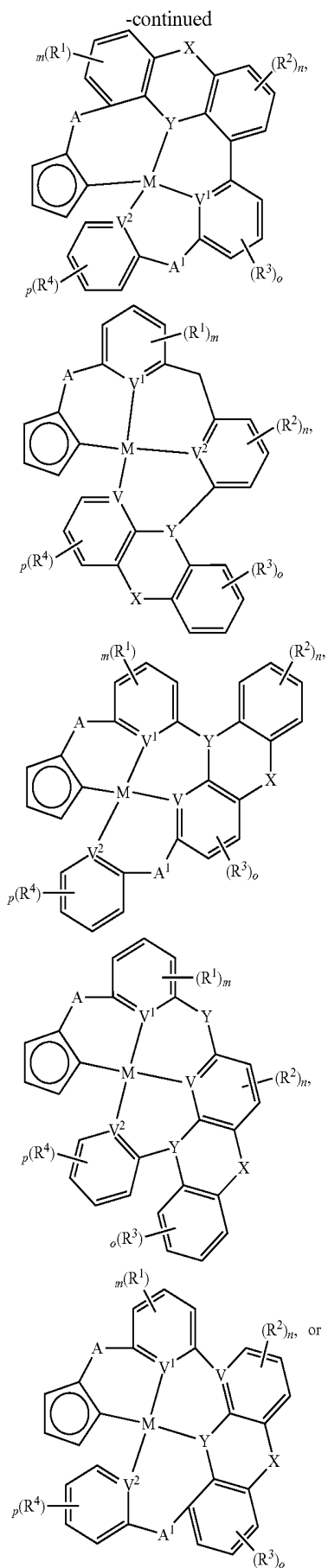


[0035] In another aspect, the compound can have the structure:



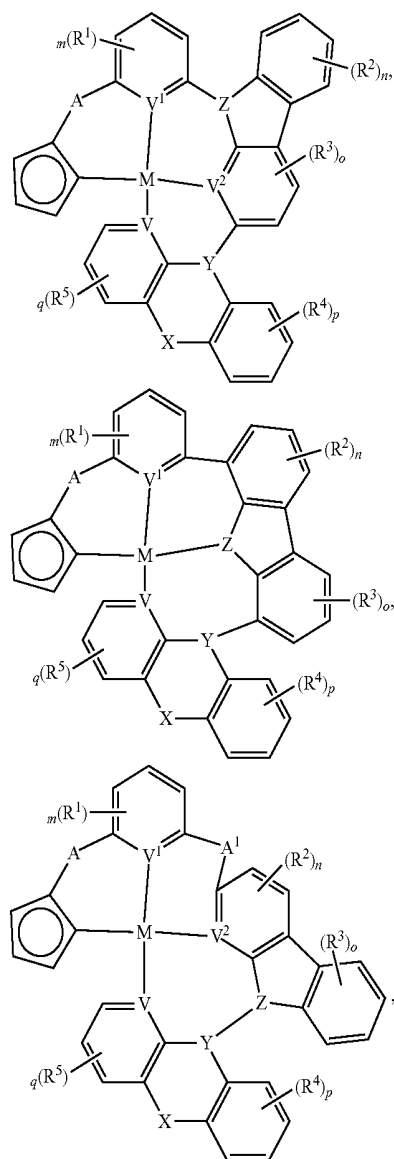
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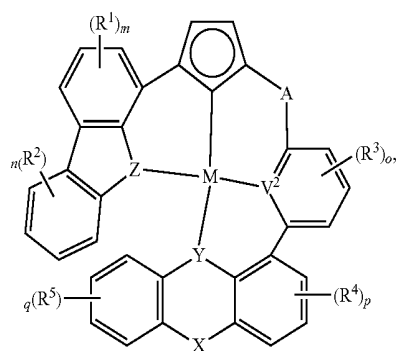
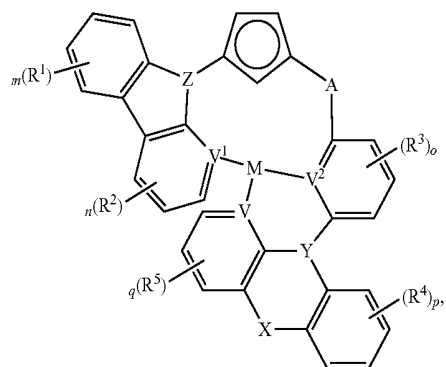
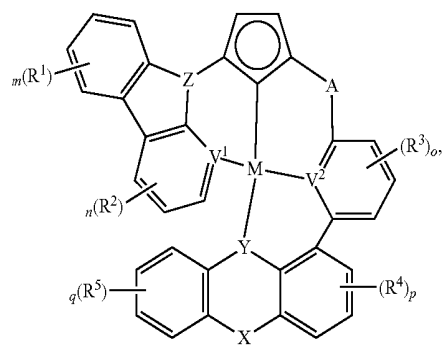
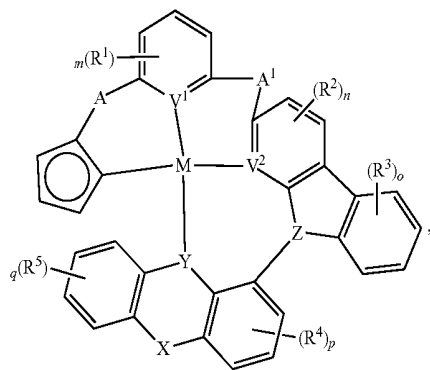
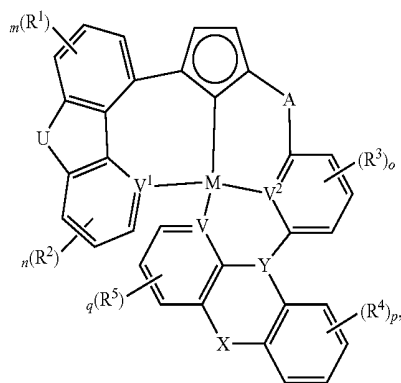
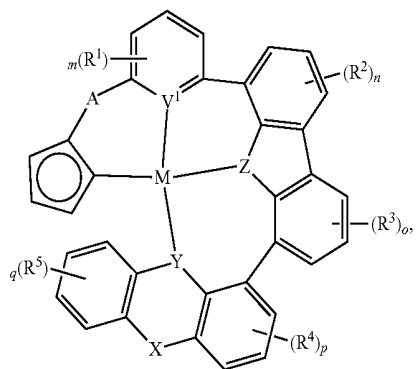
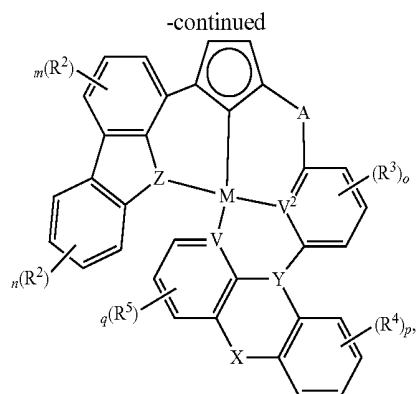
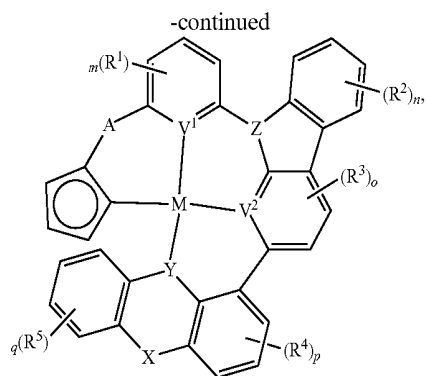


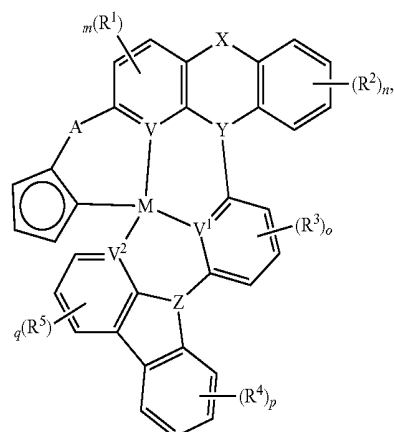
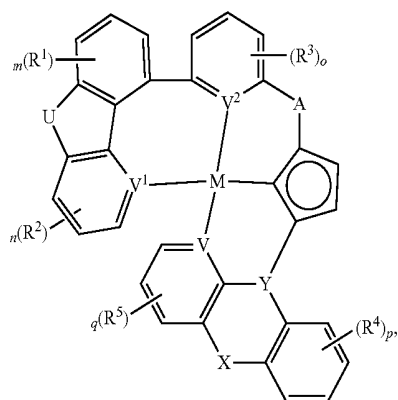
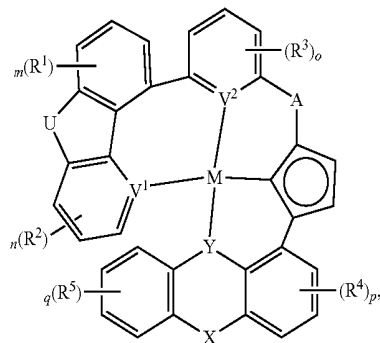
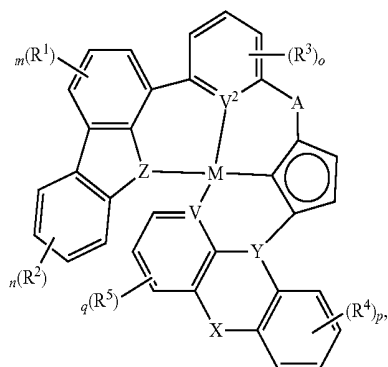
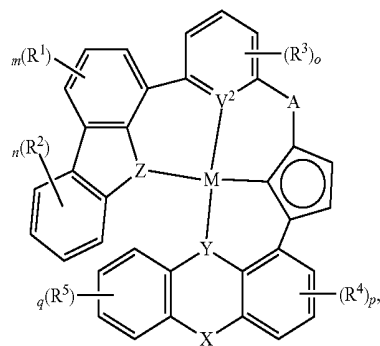
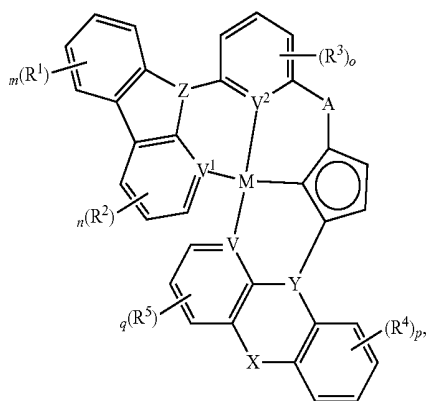
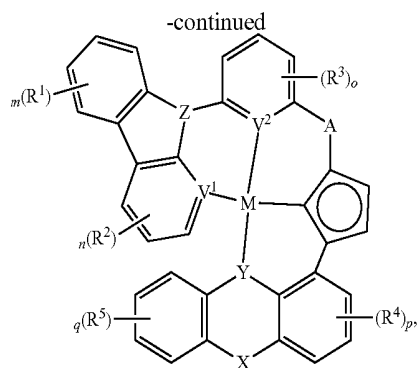
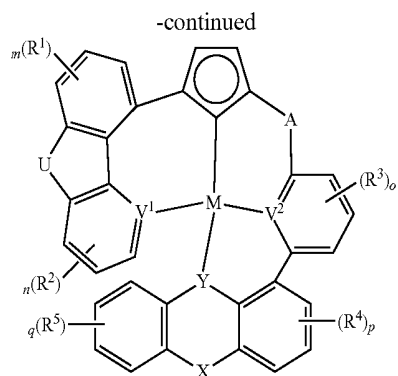


wherein each of o and p independently is an integer of 0 to 4.

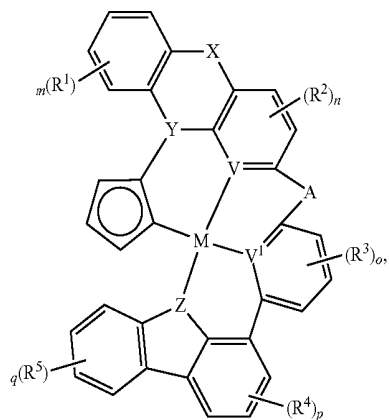
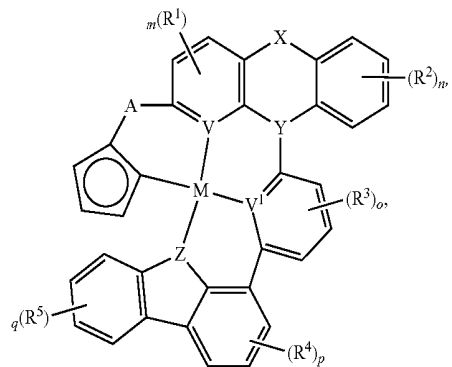
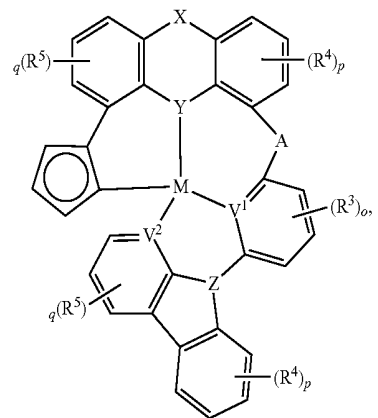
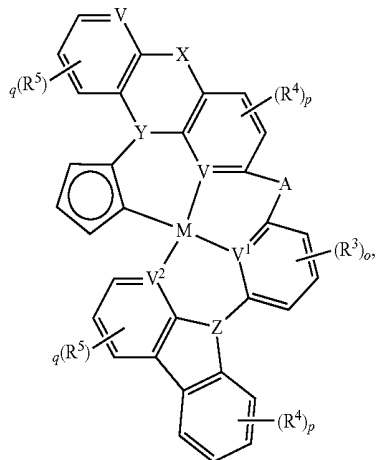
[0036] In another aspect, the compound can have the structure:



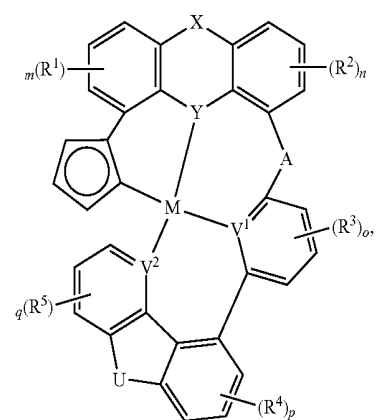
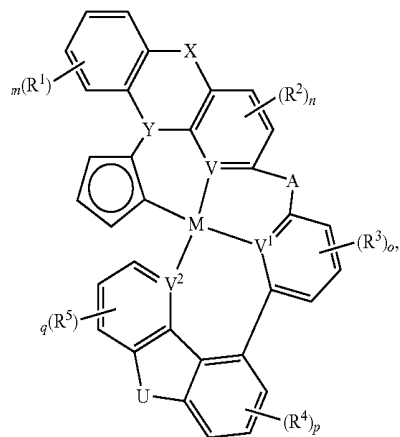
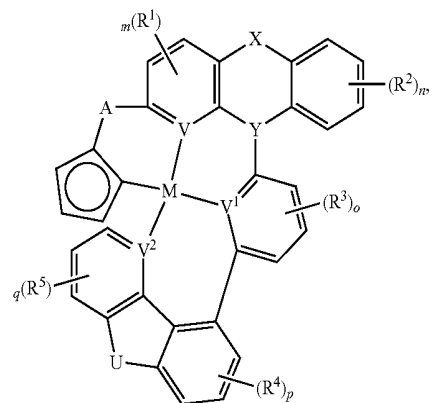
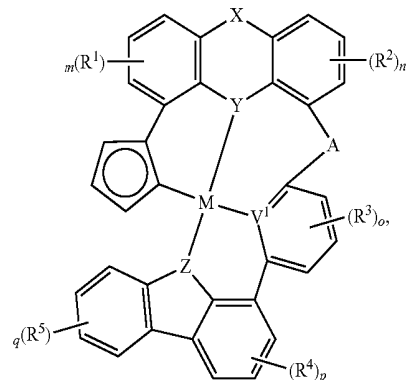




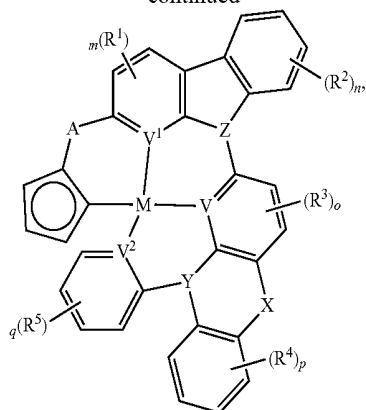
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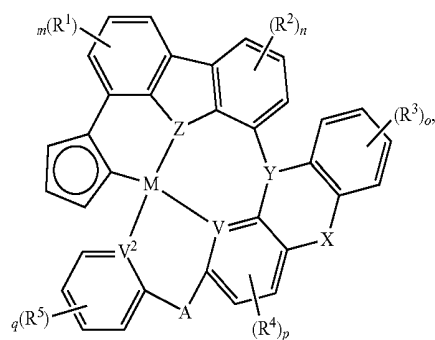
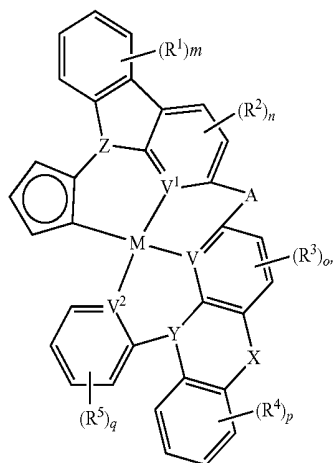
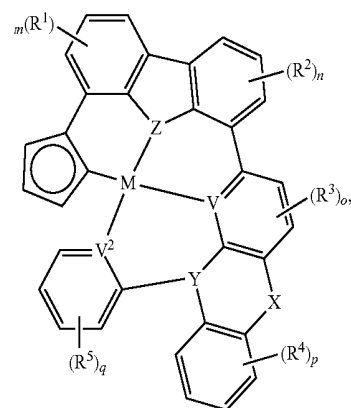
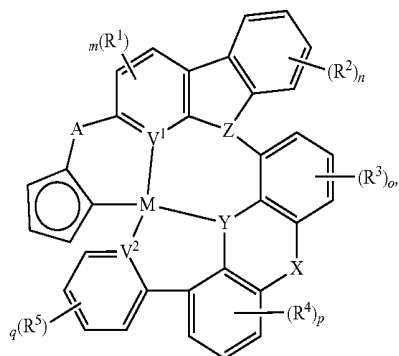
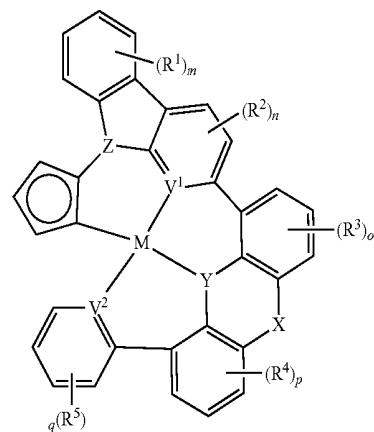
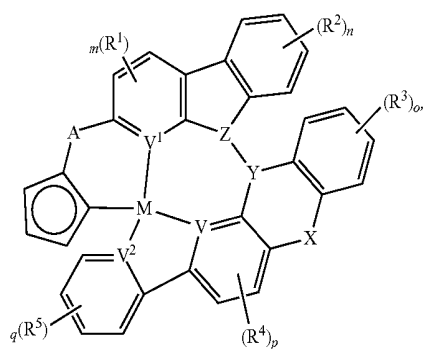
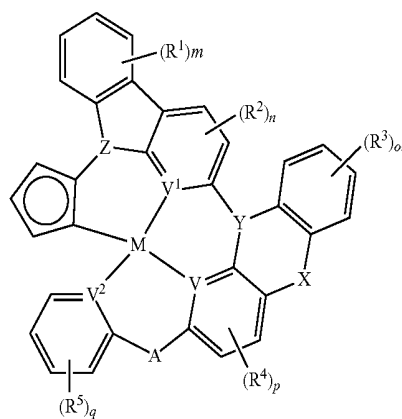
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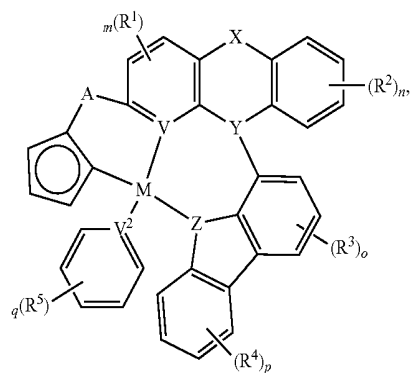
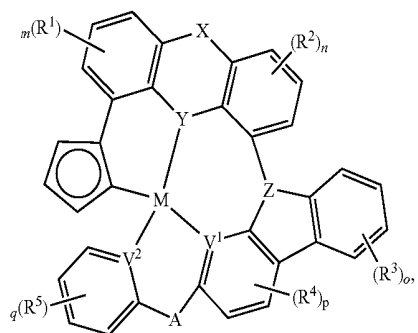
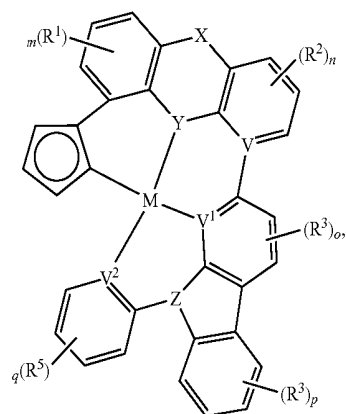
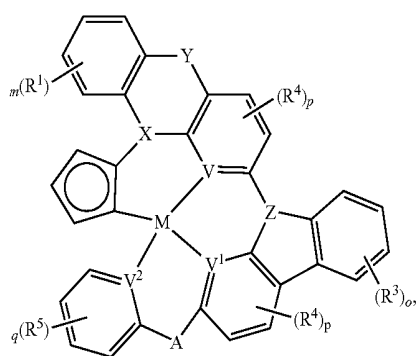
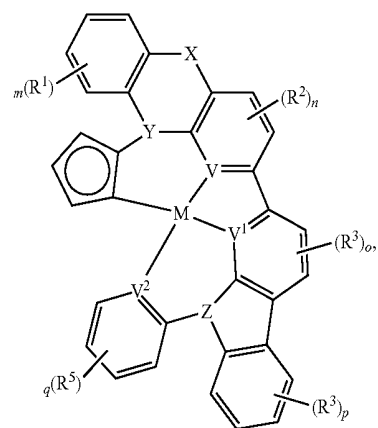
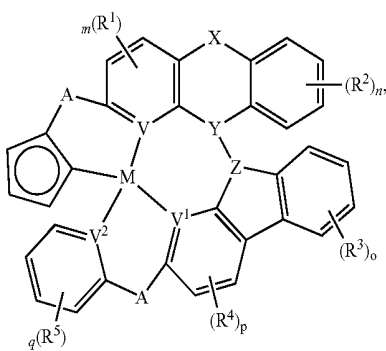
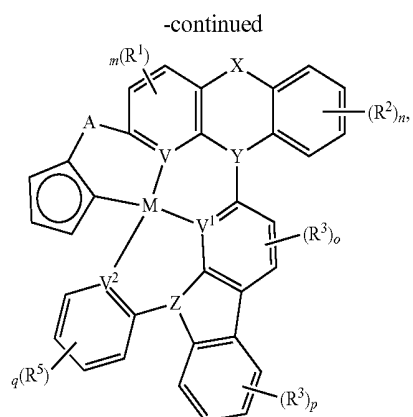
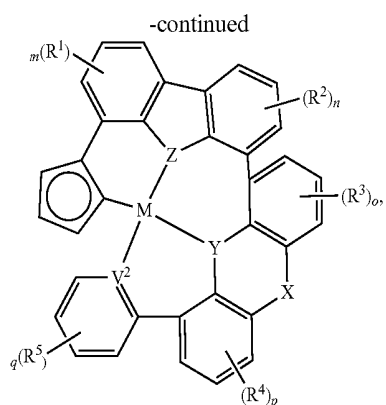


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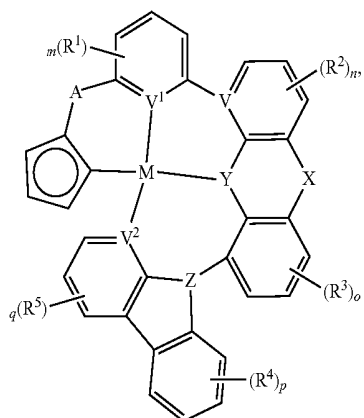
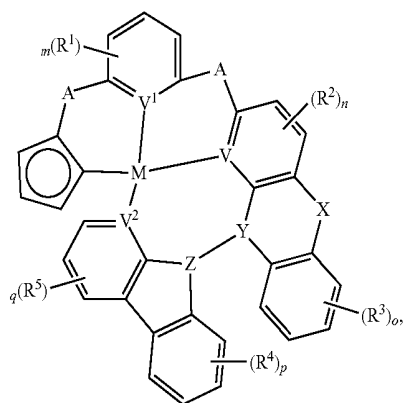
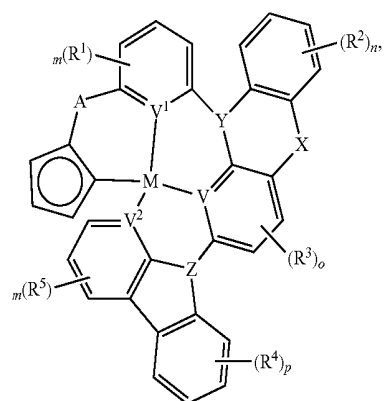
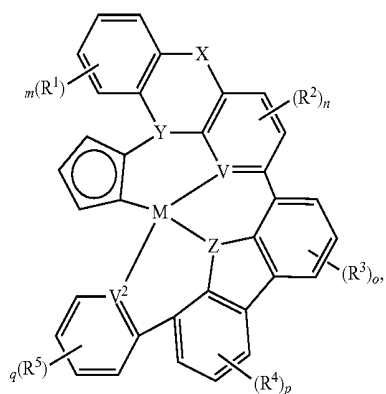


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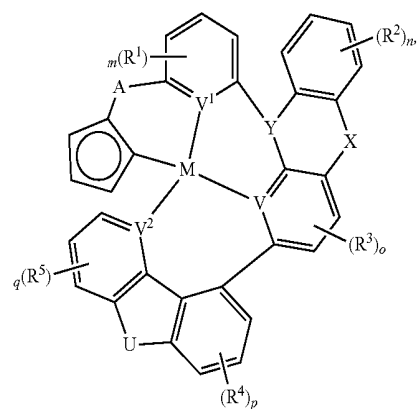
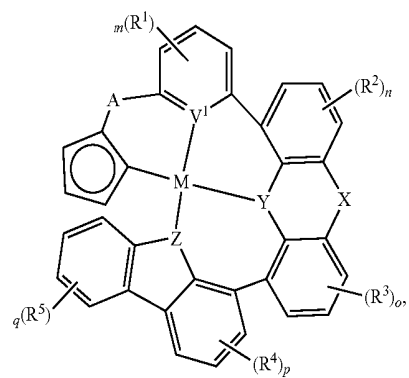
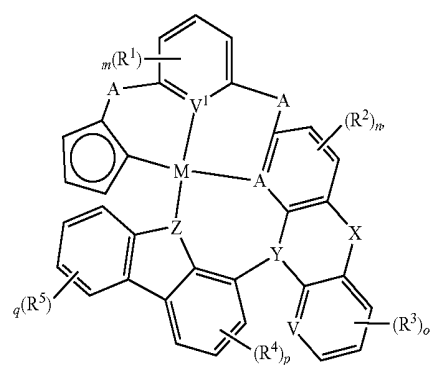
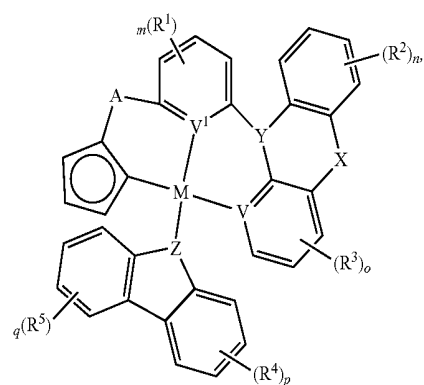




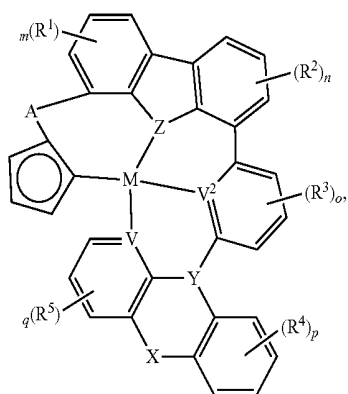
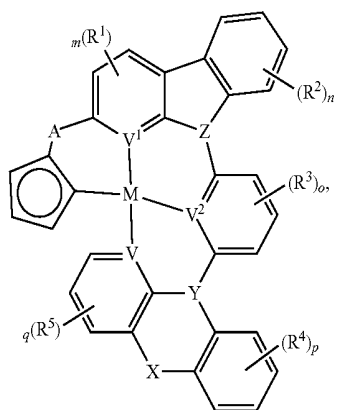
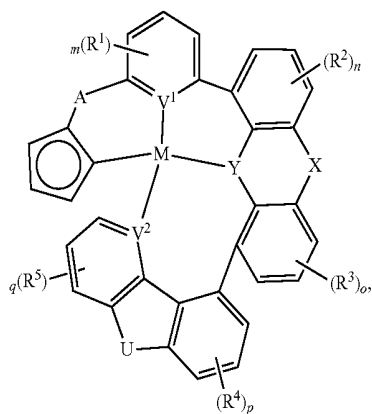
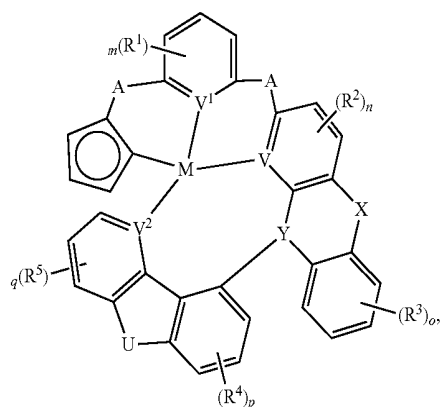
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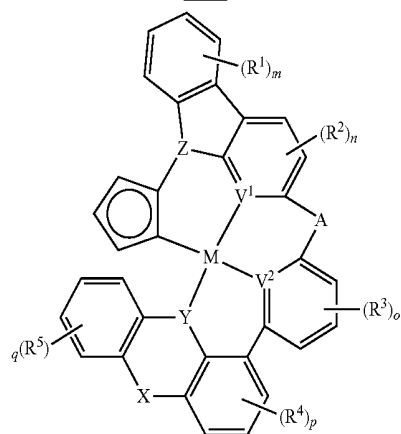
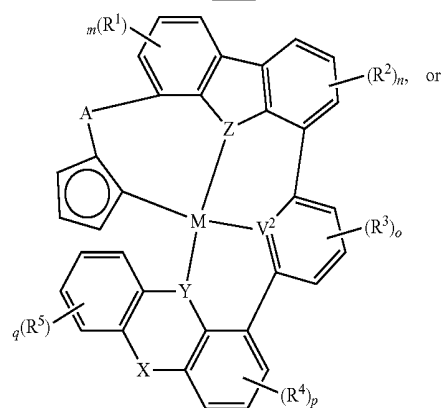
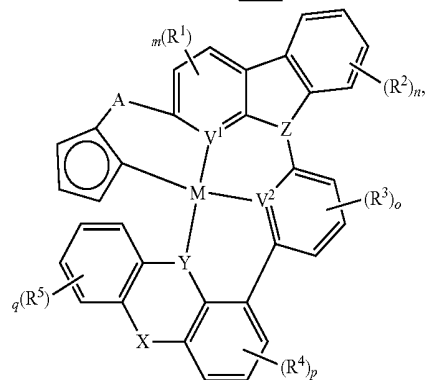
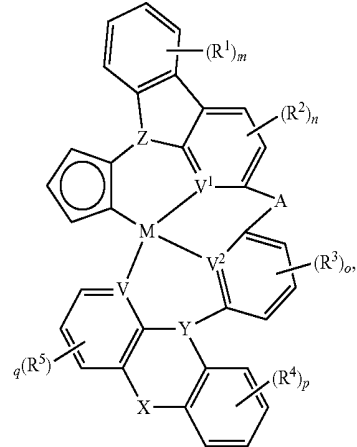
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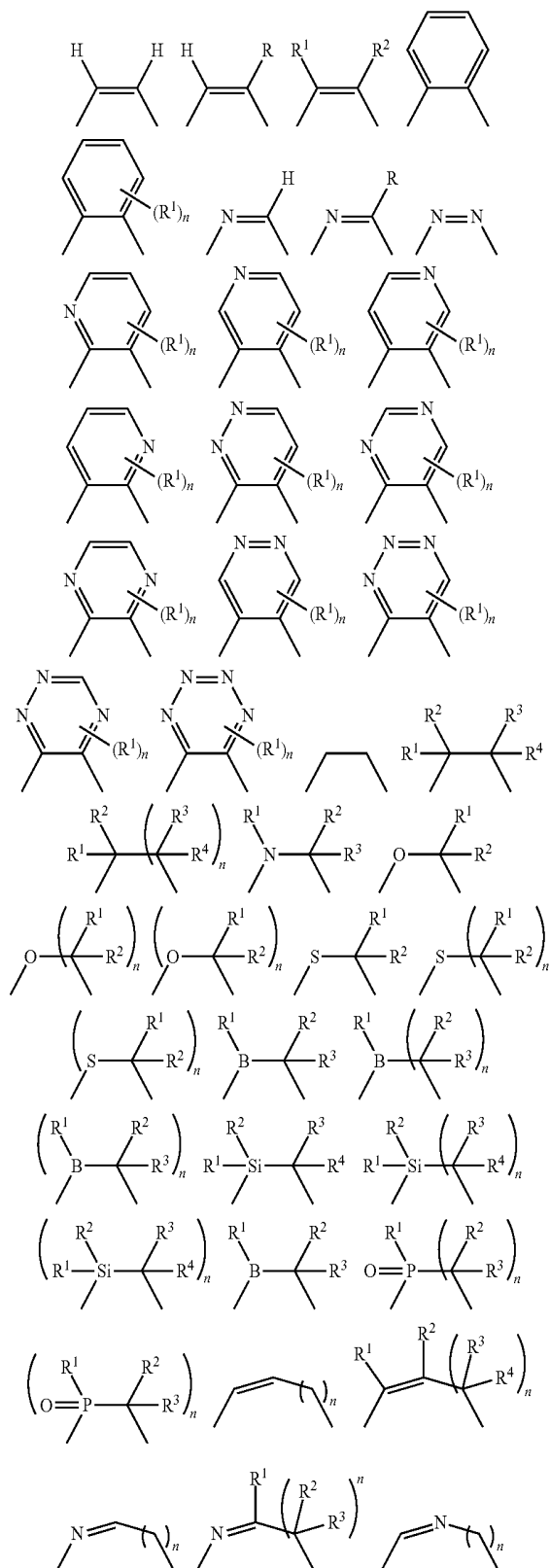


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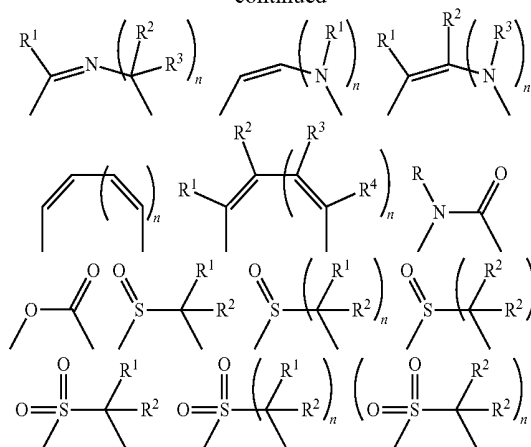


wherein each of o, p, and q independently is an integer of 0 to 4,

wherein each of Z and U independently is V^1 , V^2 , V^3 , V^4 , O, S, $S=O$, SO_2 , Se, NR, PR^3 , $R^1P=O$, CR^1R^2 , $C=O$, SiR^1R^2 , GeR^1R^2 , BH, $P(O)H$, PH, NH, CR^1H , CH_2 , SiH_2 , $SiHR^1$, BH, or BR^3 , or any one of



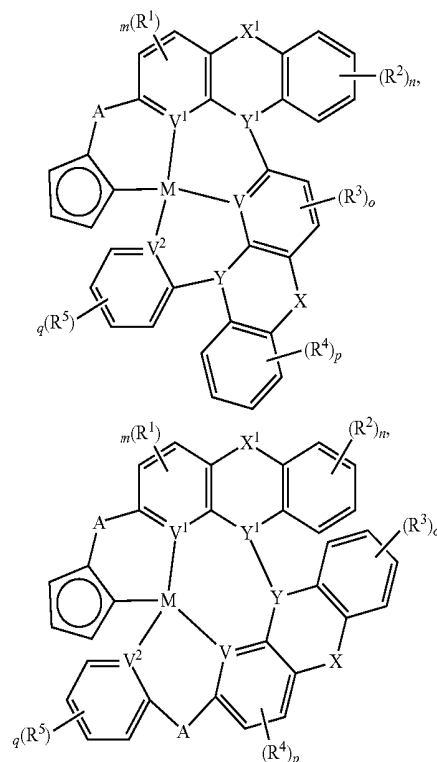
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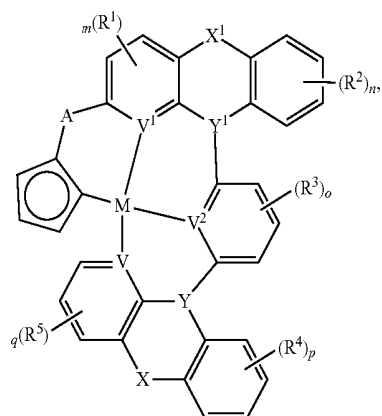
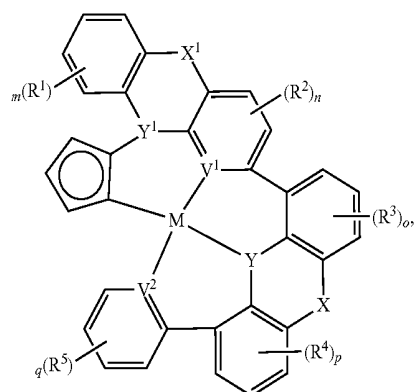
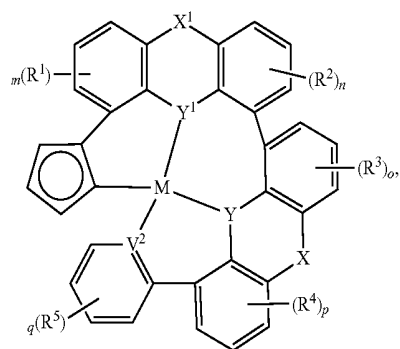
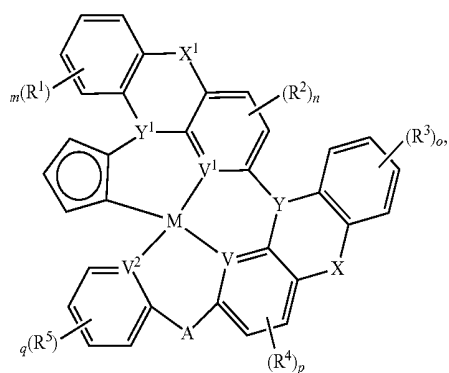
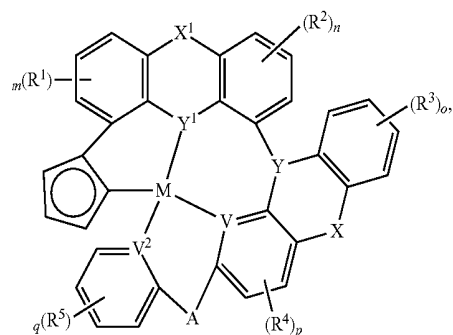
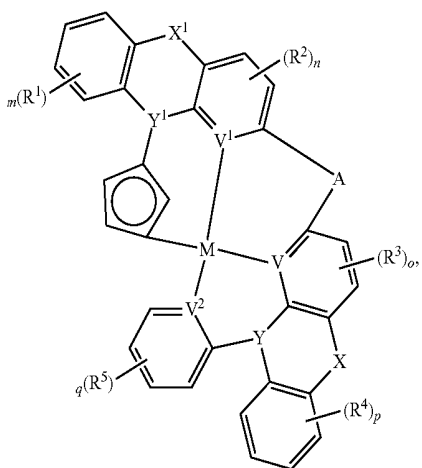
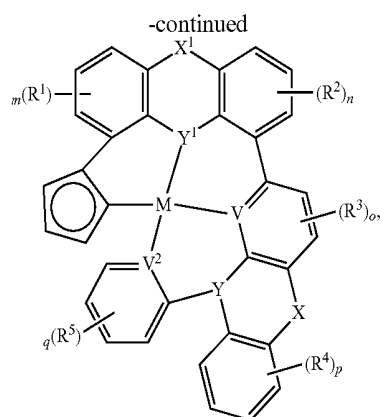
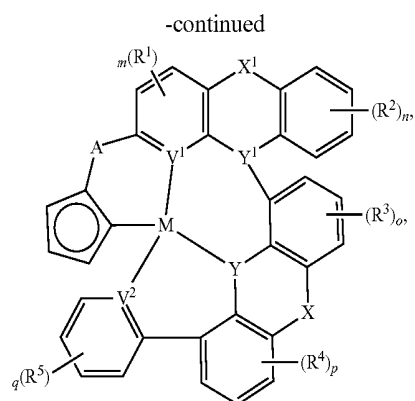


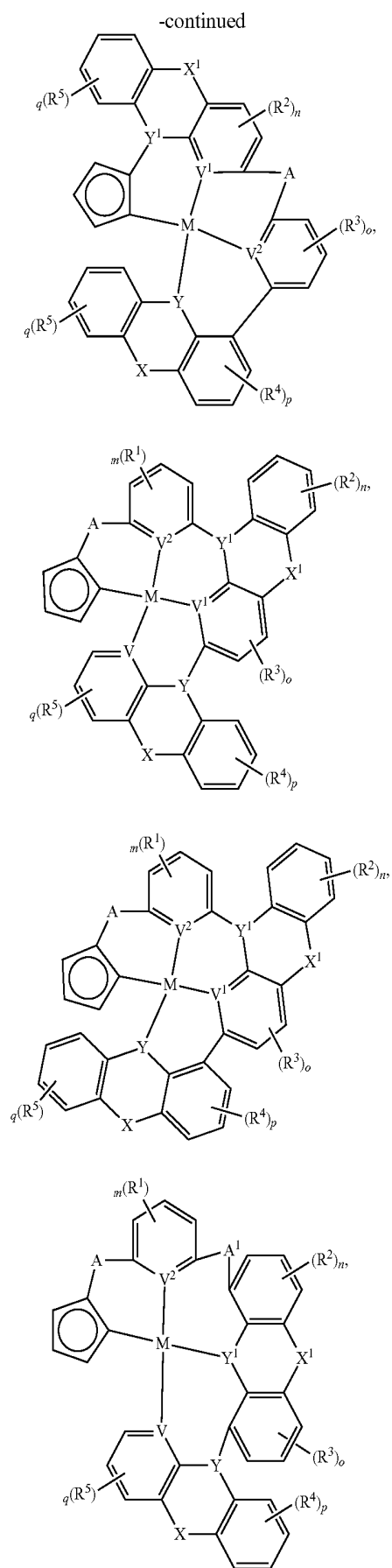
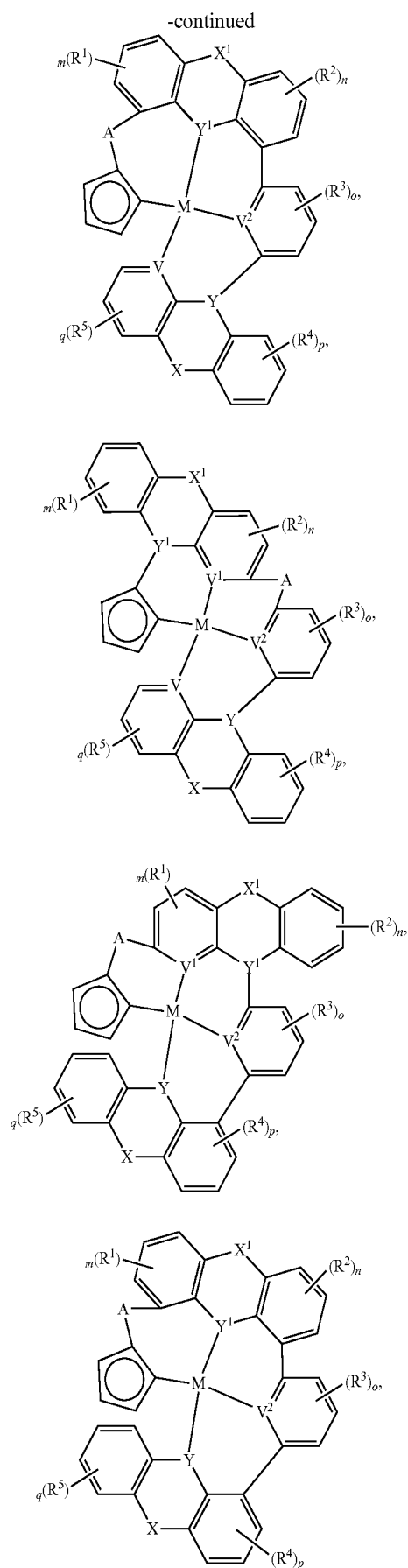
wherein at least one of U, V, X, Y, and Z is V^1 , V^2 , V^3 , or V^4 .

wherein each R^5 independently is hydrogen, aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, alkyl, alkenyl, alkynyl, deuterium, halogen, hydroxyl, thiol, nitro, cyano, amino, a mono- or di-alkylamino, a mono- or diaryl amino, alkoxy, aryloxy, haloalkyl, aralkyl, ester, nitrile, isonitrile, alkoxycarbonyl, acylamino, alkoxycarbonylamino, aryloxycarbonylamino, sulfonylamino, sulfamoyl, carbamoyl, alkylthio, sulfinyl, ureido, phosphoramidate, amercapto, sulfo, carboxyl, hydrazino, substituted silyl, or polymerizable, or any conjugate or combination thereof.

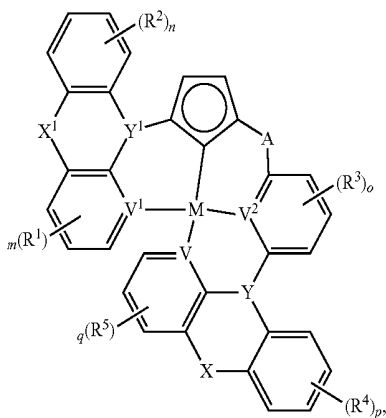
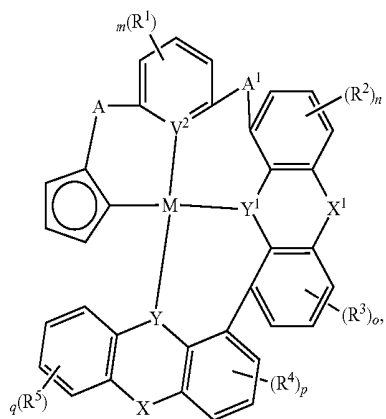
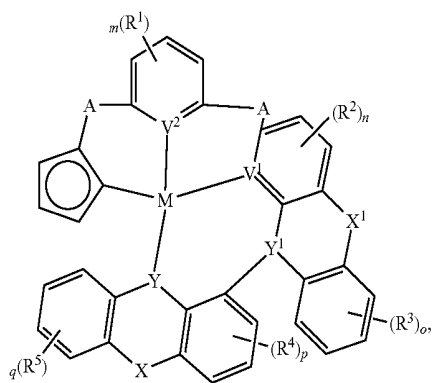
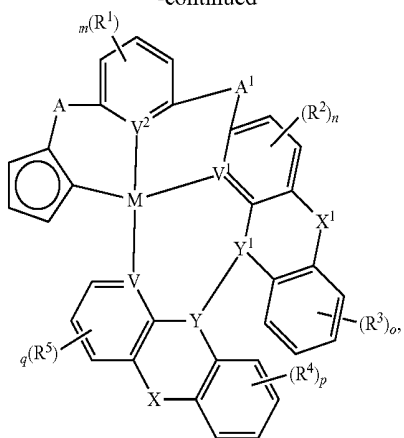
[0037] In one aspect, the compound can have the structure:



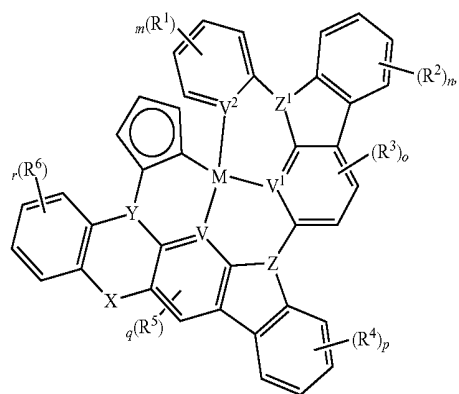
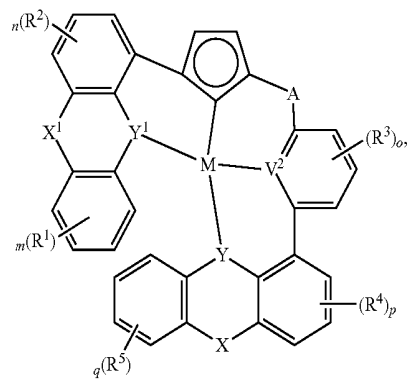
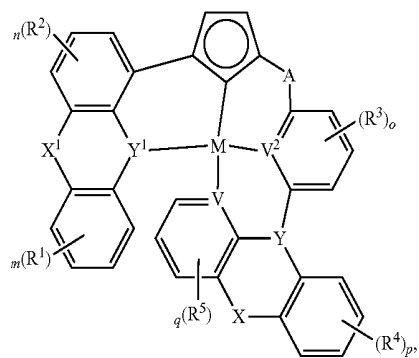
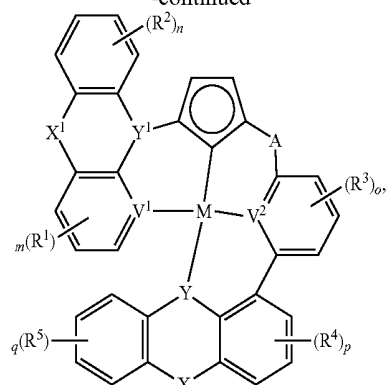


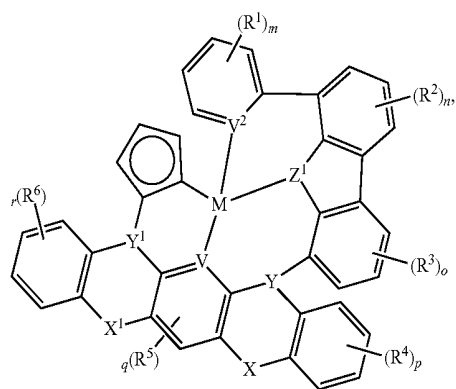
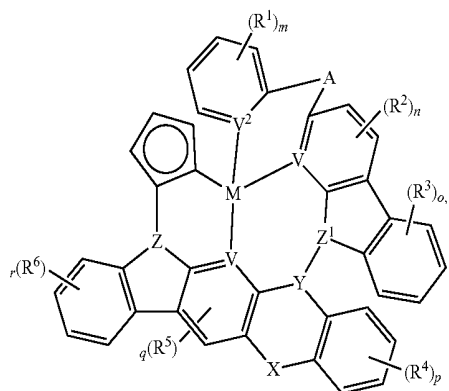
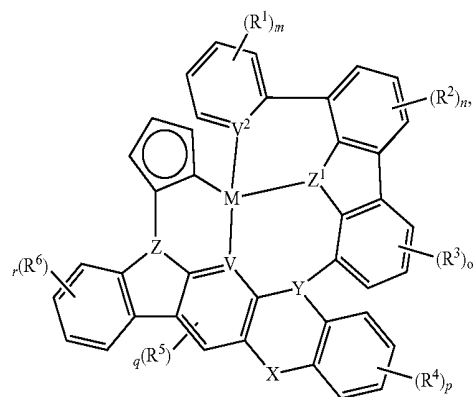
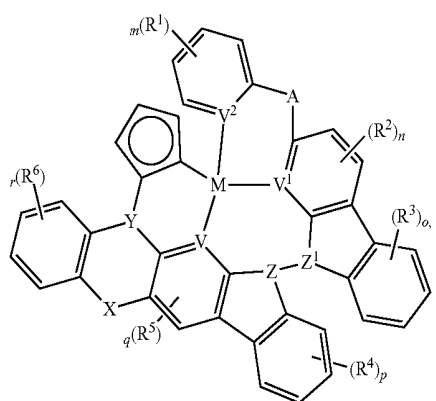
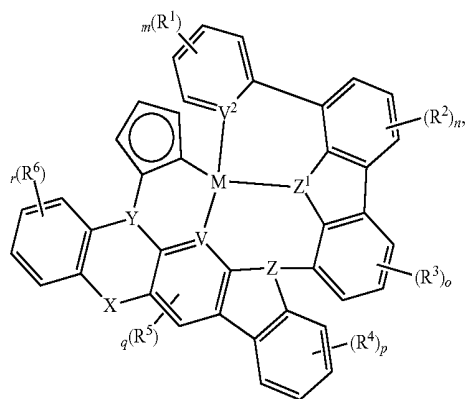
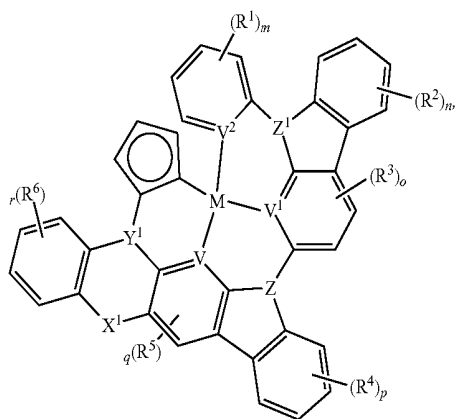
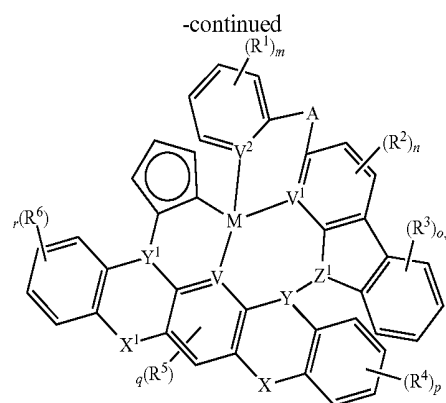
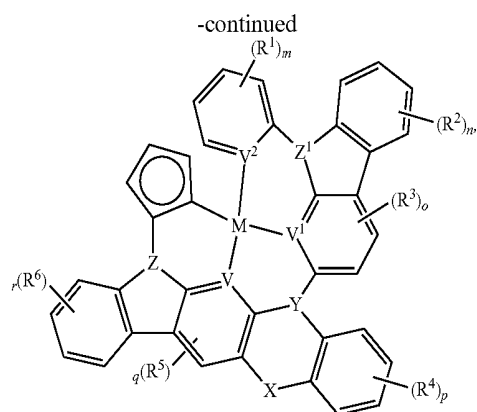


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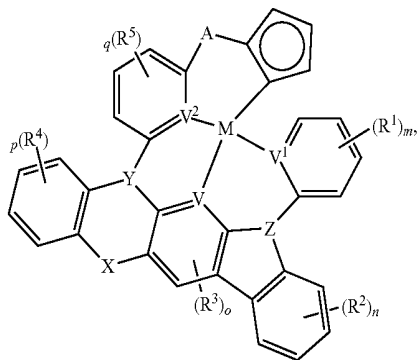


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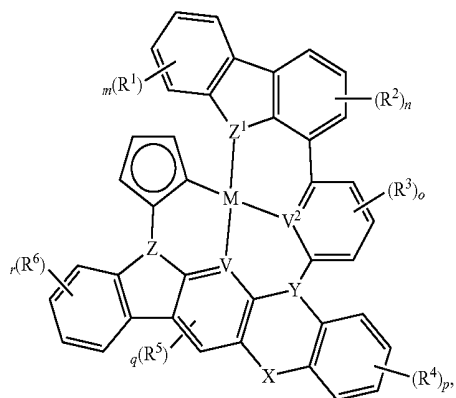
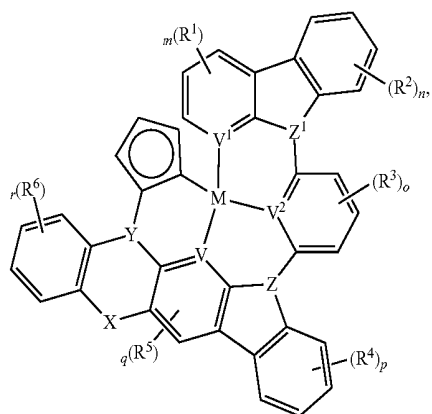
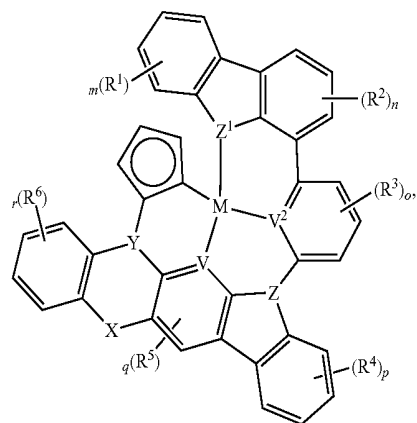
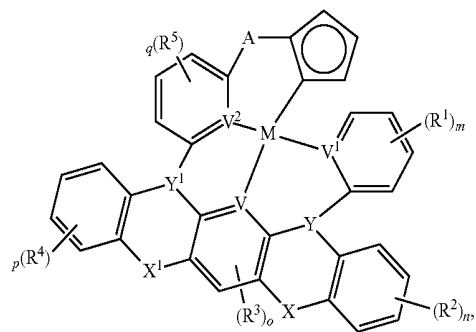
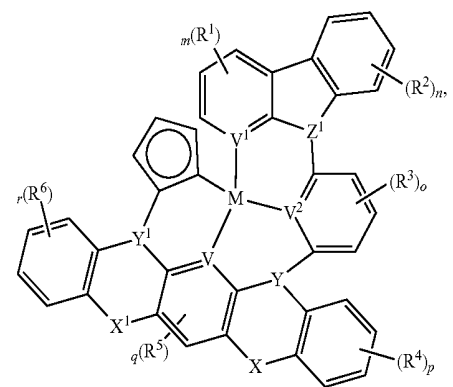
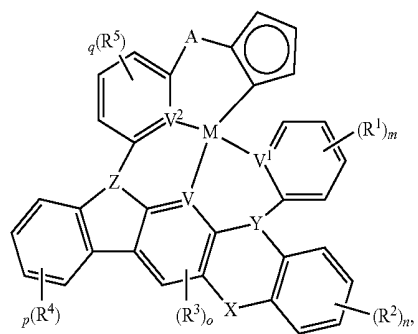
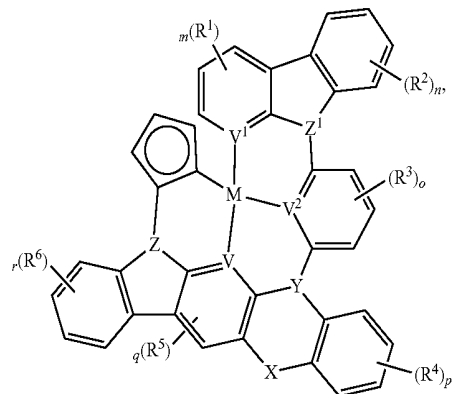


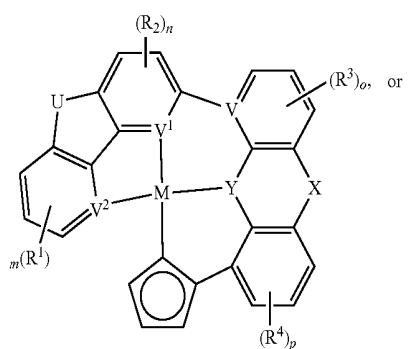
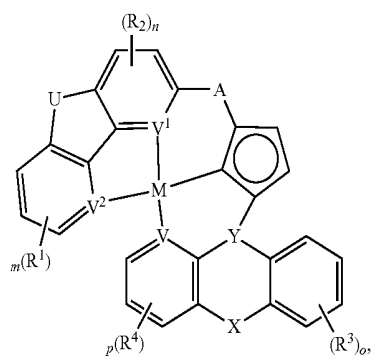
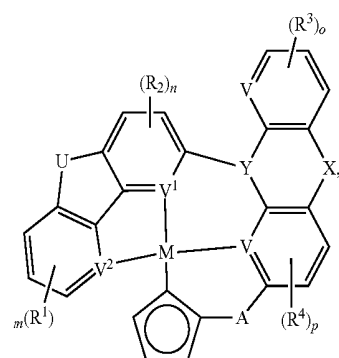
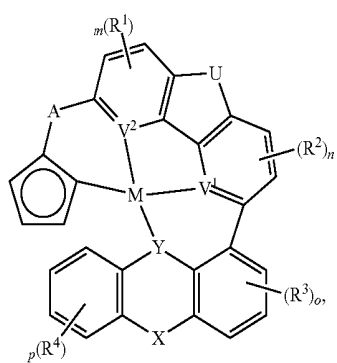
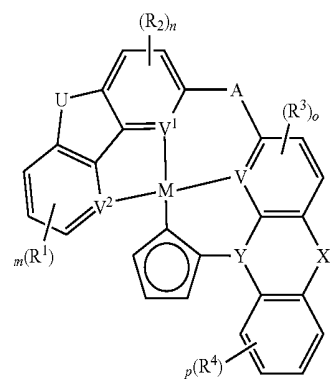
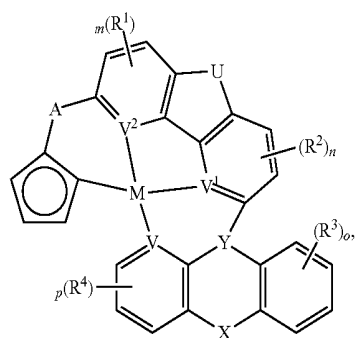
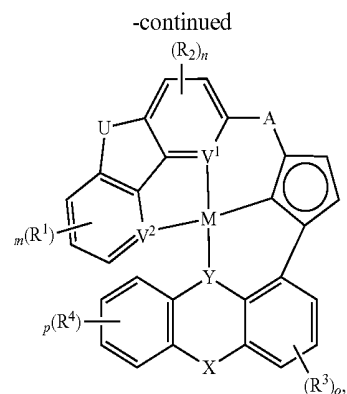
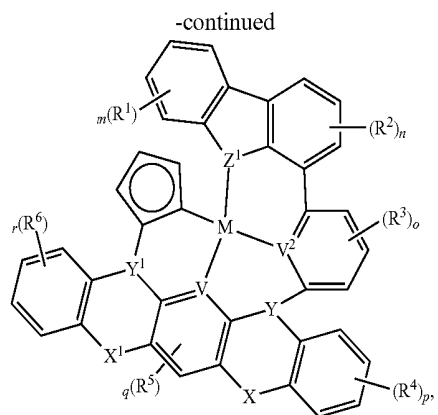


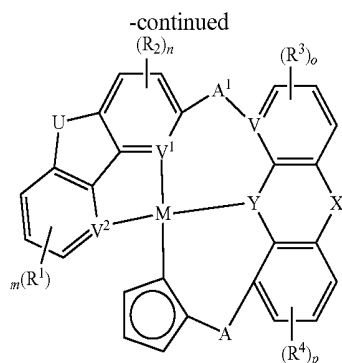
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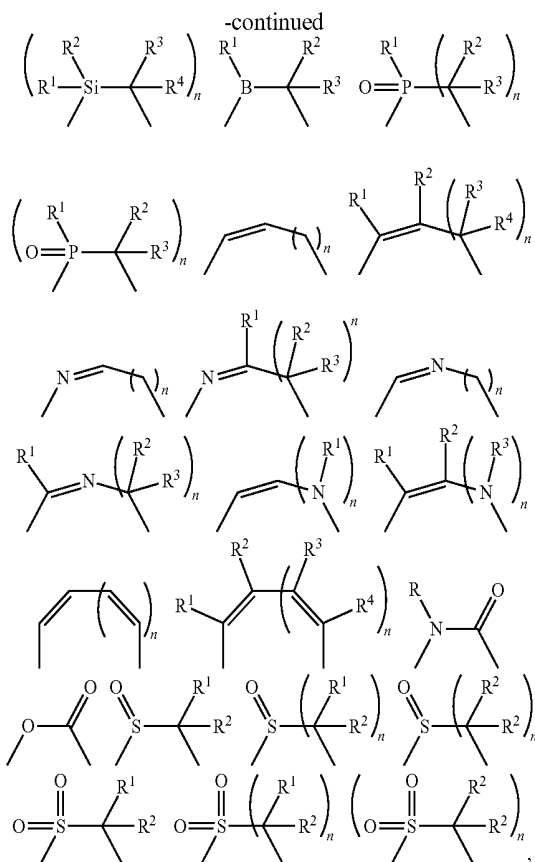
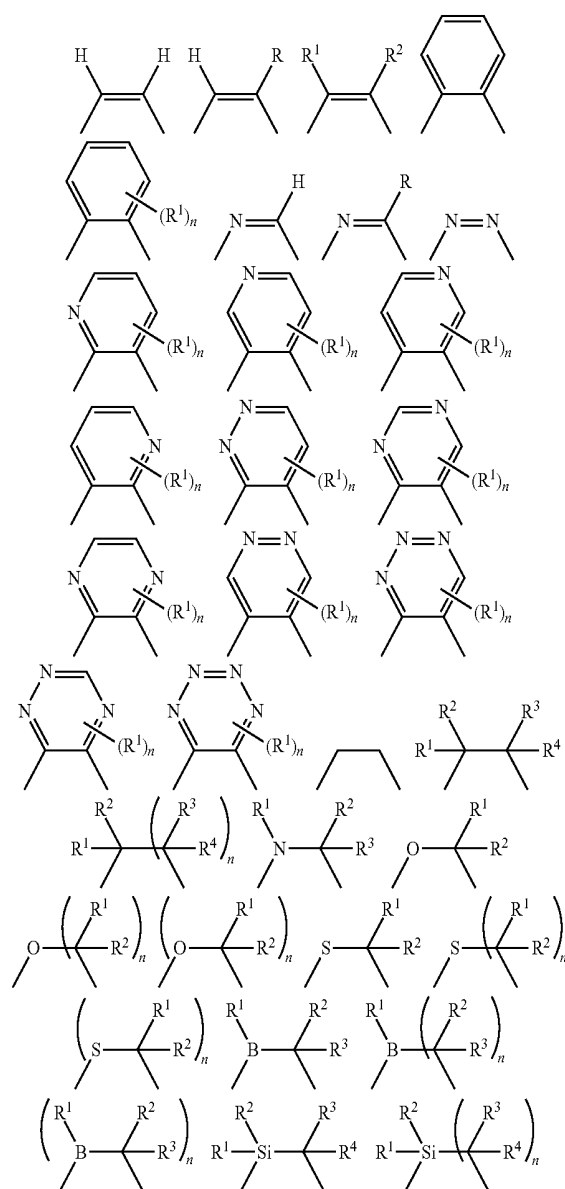
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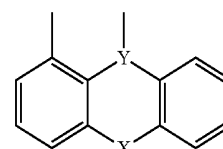
wherein each of o, p, q, and r independently is an integer of 0 to 4, wherein each of U, Y¹, X¹, Z, and Z¹, independently is V¹, V², V³, V⁴, O, S, S=O, SO₂, Se, NR³, PR³, R¹P=O, CR¹R², C=O, SiR¹R², GeR¹R², BH, P(O)H, PH, NH, CR¹H, CH₂, SiH₂, SiHR¹, BH, or BR³, or any one of



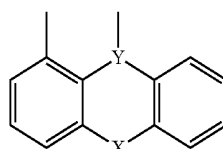
wherein at least one of U, V, X, Y, and Z is V¹, V², V³, or V⁴,

wherein each R⁵ independently is hydrogen, aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, alkyl, alkenyl, alkynyl, deuterium, halogen, hydroxyl, thiol, nitro, cyano, amino, a mono- or di-alkylamino, a mono- or diaryl amino, alkoxy, aryloxy, haloalkyl, aralkyl, ester, nitrile, isonitrile, alkoxycarbonyl, acylamino, alkoxycarbonylamino, aryloxy carbonylamino, sulfonylamino, sulfamoyl, carbamoyl, alkylthio, sulfinyl, ureido, phosphoramidate, amercapto, sulfo, carboxyl, hydrazino, substituted silyl, or polymerizable, or any conjugate or combination thereof.

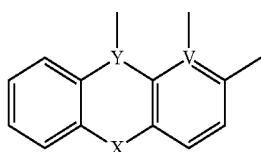
[0038] In one aspect, in the disclosed compounds,



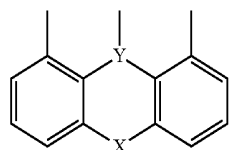
can be



[0039] In one aspect, in the disclosed compounds,



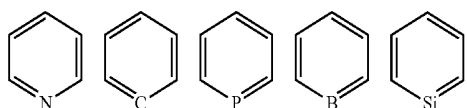
can be



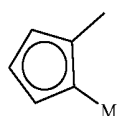
[0040] In one aspect, at least two of L^1 , L^2 , L^3 , and L^4 is substituted or unsubstituted aryl or heteroaryl.

[0041] In one aspect, at least one of V^1 , V^2 , V^3 , and V^4 is N. In another aspect, at least one of V^1 , V^2 , V^3 , and V^4 is C. In yet another aspect, at least one of V^1 , V^2 , V^3 , and V^4 is N and at least one of V^1 , V^2 , V^3 , and V^4 is C. In yet another aspect, at least one of V^1 , V^2 , V^3 , and V^4 is N and at least two of V^1 , V^2 , V^3 , and V^4 is C.

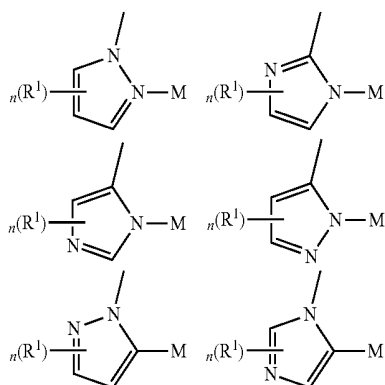
[0042] In one aspect, at least one of L , L^2 , L^3 , and L^4 together with V^1 , V^2 , V^3 , and V^4 respectively can comprise one or more of the following structures:



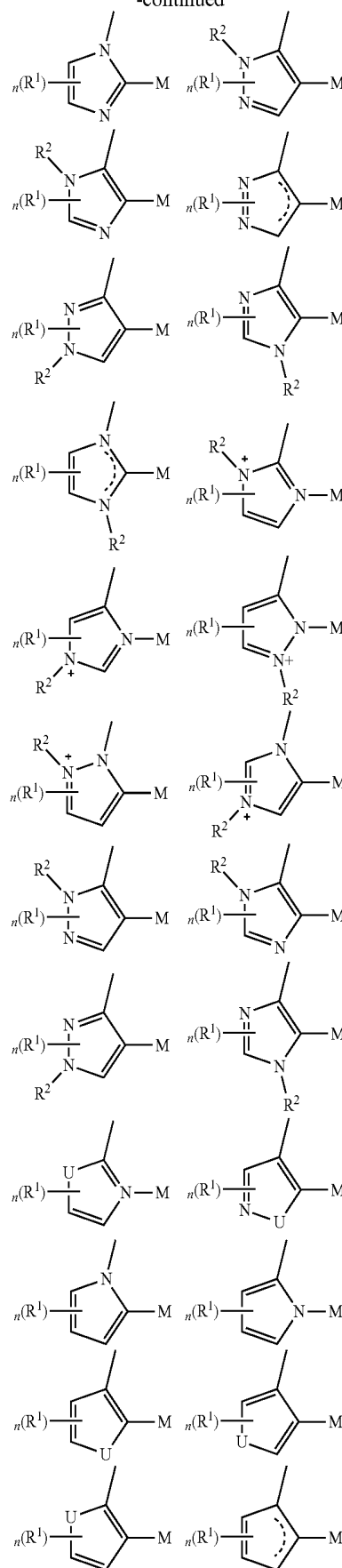
[0043] In another aspect, at least one of L^1 , L^2 , L^3 , and L^4 together with V^1 , V^2 , V^3 , and V^4 respectively or



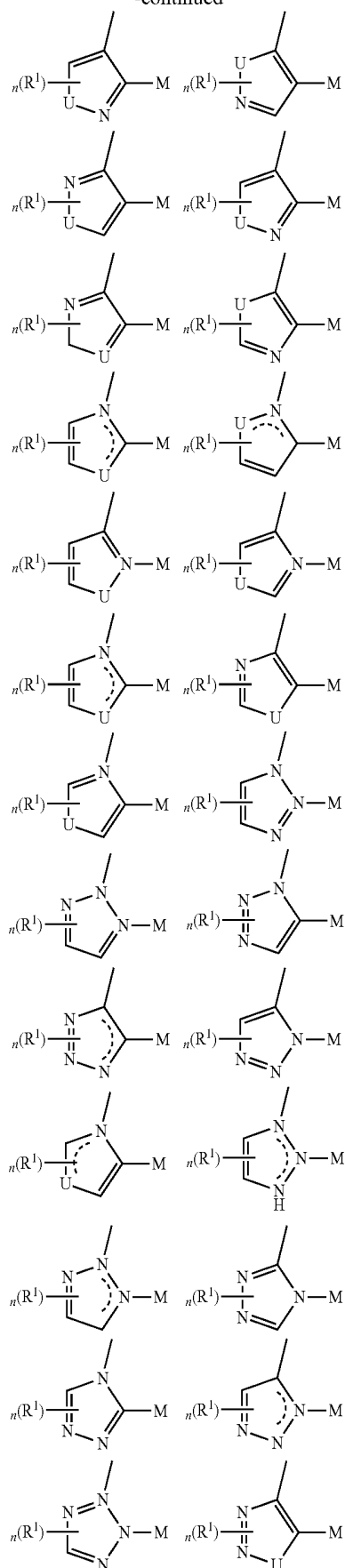
can comprise one or more of the following structures:



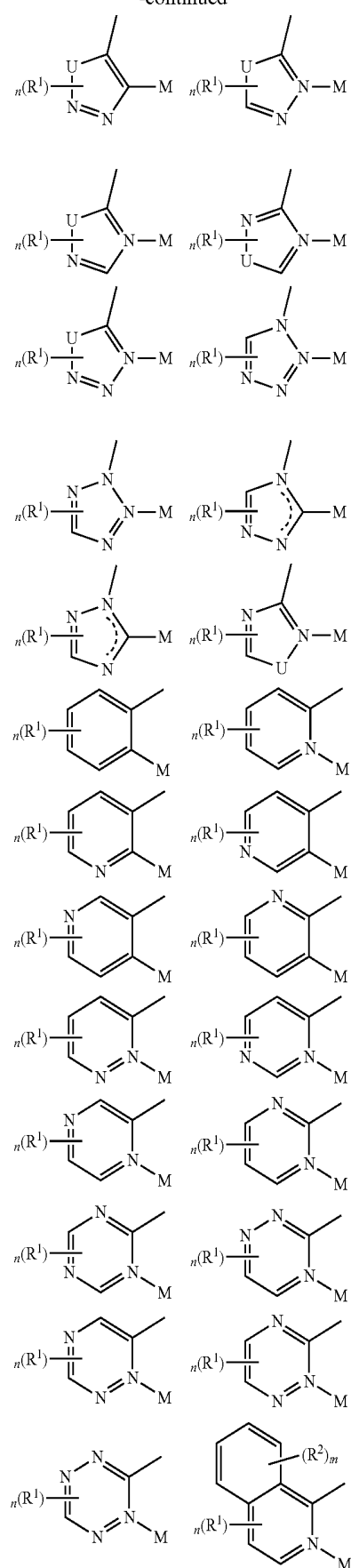
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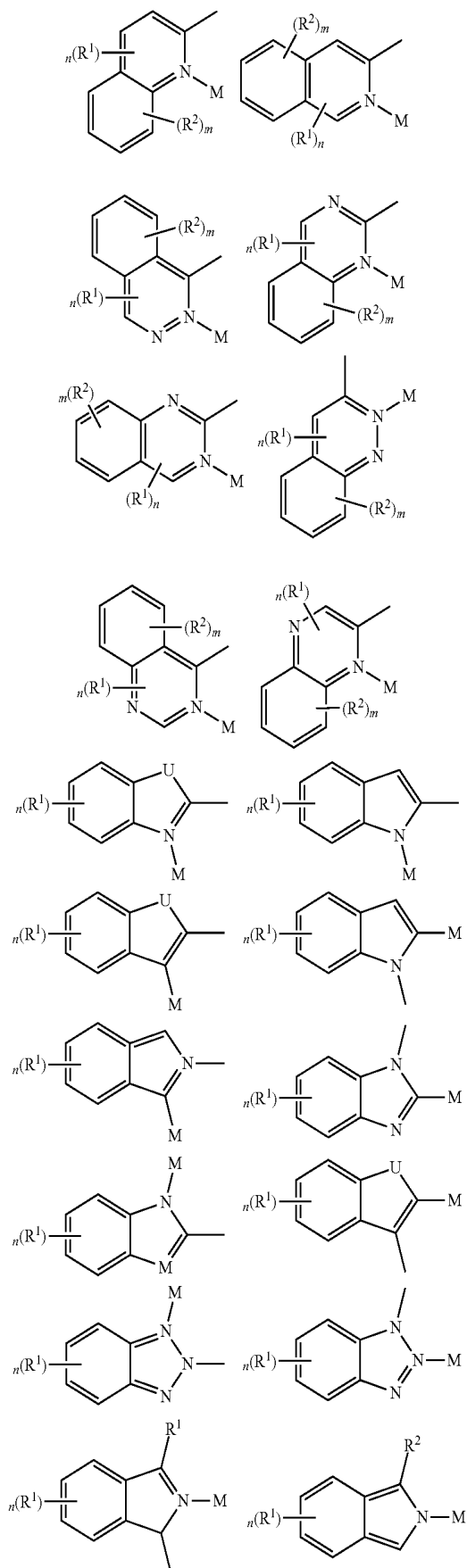
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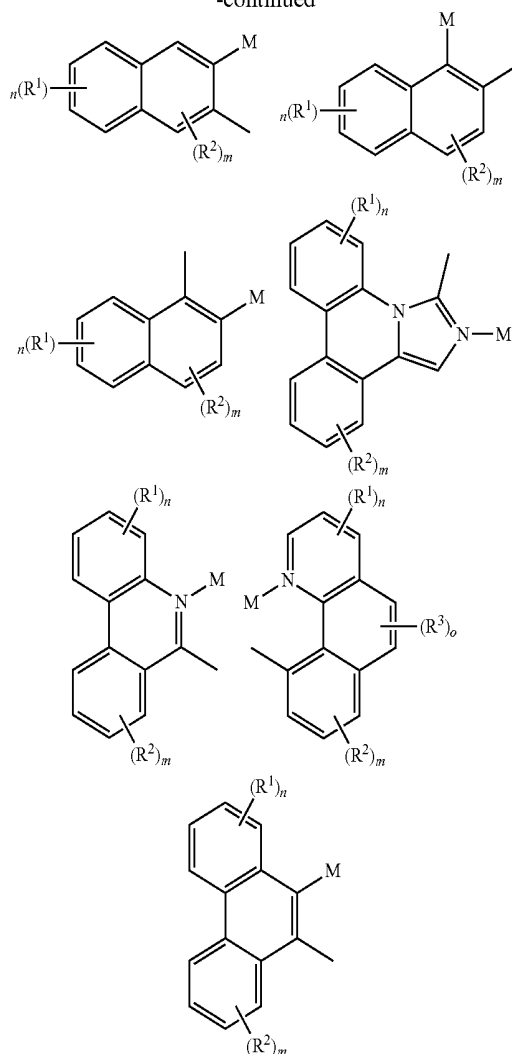
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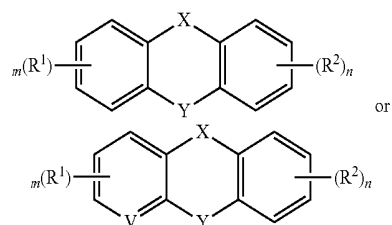
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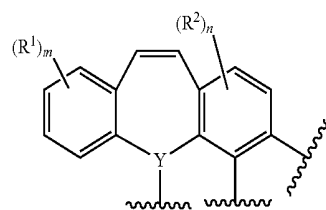
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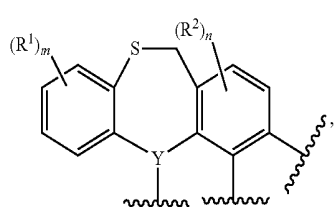
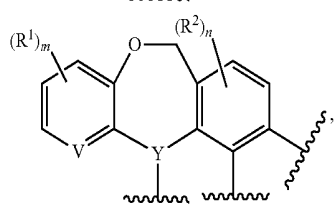
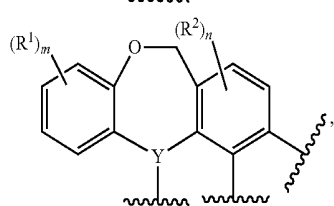
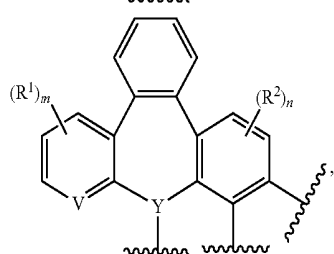
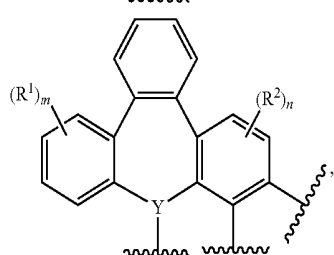
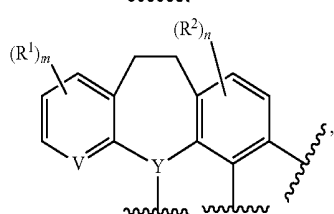
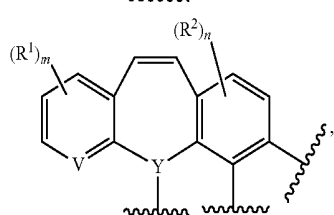
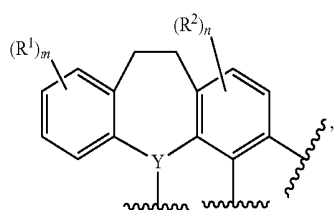
[0044] In one aspect,



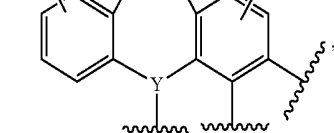
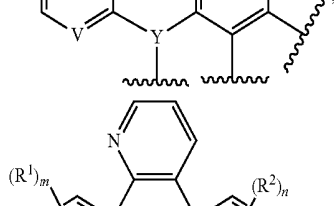
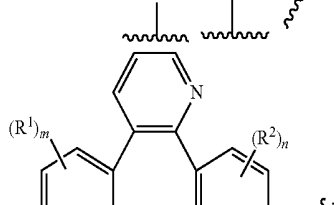
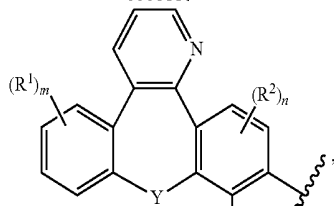
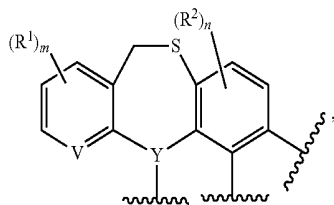
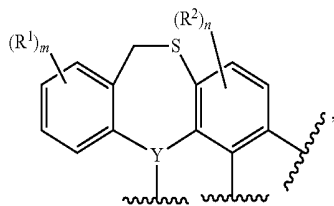
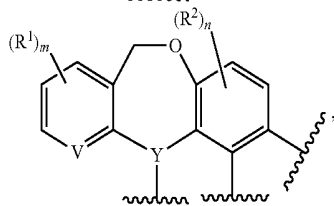
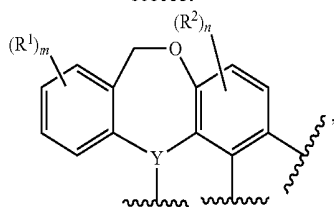
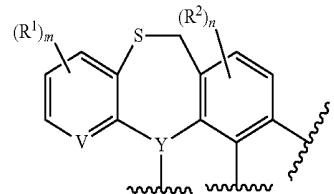
can have the structure

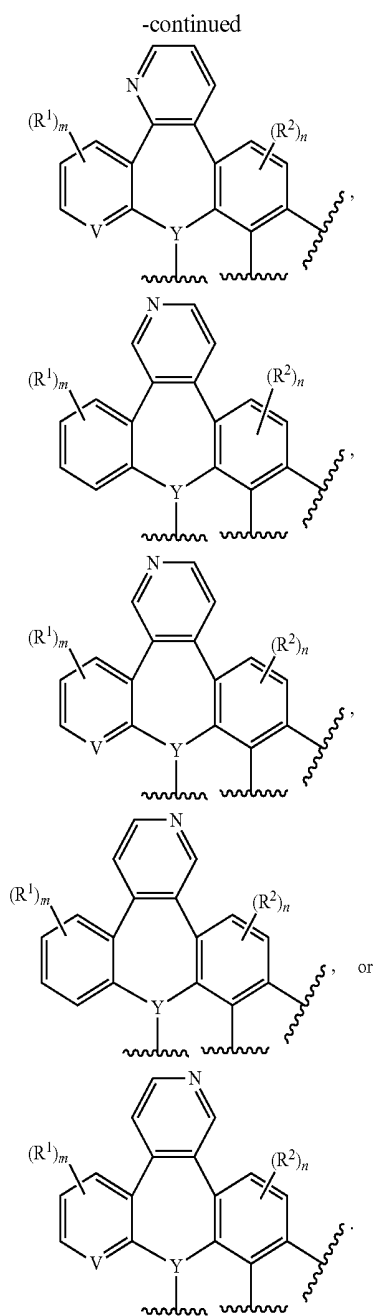


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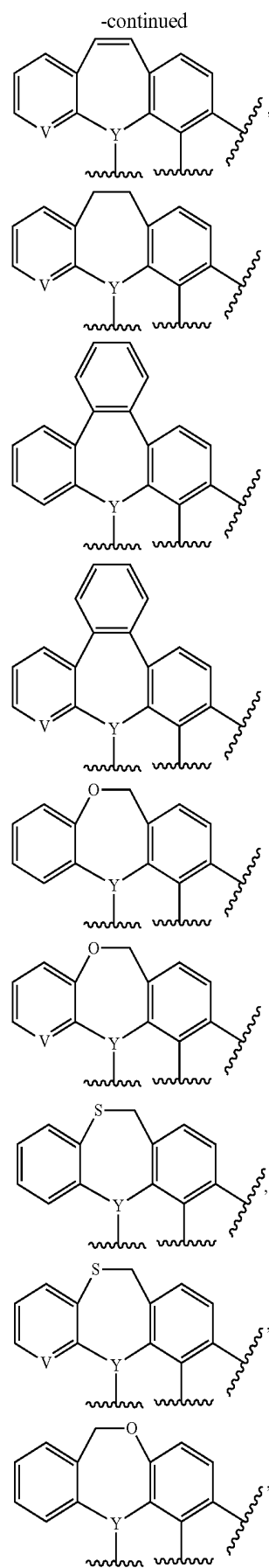
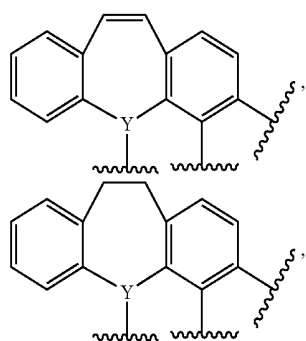


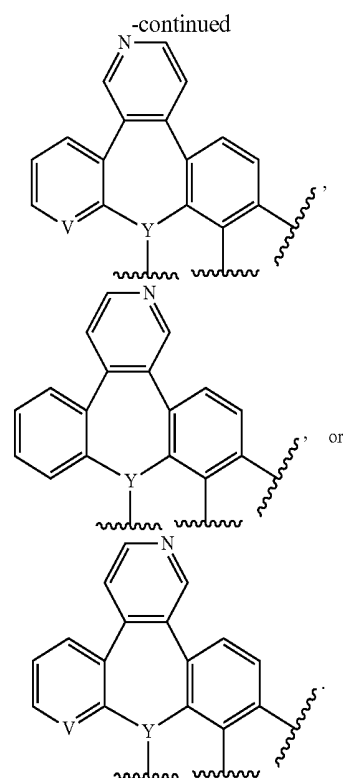
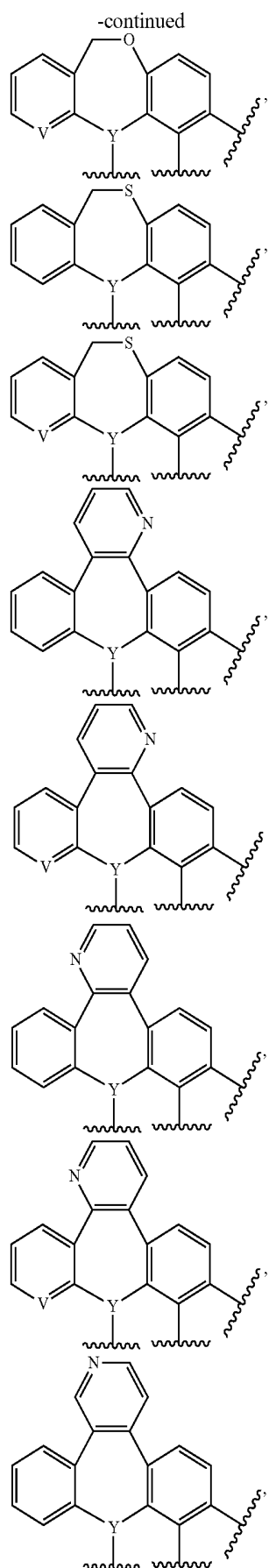
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Such as for example,

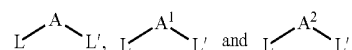




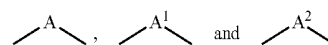
[0045] In one aspect, each of m, n, o, p, q, or r can independently be 0. In another aspect, each of m, n, o, p, q, or r can independently be 0 or 1. In another aspect, each of m, n, o, p, q, or r can independently be 1.

[0046] The compound of any one claims 1-10, wherein the compound comprises at least one phenyl and at least one pyridine.

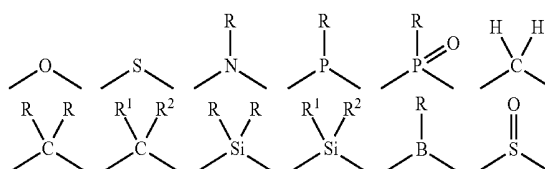
[0047] In one aspect, for any of the formulas illustrated in this disclosure,

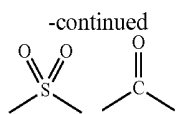


are intended to mean that L and L' are either linked directly or that L and L' are linked by a linkage group, wherein each of the linkage groups can independently be an oxygen (O), sulfur (S), nitrogen (N), phosphor (P), carbon (C), silicon (Si) or boron (B). In another aspect,



can each independently represent one or more of the following structures:

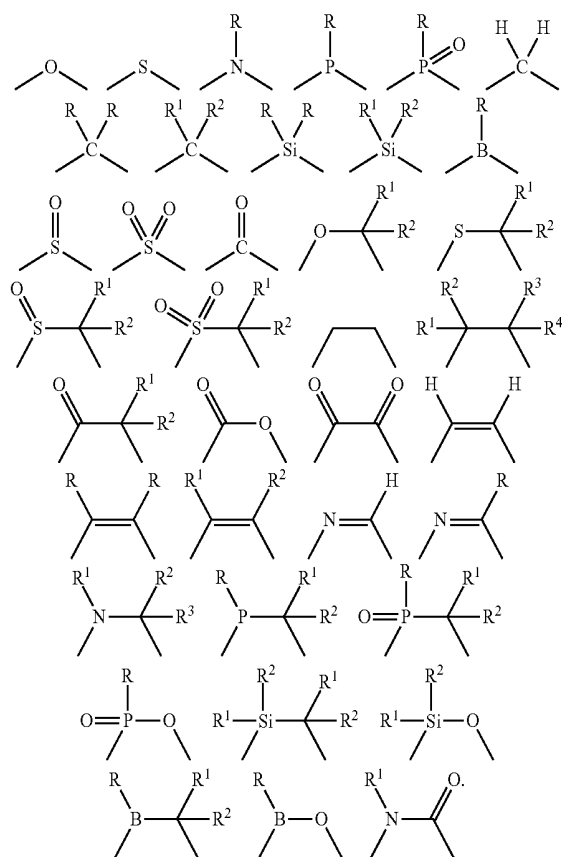




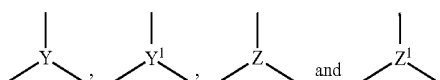
[0048] In another aspect, for any of the formulas illustrated in this disclosure,



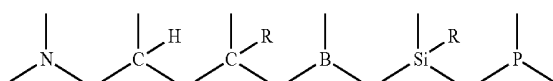
can represent one or more of the following structures:



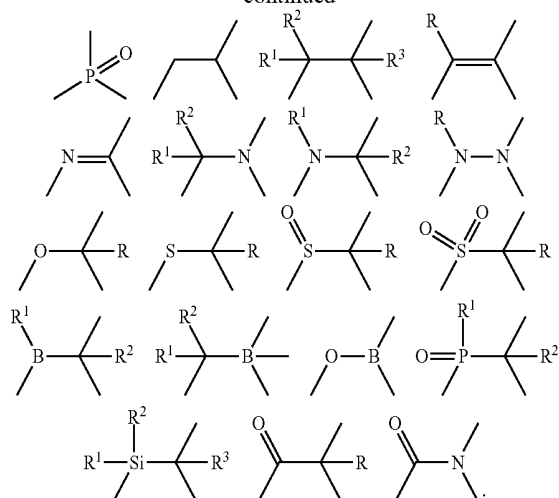
[0049] In yet another aspect, for any of the formulas illustrated in this disclosure,



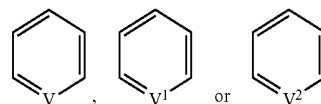
can each independently represent one or more of the following structures:



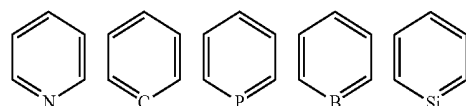
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[0050] In still another aspect,



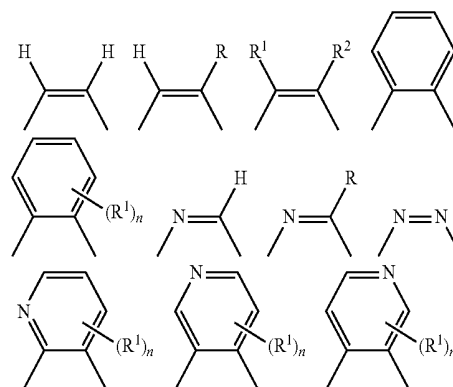
as shown in the disclosed compounds can be one of the following structures:

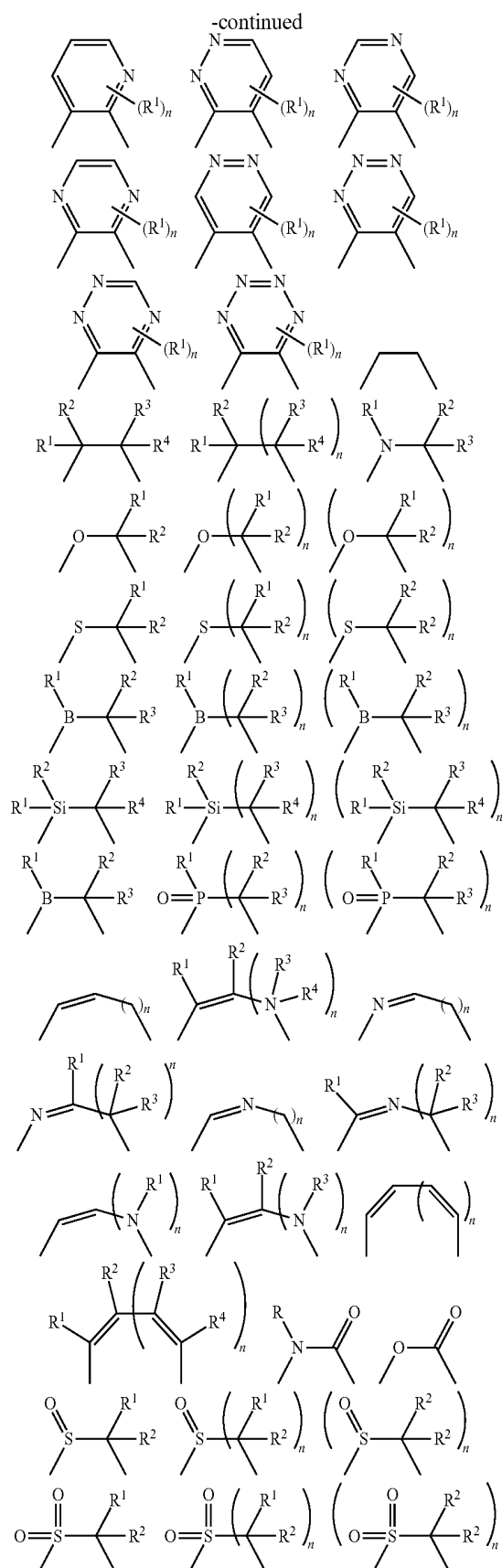


[0051] In still another aspect, for any of the formulas illustrated in this disclosure,



can represent one or more of the following structures:

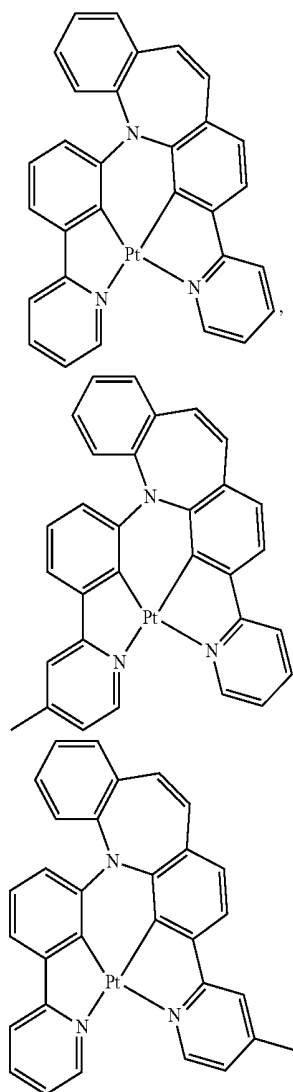


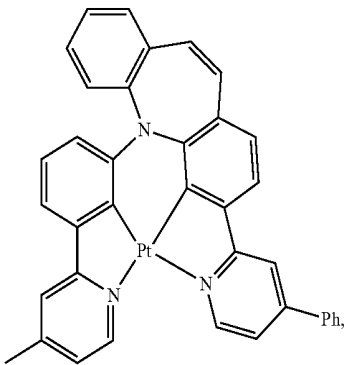
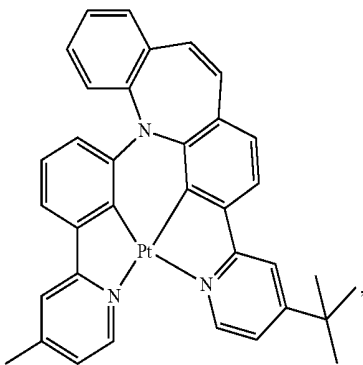
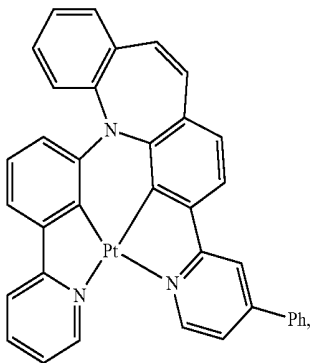
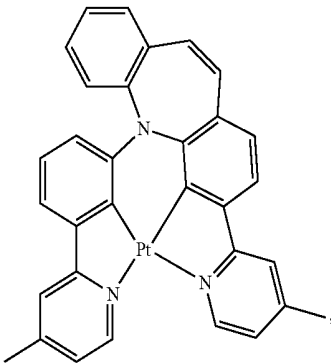
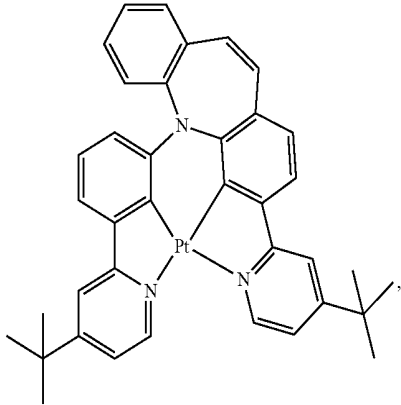
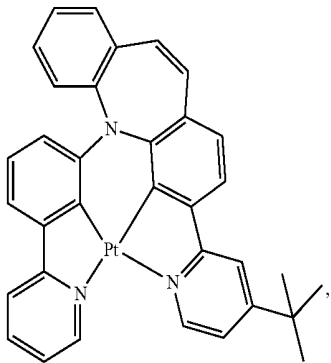
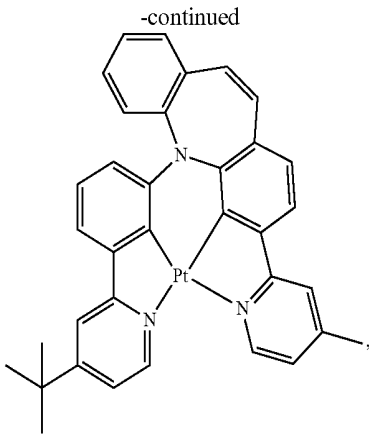
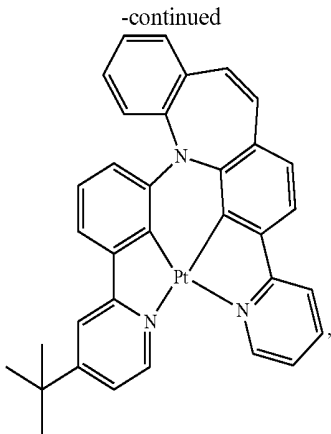


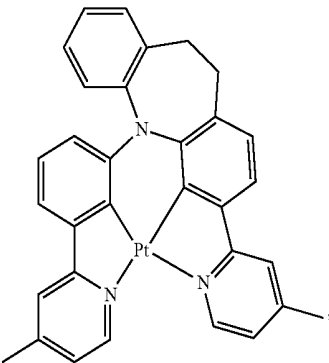
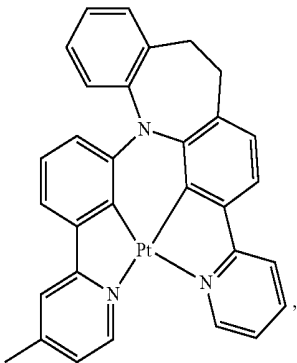
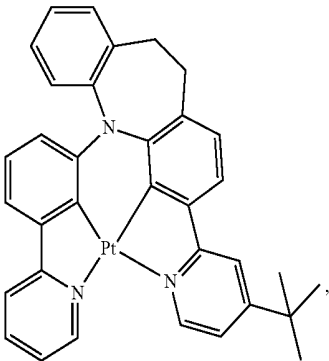
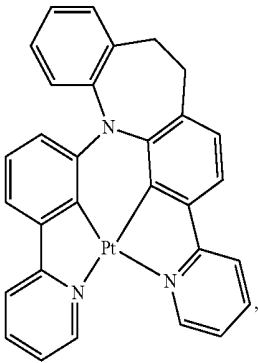
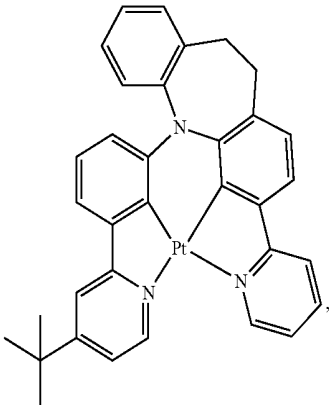
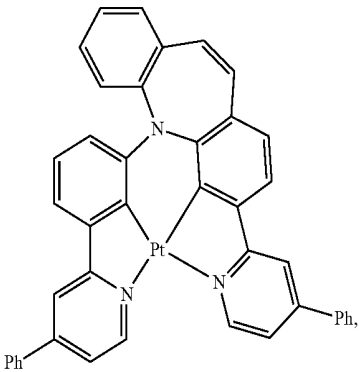
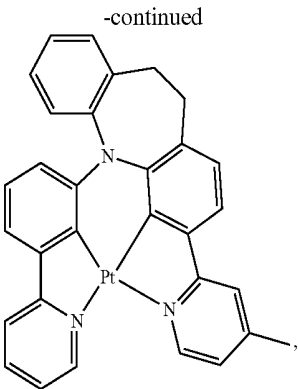
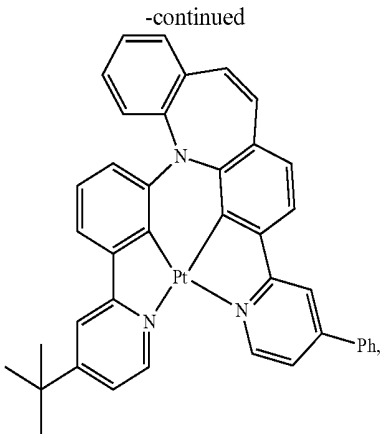
[0052] In one aspect, each of R^1 , R^2 , R^3 , R^4 , and R^5 can independently be hydrogen, aryl, cycloalkyl, cycloalkenyl,

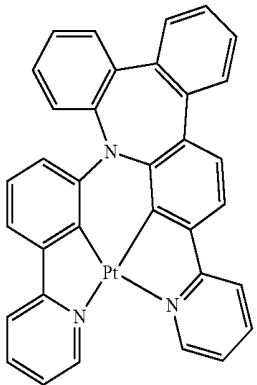
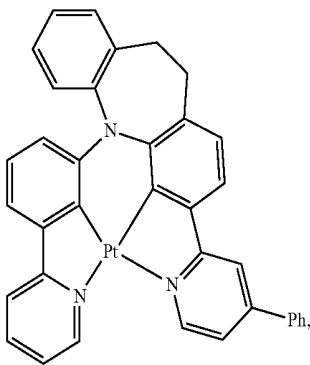
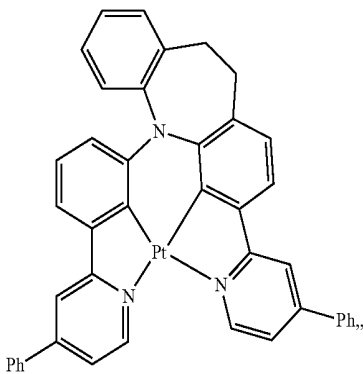
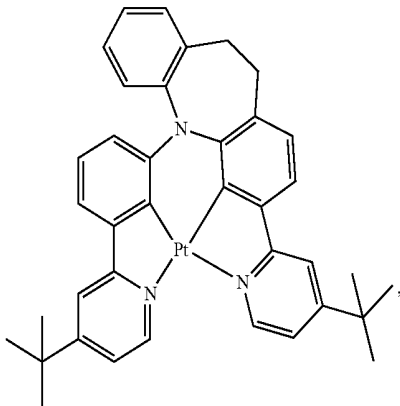
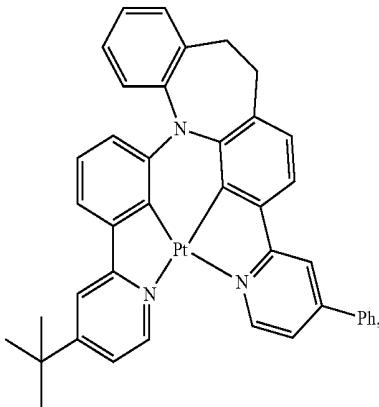
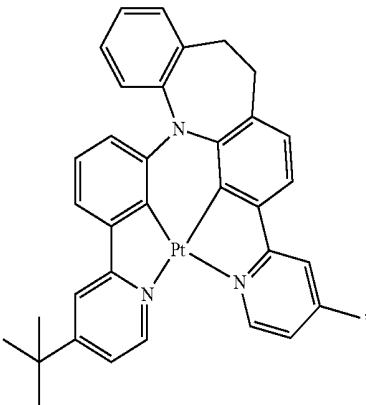
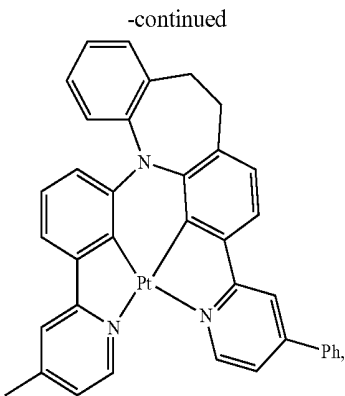
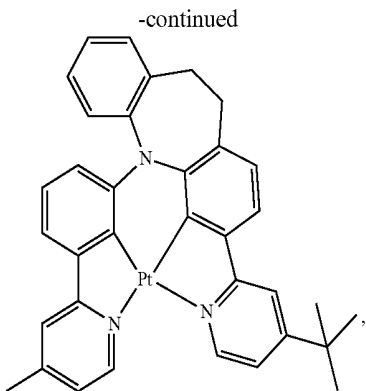
heterocyclyl, heteroaryl, alkyl, alkenyl, alkynyl, deuterium, halogen, hydroxyl, thiol, nitro, cyano, amino, a mono- or di-alkylamino, a mono- or diaryl amino, alkoxy, aryloxy, haloalkyl, aralkyl, ester, nitrile, isonitrile, alkoxycarbonyl, acylamino, alkoxycarbonylamino, aryloxycarbonylamino, sulfonylamino, sulfamoyl, carbamoyl, alkylthio, sulfinyl, ureido, phosphoramidate, mercapto, sulfo, carboxyl, hydrazino, substituted silyl, or polymerizable, or any conjugate or combination thereof. In another aspect, each of R^1 , R^2 , R^3 , R^4 , and R^5 can independently be aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, alkyl, alkenyl, alkynyl, halogen, hydroxyl, thiol, nitro, amino, haloalkyl, or any conjugate or a combination thereof. In yet another aspect, each of R^1 , R^2 , R^3 , R^4 , and R^5 can independently be aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, alkyl, alkenyl, halogen, or hydroxyl, or a combination thereof.

[0053] In one aspect, for any of the metal complexes illustrated in this disclosure, can comprise one or more of the following structures. In another aspect, they can also comprise other structures or portions thereof not specifically recited herein, and the present invention is not intended to be limited to those structures or portions thereof specifically recited.

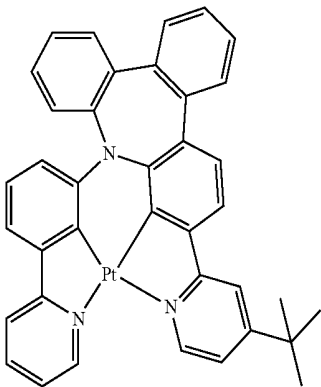
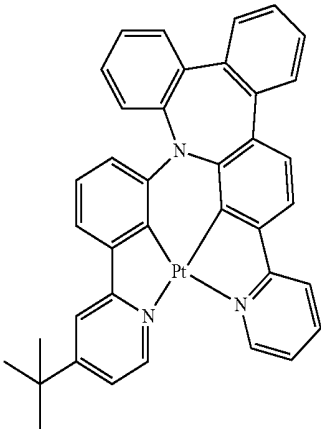
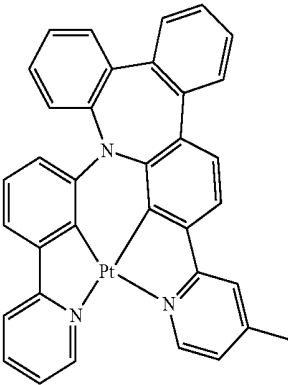
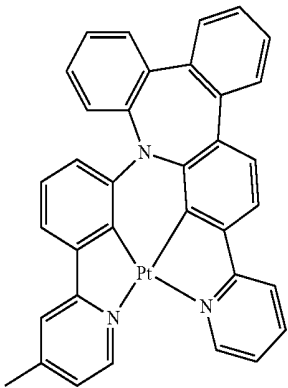




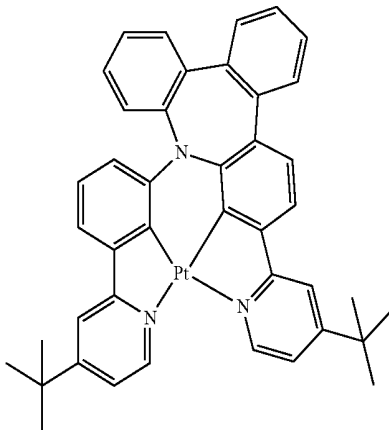
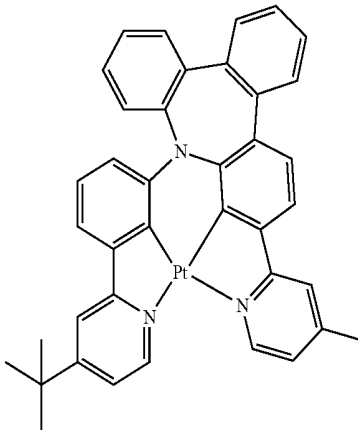
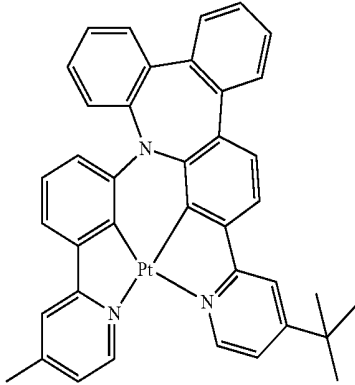
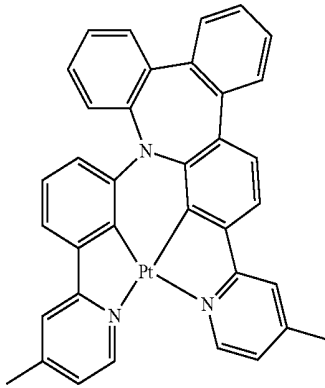




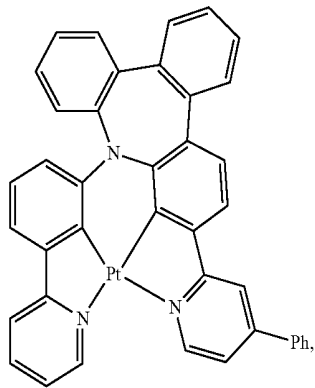
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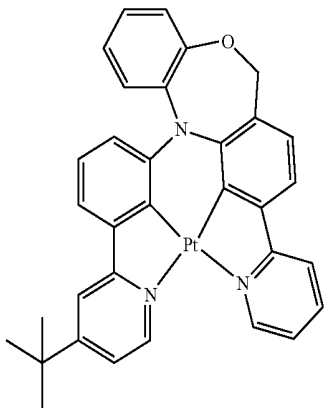
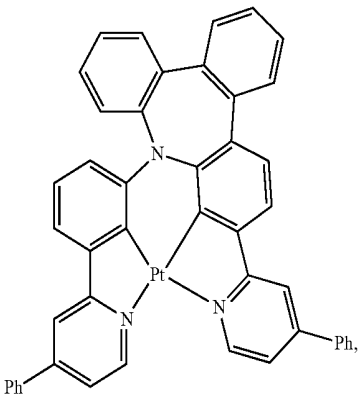
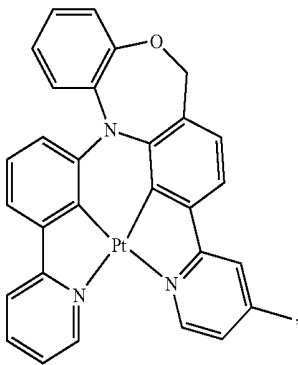
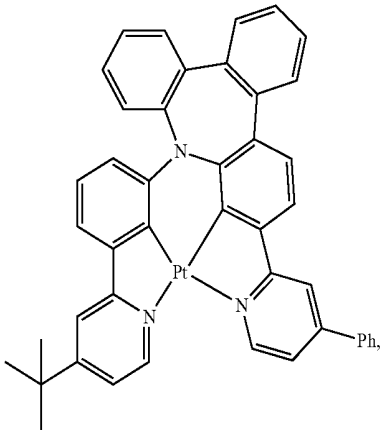
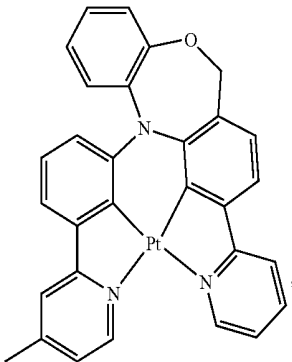
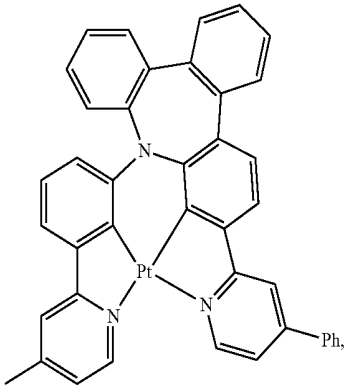
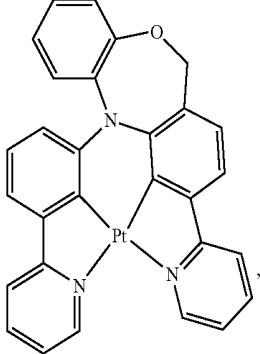
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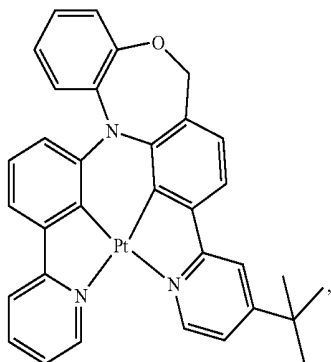
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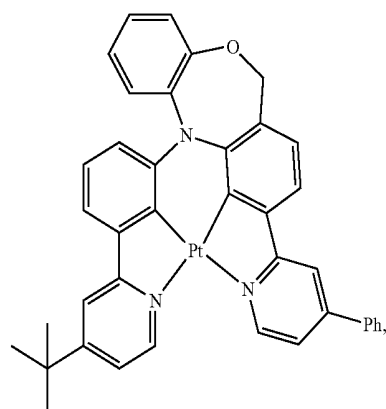
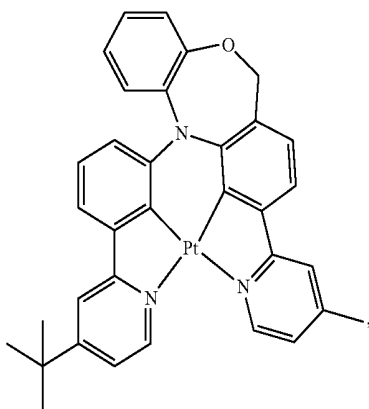
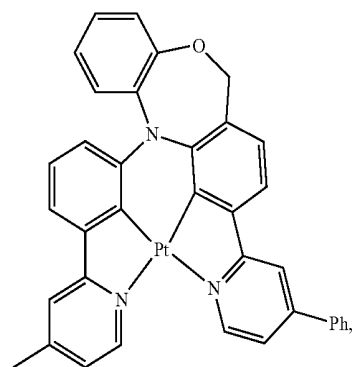
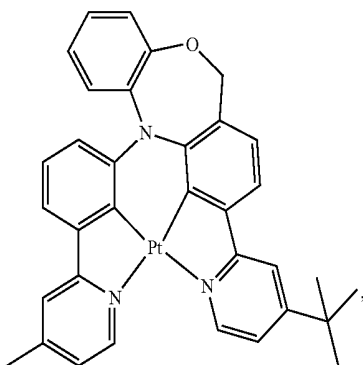
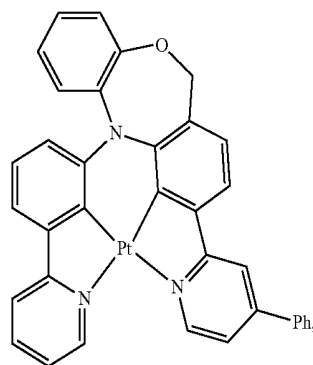
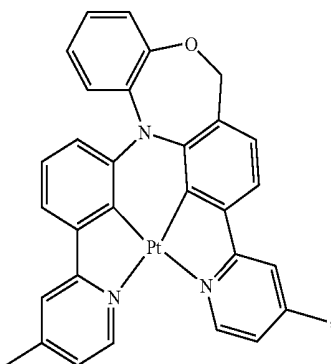
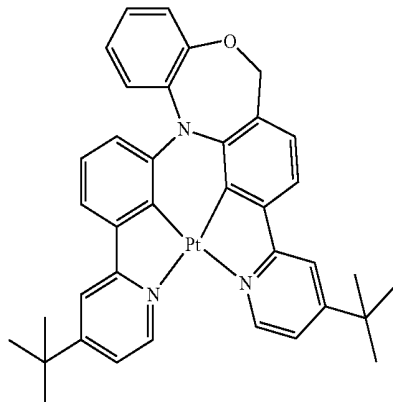
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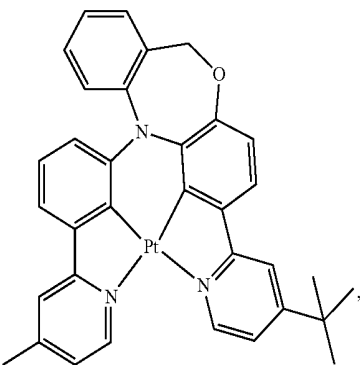
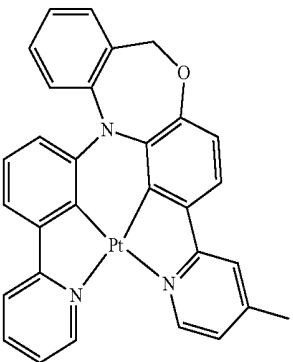
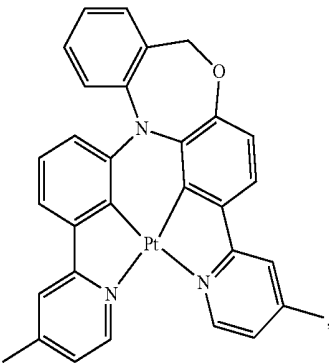
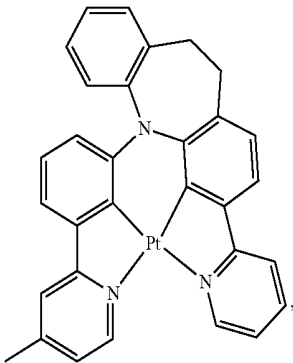
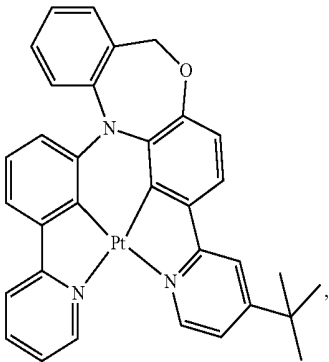
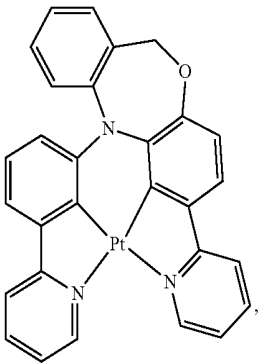
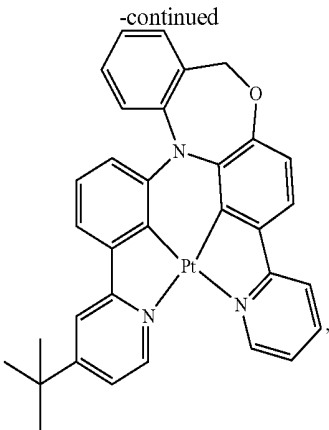
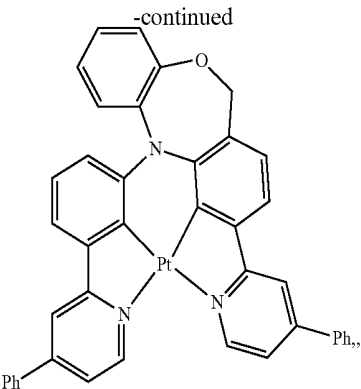


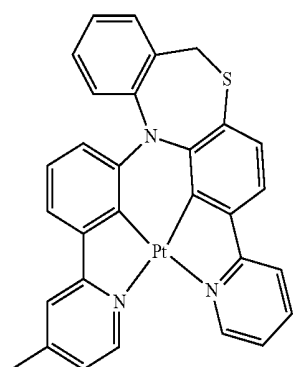
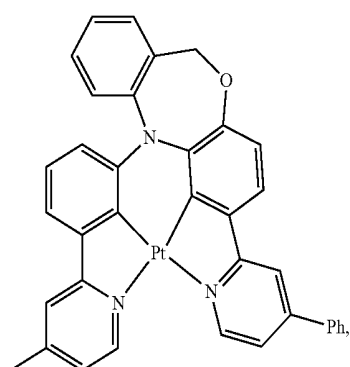
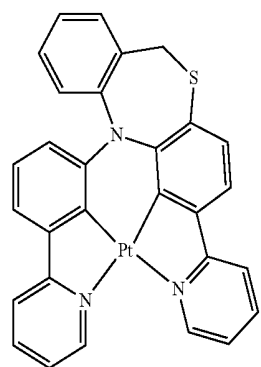
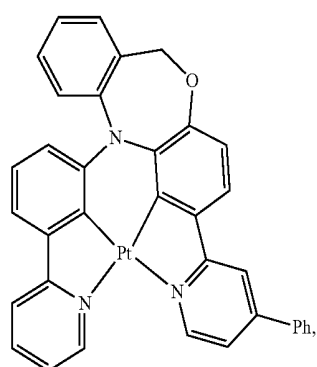
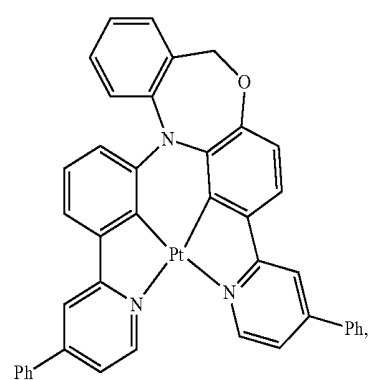
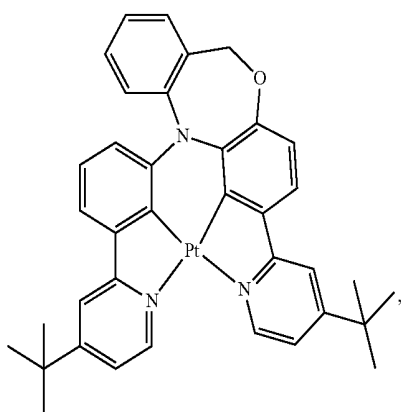
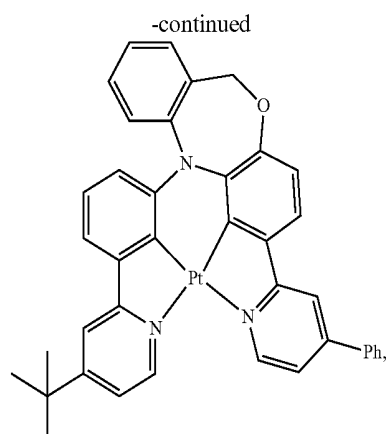
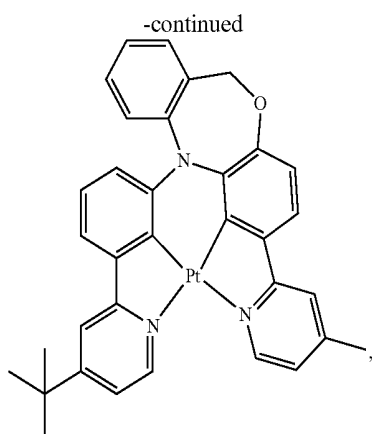
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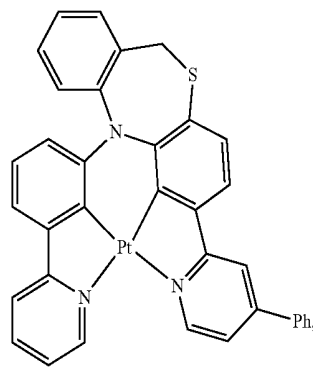
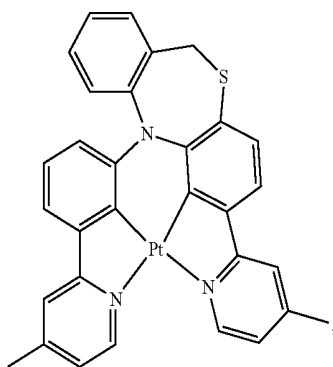
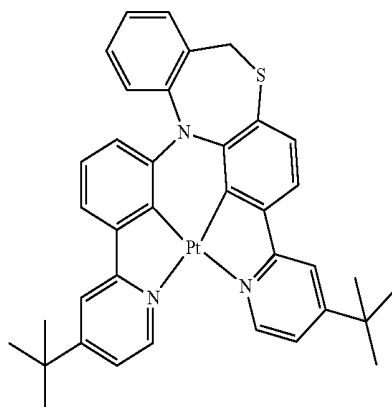
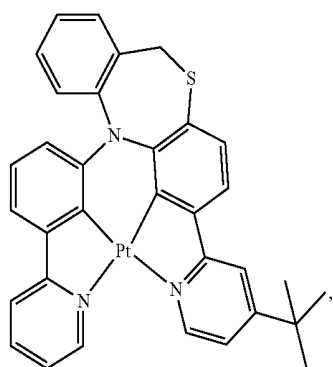
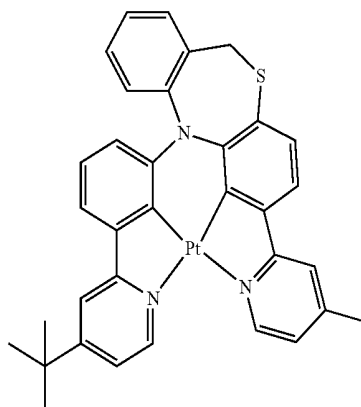
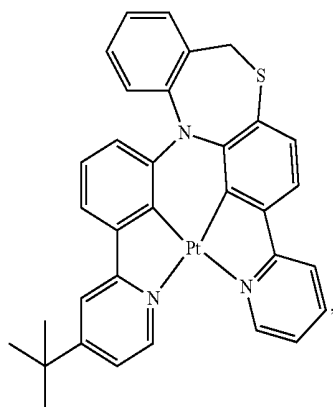
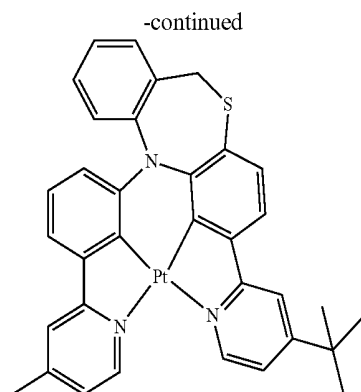
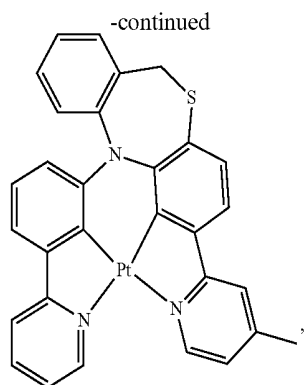


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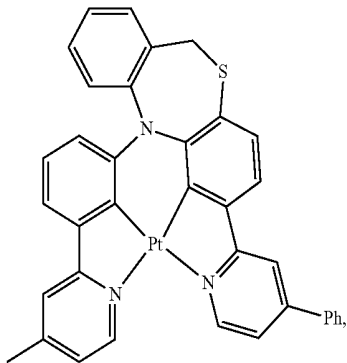




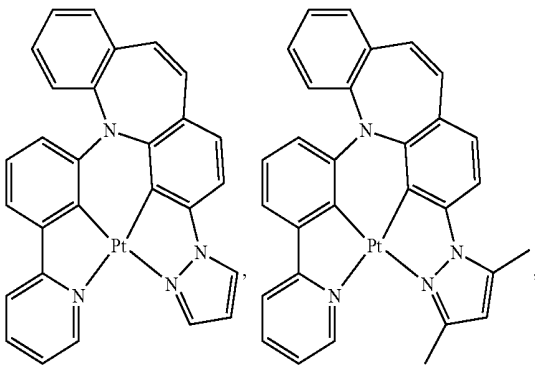
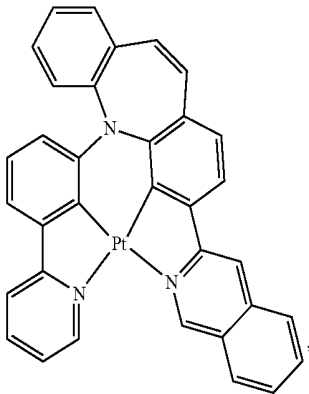
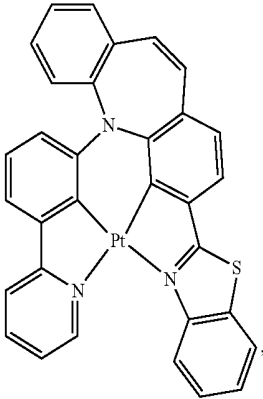
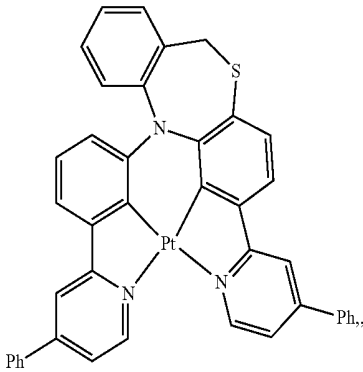
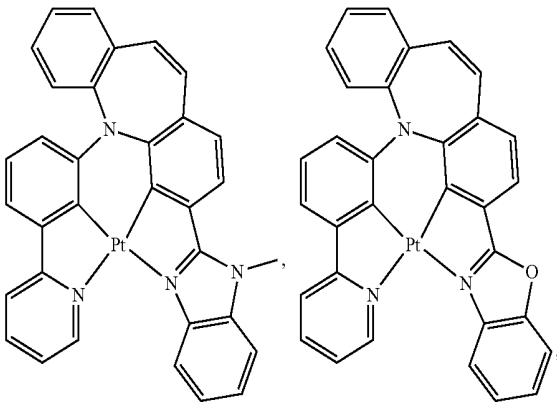
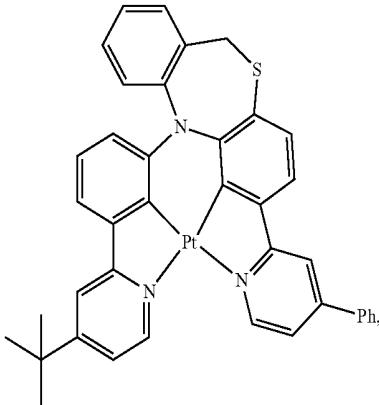
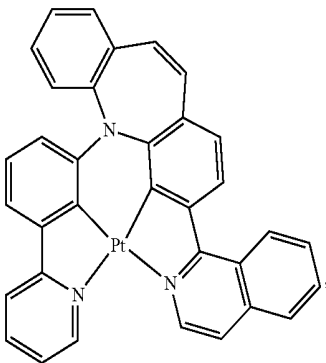




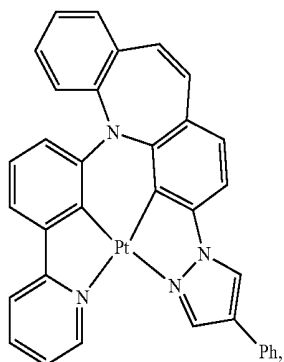
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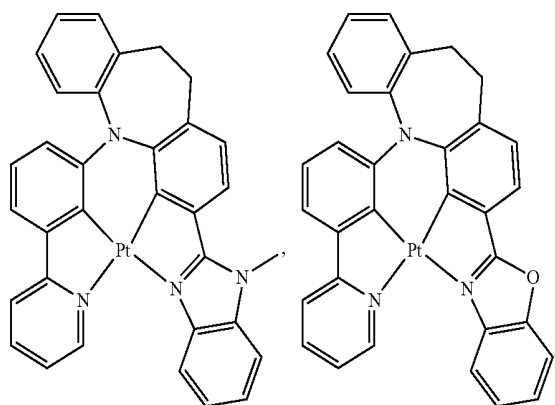
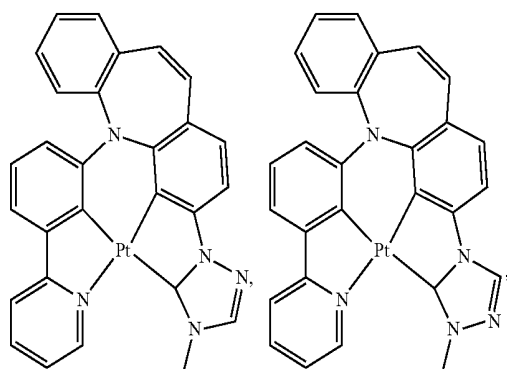
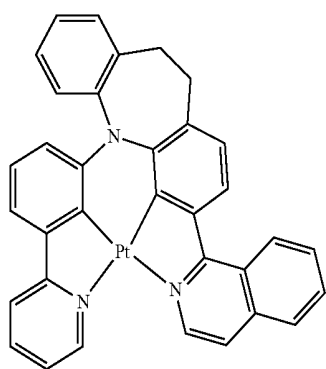
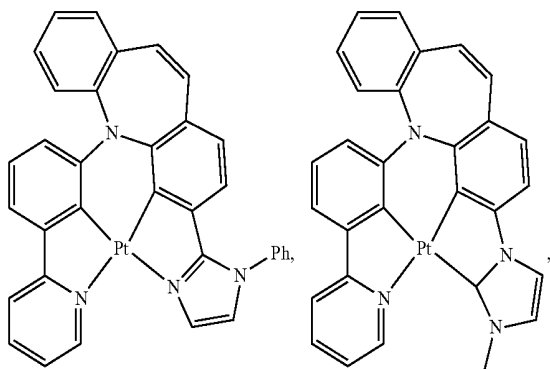
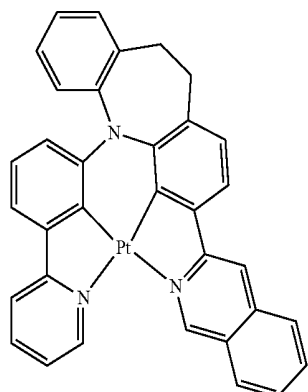
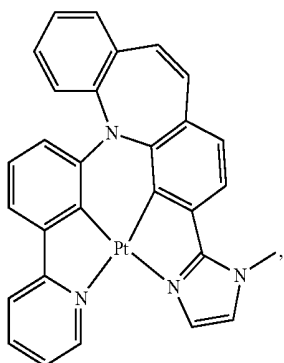
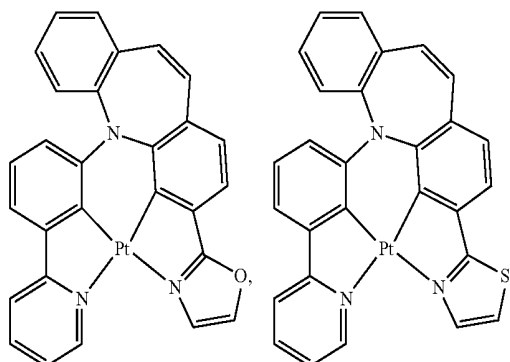
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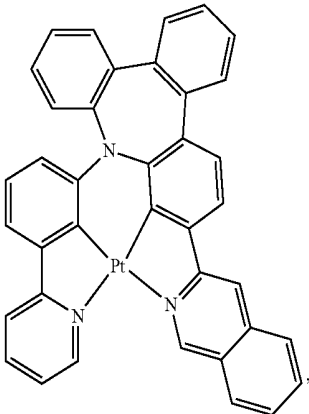
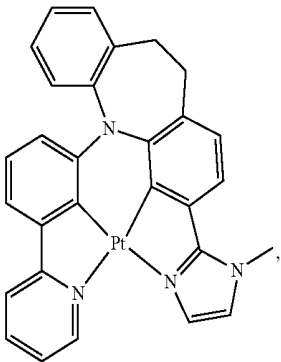
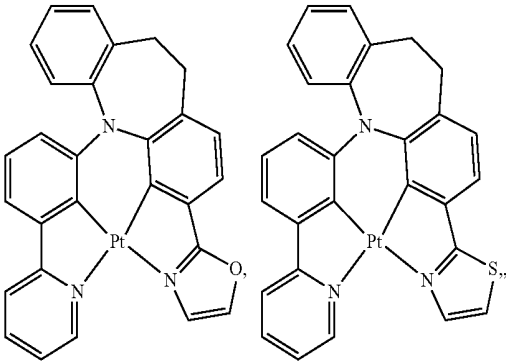
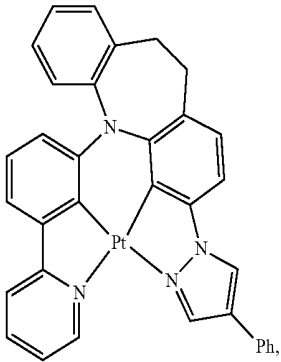
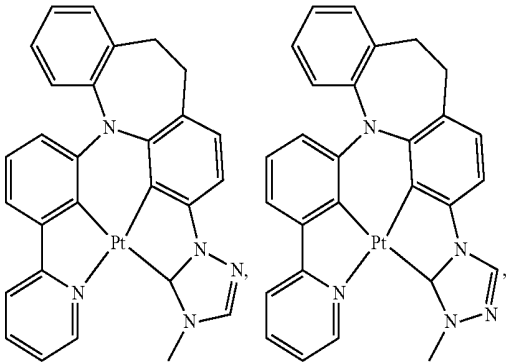
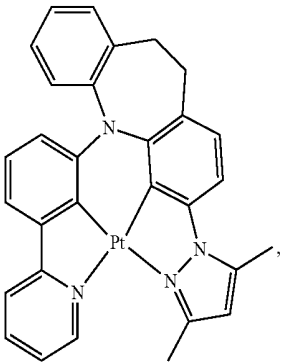
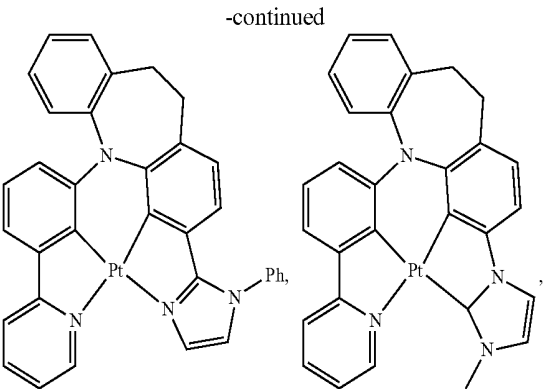
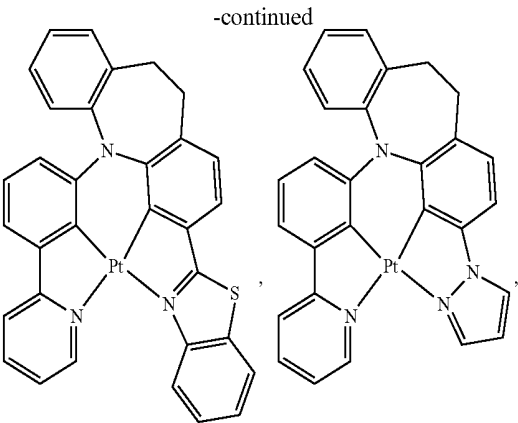


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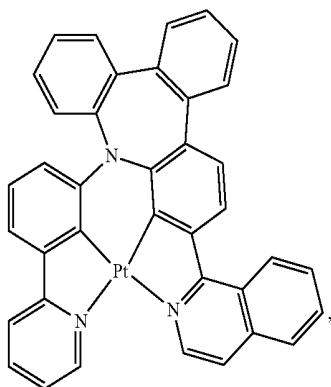


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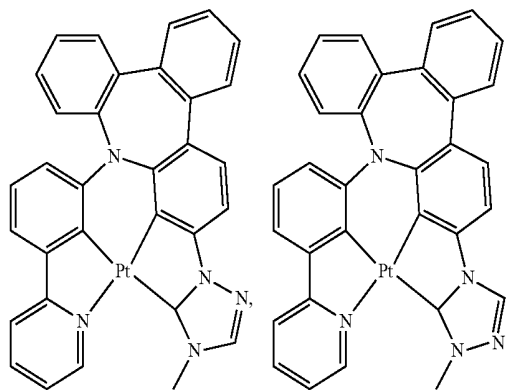
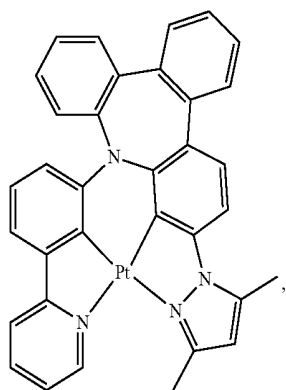
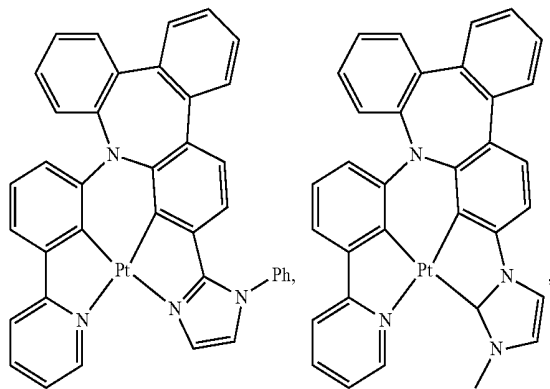
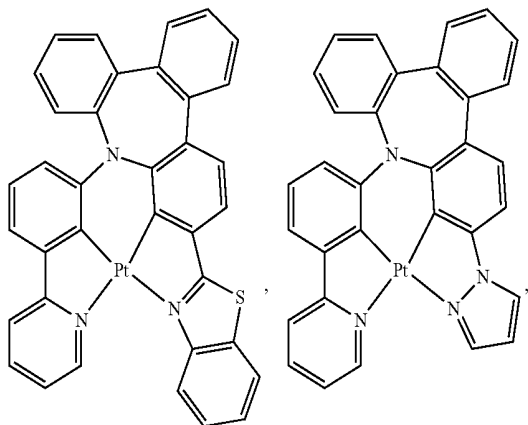
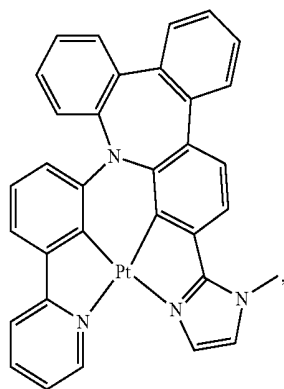
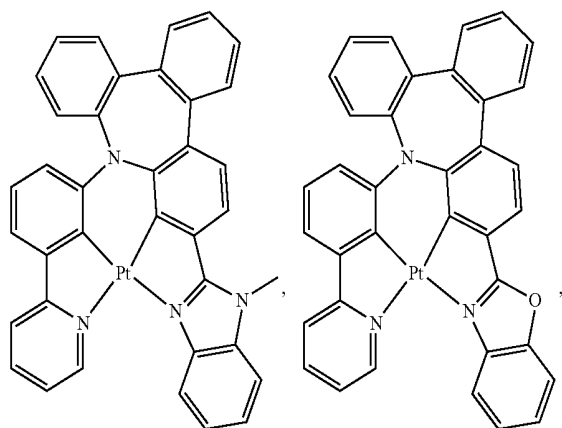
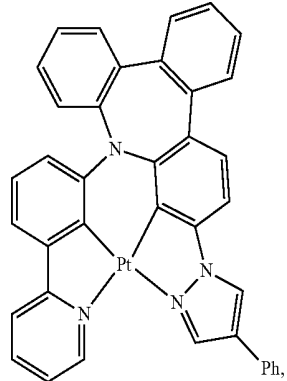




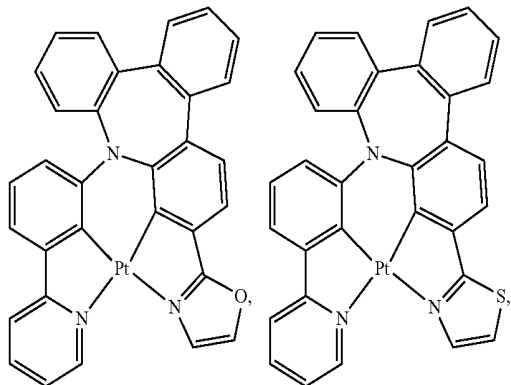
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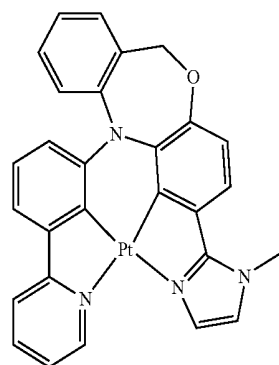
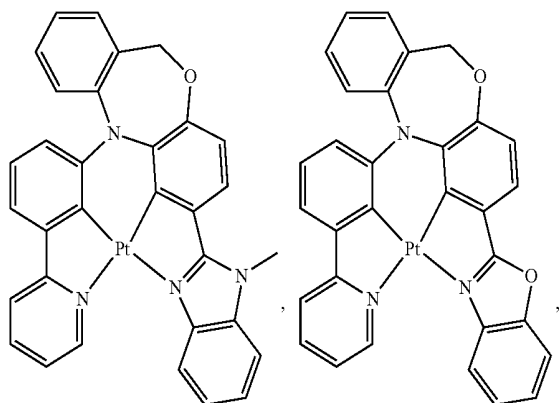
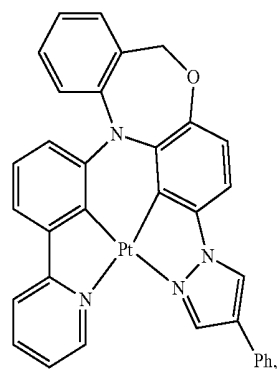
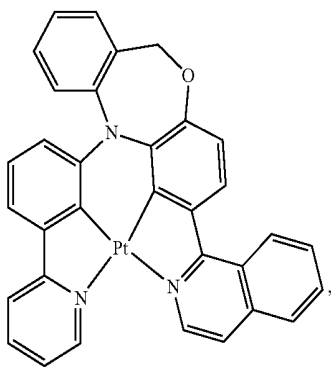
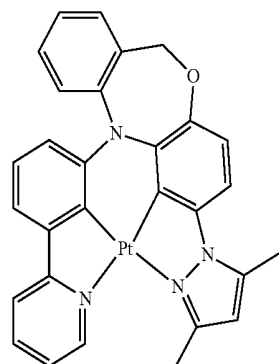
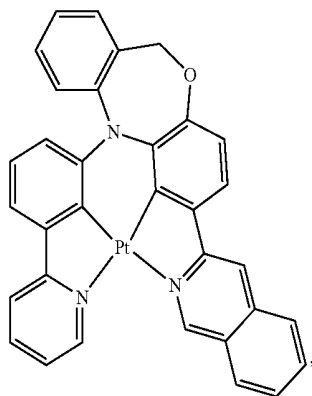
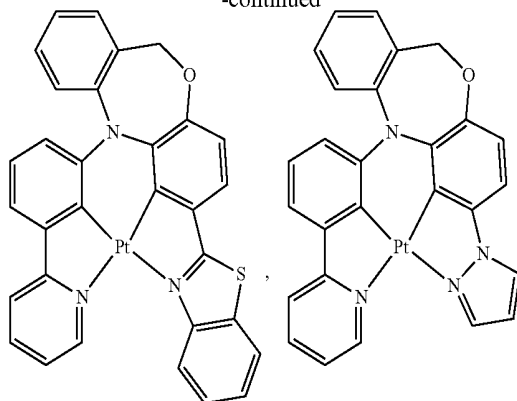
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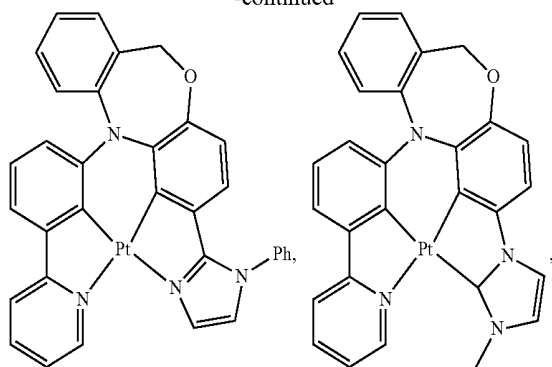
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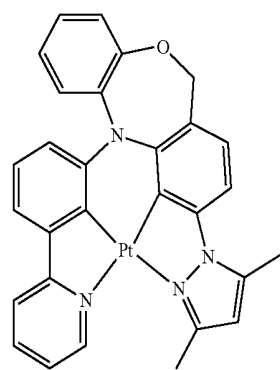
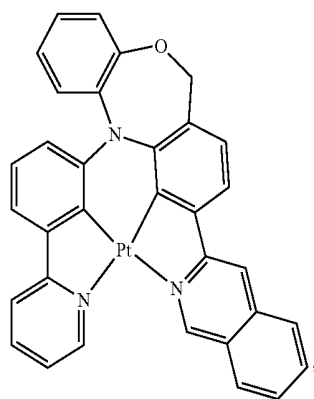
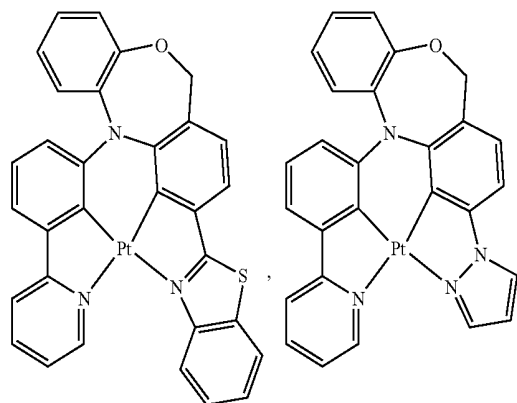
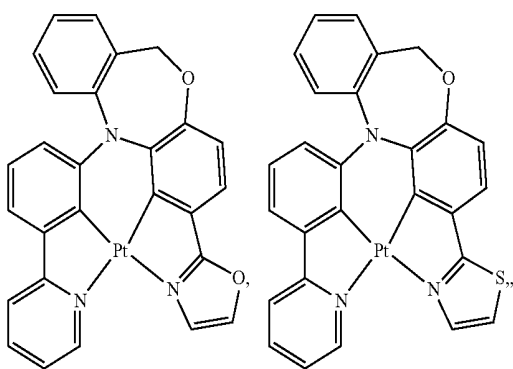
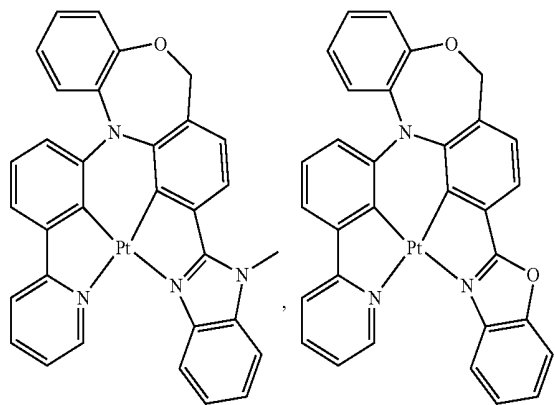
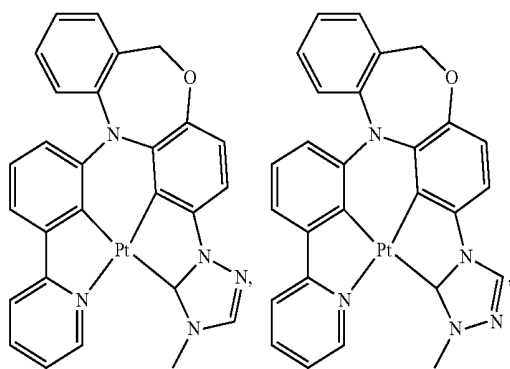
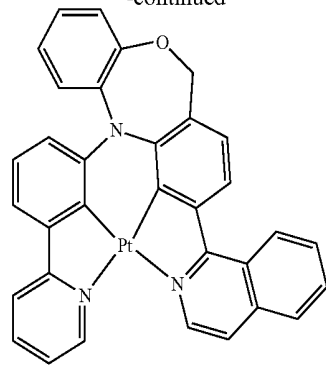
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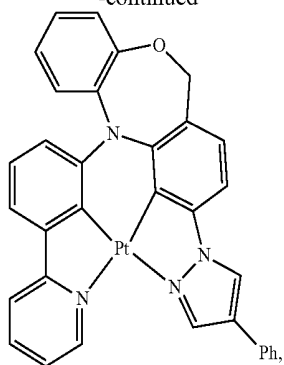
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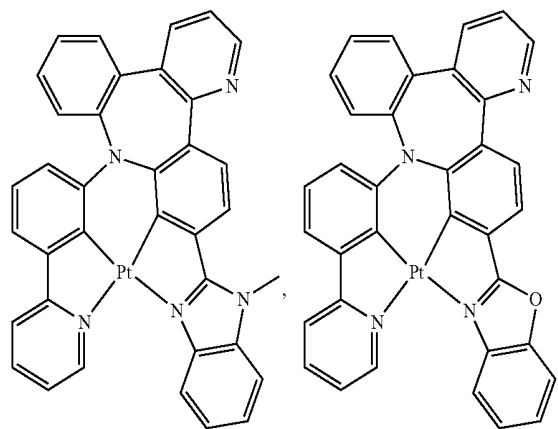
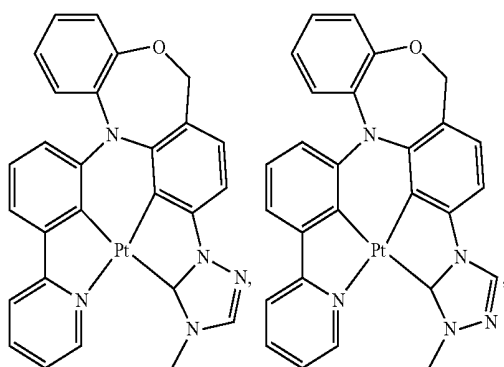
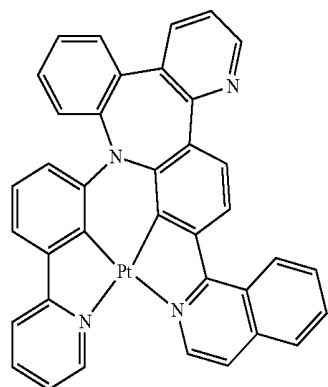
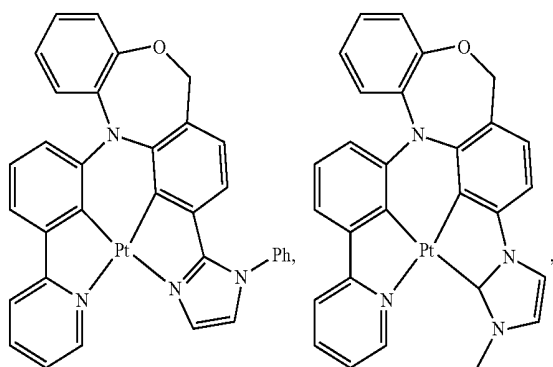
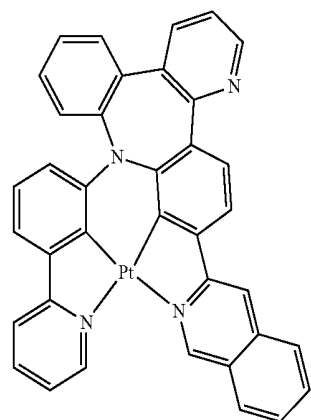
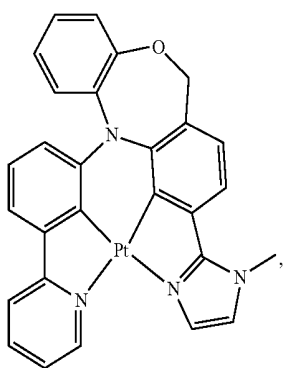
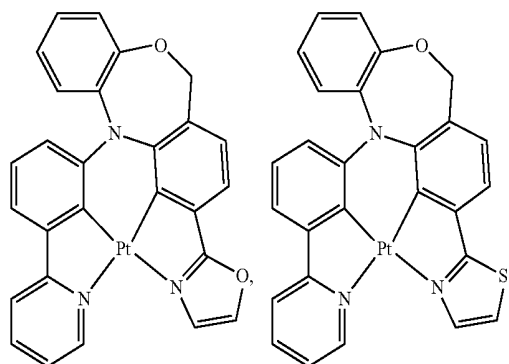
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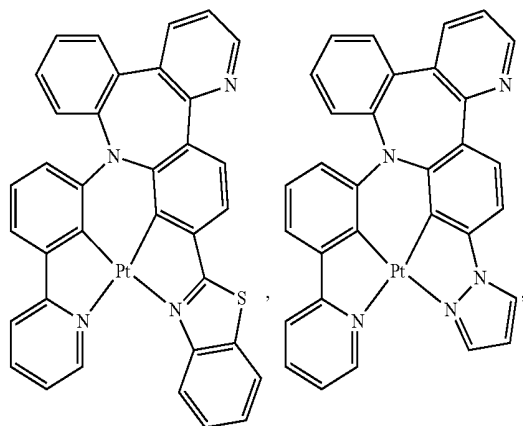
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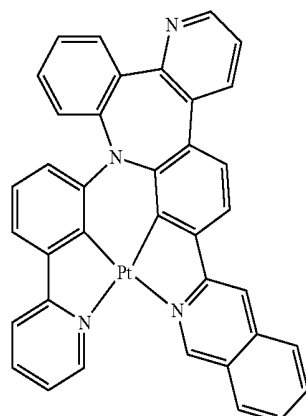
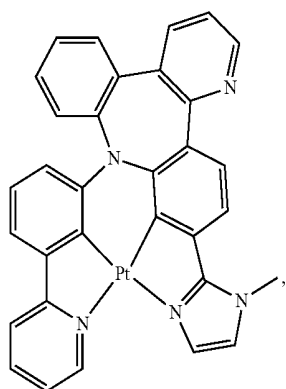
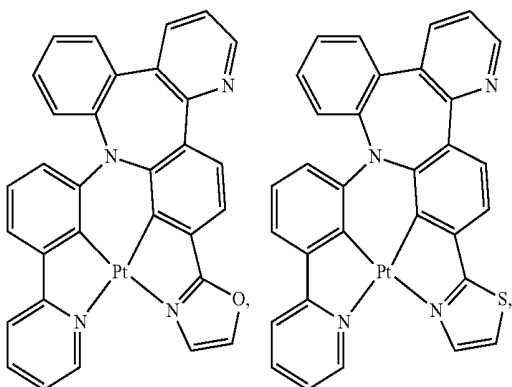
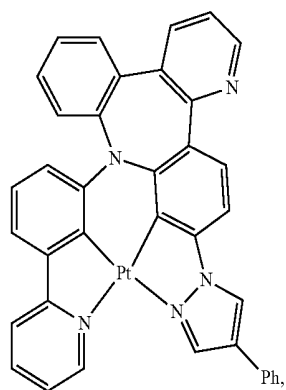
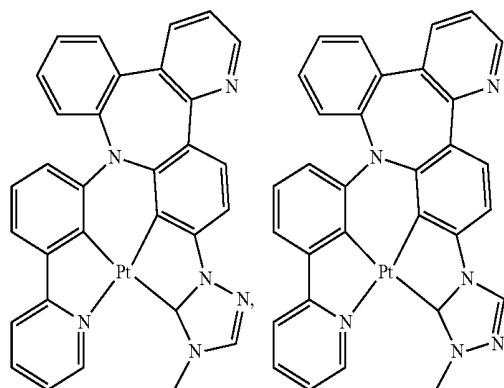
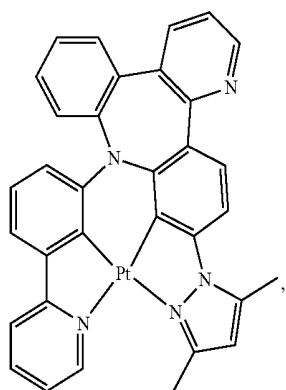
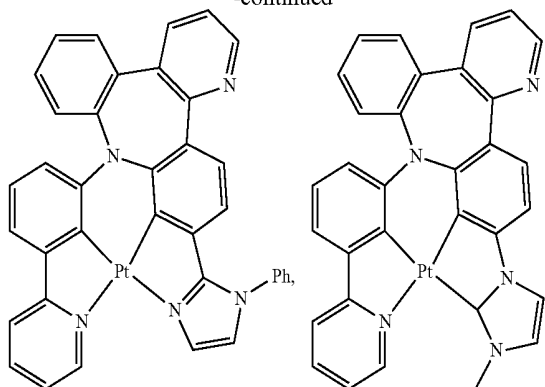
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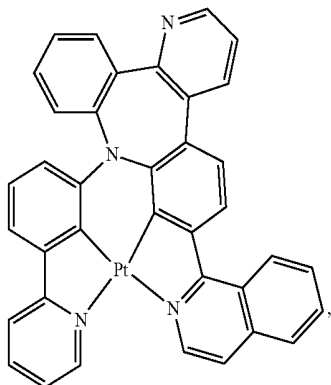
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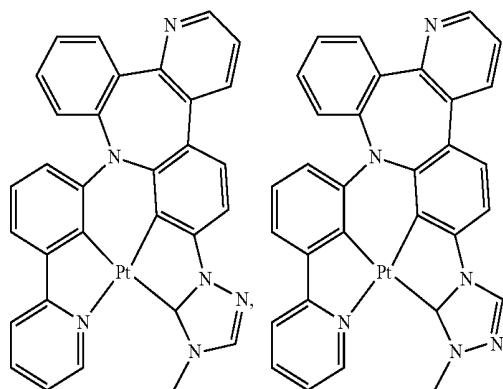
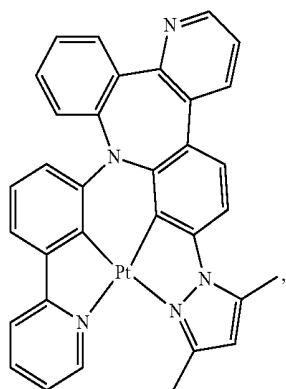
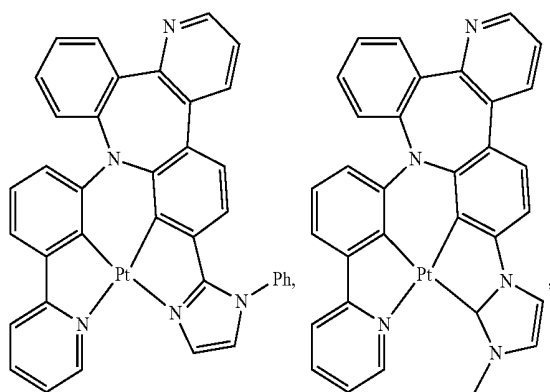
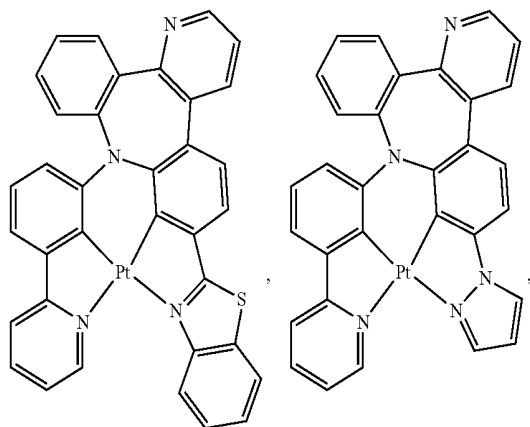
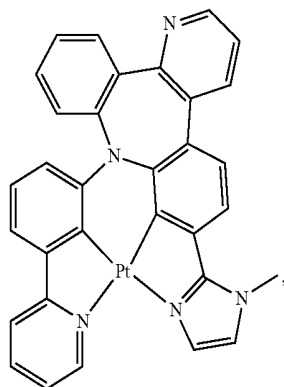
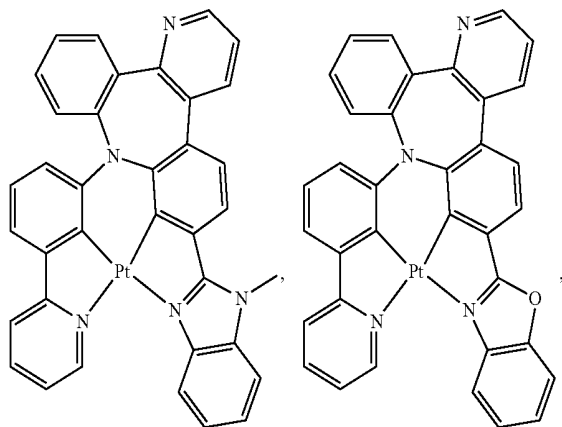
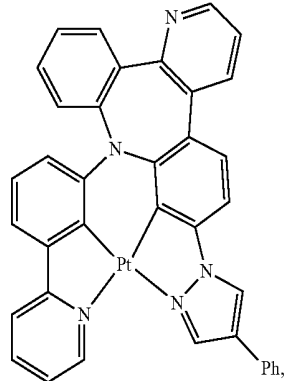
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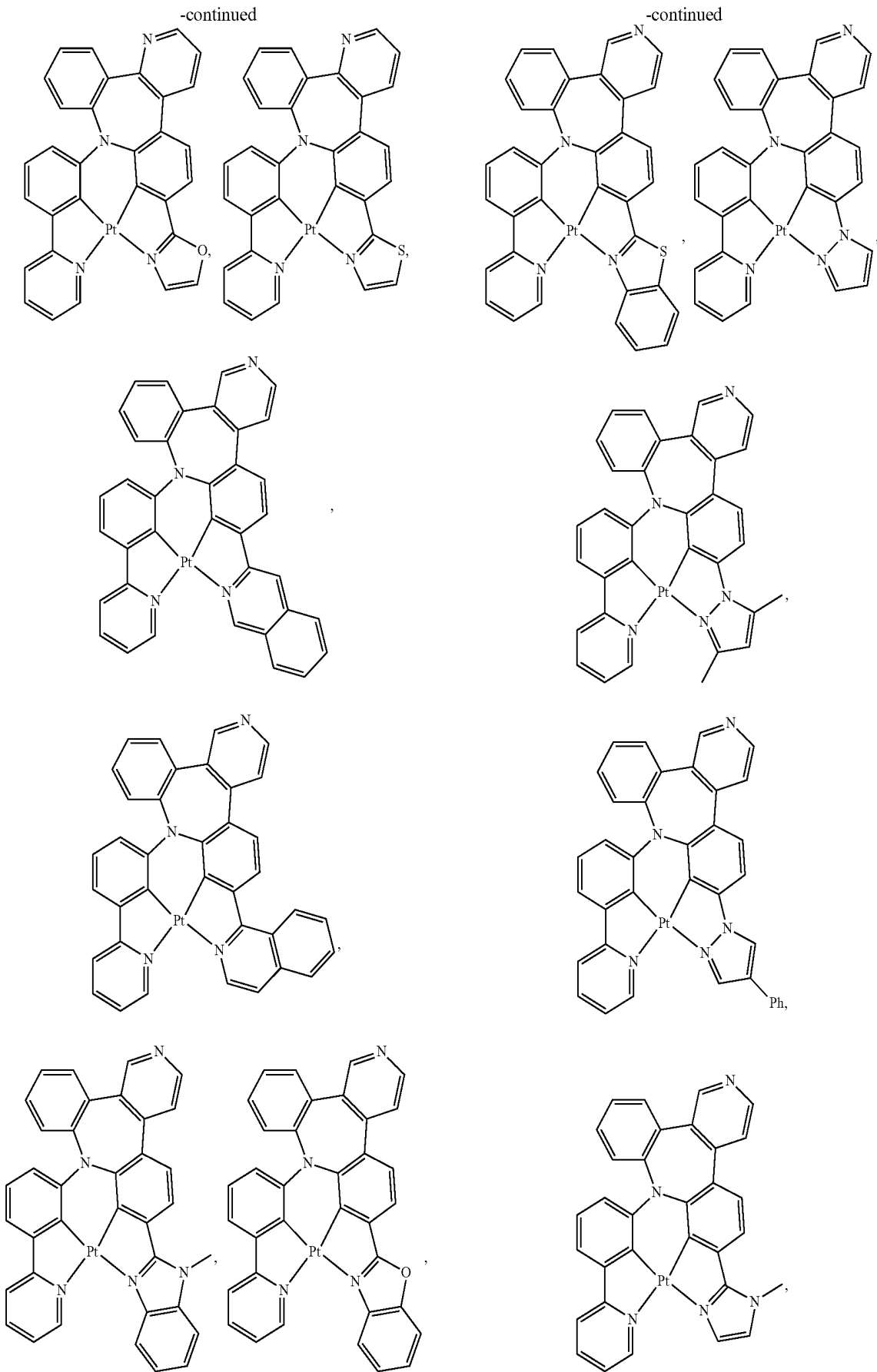


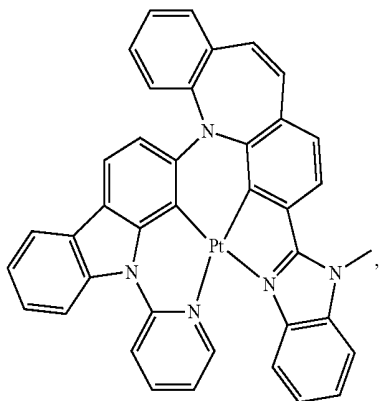
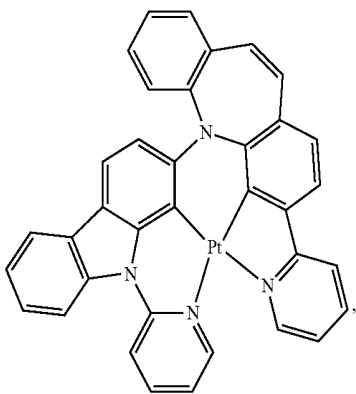
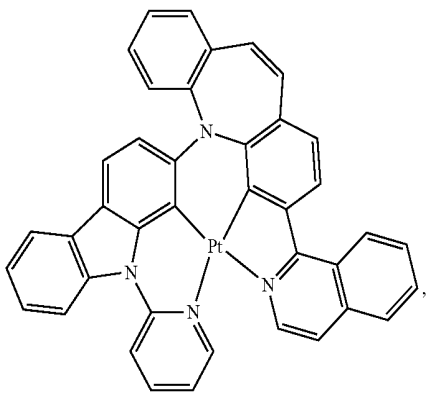
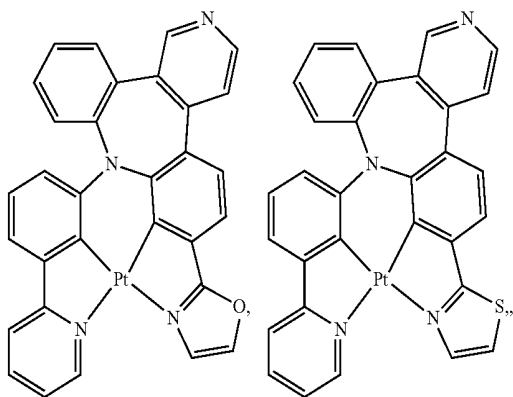
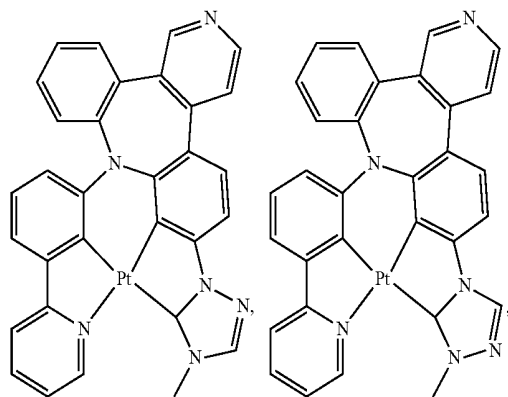
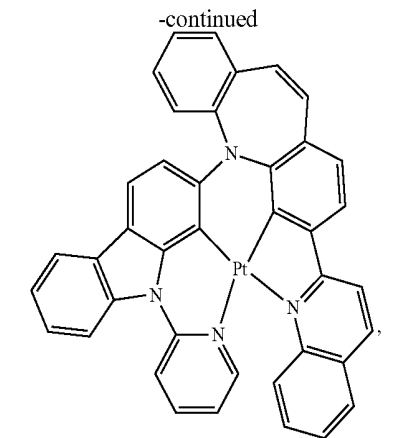
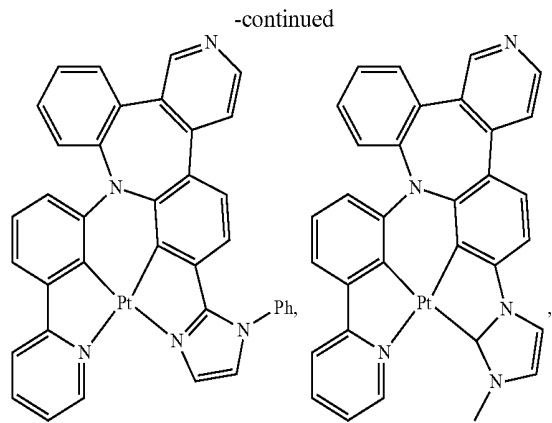
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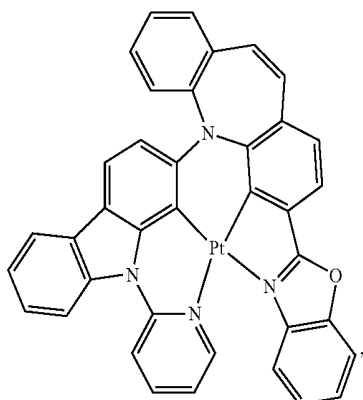
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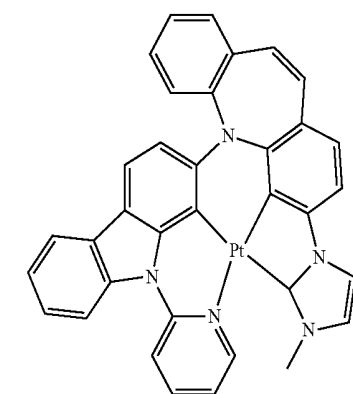
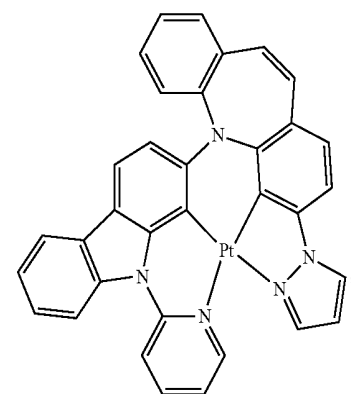
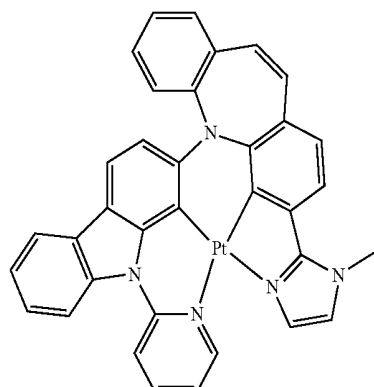
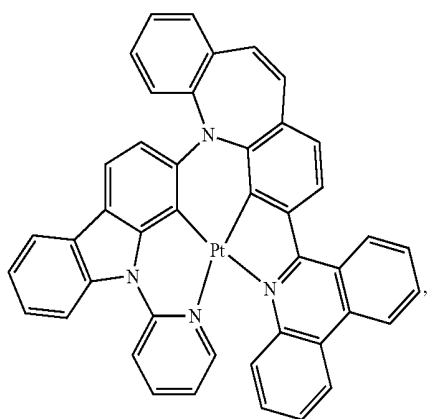
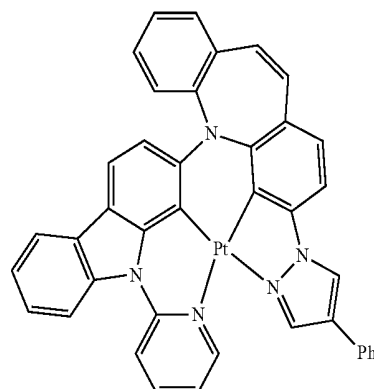
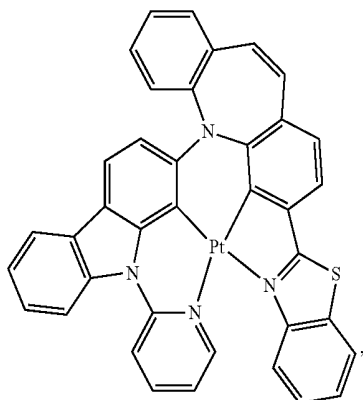
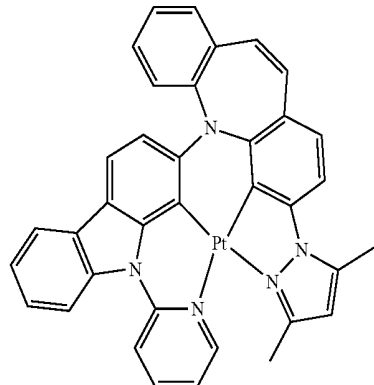




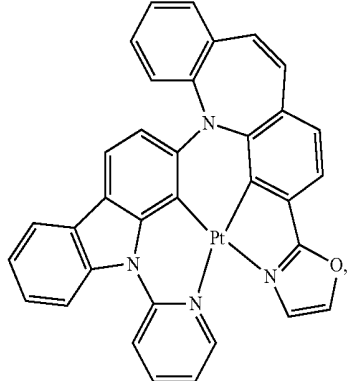
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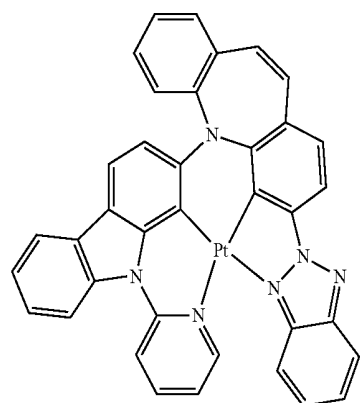
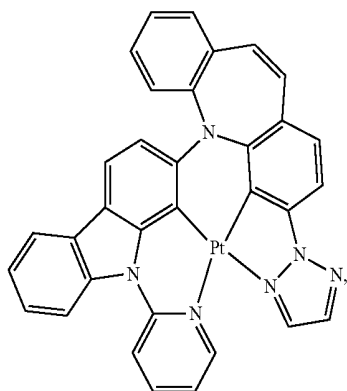
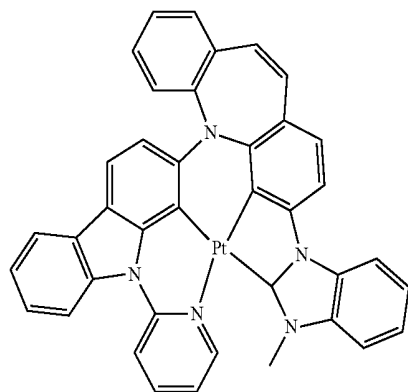
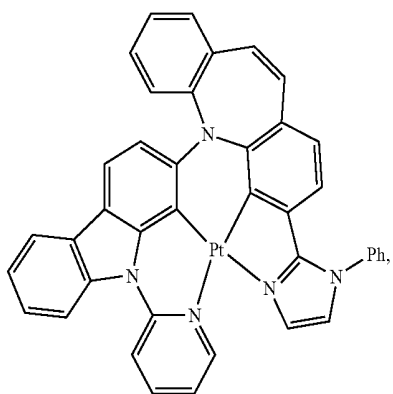
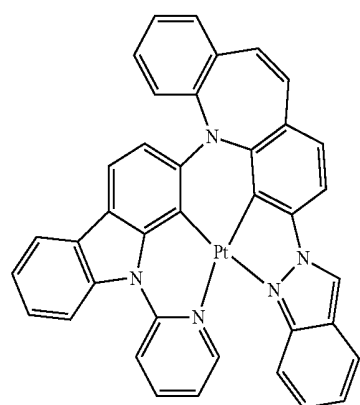
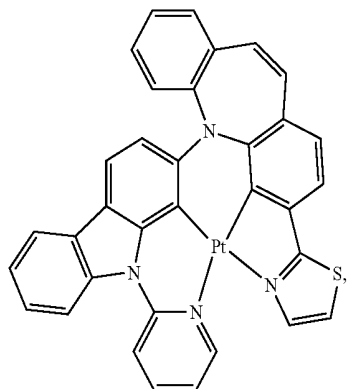
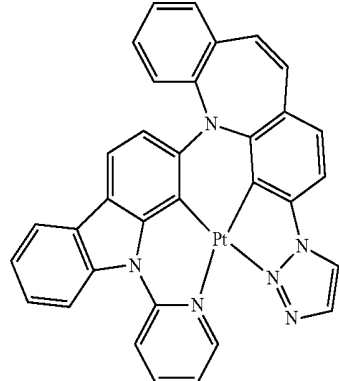
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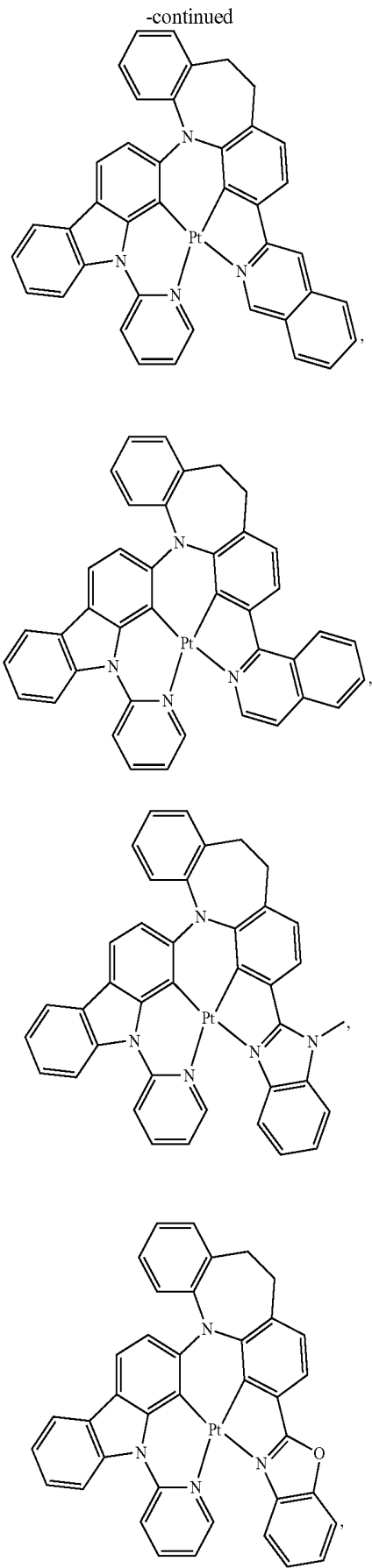
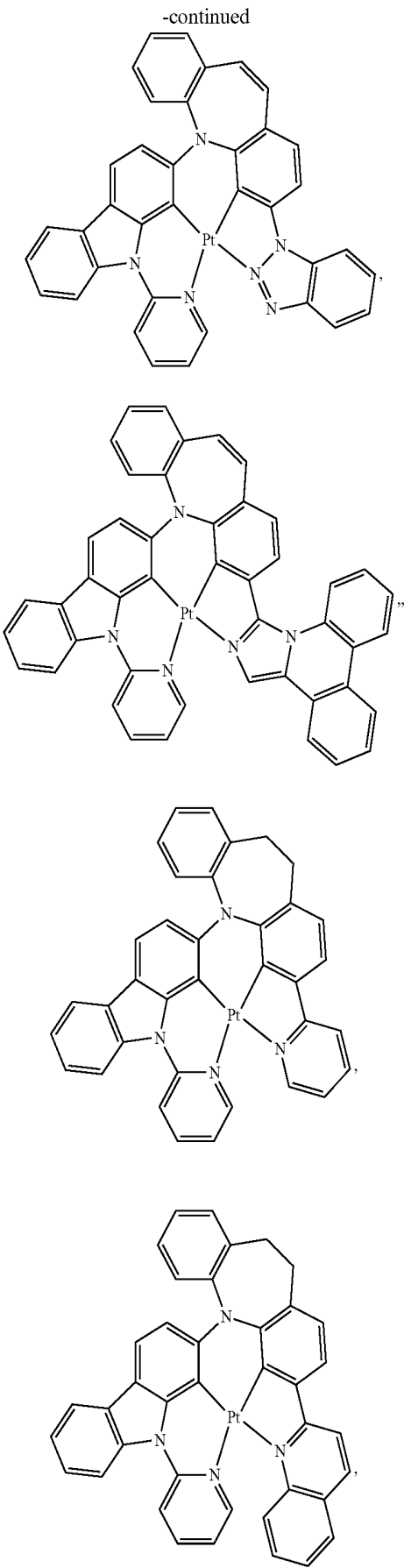


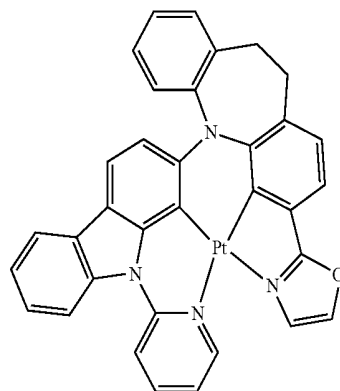
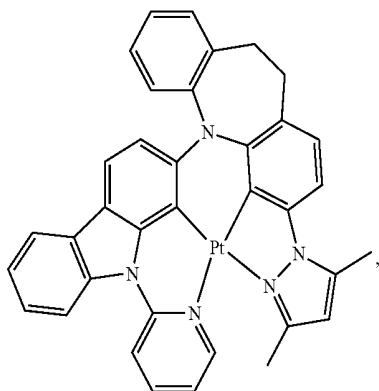
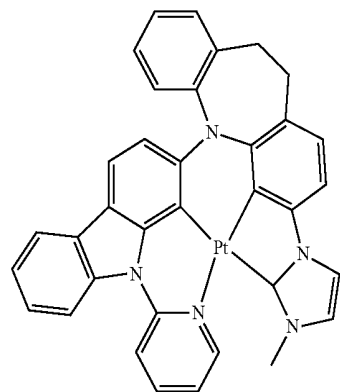
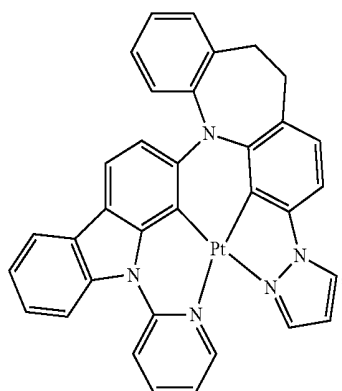
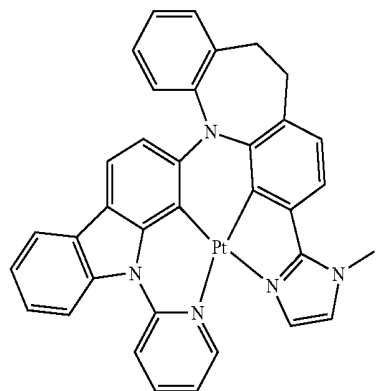
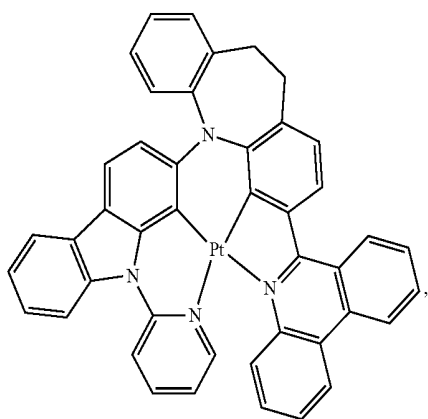
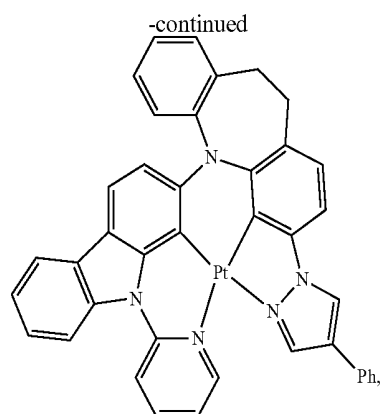
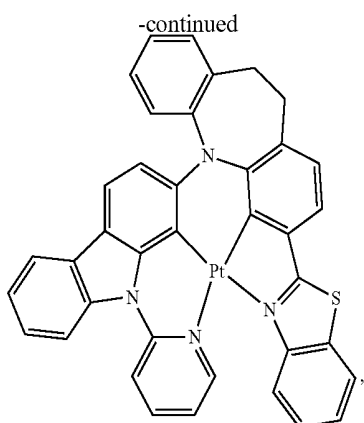
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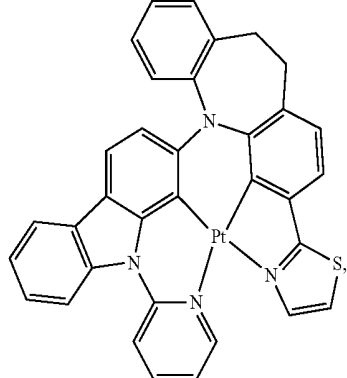
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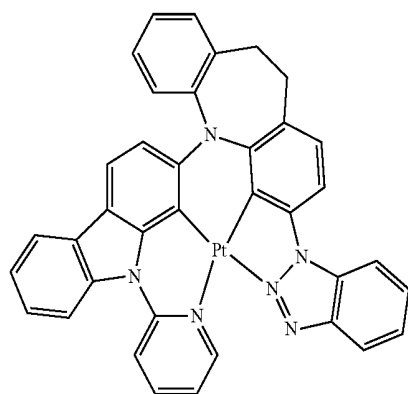
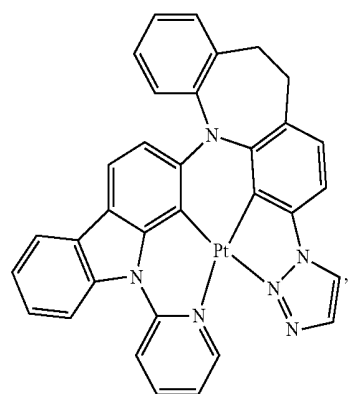
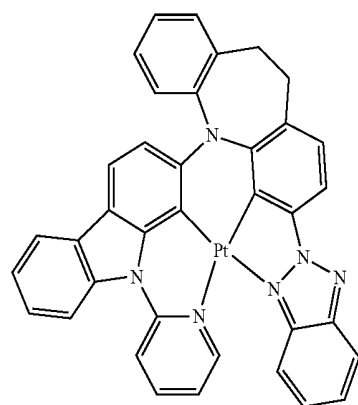
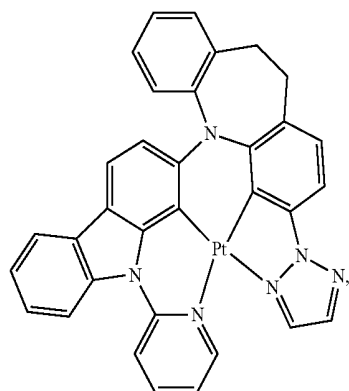
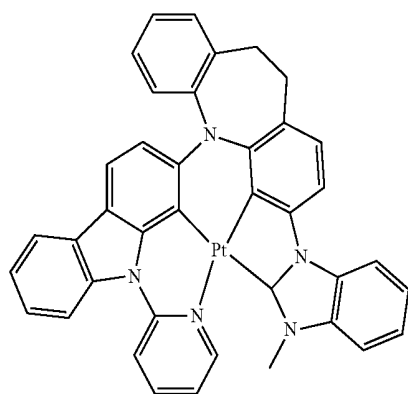
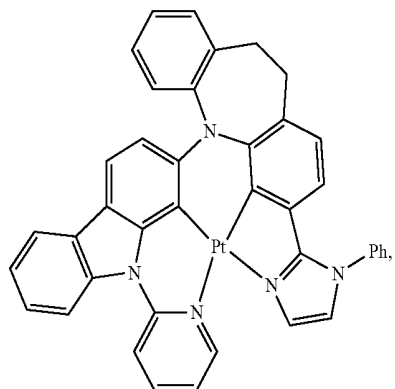
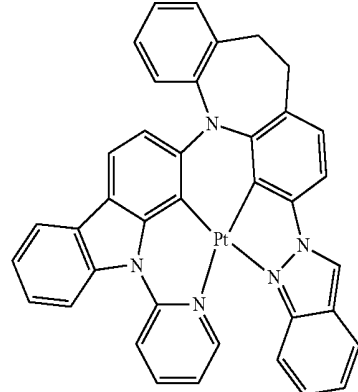


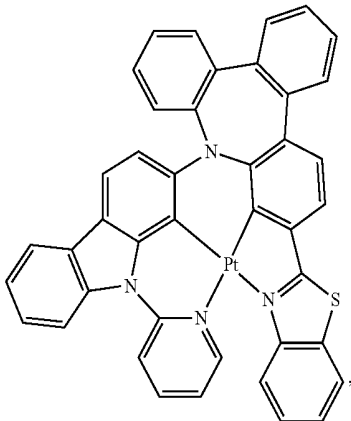
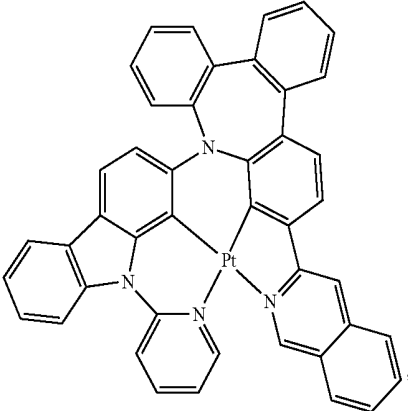
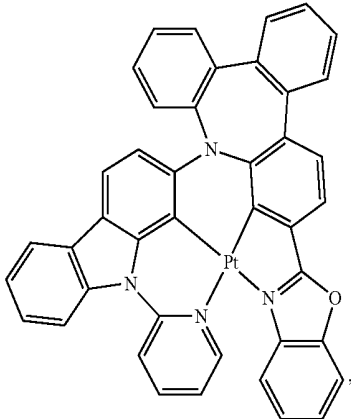
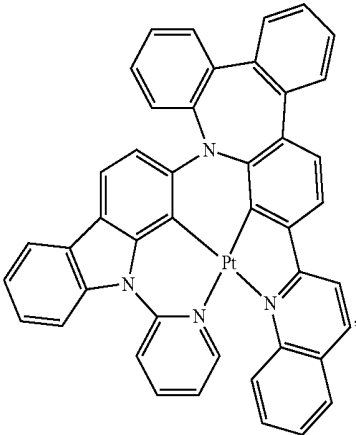
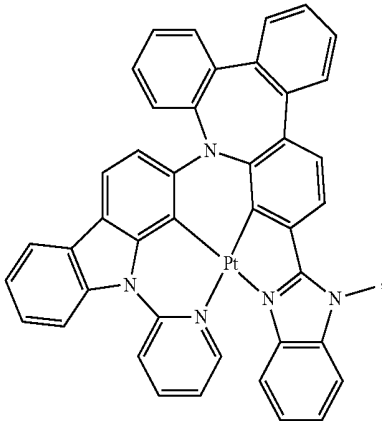
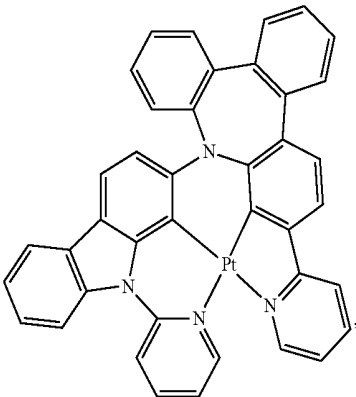
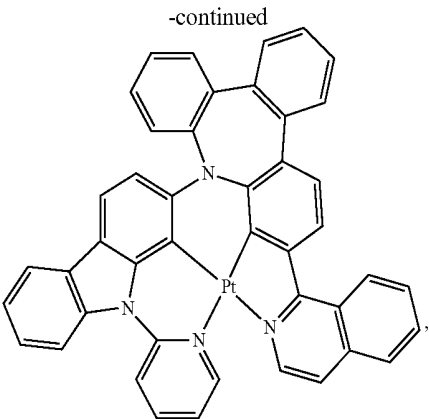
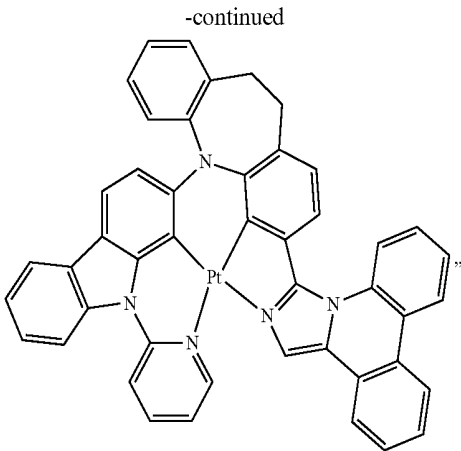


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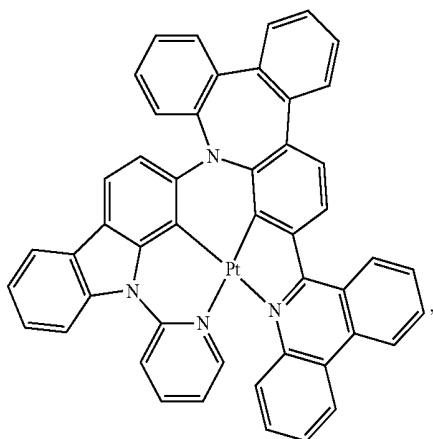


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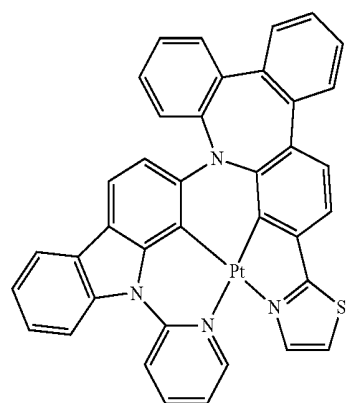
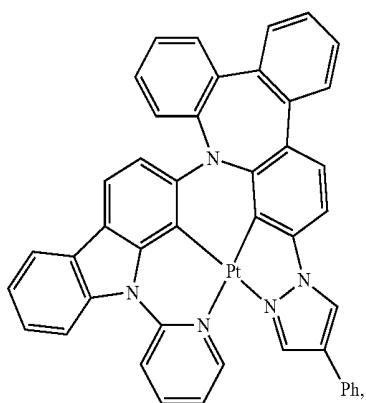
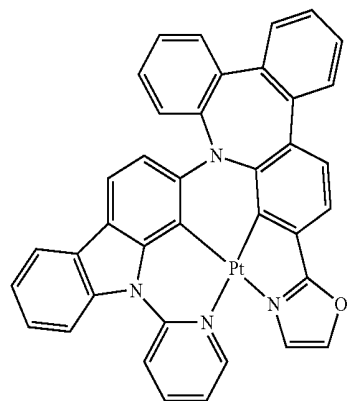
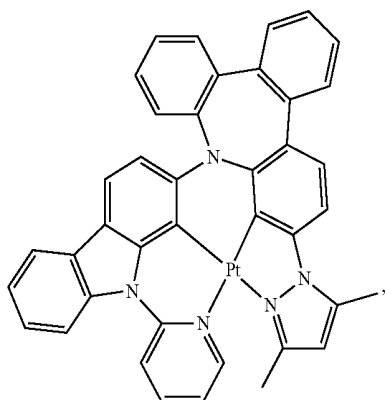
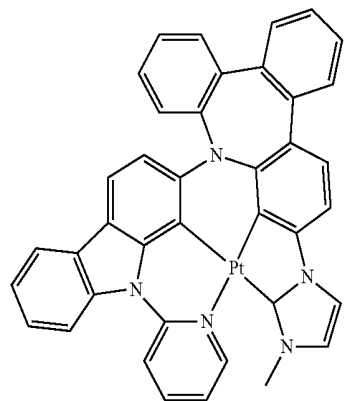
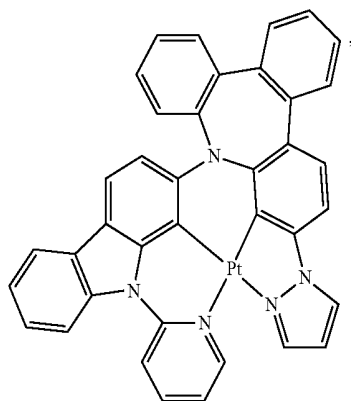
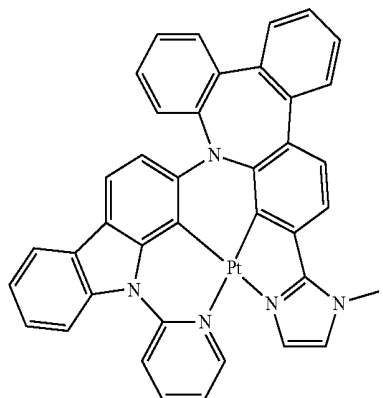




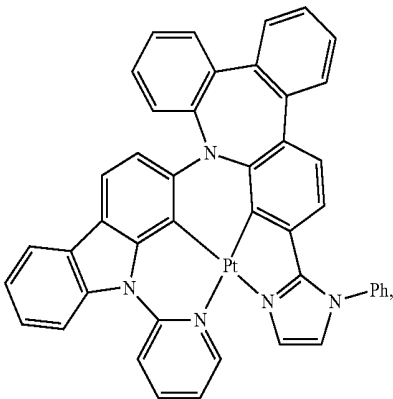
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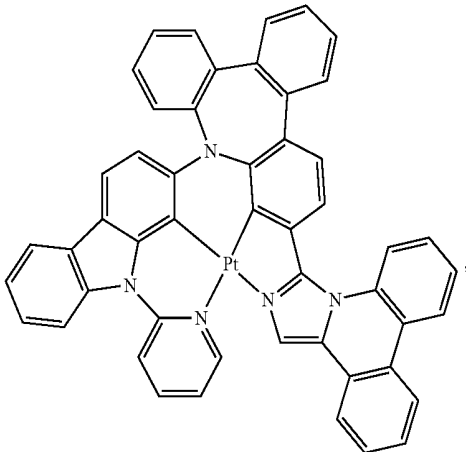
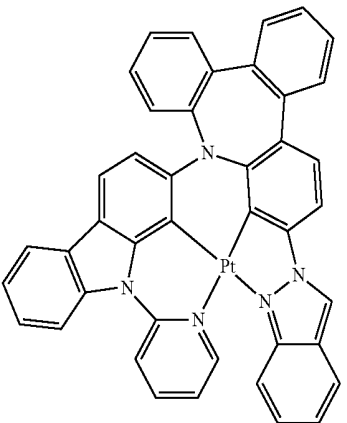
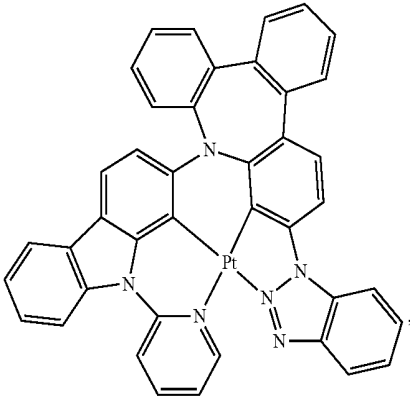
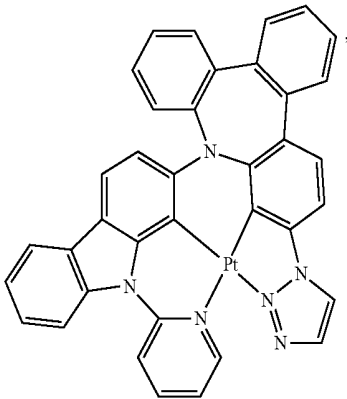
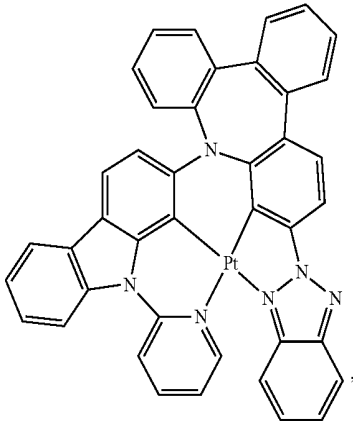
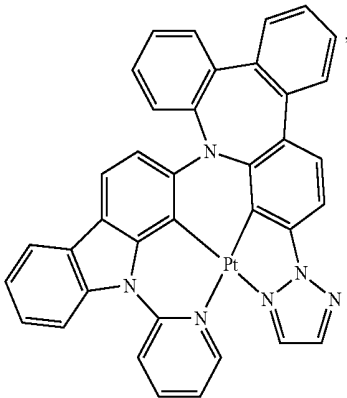
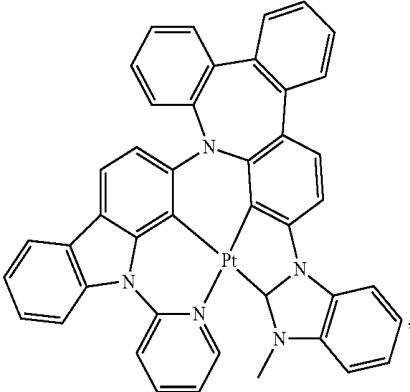
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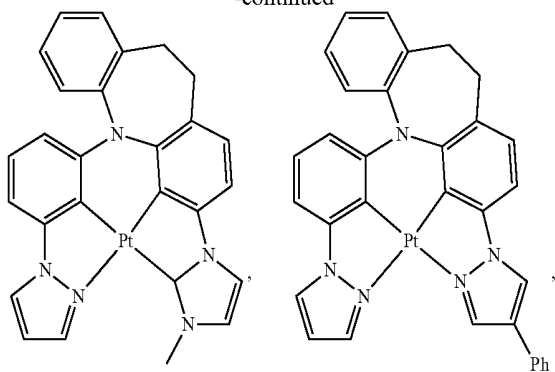
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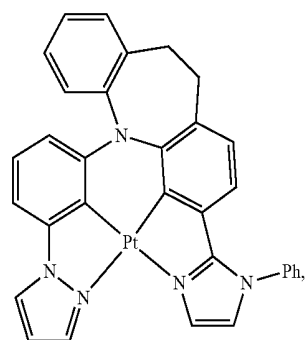
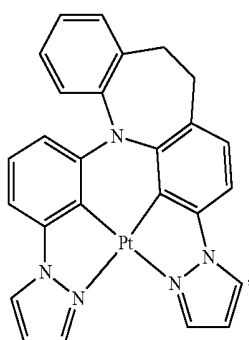
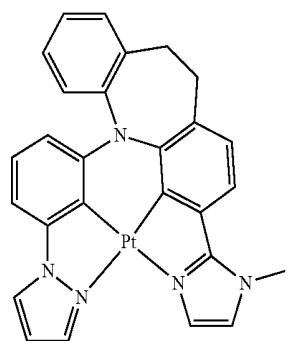
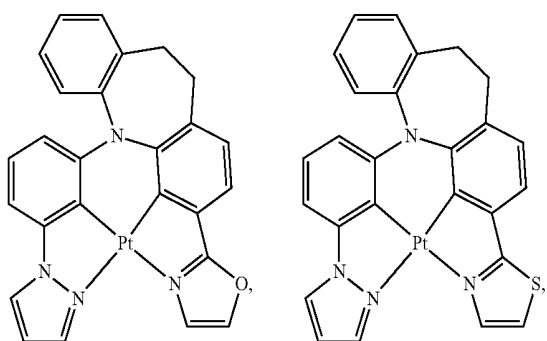
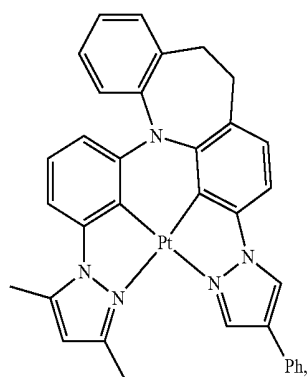
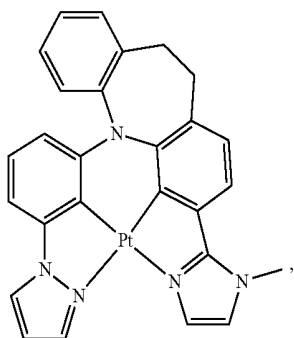
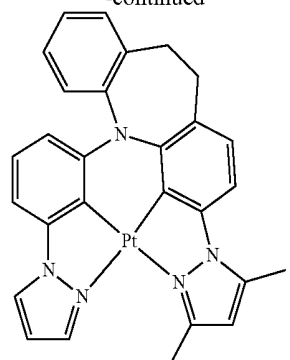
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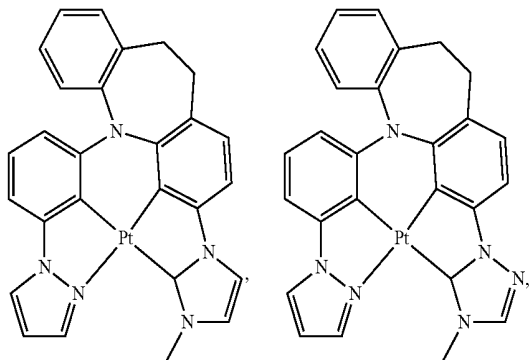
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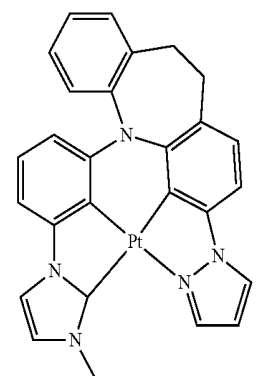
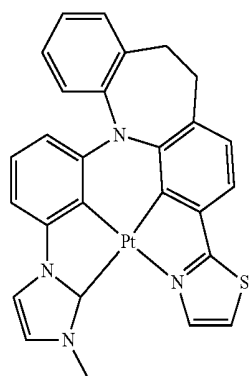
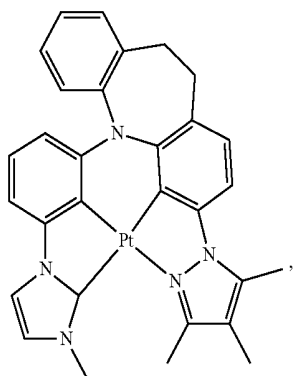
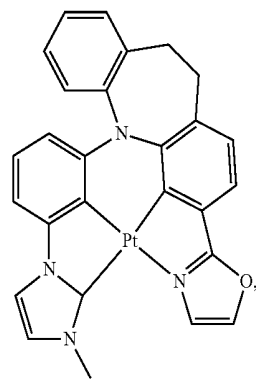
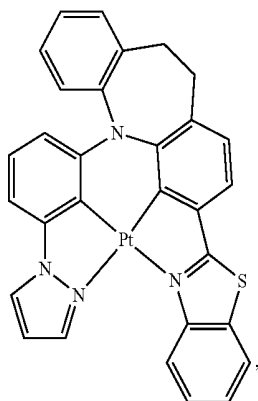
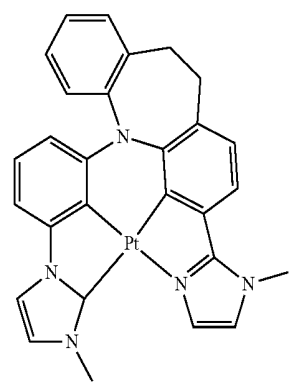
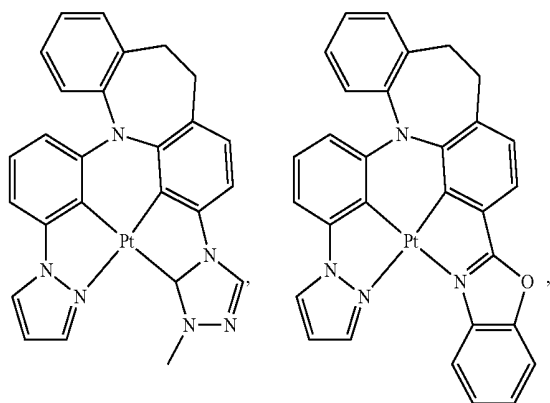
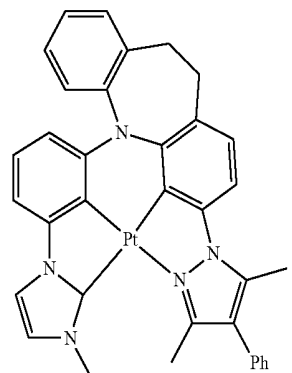
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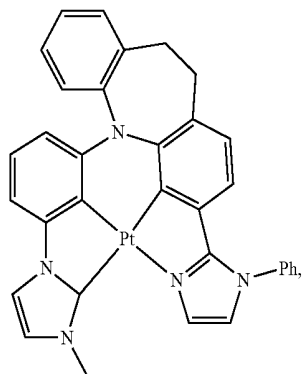
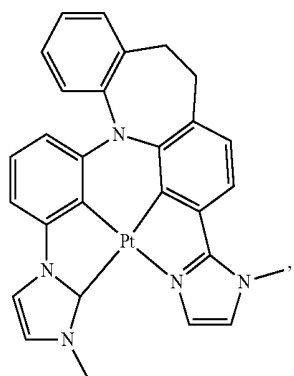
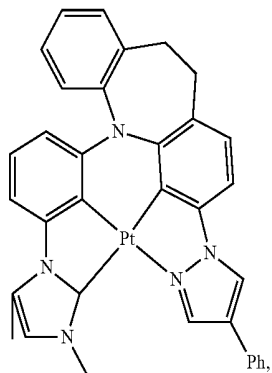
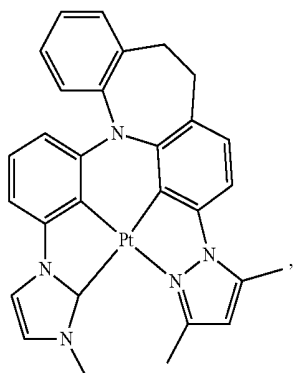
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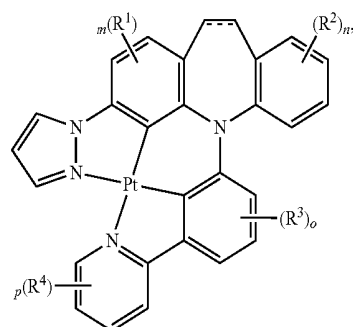
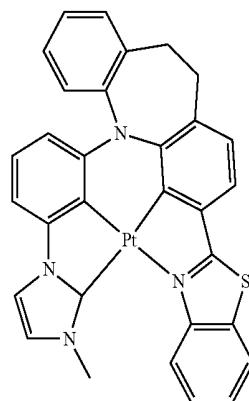
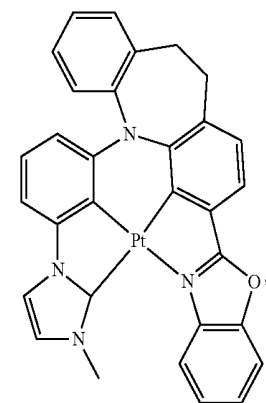
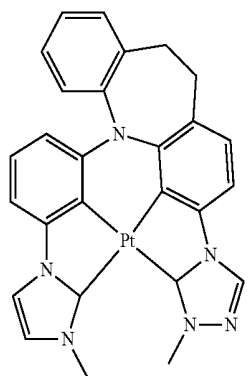
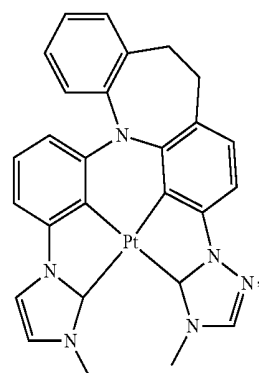
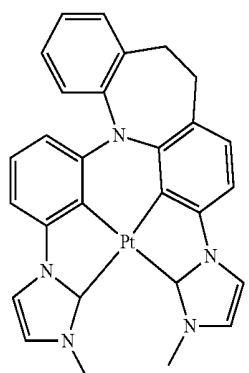
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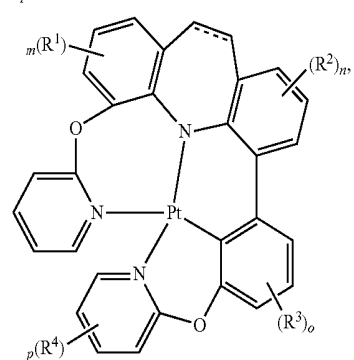
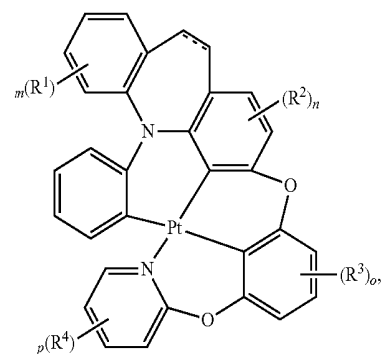
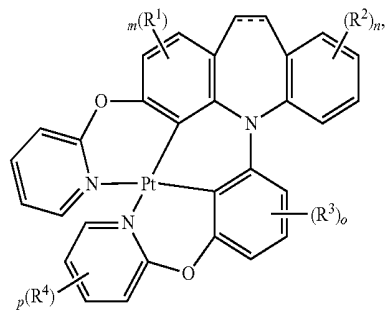
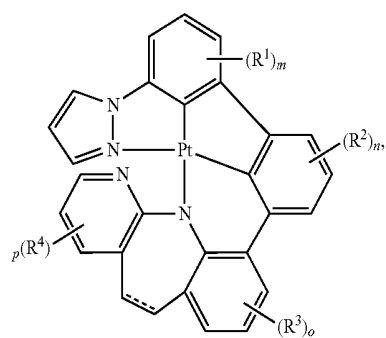
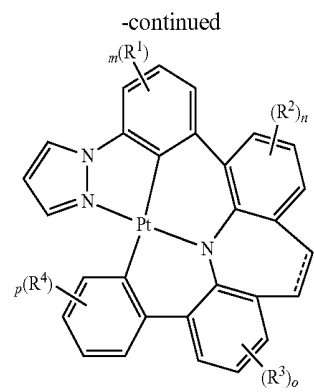
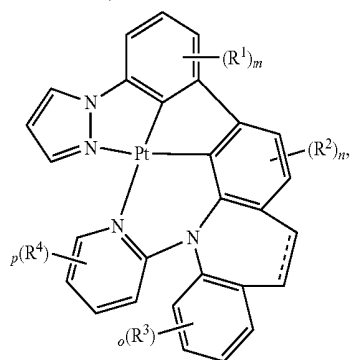
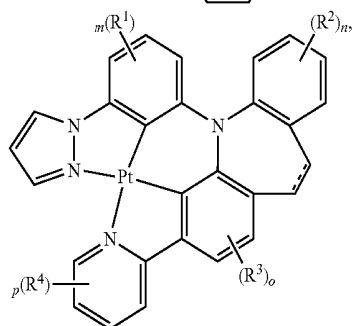
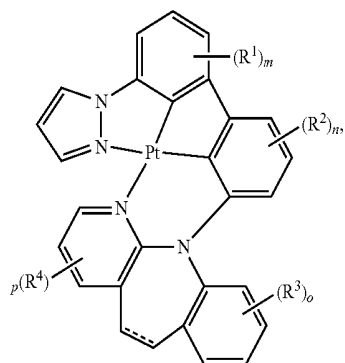
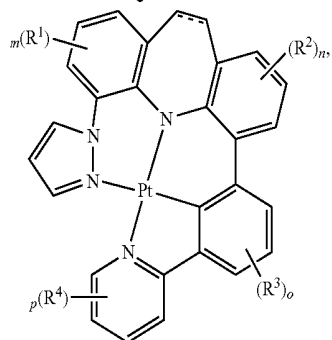
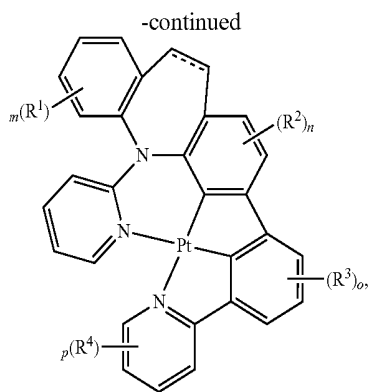


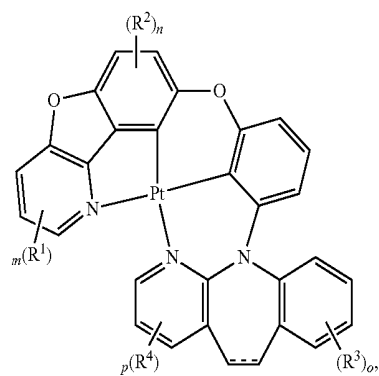
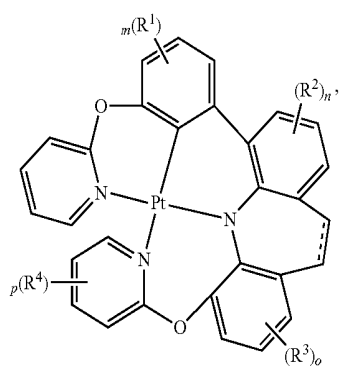
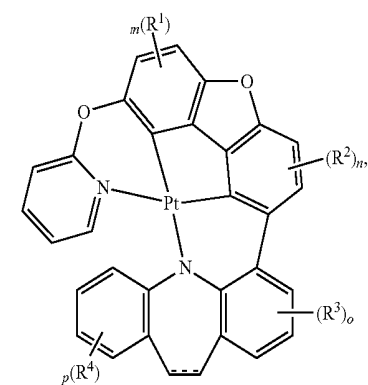
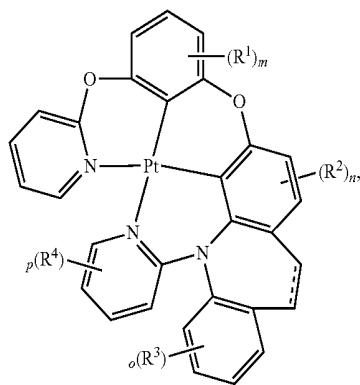
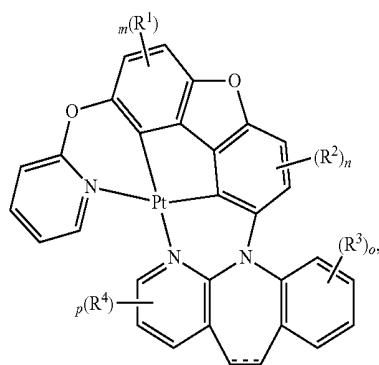
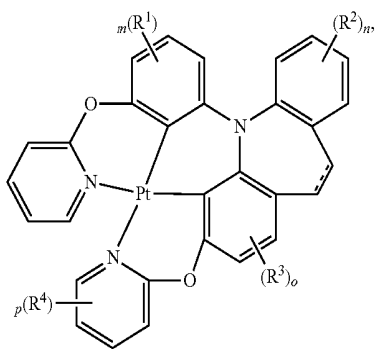
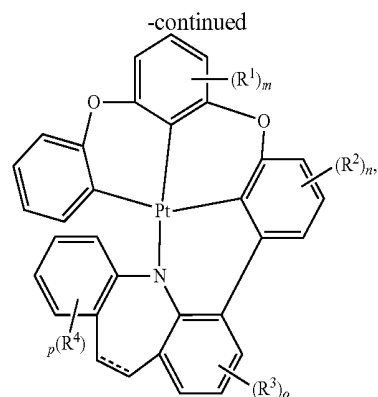
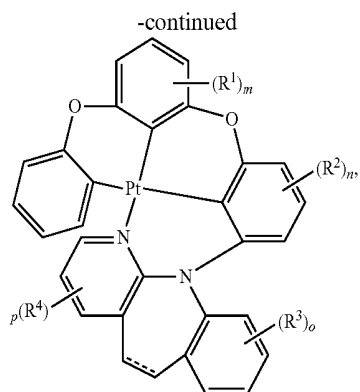
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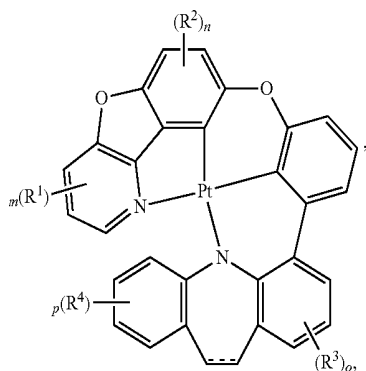
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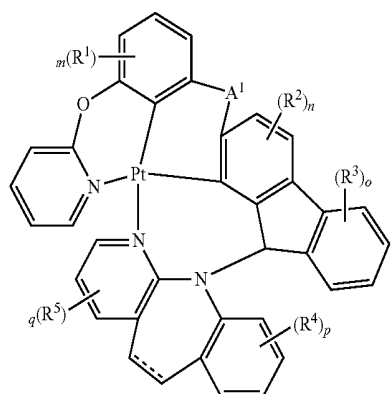
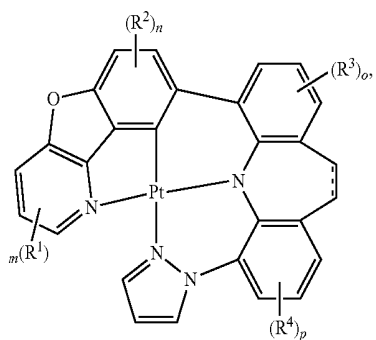
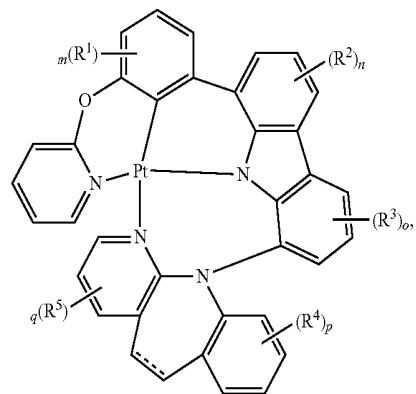
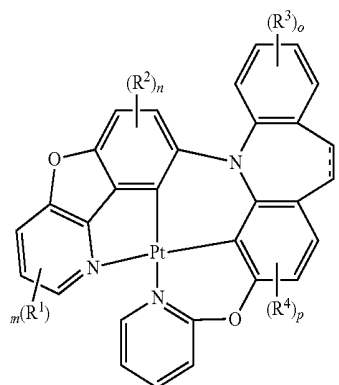
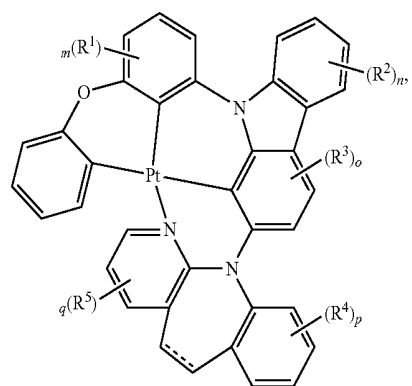
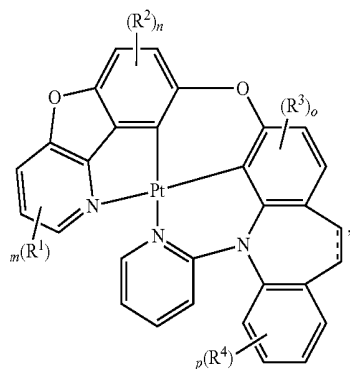
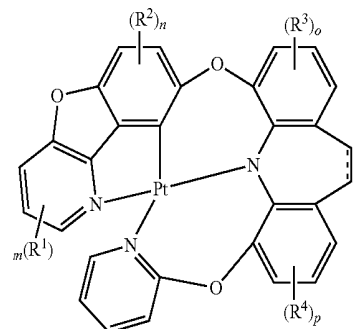


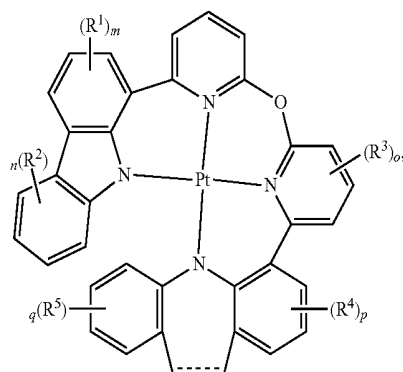
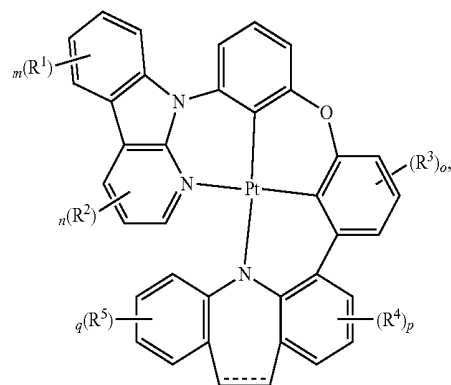
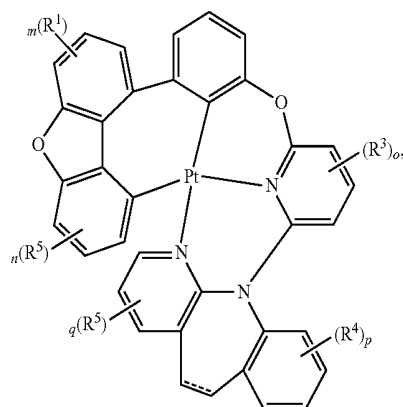
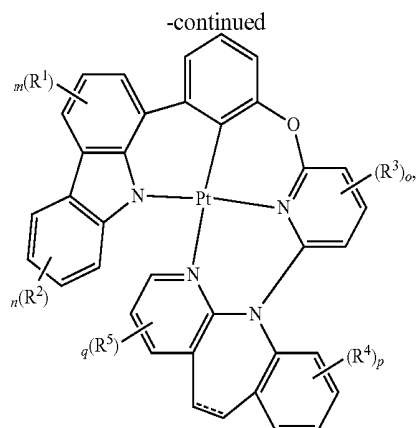
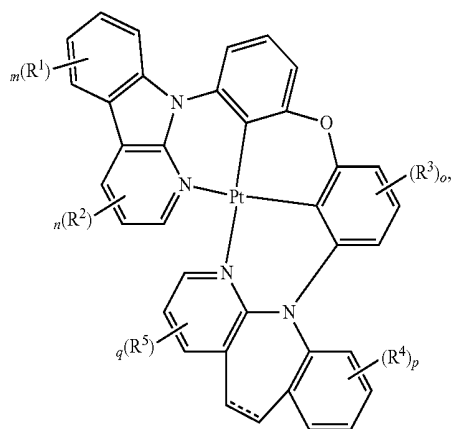
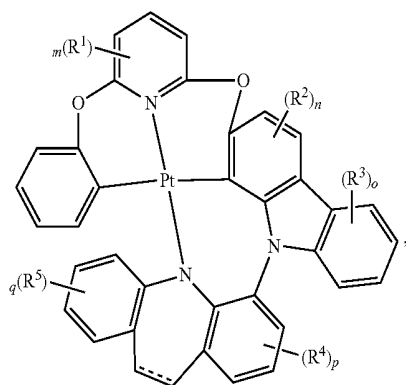
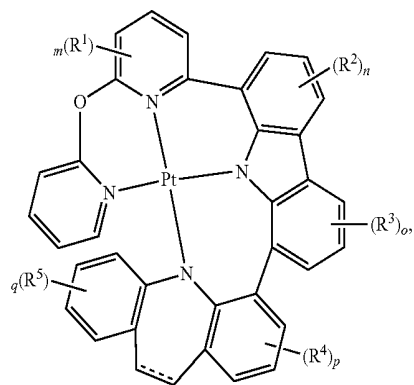
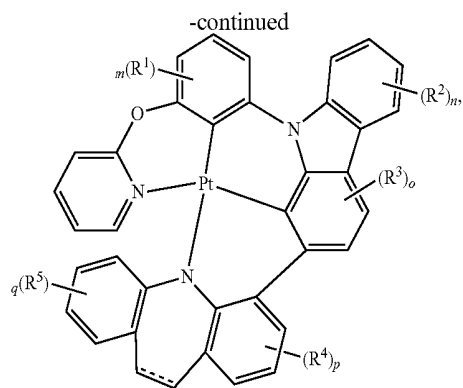


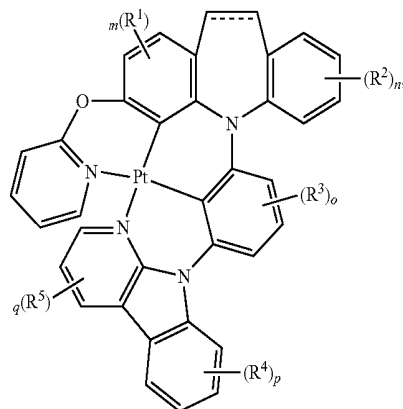
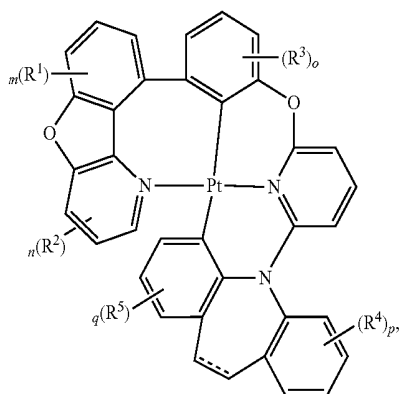
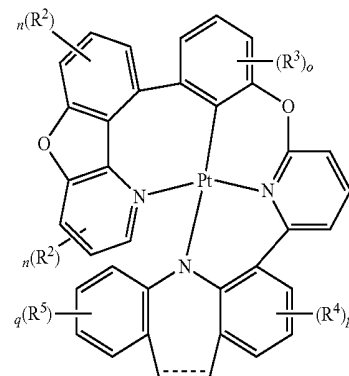
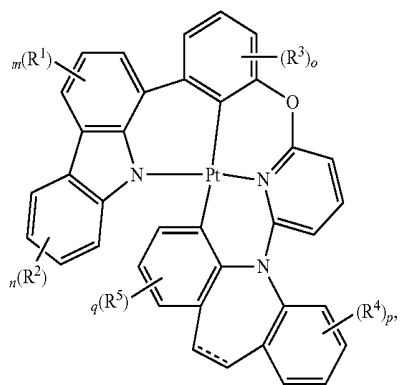
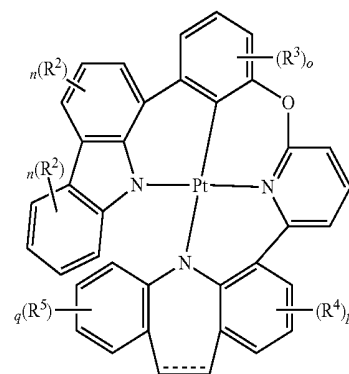
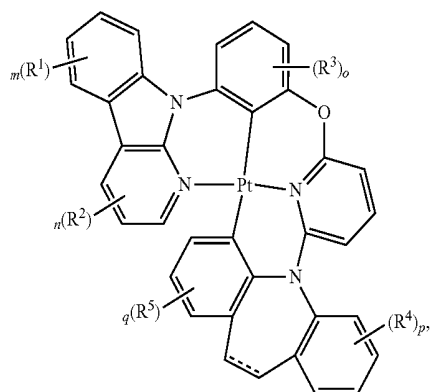
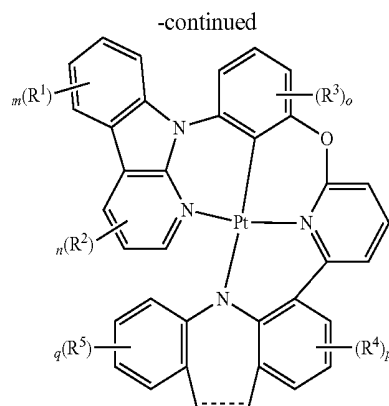
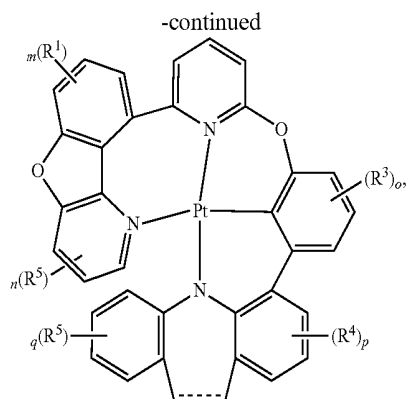
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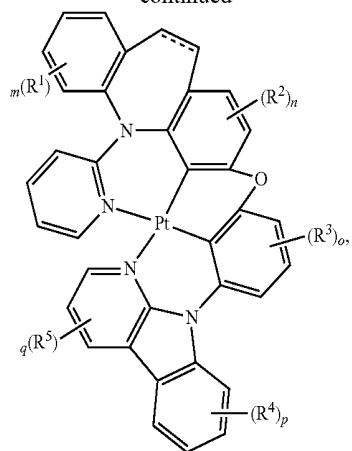
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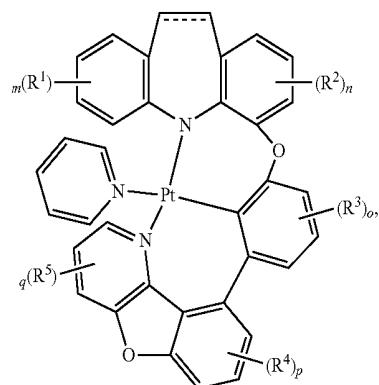
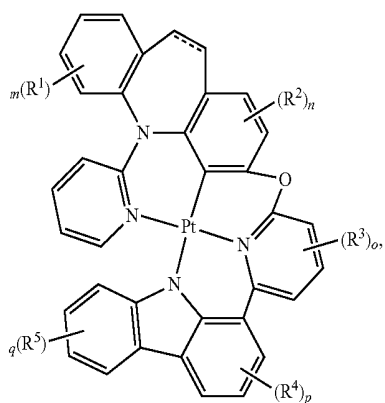
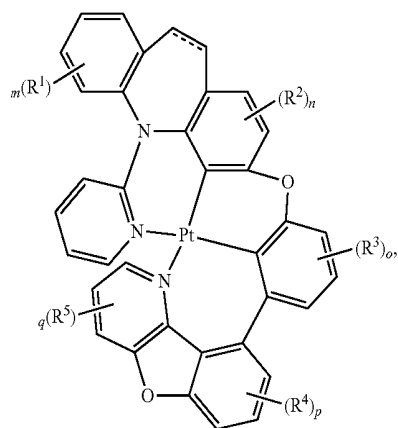
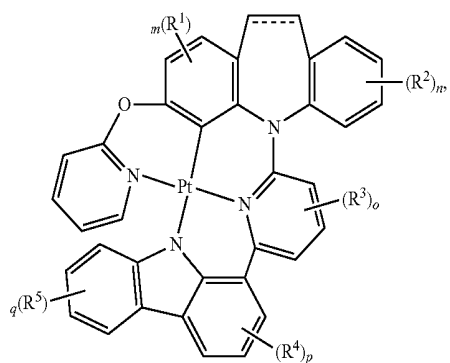
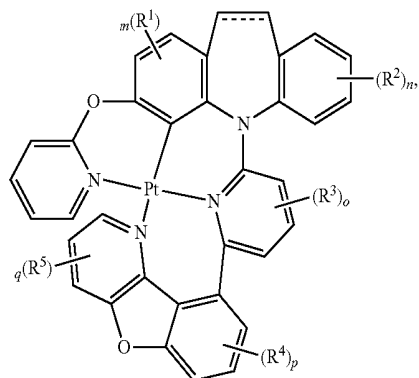
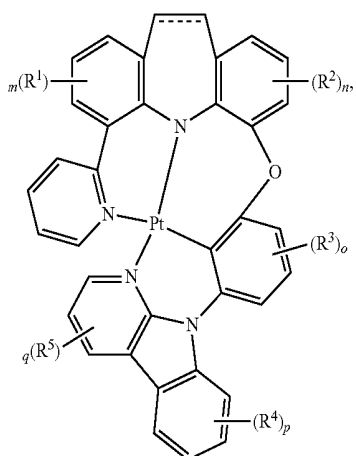
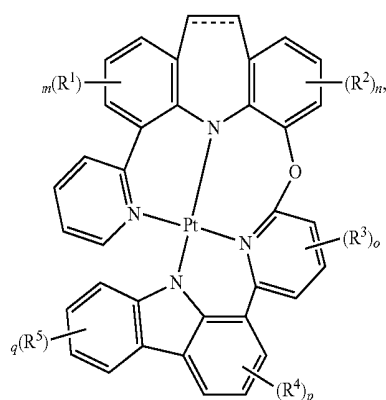




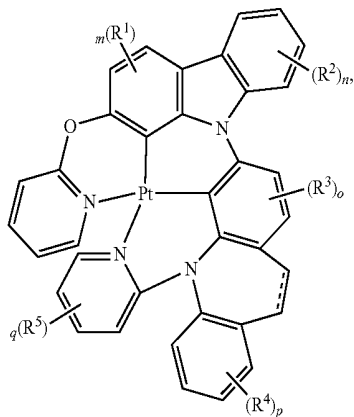
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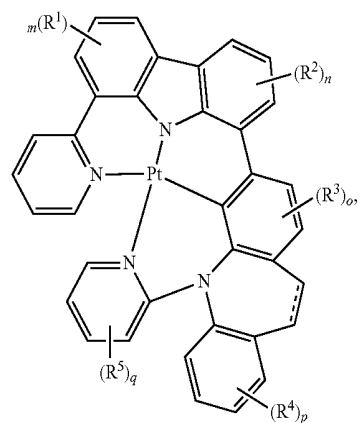
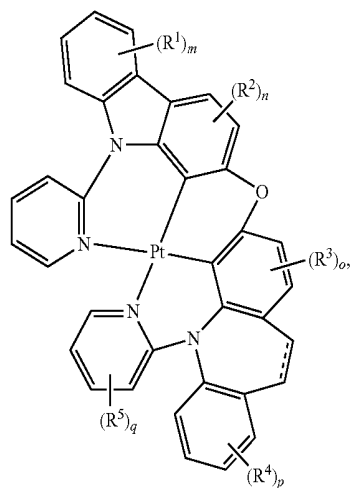
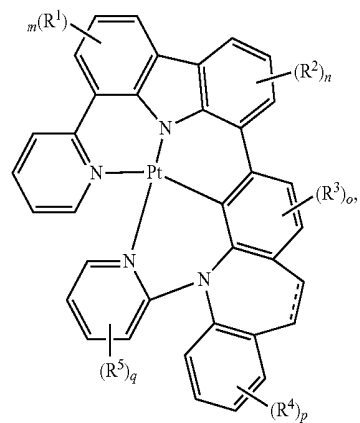
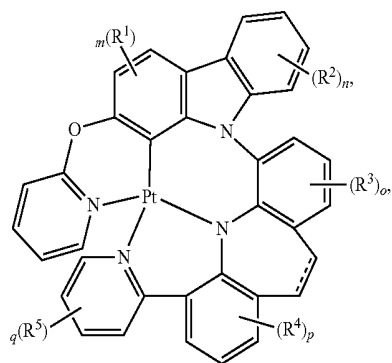
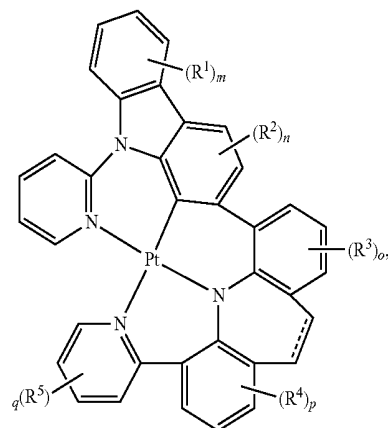
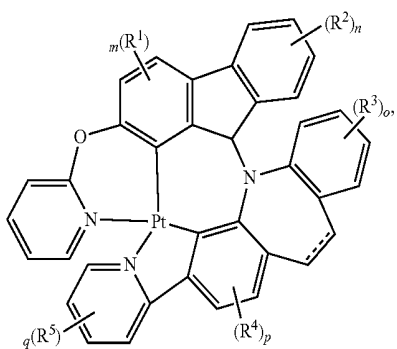
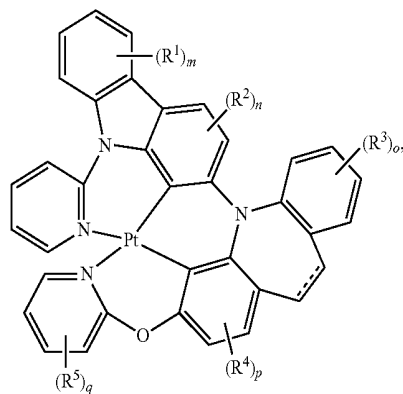
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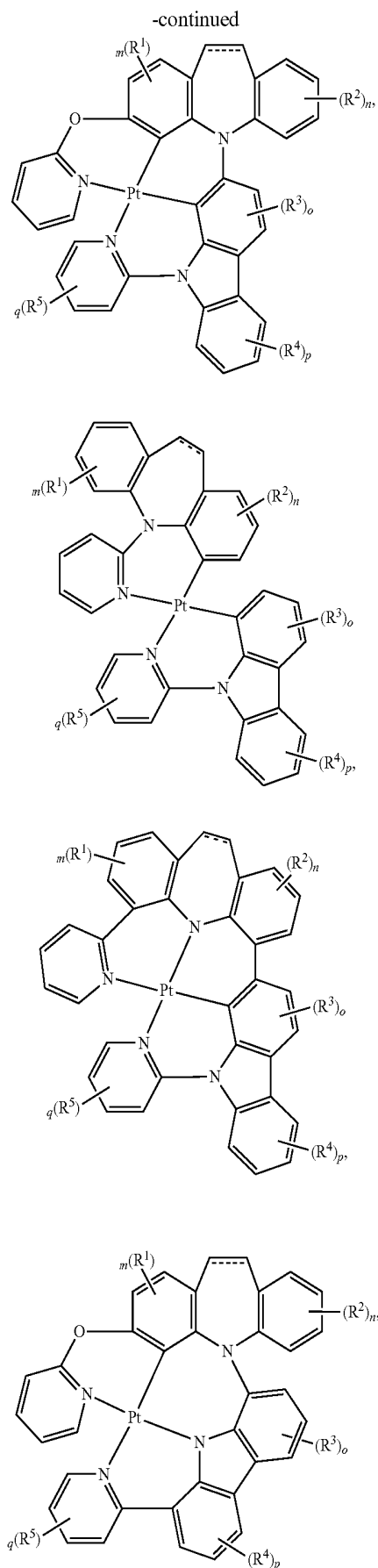
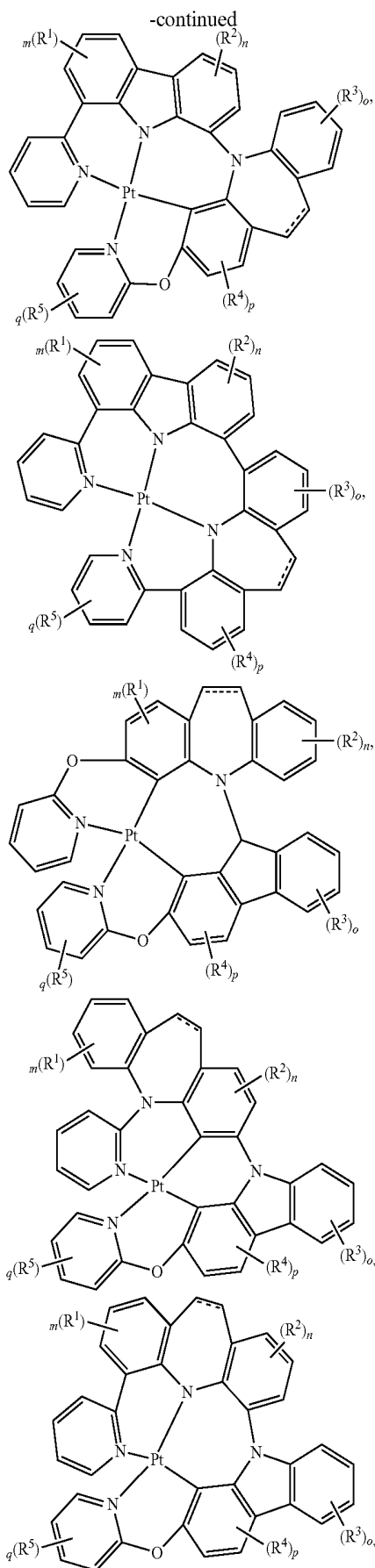


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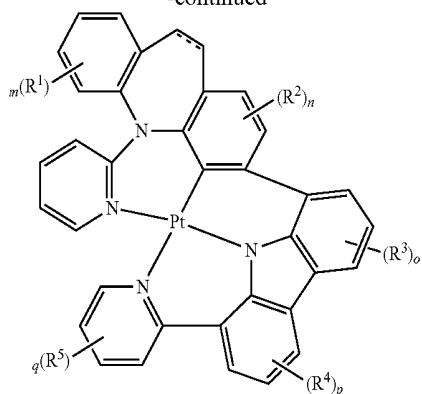


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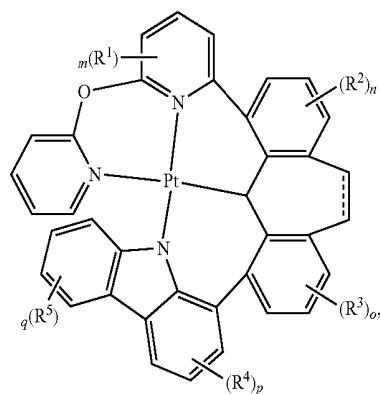
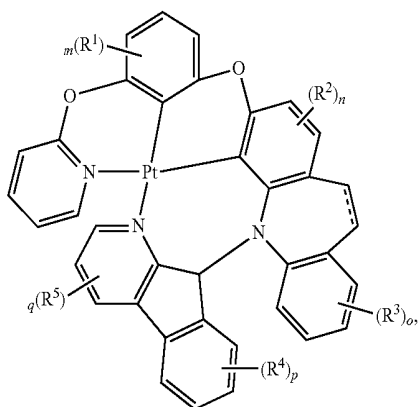
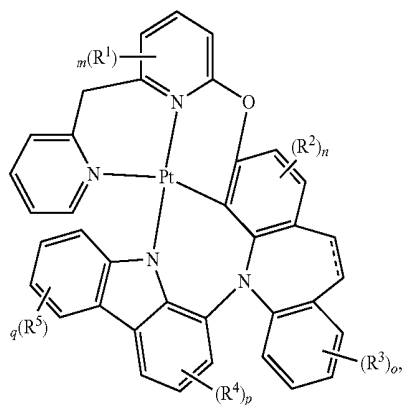
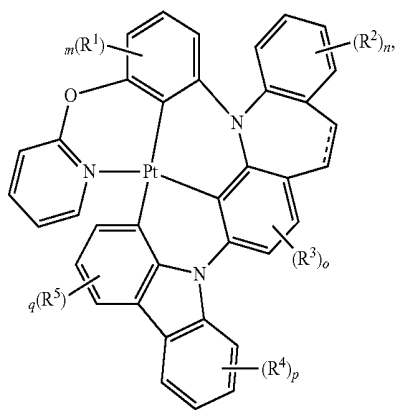
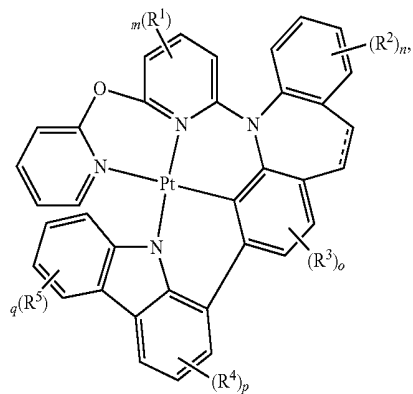
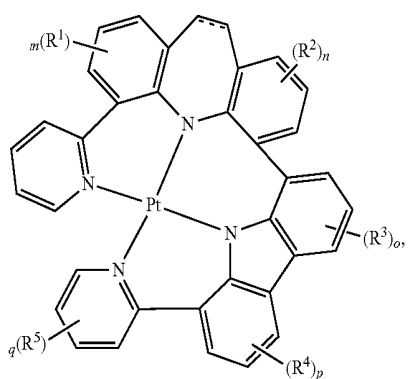
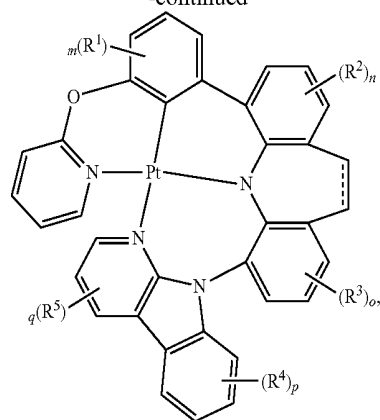


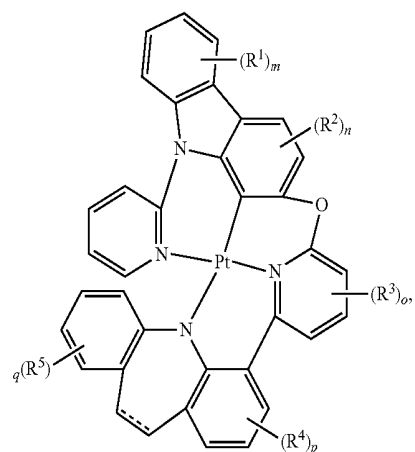


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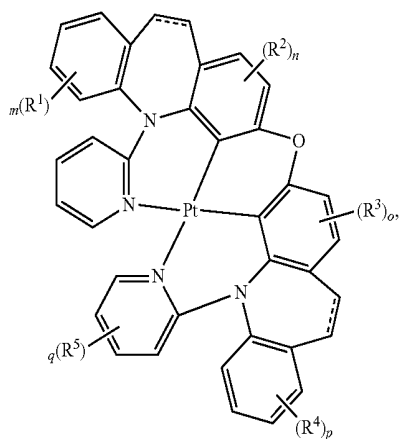
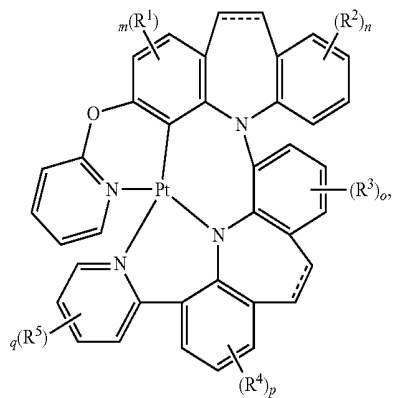
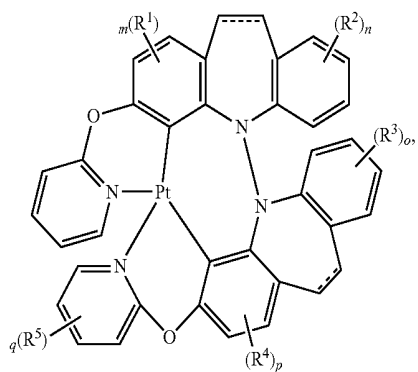
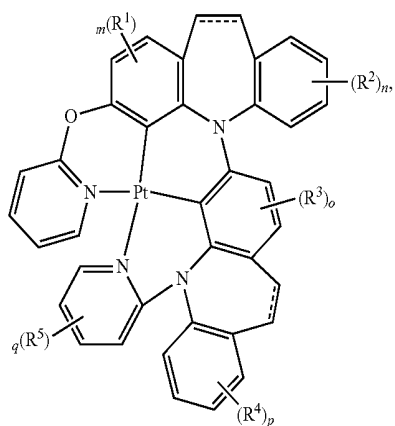


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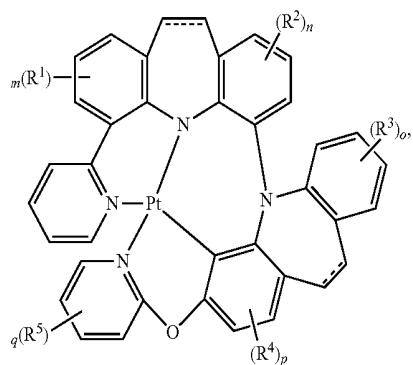
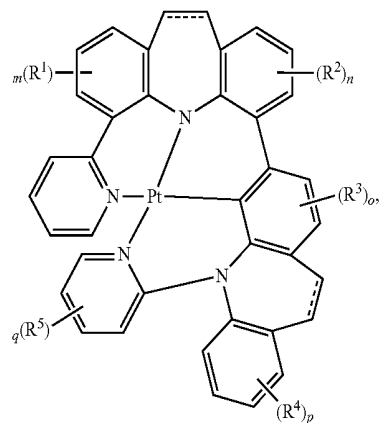
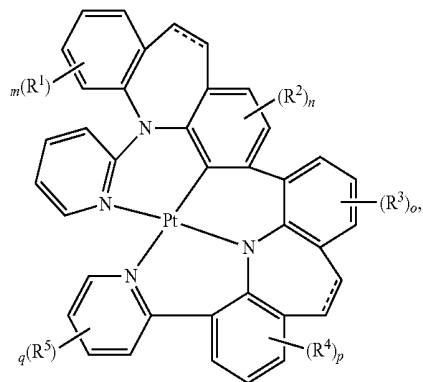
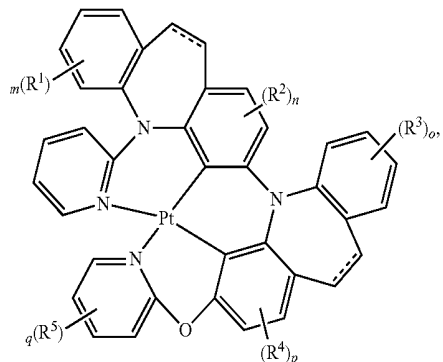


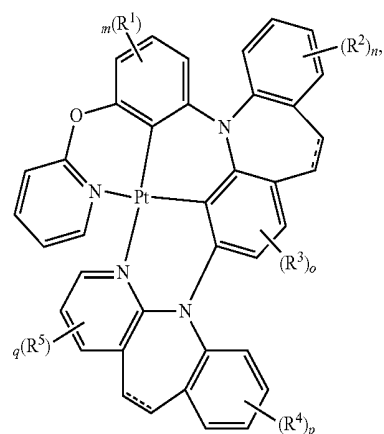
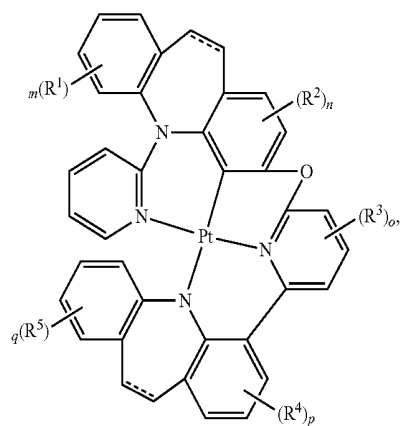
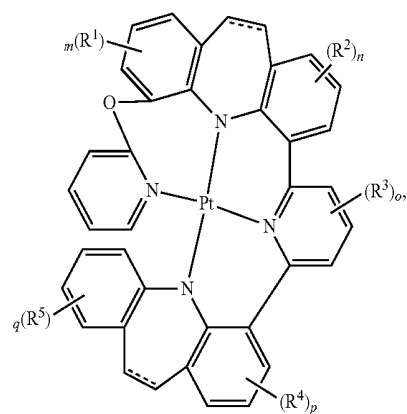
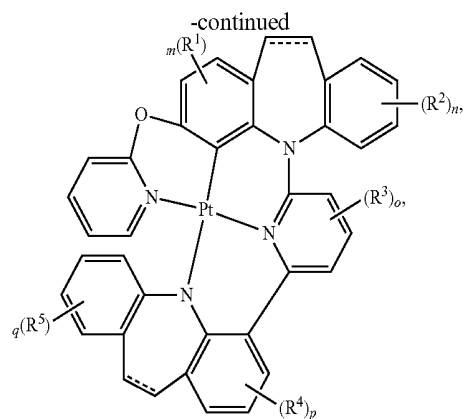
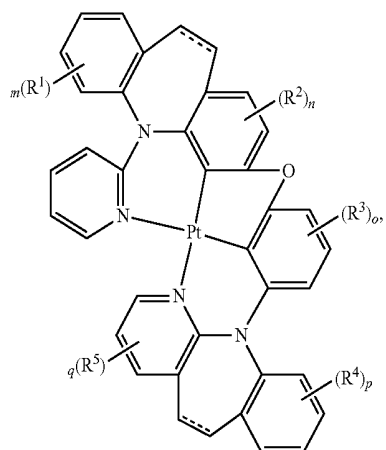
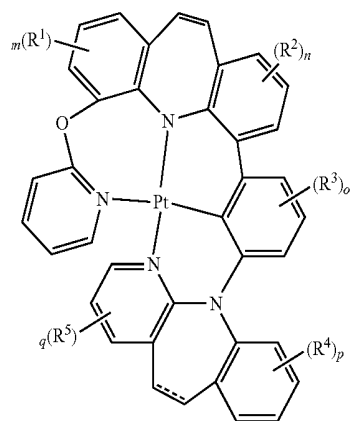
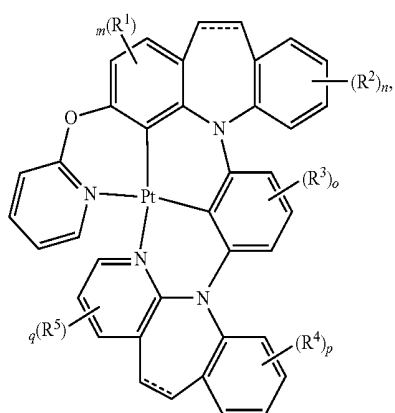
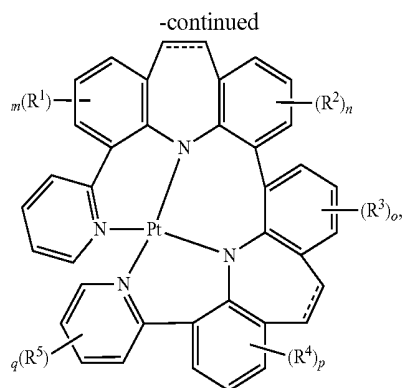


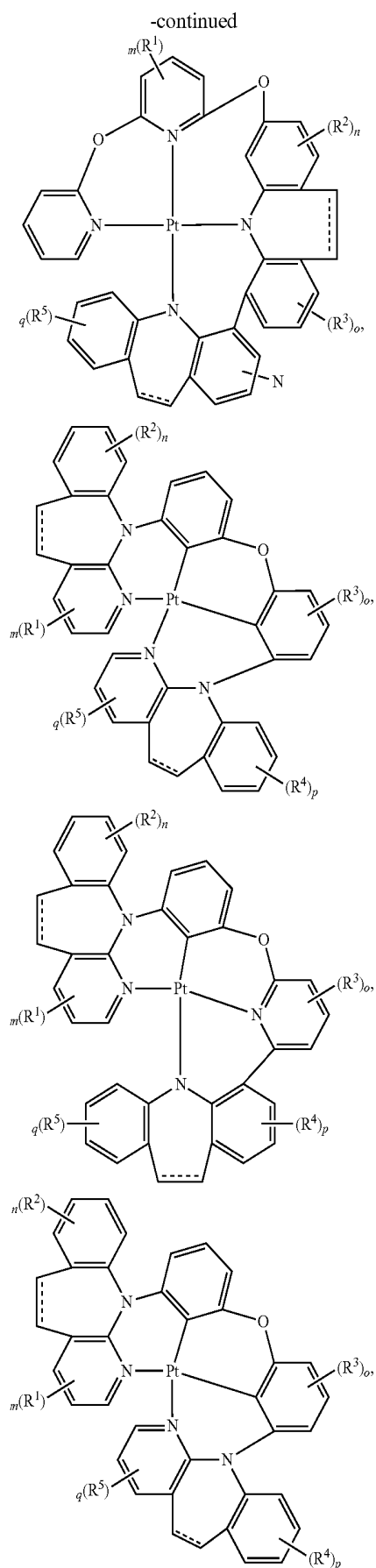
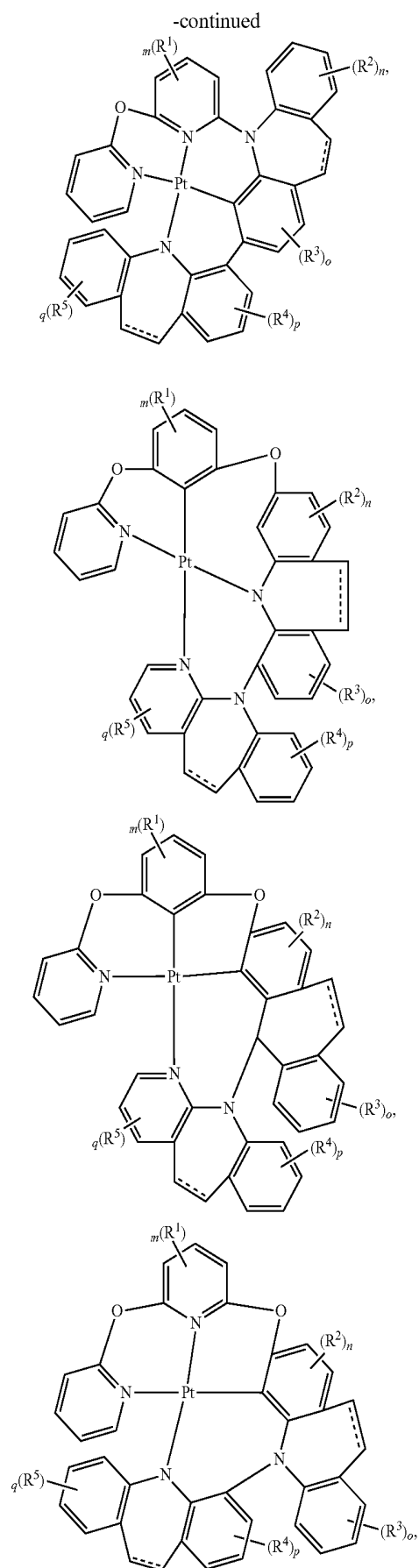
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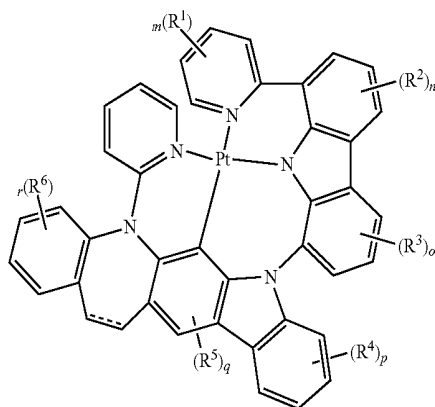
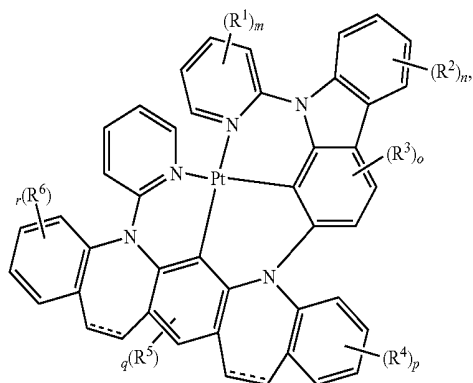
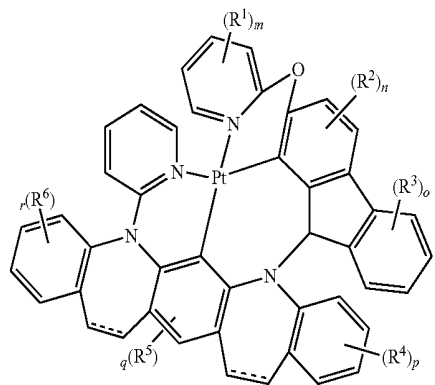
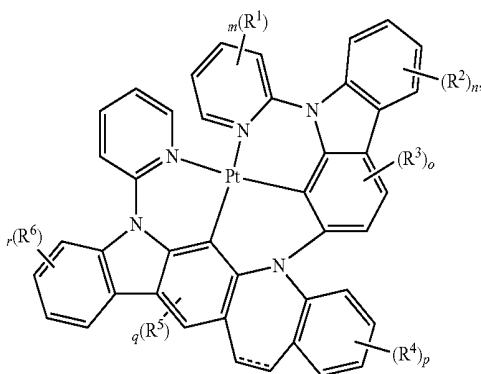
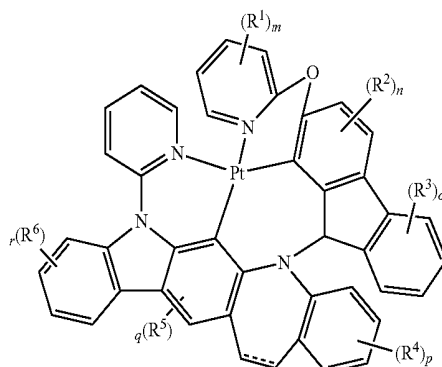
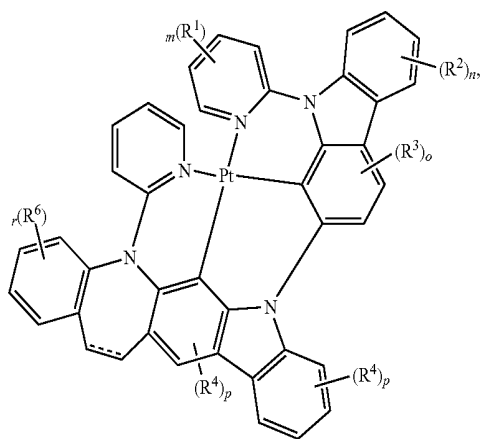
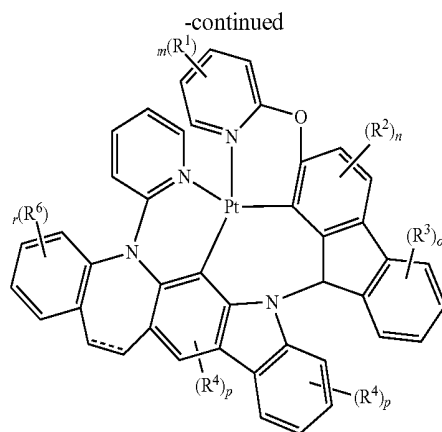
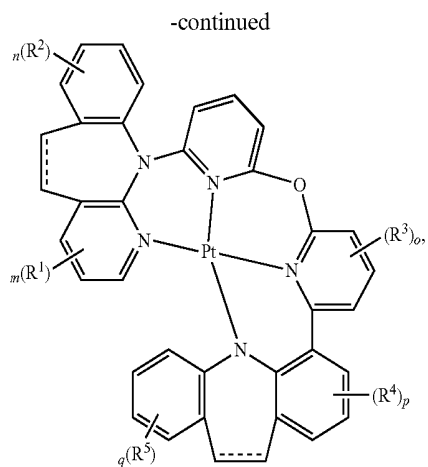


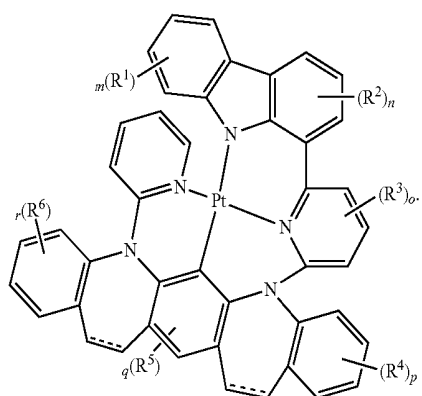
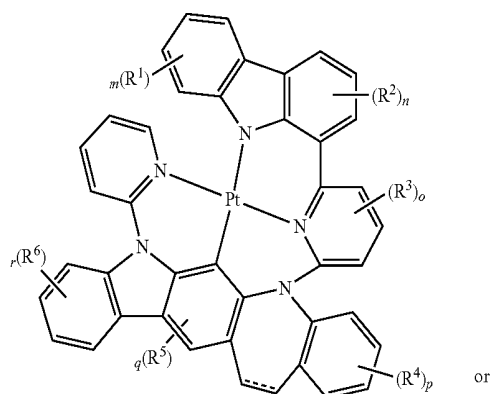
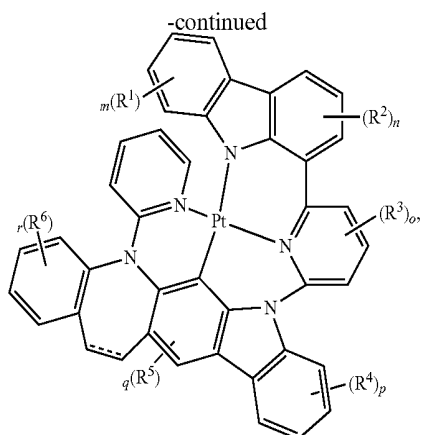
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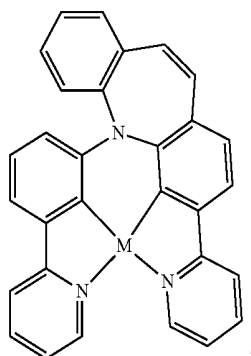




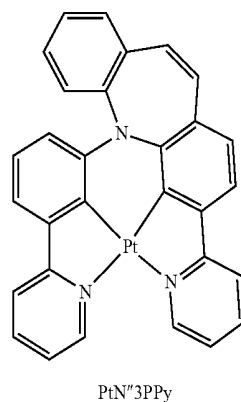




[0054] In one aspect, the compound has the structure:



such as, for example

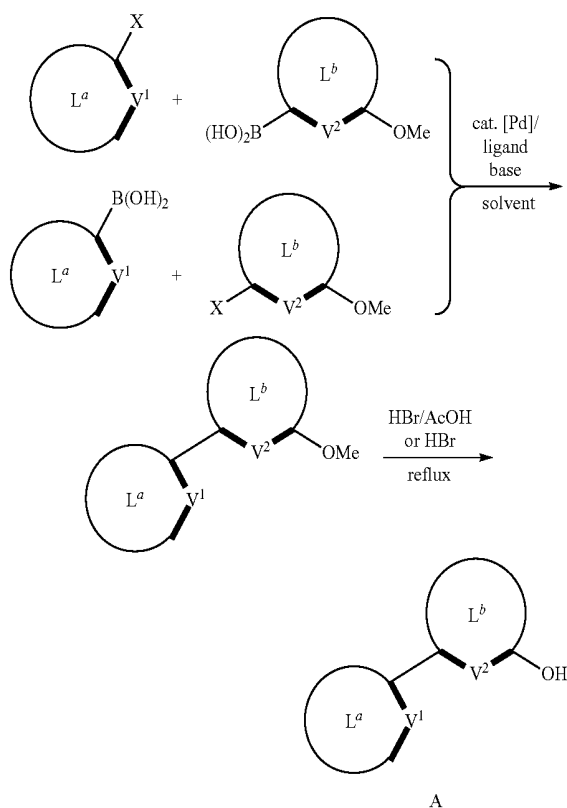


[0055] It should be noted that each of multiple recitations or illustrations of a particular structure, reference letter or numeral, etc., appearing in one or more structures of this disclosure can represent the same or different species, and that both aspects wherein they represent the same and wherein they represent different species are intended to be disclosed.

3. Methods

[0056] The disclosed compound can be made by one or more of the methods of synthesizing disclosed herein.

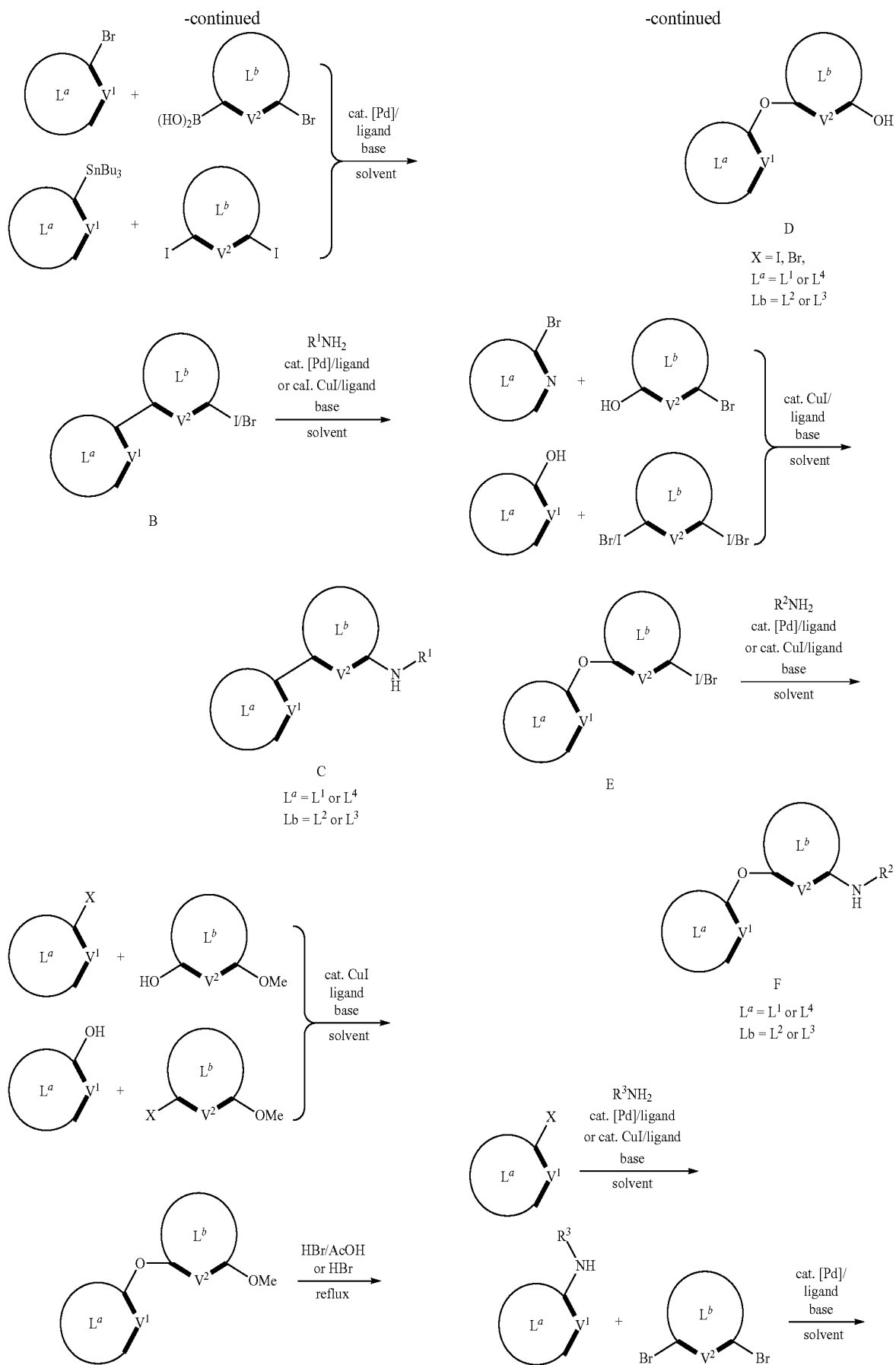
[0057] In one aspect, the disclosed compounds can be made by a synthesis comprising one or more of the following reactions or steps:



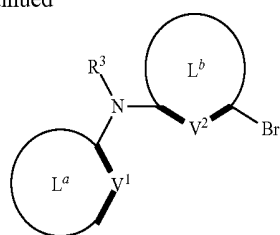
X = I, Br, Cl, OTf

L^a = L¹ or L⁴

L^b = L² or L³



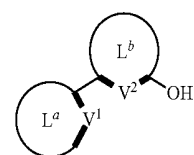
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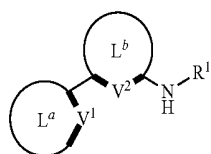
G

 $L^a = L^1 \text{ or } L^4$ $L^b = L^2 \text{ or } L^3$

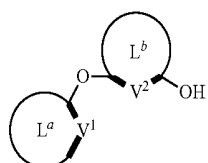
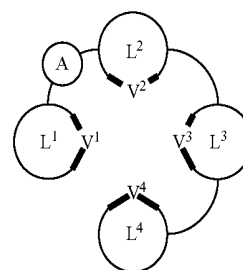
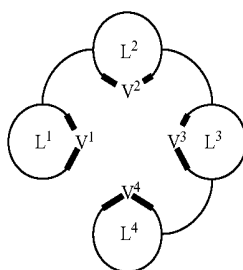
[0058] In another aspect, the disclosed compounds can be made by a synthesis comprising one or more of the following reactions or steps:



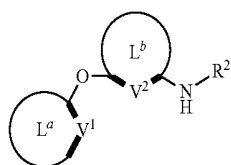
A



C

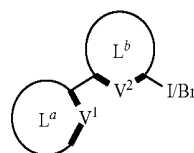


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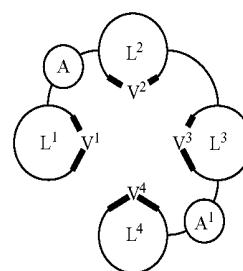
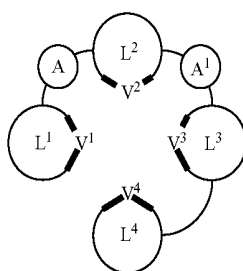
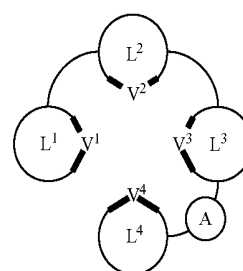
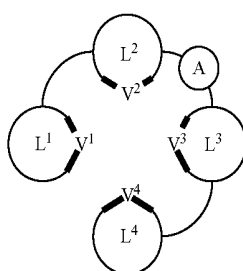
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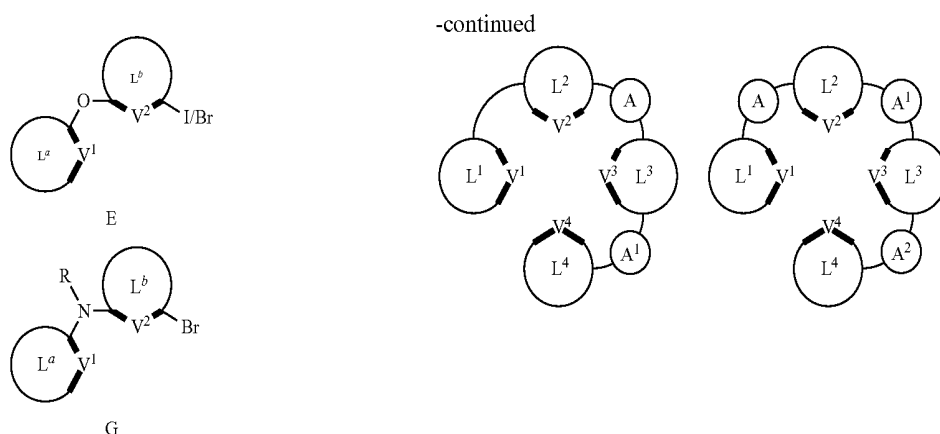
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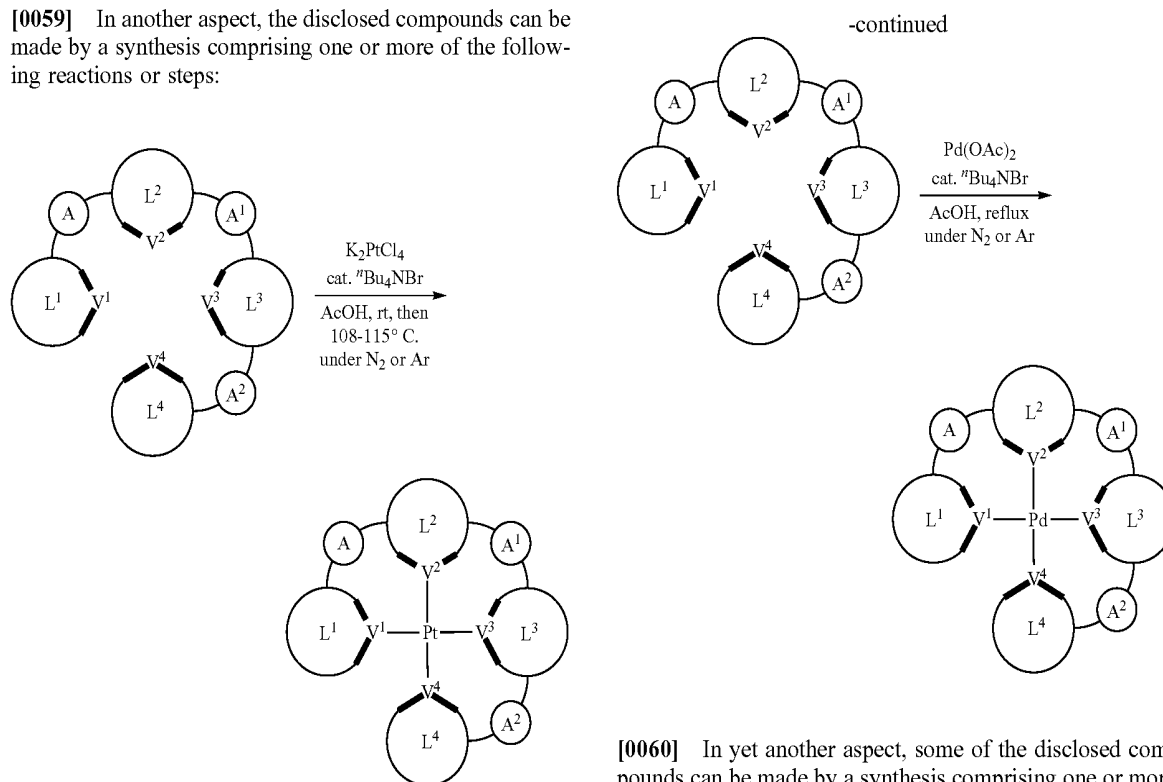
B

cat. [Pd]/ligand
base
or
cat. CuI/ligand
base
→
solvent
 $L^a = L^1 \text{ or } L^4$
 $L^b = L^2 \text{ or } L^3$
 $A, A^1, A^2 = O, NR$

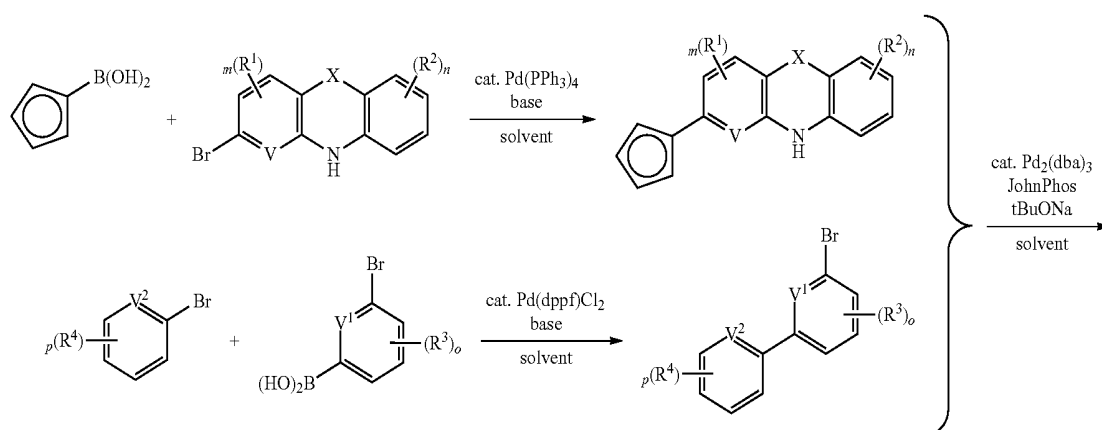




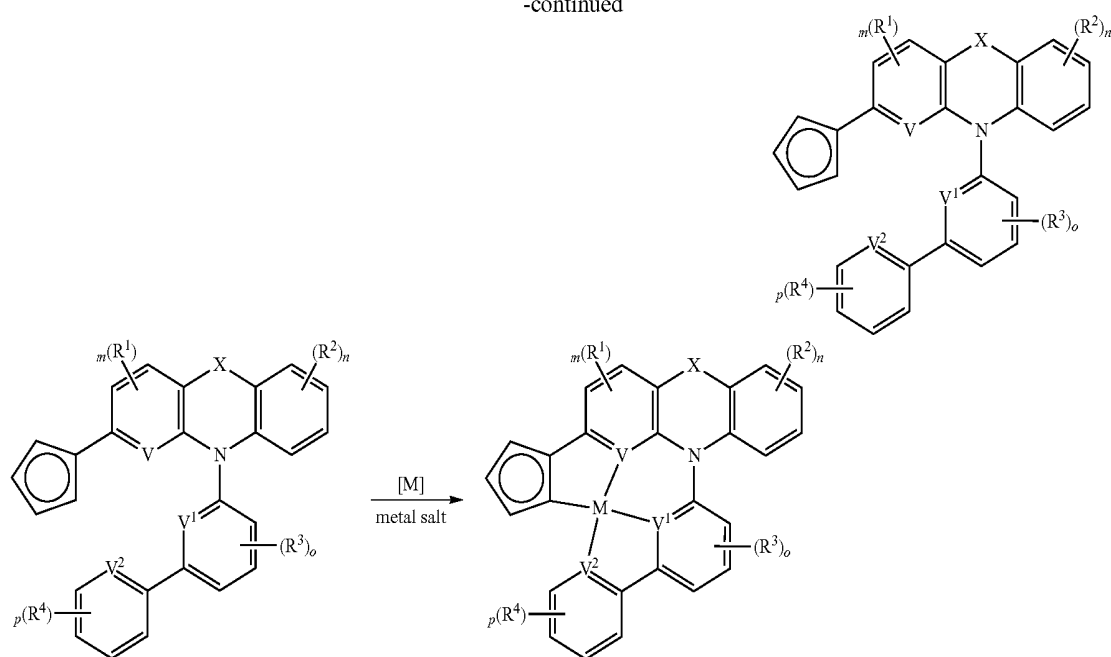
[0059] In another aspect, the disclosed compounds can be made by a synthesis comprising one or more of the following reactions or steps:



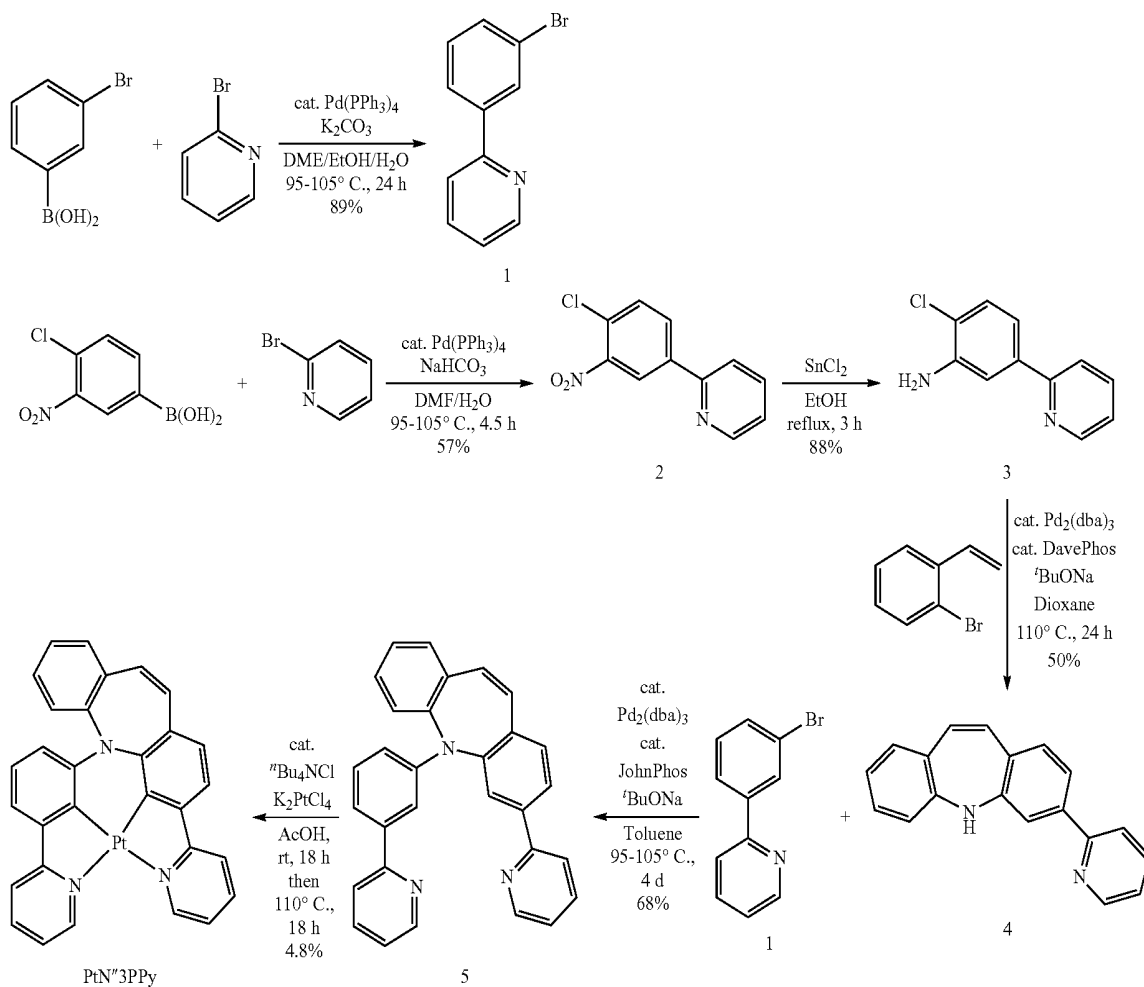
[0060] In yet another aspect, some of the disclosed compounds can be made by a synthesis comprising one or more of the following reactions or steps:



-continued



[0061] For example, a platinum compound can be made by a reaction comprising one or more of the following reactions or steps:



4. Devices

[0062] As briefly described above, the present invention is directed to metal compounds. In one aspect, the compounds disclosed herein can be used as host materials for OLED applications, such as full color displays.

[0063] The disclosed compounds can be useful in a wide variety of applications, such as, for example, lighting devices. In a particular aspect, one or more of the compounds can be host materials for an organic light emitting display device.

[0064] The compounds are useful in a variety of applications. As light emitting materials, the compounds can be useful in organic light emitting diodes (OLED)s, luminescent devices and displays, and other light emitting devices. The device can be a phosphorescent OLED device. The device can also be a fluorescent OLED device.

[0065] In one aspect, the device is a photovoltaic device. In another aspect, the device is a luminescent display device. In yet another aspect, the device is a light emitting device.

[0066] The energy profile of the compounds can be tuned by varying the structure of the ligand surrounding the metal center. For example, compounds having a ligand with electron withdrawing substituents will generally exhibit different properties, than compounds having a ligand with electron donating substituents. Generally, a chemical structural change affects the electronic structure of the compound, which thereby affects the electrical transport and transfer functions of the material. Thus, the compounds of the present invention can be tailored or tuned to a specific application that desires an energy or transport characteristic.

[0067] In another aspect, the disclosed compounds can provide improved efficiency and/or operational lifetimes in lighting devices, such as, for example, organic light emitting devices, as compared to conventional materials.

[0068] In other various aspects, the disclosed compounds can be useful as, for example, host materials for organic light emitting diodes, lighting applications, and combinations thereof.

[0069] In one embodiment, the compounds can be used in an OLED. FIG. 1 shows a cross-sectional view of an OLED 100, which includes substrate 102 with an anode 104, which is typically a transparent material, such as indium tin oxide, a layer of hole-transporting material(s) (HTL) 106, a layer of light processing material 108, such as an emissive material (EML) including an emitter and a host, a layer of electron-transporting material(s) (ETL) 110, and a metal cathode layer 112.

[0070] In one aspect, a light emitting device, such as, for example, an OLED, can comprise one or more layers. In various aspects, any of the one or more layers can comprise indium tin oxide (ITO), poly(3,4-ethylenedioxythiophene) (PEDOT), polystyrene sulfonate (PSS), N,N'-di-1-naphthyl-N,N'-diphenyl-1,1'-biphenyl-4,4'-diamine (NPD), 1,1-bis((di-4-tolylamino)phenyl) cyclohexane (TAPC), 2,6-Bis(N-carbazolyl)pyridine (mCpy), 2,8-bis(diphenylphosphoryl) dibenzothiophene (PO15), LiF, Al, or a combination thereof.

[0071] In this embodiment, the layer of light processing material 108 can comprise one or more of the disclosed compounds optionally together with a host material. The host material can be any suitable host material known in the art. The emission color of an OLED is determined by the emission energy (optical energy gap) of the light processing material 108, which as discussed above can be tuned by tuning the electronic structure of the emitting compounds and/or the host material. Both the hole-transporting material in the HTL layer 106 and the electron-transporting material (s) in the ETL layer 110 can comprise any suitable hole-transporter known in the art. A selection of which is well within the purview of those skilled in the art.

[0072] It will be apparent that the compounds of the present invention can exhibit phosphorescence. Phosphorescent OLEDs (i.e., OLEDs with phosphorescent emitters) typically have higher device efficiencies than other OLEDs, such as fluorescent OLEDs. Light emitting devices based on electrophosphorescent emitters are described in more detail in WO2000/070655 to Baldo et al., which is incorporated herein by this reference for its teaching of OLEDs, and in particular phosphorescent OLEDs.

[0073] The compounds of the invention can be made using a variety of methods, including, but not limited to those recited in the examples provided herein. In other aspects, one of skill in the art, in possession of this disclosure, could readily determine an appropriate method for the preparation of an iridium complex as recited herein.

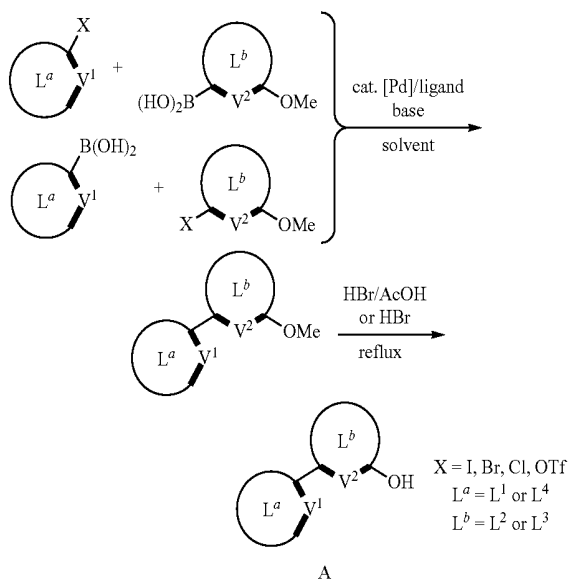
Examples

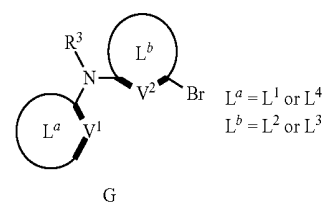
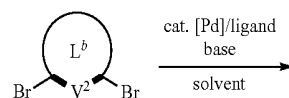
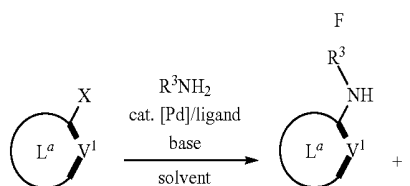
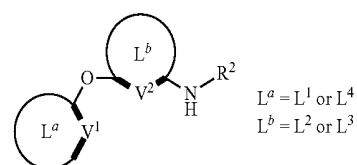
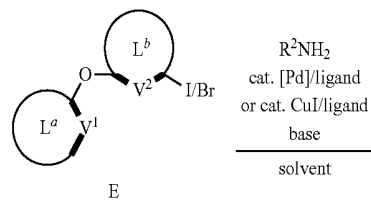
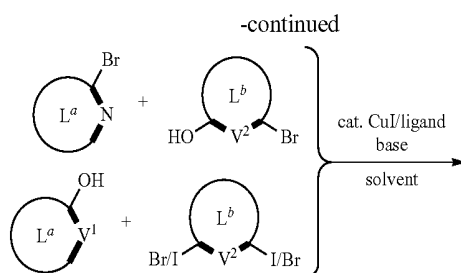
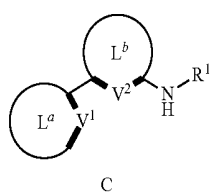
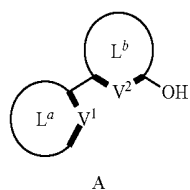
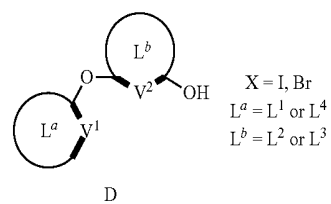
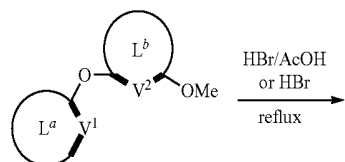
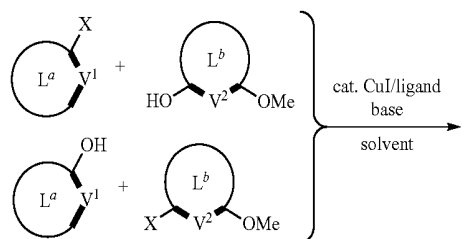
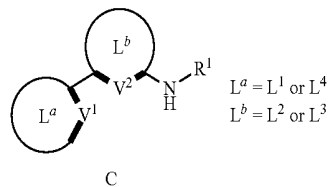
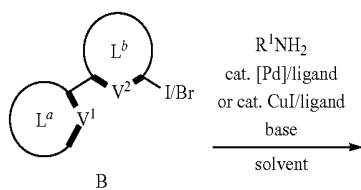
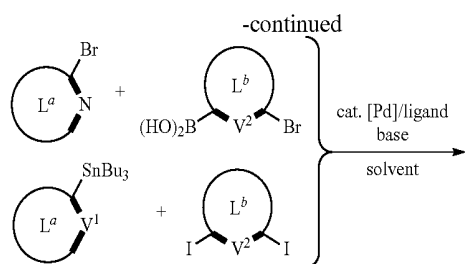
[0074] The following examples are put forth so as to provide those of ordinary skill in the art with a complete disclosure and description of how the compounds, compositions, articles, devices and/or methods claimed herein are made and evaluated, and are intended to be purely exemplary of the invention and are not intended to limit the scope of what the inventors regard as their invention. Efforts have been made to ensure accuracy with respect to numbers (e.g., amounts, temperature, etc.), but some errors and deviations should be accounted for. Unless indicated otherwise, parts are parts by weight, temperature is in ° C. or is at ambient temperature, and pressure is at or near atmospheric.

[0075] Hereinafter, the preparation method of the compounds for the displays and lighting applications will be illustrated. However, the following embodiments are only exemplary and do not limit the scope of the present invention. Temperatures, catalysts, concentrations, reactant compositions, and other process conditions can vary, and one of skill in the art, in possession of this disclosure, could readily select appropriate reactants and conditions for a desired complex.

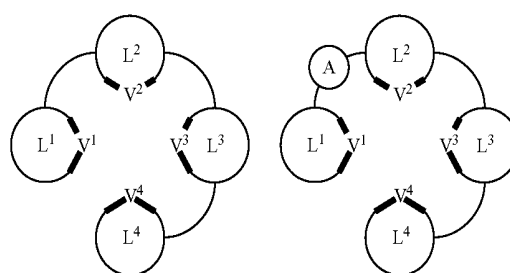
[0076] 1. Prophetic Synthetic Routes

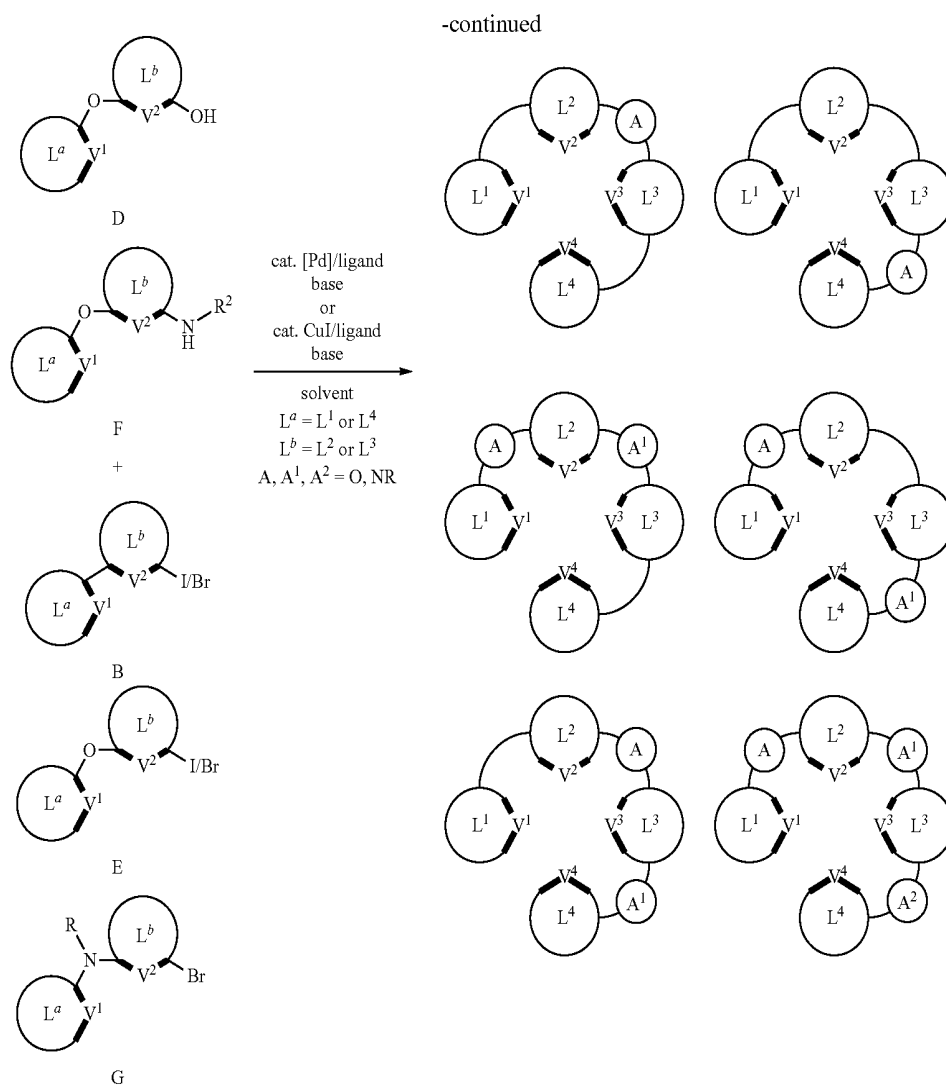
[0077] A general proposed synthetic route for the compounds disclosed herein includes:



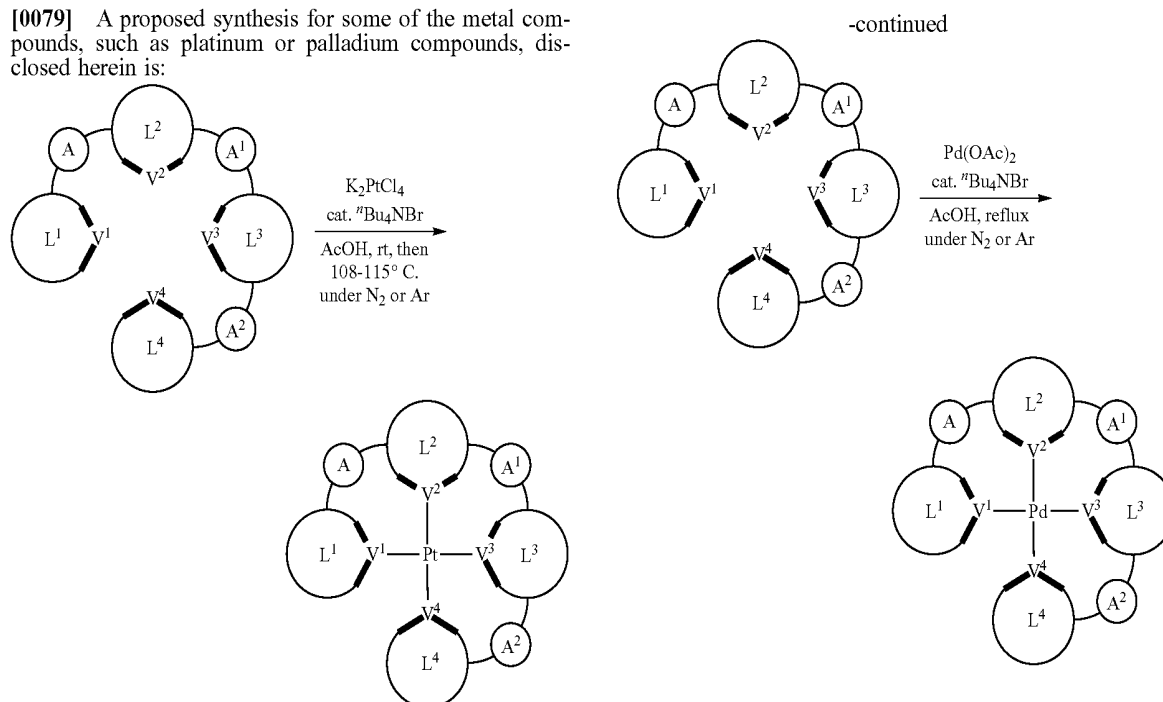


[0078] A proposed synthesis for the disclosed compounds herein also includes:

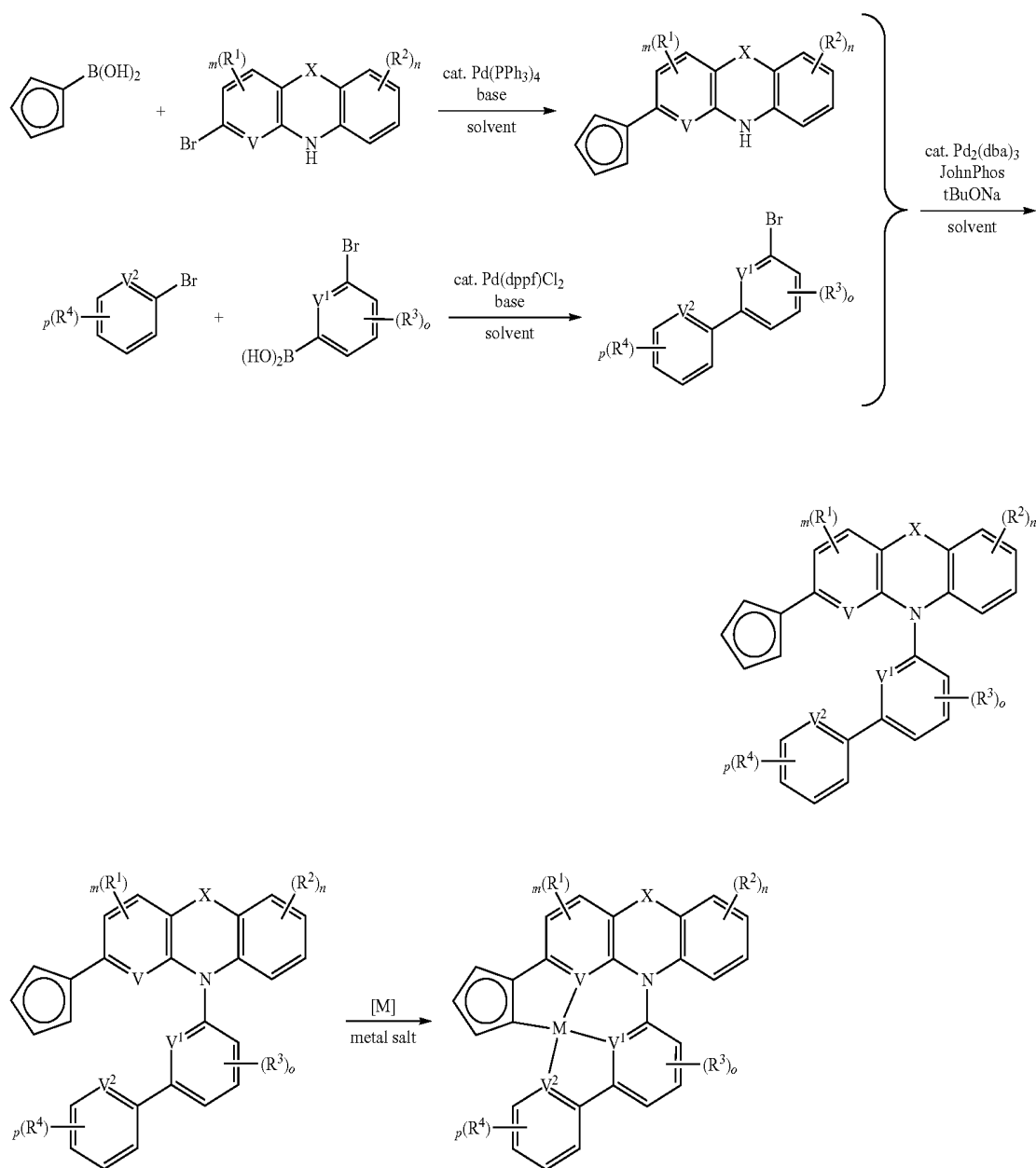




[0079] A proposed synthesis for some of the metal compounds, such as platinum or palladium compounds, disclosed herein is:

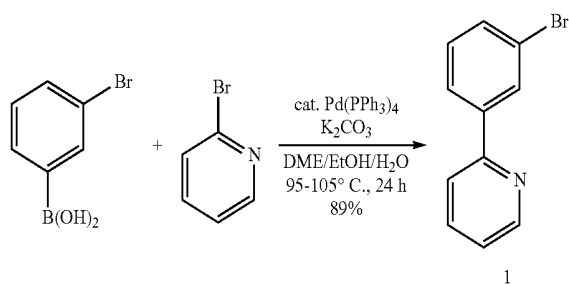


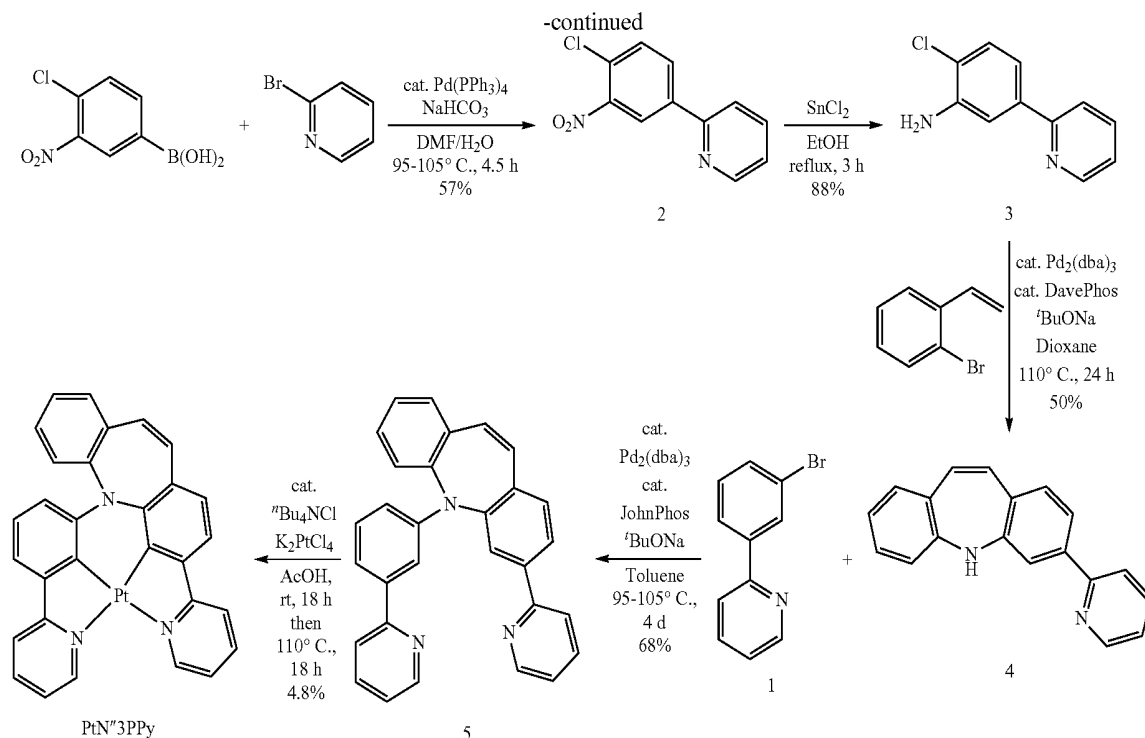
[0080] A proposed synthesis for a metal compound is:



[0081] 2. Synthesis of PtN3"PPy

[0082] Platinum complex PtN3"PPy was prepared according to the following scheme:





[0083] Synthesis of 2-(3-bromophenyl)pyridine 1: To a three-necked flask equipped with a magnetic stir bar and a condenser, added 3-bromophenylboronic acid (8033 mg, 40 mmol) and K₂CO₃ (12.16 g, 88 mmol). The flask was sealed and then evacuated and backfilled with nitrogen. Repeated the evacuation and backfill procedure four additional times. After that solvents DME (100 mL), EtOH (32 mL) and H₂O (44 mL) were added independently by syringe. The mixture was bubbled with nitrogen through a needle for 30 minutes. Then 2-bromopyridine (3.89 mL, 40 mmol) and Pd(PPh₃)₄ (924 mg, 0.8 mmol) were added under the atmosphere of nitrogen. The mixture was heated to reflux (about 95-105° C.) in an oil bath. The reaction was monitored by TLC and about 24 hours later the reaction was completed. Then cooled down to ambient temperature and water was added until the salt dissolved completely. Then the organic layer was separated and the aqueous layer was extracted with ethyl acetate for twice. The combined organic layer was dried over sodium sulfate. Then filtered and washed with ethyl acetate. The filtrate was concentrated under reduced pressure using a rotary evaporator and the residue was purified through column chromatography on silica gel using hexane and ethyl acetate (20:1) as eluent to afford the desired product 2-(3-bromophenyl)pyridine 1 as a colorless liquid 8.33 g in 89% yield. ¹H NMR (CDCl₃, Me₄Si, 400 MHz): δ 7.25-7.28 (m, 1H), 7.34 (td, J=8.0, 1.2 Hz, 1H), 7.53-7.56 (m, 1H), 7.71 (dd, J=8.0, 0.8 Hz, 1H), 7.75-7.80 (m, 1H), 7.90-7.93 (m, 1H), 8.18-8.19 (m, 1H), 8.70-8.81 (m, 1H). ¹H NMR (DMSO-d₆, 400 MHz): δ 7.39-7.42 (m, 1H), 7.47 (t, J=8.0 Hz, 1H), 7.65 (dt, J=8.4, 0.8 Hz, 1H), 7.91 (tt, J=7.2, 0.8 Hz, 1H), 8.02-8.04 (m, 1H), 8.09-8.12 (m, 1H), 8.29-8.30 (m, 1H), 8.68-8.70 (m, 1H).

[0084] Synthesis of 2-(4-chloro-3-nitrophenyl)pyridine 2: To a three-necked flask equipped with a magnetic stir bar and a condenser, added 4-chloro-3-nitrophenylboronic acid (3968 mg, 19.7 mmol) and NaHCO₃ (3310 mg, 39.4 mmol). The flask was sealed and then evacuated and backfilled with

nitrogen. Repeated the evacuation and backfill procedure four additional times. After that solvents DMF (40 mL) and H₂O (20 mL) were added independently by syringe. The mixture was bubbled with nitrogen through a needle for 30 minutes. Then 2-bromopyridine (3.11 mL, 21.7 mmol) and Pd(PPh₃)₄ (1138 mg, 0.99 mmol) were added under the atmosphere of nitrogen. The mixture was heated to reflux (about 95-105° C.) in an oil bath. The reaction was monitored by TLC and after about 4.5 hours the reaction was completed. Then cooled down to ambient temperature and water was added. Then the mixture was filtered and washed with ethyl acetate. The organic layer of the filtrate was separated and the aqueous layer was extracted with ethyl acetate for twice. The combined organic layer was dried over sodium sulfate. Then filtered and washed with ethyl acetate. The filtrate was concentrated under reduced pressure using a rotary evaporator and the residue was purified through column chromatography on silica gel using hexane and ethyl acetate (20:1-15:1-10:1-5:1) as eluent to afford the desired product 2-(4-chloro-3-nitrophenyl)pyridine 2 as a grey solid 2.65 g in 57% yield. ¹H NMR (DMSO-d₆, 400 MHz): δ 7.47-7.50 (m, 1H), 7.92 (d, J=8.4 Hz, 1H), 7.99 (td, J=7.6, 0.8 Hz, 1H), 8.16 (d, J=8.0 Hz, 1H), 8.44 (dd, J=8.0, 0.8 Hz, 1H), 8.74-8.75 (m, 1H), 8.77 (d, J=2.0 Hz, 1H).

[0085] Synthesis of 2-chloro-5-(pyridin-2-yl)benzenamine 3: To a single-necked flask equipped with a magnetic stir bar, added 2-(4-chloro-3-nitrophenyl)pyridine 2 (2.65 g, 11.29 mmol) and SnCl₂ (21.41 g, 112.90 mmol). Then a condenser was equipped and the system was evacuated and backfilled with nitrogen. Repeated the evacuation and backfill procedure twice additional times. After that solvent EtOH (150 mL) was added under the atmosphere of nitrogen. The mixture was heated to 60-70° C. in an oil bath. The reaction was monitored by TLC and after about 3 hours the reaction was completed. Then cooled down to ambient temperature and quenched with water. Then the pH of the mixture was adjusted to 9-10 using 1 N NaOH aqueous

solution. Filtered and washed with ethyl acetate. The organic layer of the filtrate was separated and the aqueous layer was extracted with ethyl acetate for three times. The combined organic layer was dried over sodium sulfate. Then filtered and washed with ethyl acetate. The filtrate was concentrated under reduced pressure using a rotary evaporator and the residue was purified through column chromatography on silica gel using hexane and ethyl acetate (5:1-3:1-2:1) as eluent to afford the desired product 2-chloro-5-(pyridin-2-yl)benzenamine 3 as a colorless liquid 2.03 g in 88% yield. ¹H NMR (DMSO-d₆, 400 MHz): δ 5.48 (bs, 2H), 7.23 (dd, J=8.0, 2.0 Hz, 1H), 7.29 (d, J=8.0 Hz, 1H), 7.33-7.36 (m, 1H), 7.60 (d, J=2.0 Hz, 1H), 7.82-7.89 (m, 2H), 8.63-8.64 (m, 1H).

[0086] Synthesis of 3-(pyridin-2-yl)-5H-dibenzo[b,f]azepine 4: To an oven dried pressure vessel equipped with a magnetic stir bar, added Pd₂(dba)₃ (268 mg, 0.29 mmol), DavePhos (346 mg, 0.88 mmol) and ^tBuONa (2817 mg, 29.31 mmol). The vessel was evacuated and backfilled with nitrogen. Repeated the evacuation and backfill procedure four additional times. Then a degassed solution of 2-chloro-5-(pyridin-2-yl)benzenamine 3 (2000 mg, 9.77 mmol) and 2-bromostyrene (1.52 mL, 11.72 mmol) in dioxane (20 mL) was added under the atmosphere of nitrogen. The sealed vessel was then placed in a preheated oil bath at a temperature of 110° C. The reaction was stirred and monitored by TLC and after about 24 hours the reaction was completed. Cooled down to ambient temperature and quenched with water. Then diluted with ethyl acetate and stirred for 10 minutes, filtered and washed with ethyl acetate. The organic layer of the filtrate was separated and the aqueous layer was extracted with ethyl acetate for three times. The combined organic layer was dried over sodium sulfate. Then filtered and washed with ethyl acetate. The filtrate was concentrated under reduced pressure using a rotary evaporator and the residue was purified through column chromatography on silica gel using hexane and ethyl acetate (5:1-3:1) as eluent to afford the desired product 3-(pyridin-2-yl)-5H-dibenzo[b,f]azepine 4 as a brown-yellow solid 1.33 g in 50% yield. ¹H NMR (DMSO-d₆, 400 MHz): δ 6.10 (s, 2H), 6.65 (d, J=8.0 Hz, 1H), 6.69 (td, J=7.6, 1.2 Hz, 1H), 6.76 (dd, J=7.6, 1.6 Hz, 1H), 6.85 (d, J=8.0 Hz, 1H), 6.98 (td, J=8.0, 1.6 Hz, 1H), 7.13 (s, 1H), 7.33-7.39 (m, 2H), 7.47 (d, J=1.6 Hz, 1H), 7.82-7.89 (m, 2H), 8.64 (dd, J=4.8, 0.8 Hz, 1H).

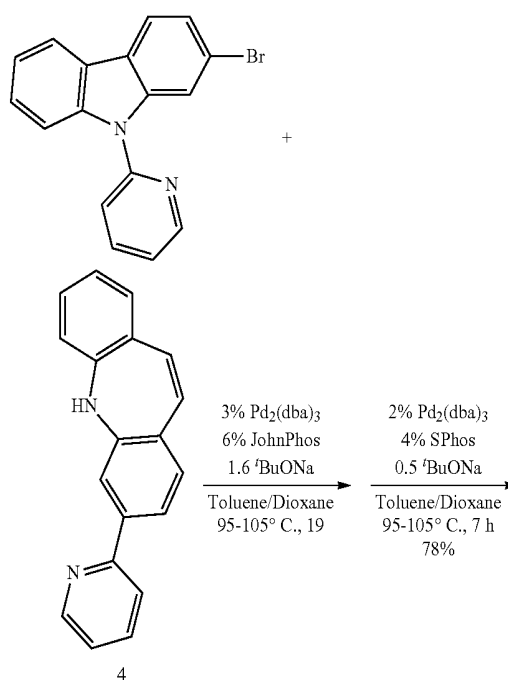
[0087] Synthesis of 3-(pyridin-2-yl)-5-(3-(pyridin-2-yl)phenyl)-5H-dibenzo[b,f]azepine 5: To an oven dried pressure vessel equipped with a magnetic stir bar, added 3-(pyridin-2-yl)-5H-dibenzo[b,f]azepine 4 (1.33 g, 4.92 mmol), Pd₂(dba)₃ (180 mg, 0.20 mmol), JohnPhos (118 mg, 0.39 mmol) and ^tBuONa (756 mg, 7.87 mmol). The vessel was evacuated and backfilled with nitrogen. Repeated the evacuation and backfill procedure four additional times. Then a degassed solution of 2-(3-bromophenyl)pyridine 1 (1728 mg, 7.38 mmol) in toluene (20 mL) was added under the atmosphere of nitrogen. The sealed vessel was then placed in a preheated oil bath at a temperature of 95-105° C. The reaction was stirred and monitored by TLC and after about 4 days the reaction was completed. Cooled down to ambient temperature. The solvent was removed under reduced pressure using a rotary evaporator and the residue was purified through column chromatography on silica gel using hexane and ethyl acetate (5:1-3:1) as eluent to afford the desired

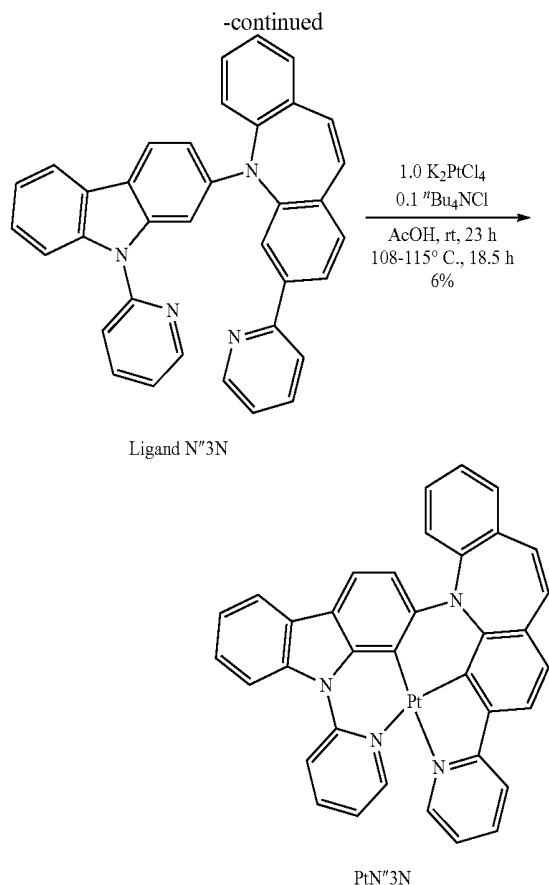
ligand 3-(pyridin-2-yl)-5-(3-(pyridin-2-yl)phenyl)-5H-dibenzo[b,f]azepine 5 as a brown solid 1.42 g in 68% yield. ¹H NMR (DMSO-d₆, 400 MHz): δ 7.2, 1.6 Hz, 1H), 7.61 (s, 1H), 7.63 (s, 1H), 7.65 (dd, J=7.2, 1.6 Hz, 1H), 7.68 (dd, J=7.6, 1.6 Hz, 1H), 7.71 (d, J=8.4 Hz, 1H), 7.78 (td, J=7.6, 2.0 Hz, 1H), 7.92 (td, J=8.0, 1.6 Hz, 1H), 8.11 (d, J=8.0, 1H), 8.20 (dd, J=8.0, 2.0 Hz, 1H), 8.33 (d, J=2.0 Hz, 1H), 8.51 (dd, J=4.8, 0.8 Hz, 1H), 8.71 (dd, J=4.8, 0.8 Hz, 1H).

[0088] Synthesis of platinum complex PtN³3PPy: To an oven dried pressure vessel equipped with a magnetic stir bar, ligand 3-(pyridin-2-yl)-5-(3-(pyridin-2-yl)phenyl)-5H-dibenzo[b,f]azepine 5 (84.7 mg, 0.20 mmol), K₂PtCl₄ (87.2 mg, 0.21 mmol) and ^tBu₄NCl (5.5 mg, 0.02 mmol). The vessel was taken into a glove box. Then solvent acetic acid (12 mL) was added. The mixture was bubbled with nitrogen for 30 minutes. Then the vessel was sealed and taken out of the glove box and stirred at ambient temperature for 18 hours. After that the mixture was heated in an oil bath at a temperature of 110° C. for another 18 hours. Cooled down to ambient temperature and water (about 24 mL) was added. The precipitate was collected through filter and washed with water for three times. The precipitate was dried in air and purified through column chromatography on silica gel using dichloromethane as eluent to obtain a crude product which was further purified by recrystallization in dichloromethane and ether at refrigerator to get the desired platinum complex PtN³3PPy as a brown-yellow solid 5.9 mg in 4.8% yield. ¹H NMR (DMSO-d₆, 400 MHz): δ 6.43 (d, J=8.0 Hz, 1H), 6.75 (d, J=8.4 Hz, 1H), 6.92 (d, J=7.2 Hz, 1H), 6.98-7.07 (m, 5H), 7.19 (dd, J=7.4, 1.6 Hz, 1H), 7.58-7.63 (m, 3H), 7.67 (d, J=8.0 Hz, 1H), 8.11 (d, J=7.2 Hz, 1H), 8.15 (d, J=7.6 Hz, 1H), 8.21 (d, J=8.0 Hz, 1H), 8.11 (d, J=8.0, 1H), 8.25 (d, J=7.6 Hz, 1H), 9.15 (d, J=5.2 Hz, 1H), 8.17 (d, J=5.2 Hz, 1H); HRMS (MALDI/DHB) for C₃₀H₂₀N₃Pt [M+H]⁺: calcd 617.13, found 617.33.

[0089] 3. Synthesis of PtN³3N

[0090] Platinum complex PtN³3N can be prepared according to the following scheme:



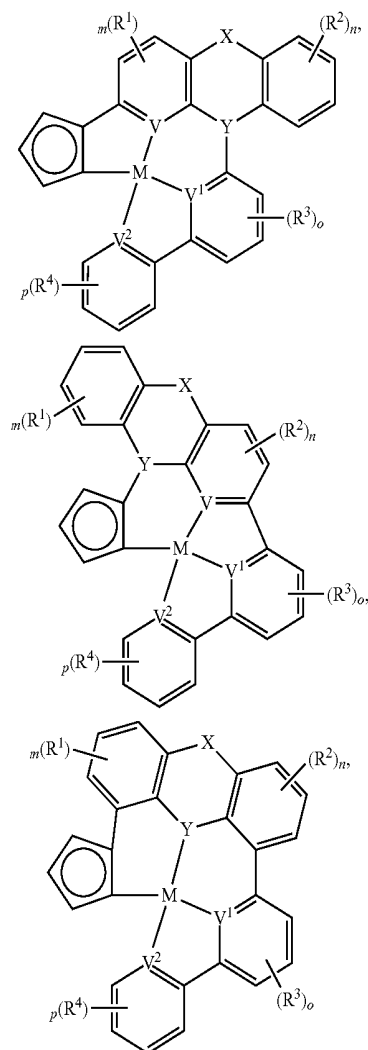


[0091] Synthesis of 3-(pyridin-2-yl)-5-(9-(pyridin-2-yl)-9H-carbazol-2-yl)-5H-dibenzo[b,f]azepine Ligand N*3N: To a dry Schlenk tube equipped with a magnetic stir bar, added of 3-(pyridin-2-yl)-5H-dibenzo[b,f]azepine 4 (92 mg, 0.34 mmol, 1.0 eq), 2-bromo-9-(pyridin-2-yl)-9H-carbazole (132 mg, 0.41 mmol, 1.2 eq), Pd₂(dba)₃ (9 mg, 0.010 mmol, 0.03 eq), JohnPhos (6 mg, 0.020 mmol, 0.06 eq) and ^tBuONa (52 mg, 0.54 mmol, 1.6 eq). The tube was evacuated and back-filled with nitrogen and this evacuation/back-fill procedure was repeated for another twice. Then dry solvents toluene (4 mL) and dioxane (3 mL) were added under the atmosphere of nitrogen, the tube was then sealed quickly. And then the mixture was stirred in an oil bath at a temperature of 95-105° C. The reaction was monitored by TLC. 19 hours later, much of the starting material 4 was not consumed. So more Pd₂(dba)₃ (6 mg, 0.0067 mmol, 0.02 eq), and SPhos (5.6 mg, 0.0134 mmol, 0.04 eq), ^tBuONa (16 mg, 0.21 mmol, 0.5 eq) were added. The mixture was stirred in an oil bath at a temperature of 95-105° C. for another 7 hours, cooled down to ambient temperature. The mixture was concentrated and the residue was purified through column chromatography on silica gel using hexane/ethyl acetate (20:1-10:1-5:1-3:1) as eluent to obtain the desired product Ligand N*3N as a brown solid 135 mg in 78% yield. ¹H NMR (DMSO-d₆, 400 MHz): δ 6.26 (dd, J=8.4, 2.4 Hz, 1H), 6.73 (d, J=1.6 Hz, 1H), 6.98 (s, 2H), 7.21 (t, J=7.6 Hz, 1H), 7.26-7.31 (m, 2H), 7.41-7.52 (m, 3H), 7.60-7.70 (m, 5H), 7.79-7.83 (m, 2H), 7.93-7.97 (m, 2H), 8.11 (d, J=7.2 Hz, 1H), 8.18 (dd, J=7.6, 2.4 Hz, 1H), 8.36 (d, J=1.6 Hz, 1H), 8.40 (dd, J=4.8, 2.0 Hz, 1H), 8.72-8.73 (m, 1H).

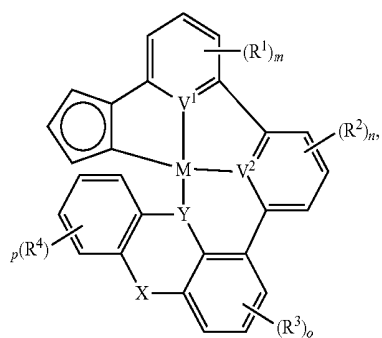
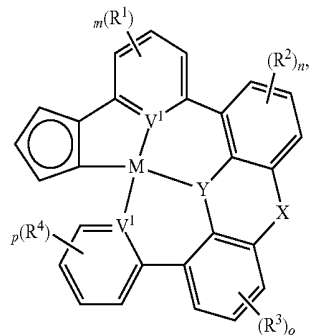
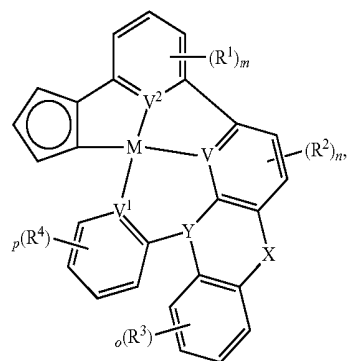
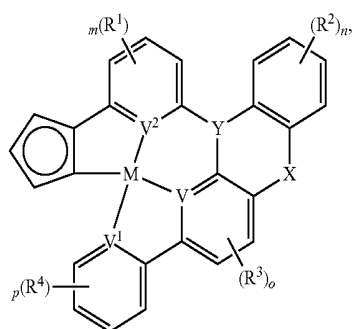
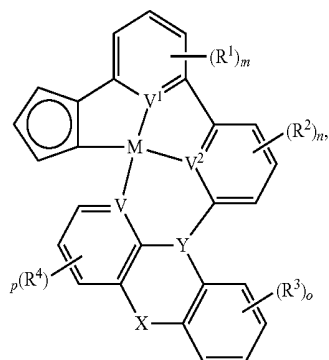
[0092] Synthesis of 3-(pyridin-2-yl)-5-(9-(pyridin-2-yl)-9H-carbazol-2-yl)-5H-dibenzo[b,f]azepine platinum complex PtN*3N: To a dry pressure tube equipped with a magnetic stir bar, added 3-(pyridin-2-yl)-5-(9-(pyridin-2-yl)-9H-carbazol-2-yl)-5H-dibenzo[b,f]azepine Ligand N*3N (51 mg, 0.1 mmol, 1.0 eq), K₂PtCl₄ (42 mg, 0.1 mmol, 1.0 eq) and ^tBu₄NCl (2.8 mg, 0.01 mmol, 0.1 eq). Then the tube was taken into a glove box. Solvent acetic acid (6 mL) was added. The mixture was bubbled with nitrogen for 30 minutes and then the tube was sealed. The tube was taken out of the glove box and the mixture was stirred in an oil bath at a temperature of 108-115° C. for 18.5 hours. Then the mixture was cooled down to ambient temperature and water (12 mL) was added slowly. The precipitate was filtered off and washed with water for three times. Then the solid was dried in air under reduced pressure. The collected solid was purified through column chromatography on silica gel using dichloromethane/hexane (1:1) as eluent to obtain the desired product PtN*3N 4.5 mg in 6% yield.

1-22. (canceled)

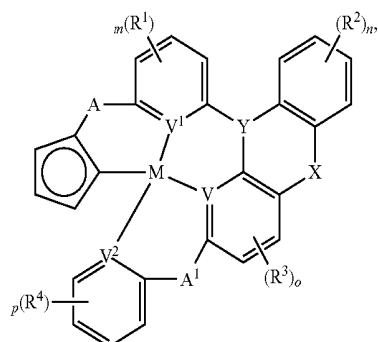
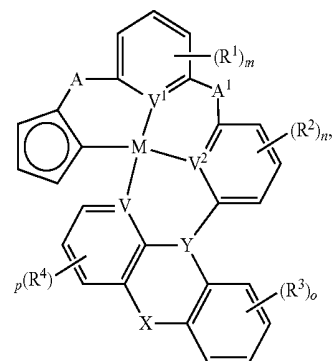
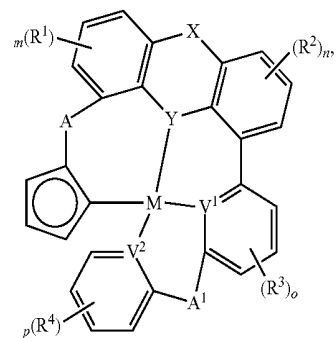
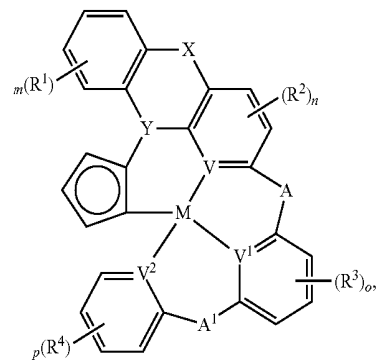
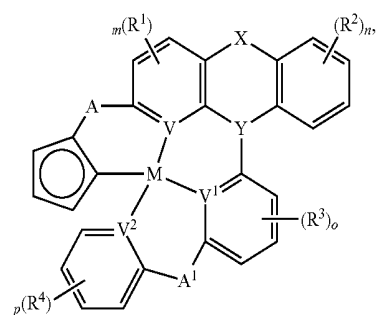
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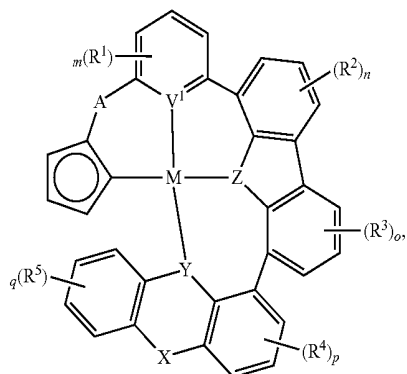
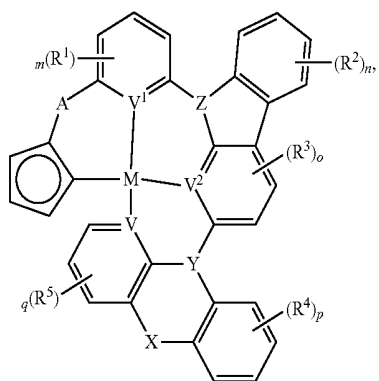
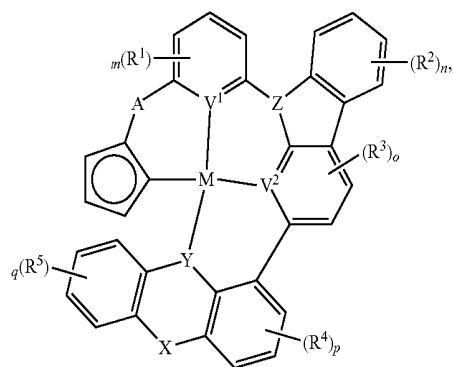
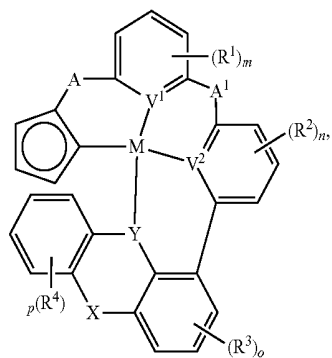
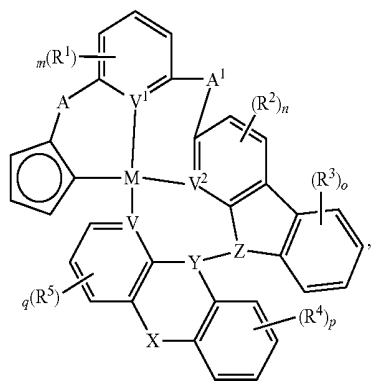
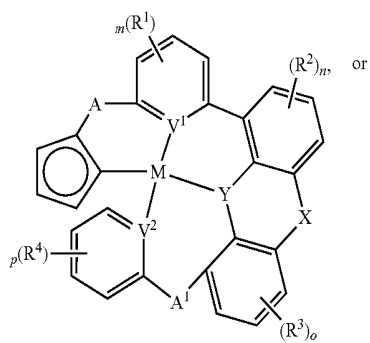
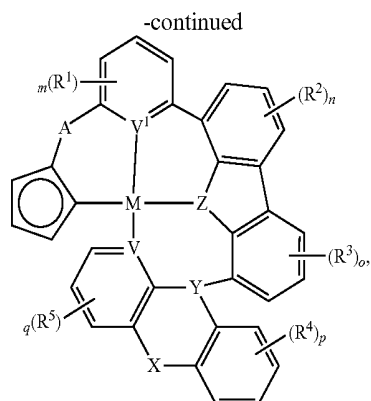
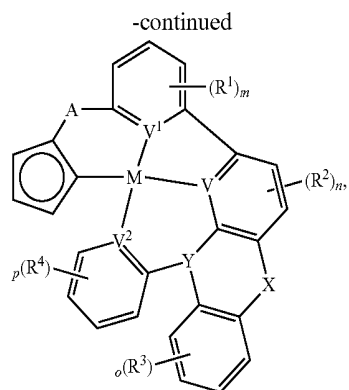


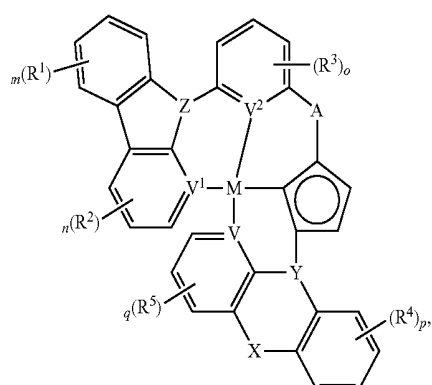
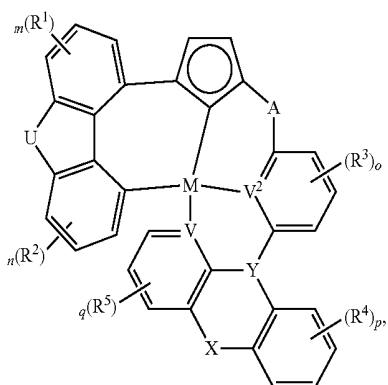
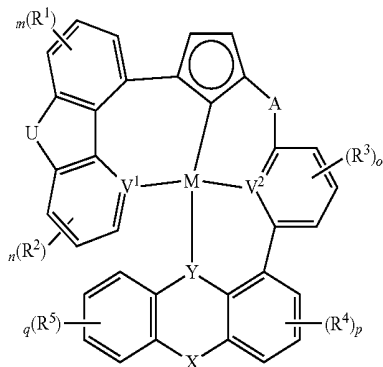
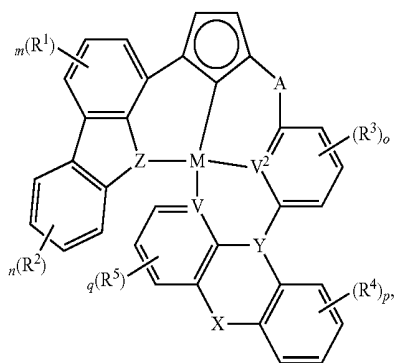
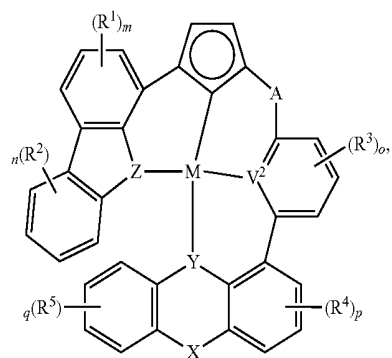
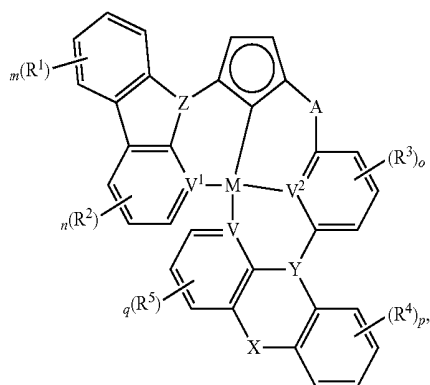
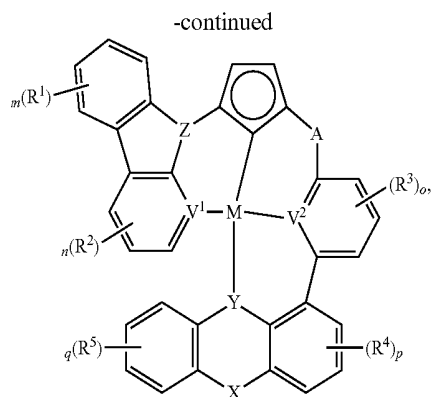
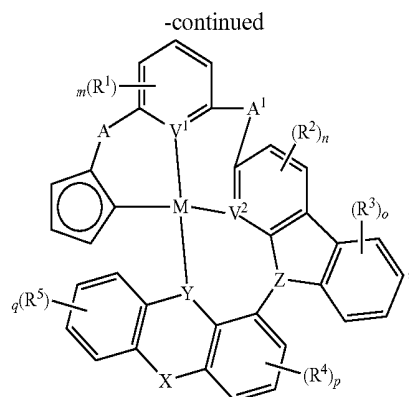
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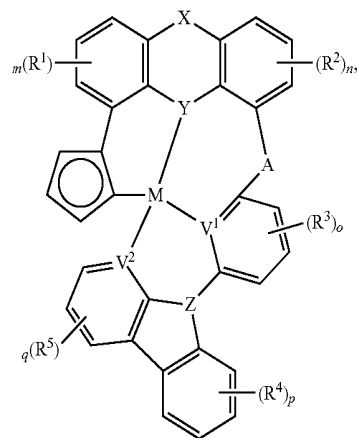
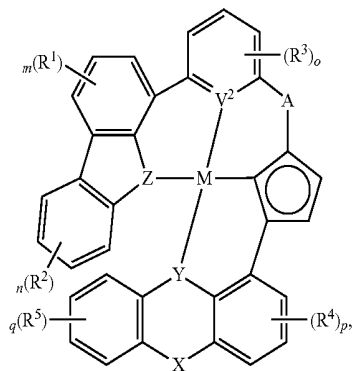
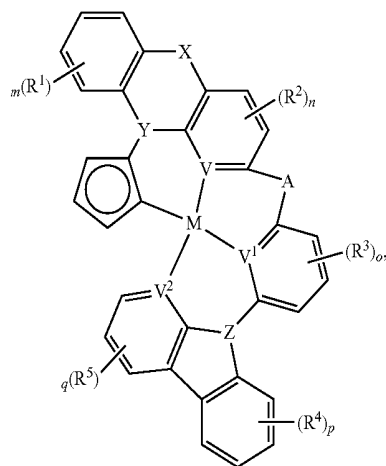
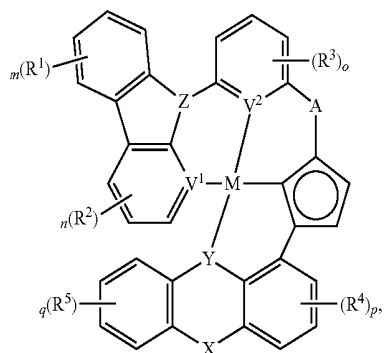
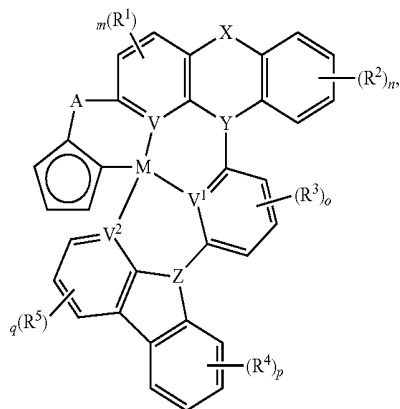
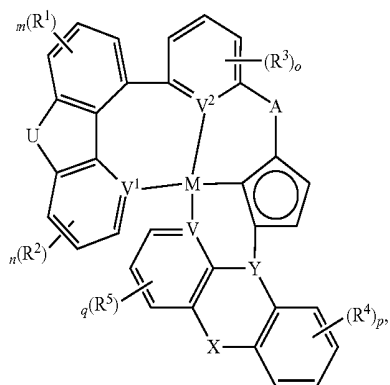
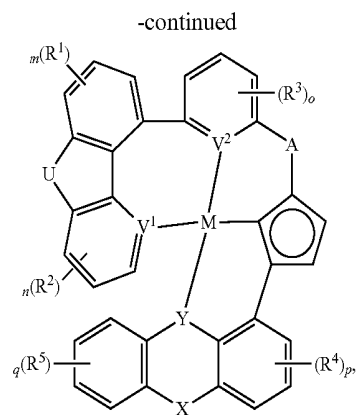
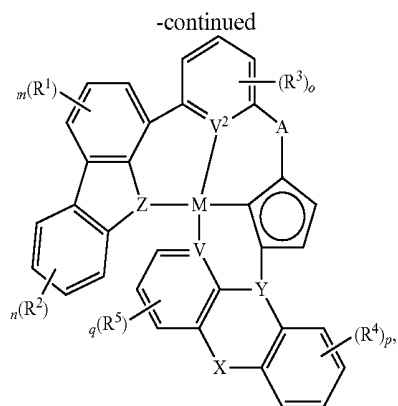


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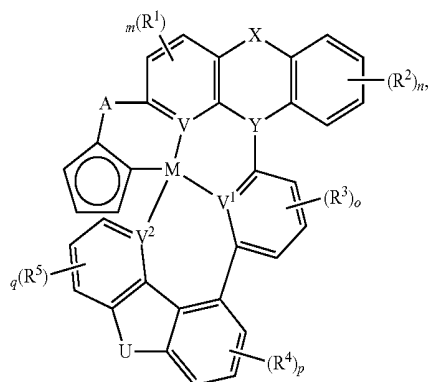
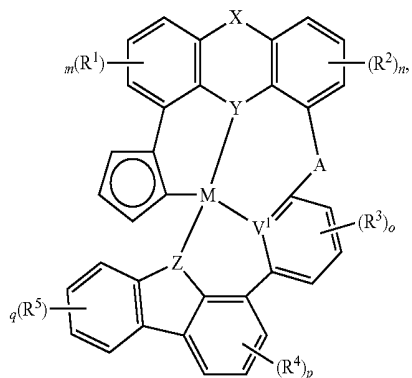
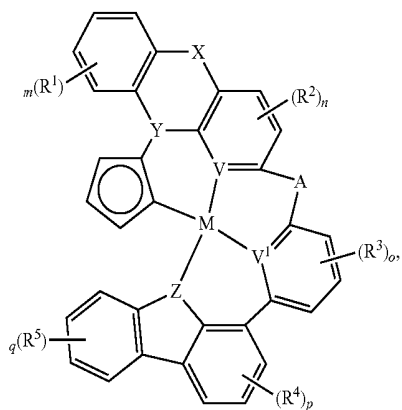
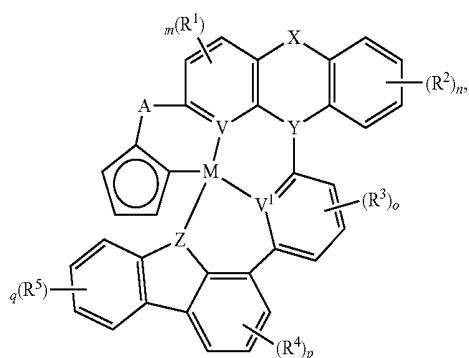




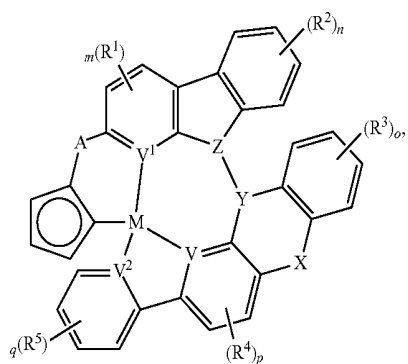
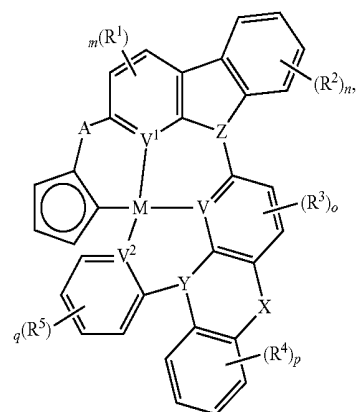
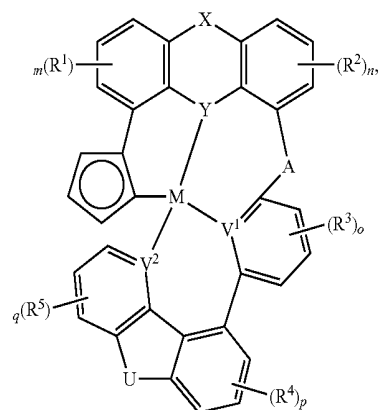
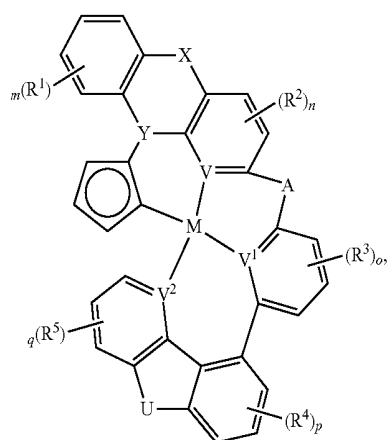




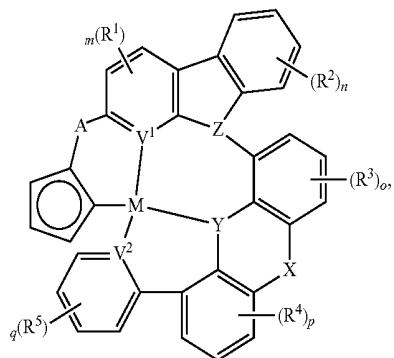
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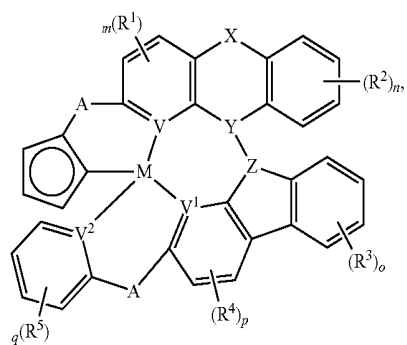
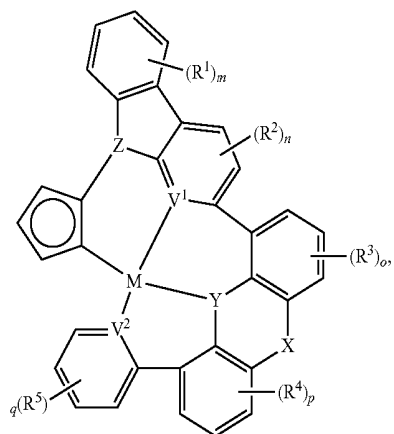
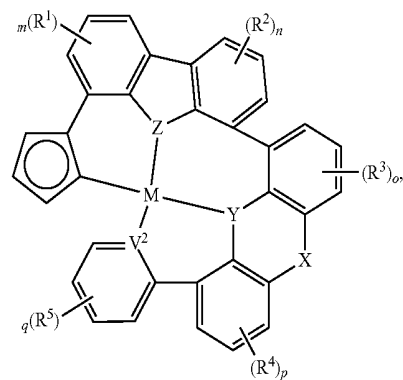
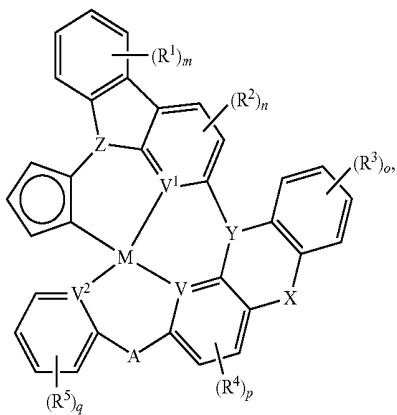
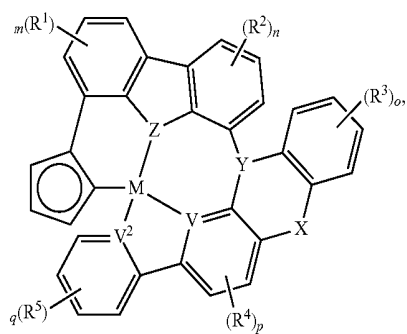
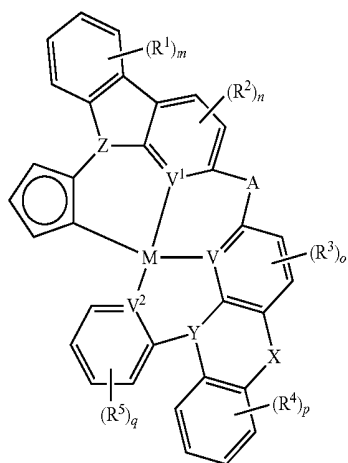
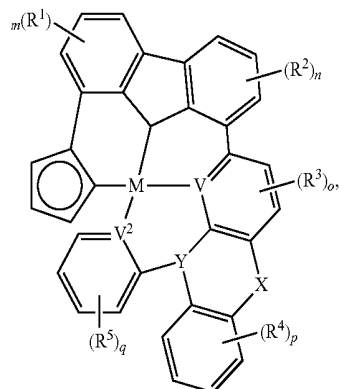
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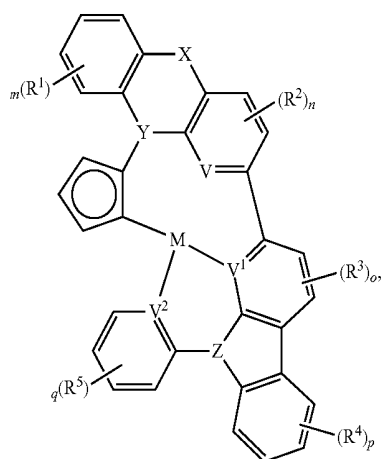
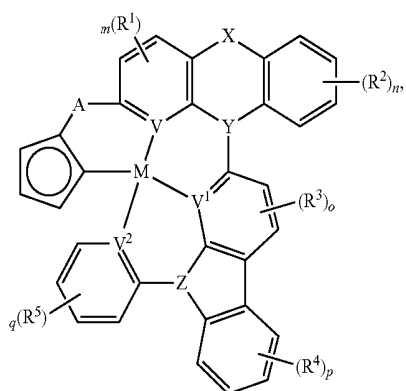
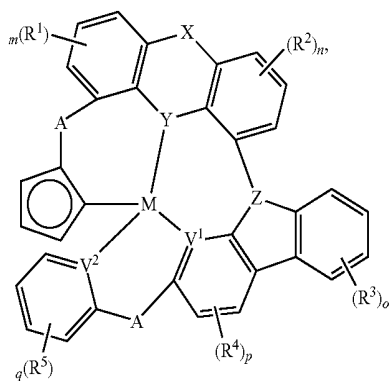
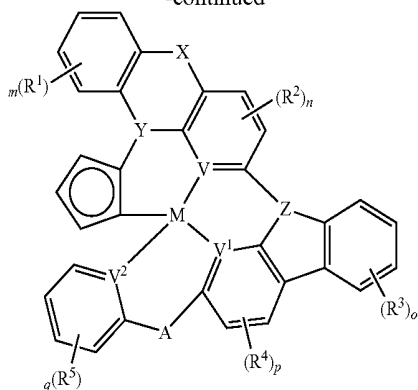
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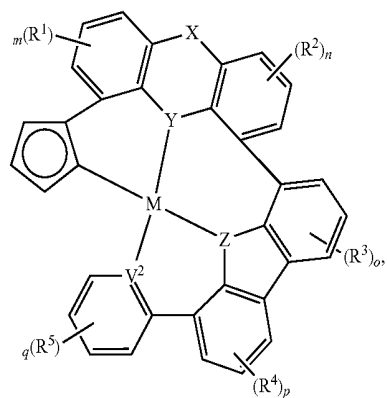
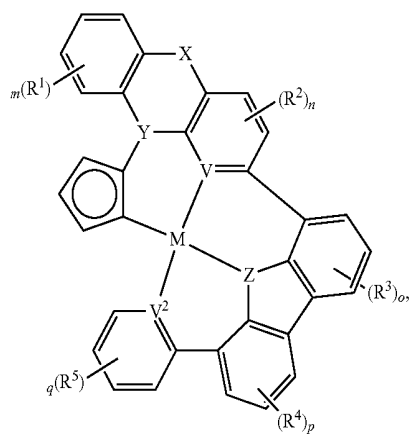
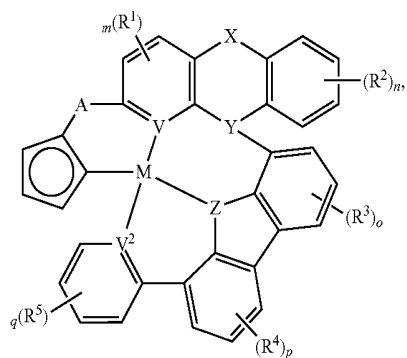
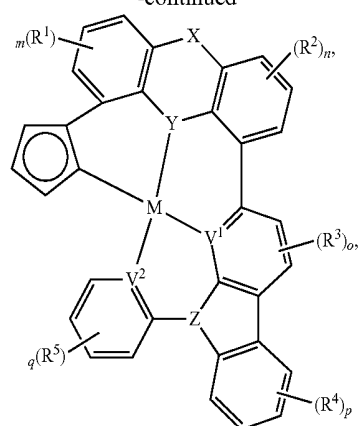
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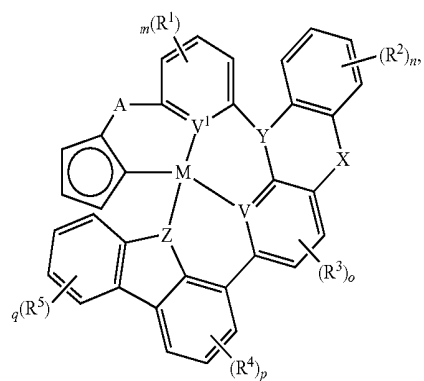
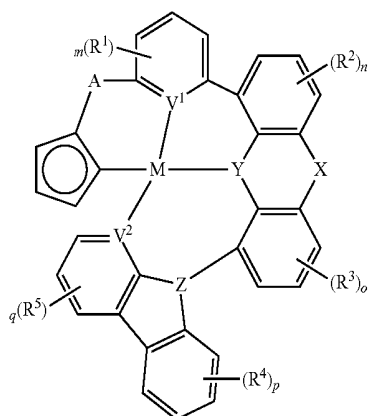
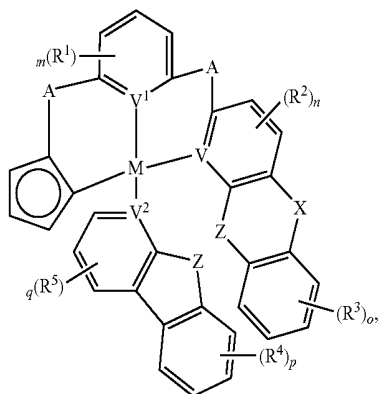
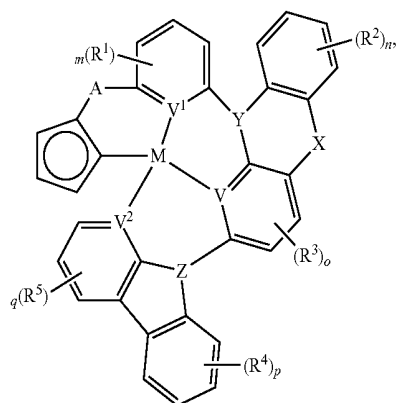
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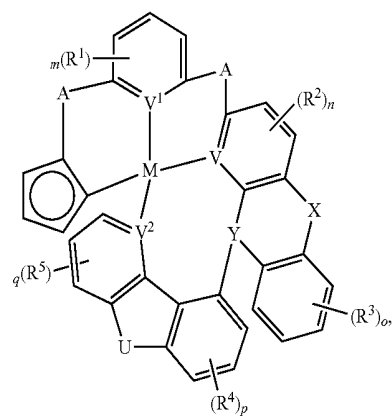
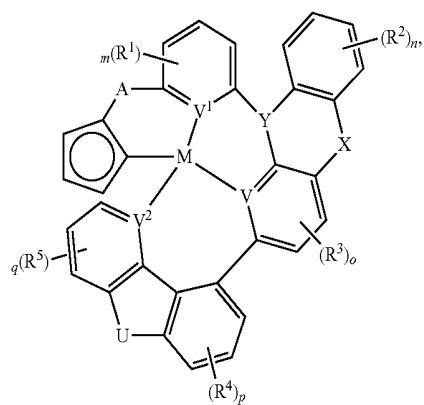
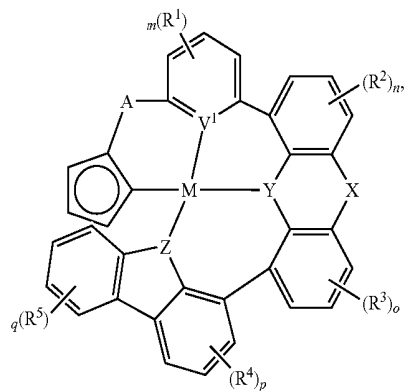
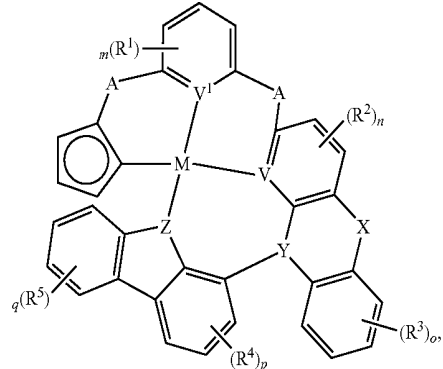
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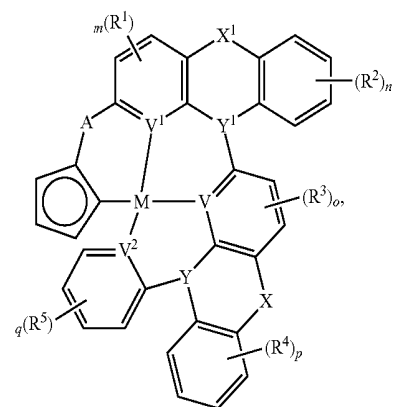
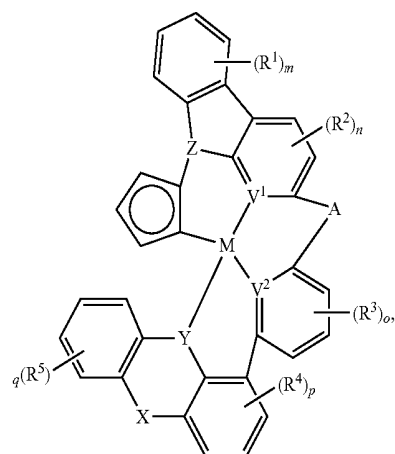
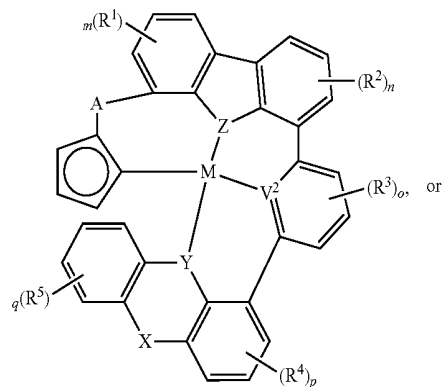
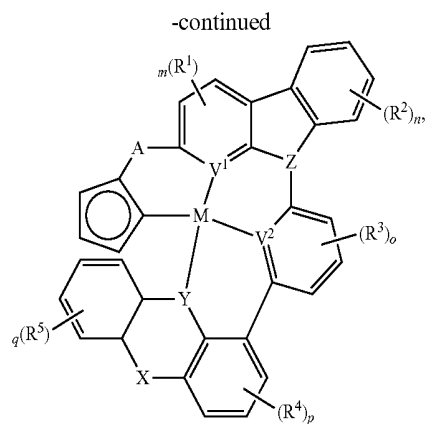
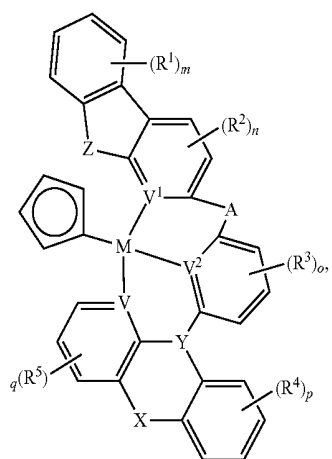
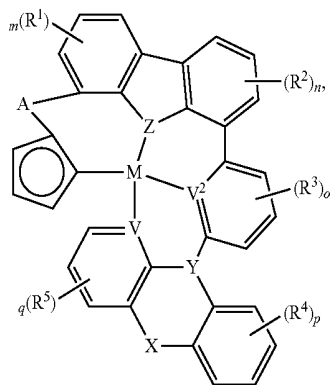
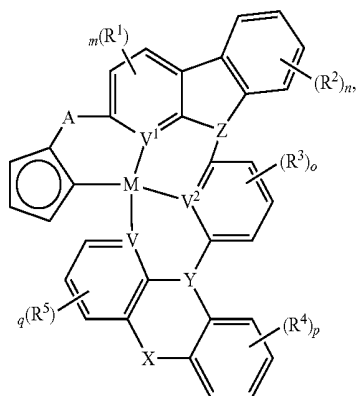
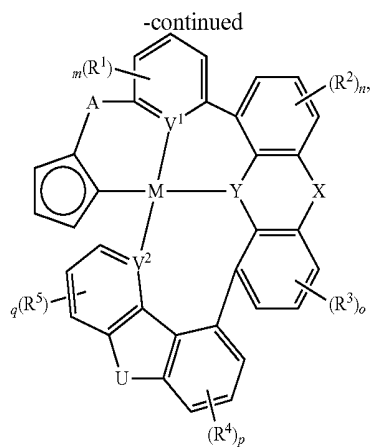


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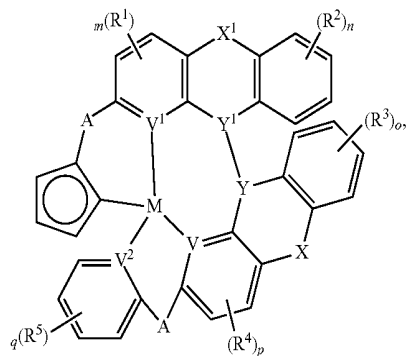


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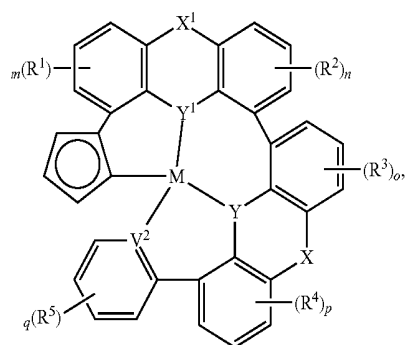
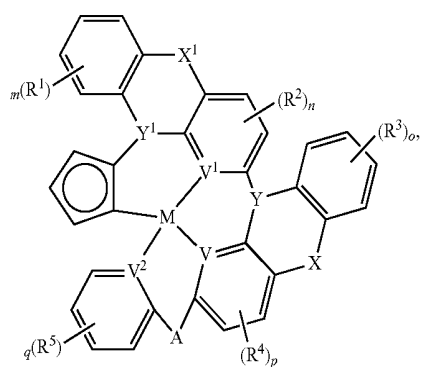
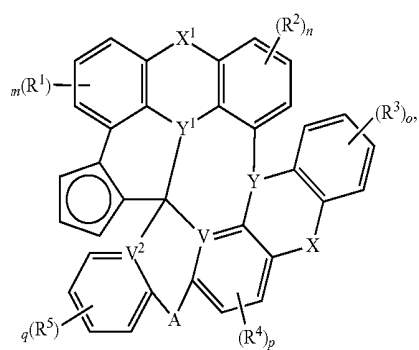
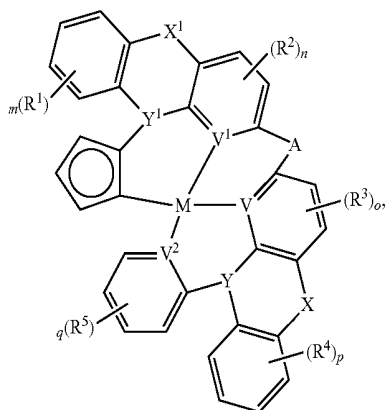
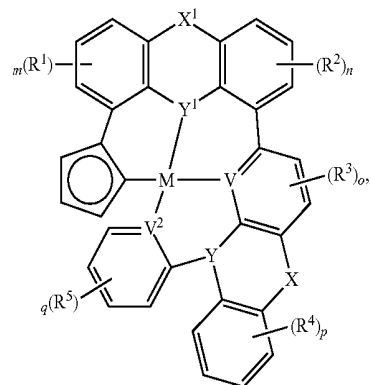
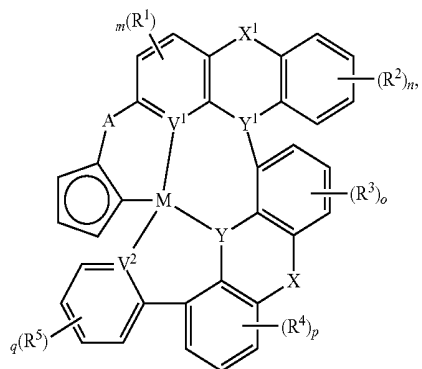
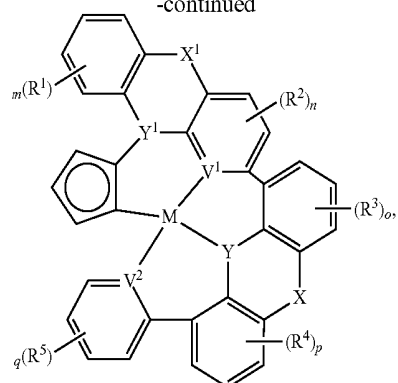




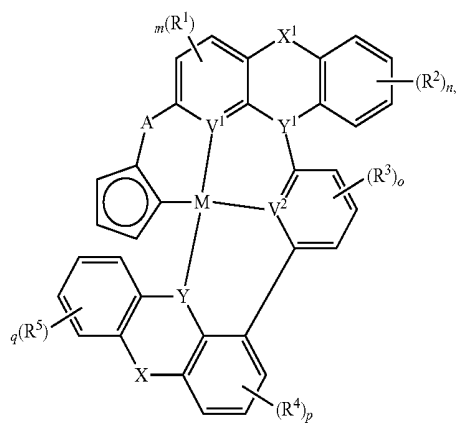
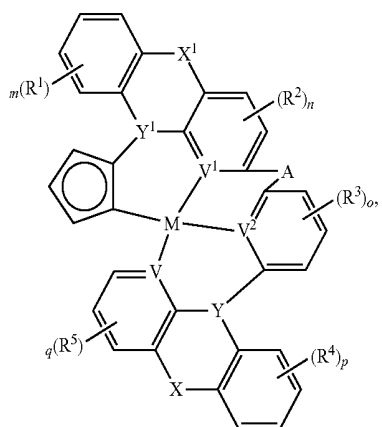
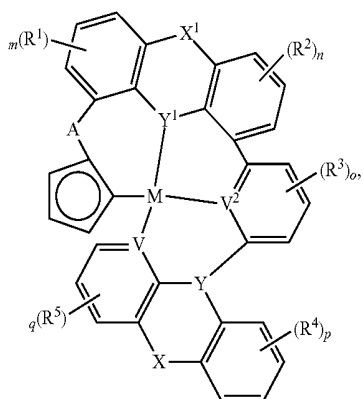
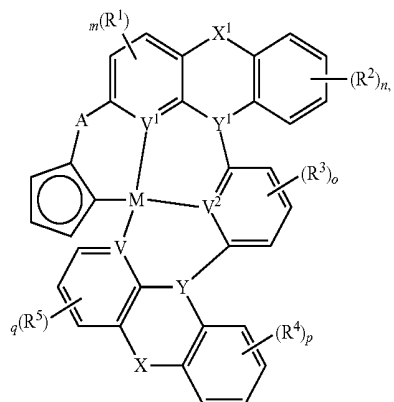
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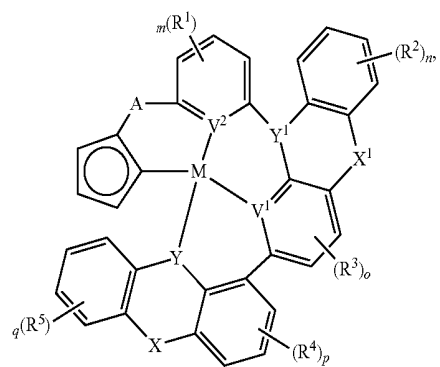
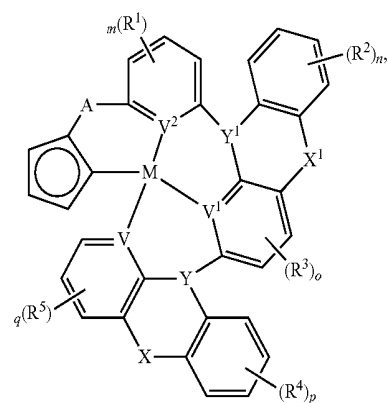
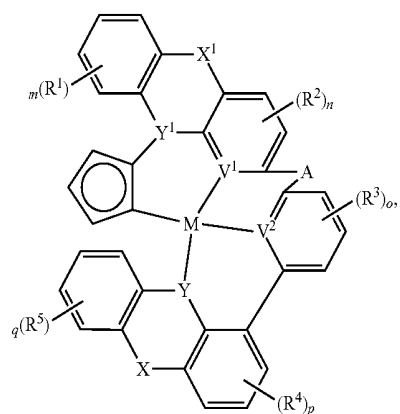
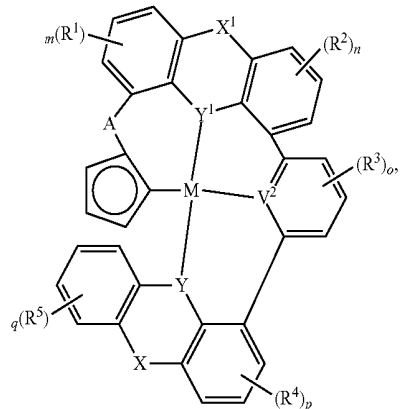
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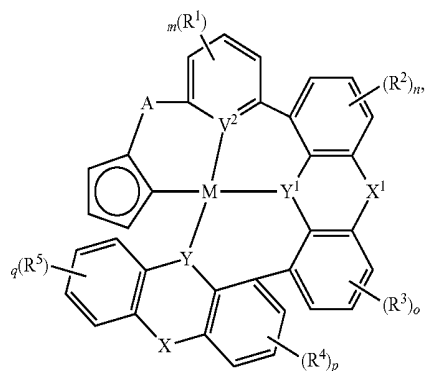
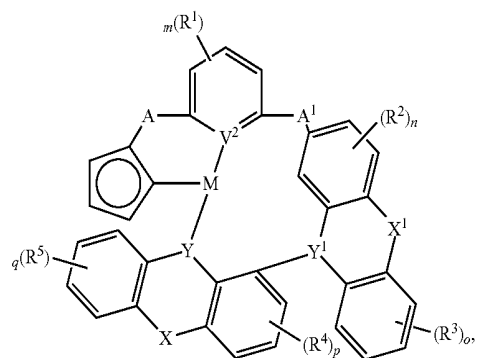
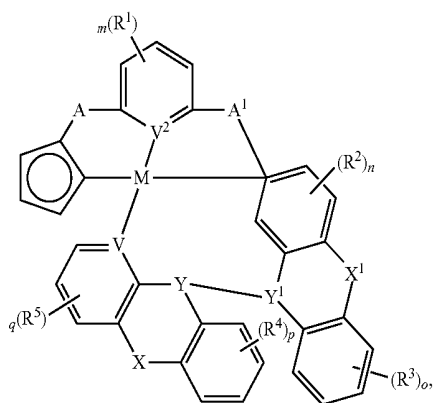
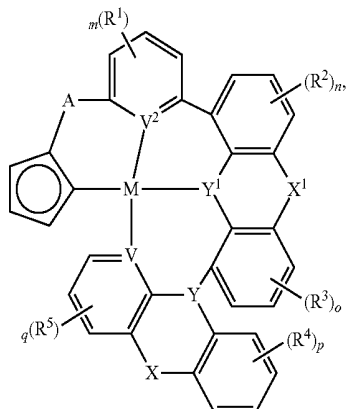
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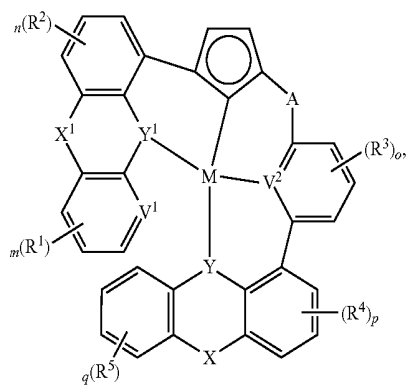
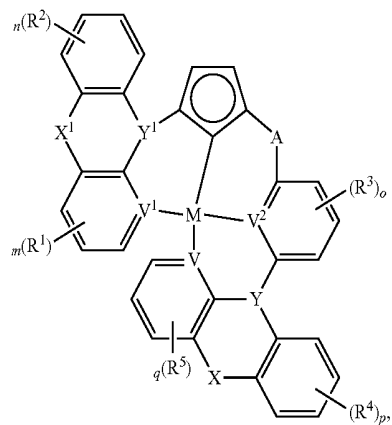
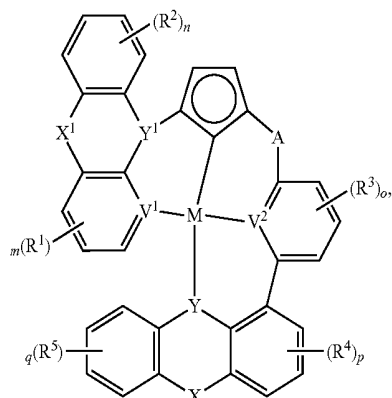
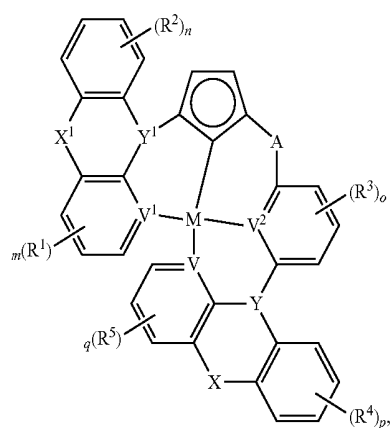
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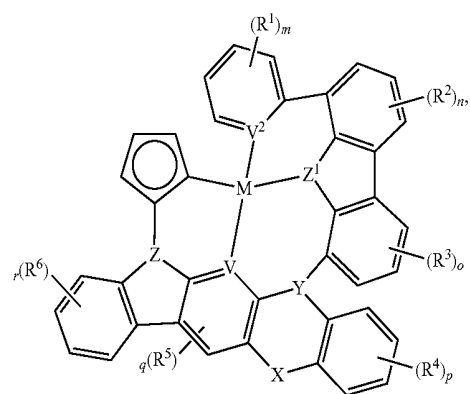
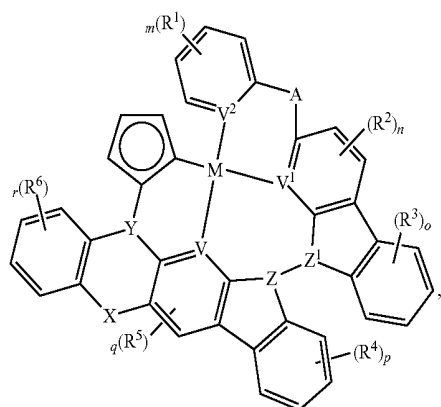
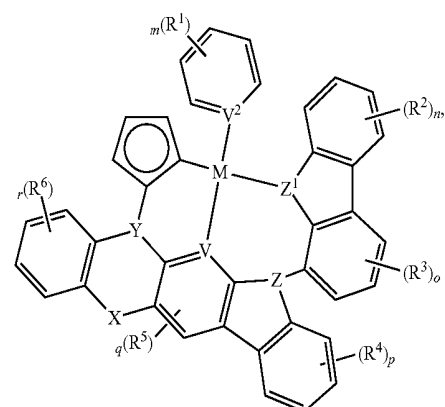
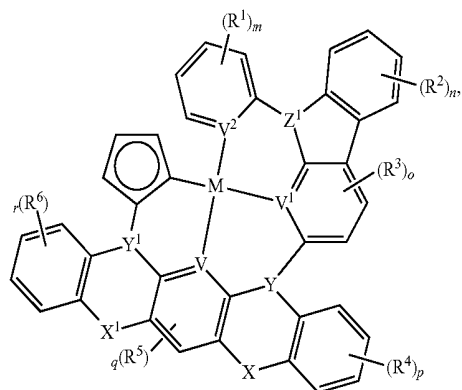
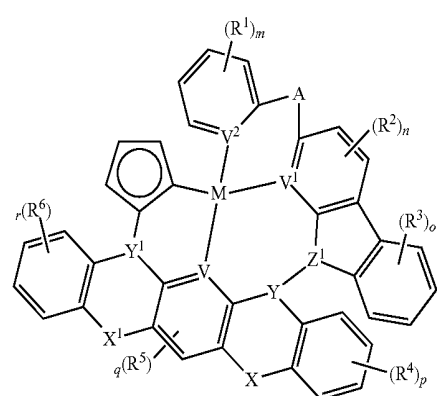
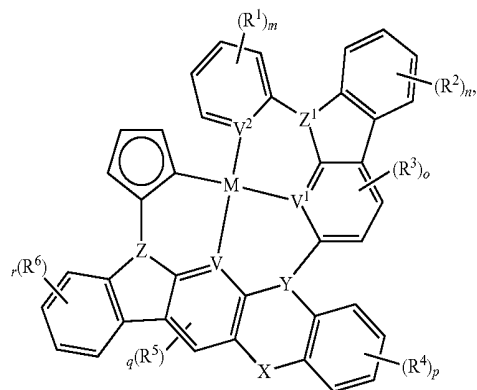
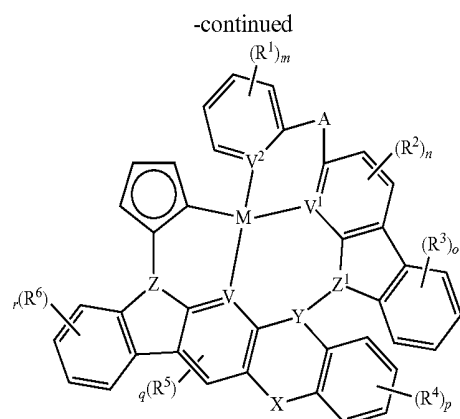
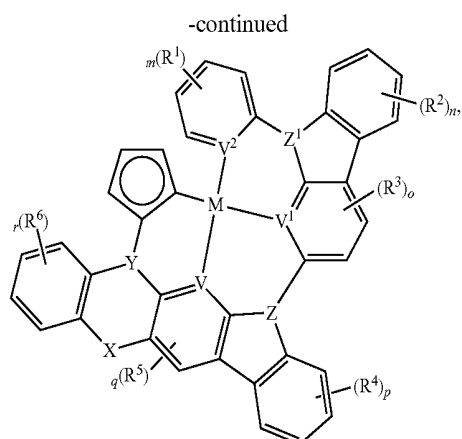


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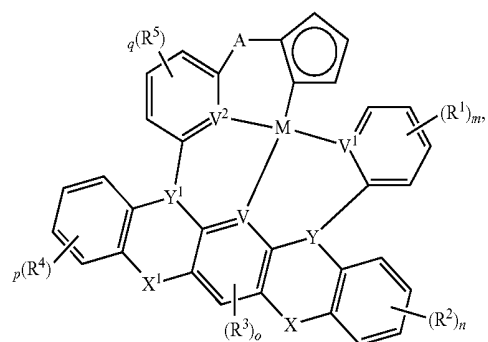
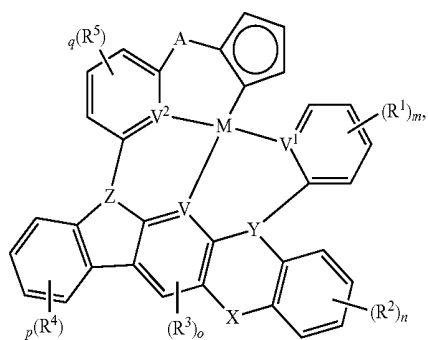
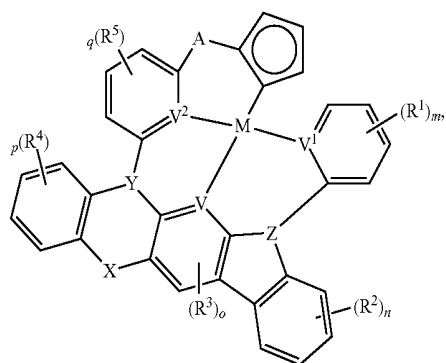
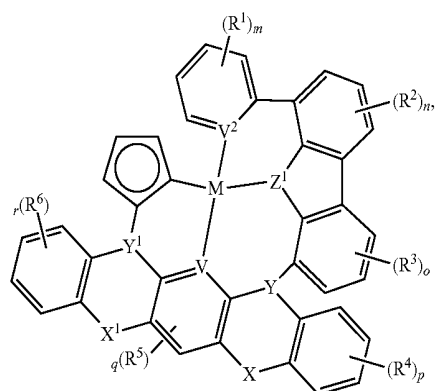


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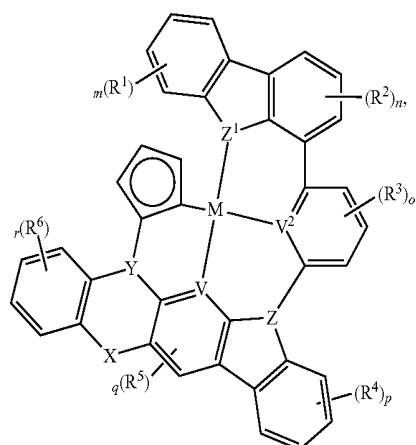
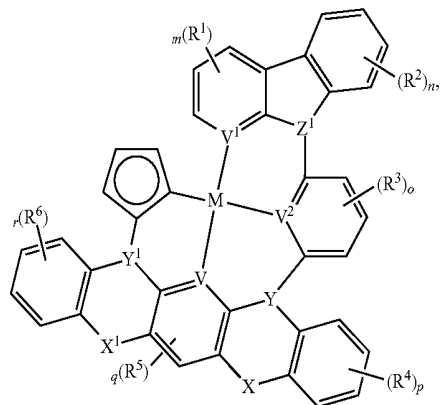
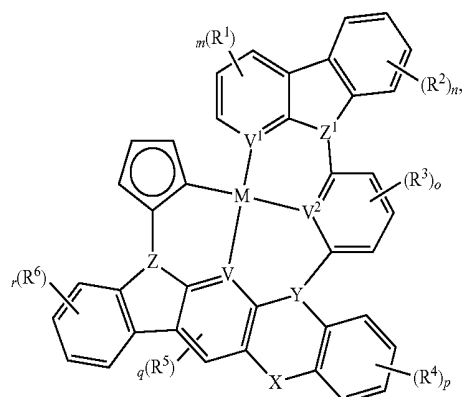
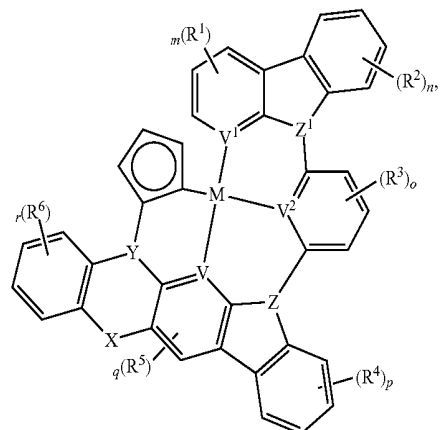


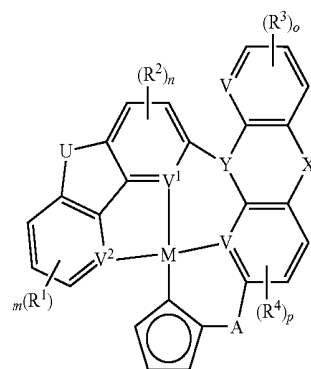
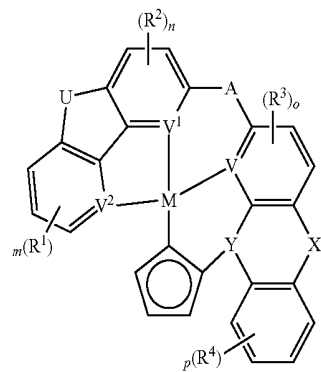
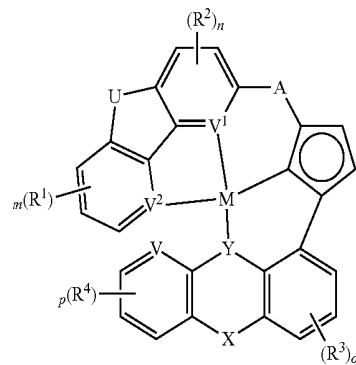
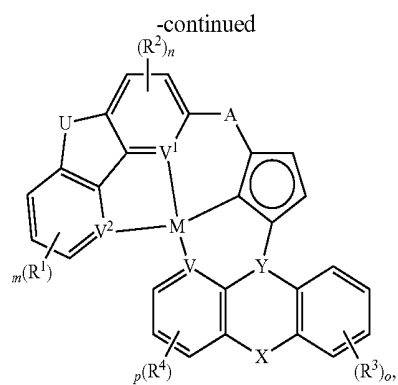
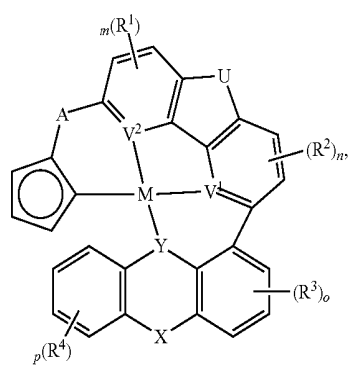
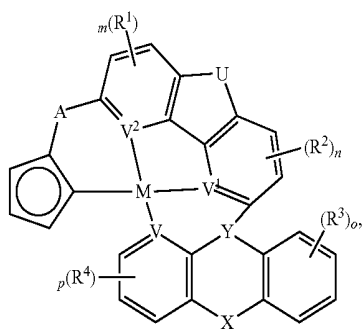
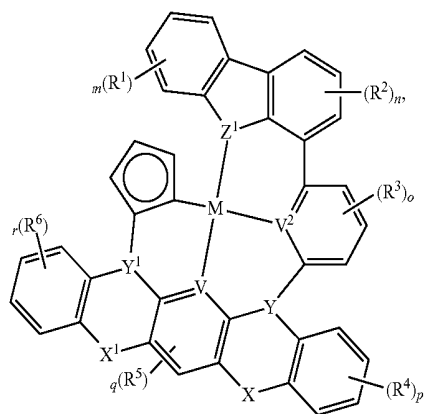
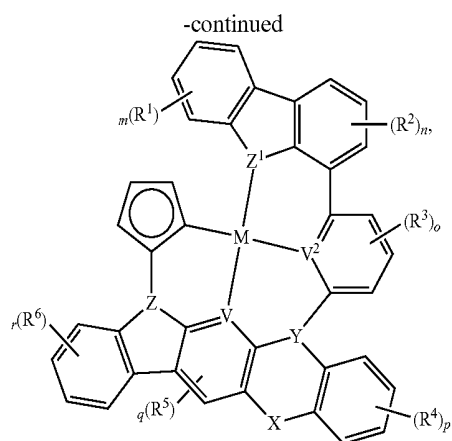


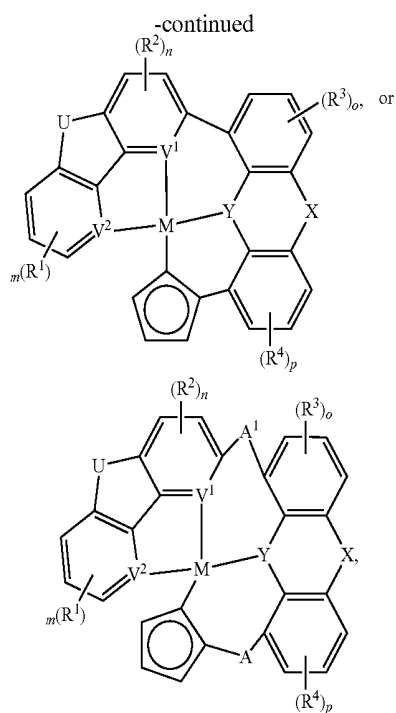
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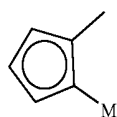




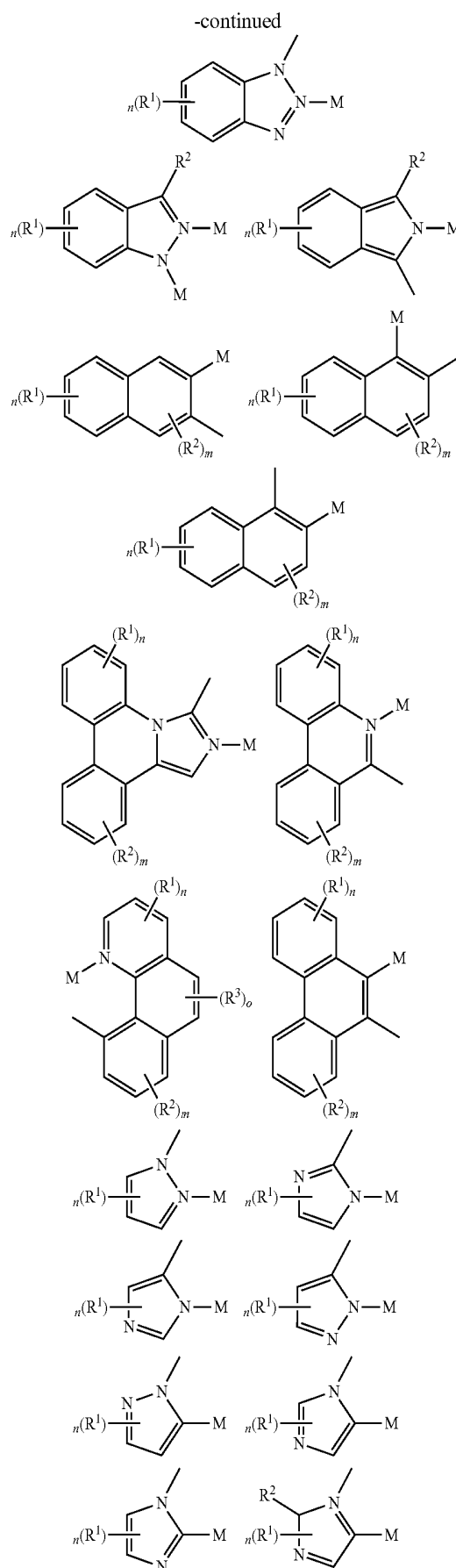
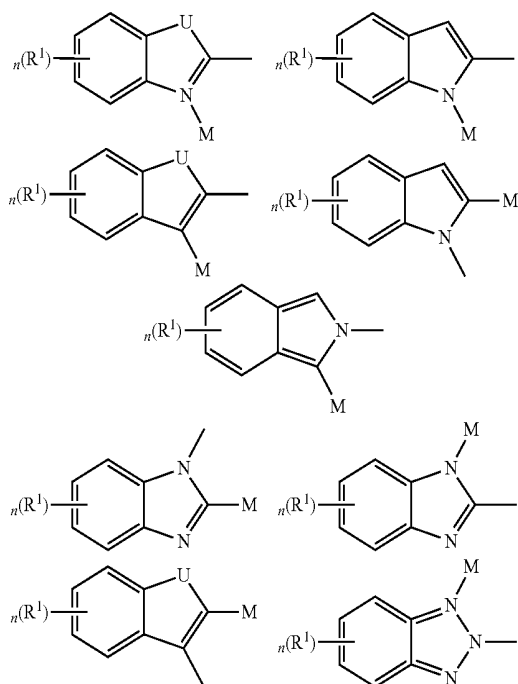


wherein:

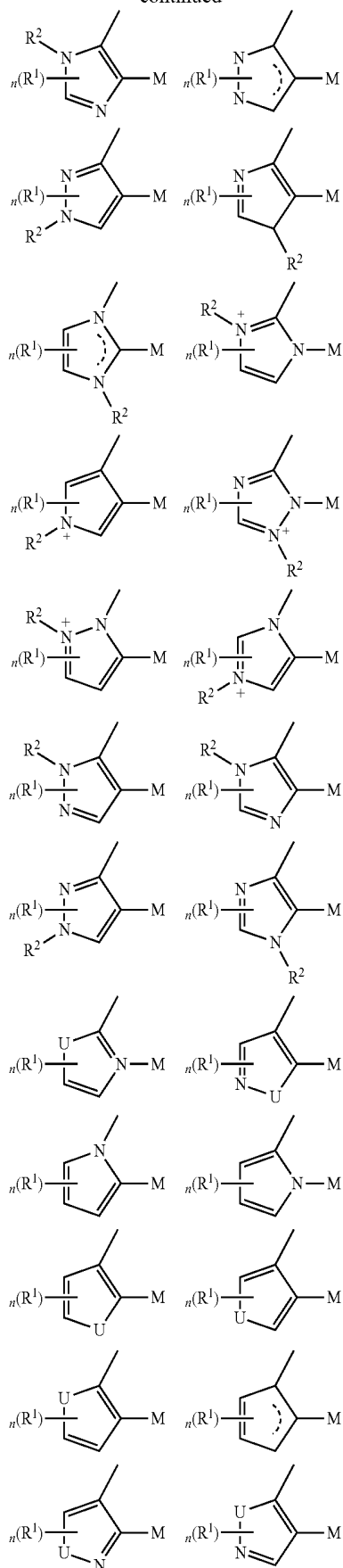
M represents Pt, Pd, Au, Ir, Rh, Ru, Fe, Co, Ni, Cu, Zn, Ag, Hg, Cd, or Zr;



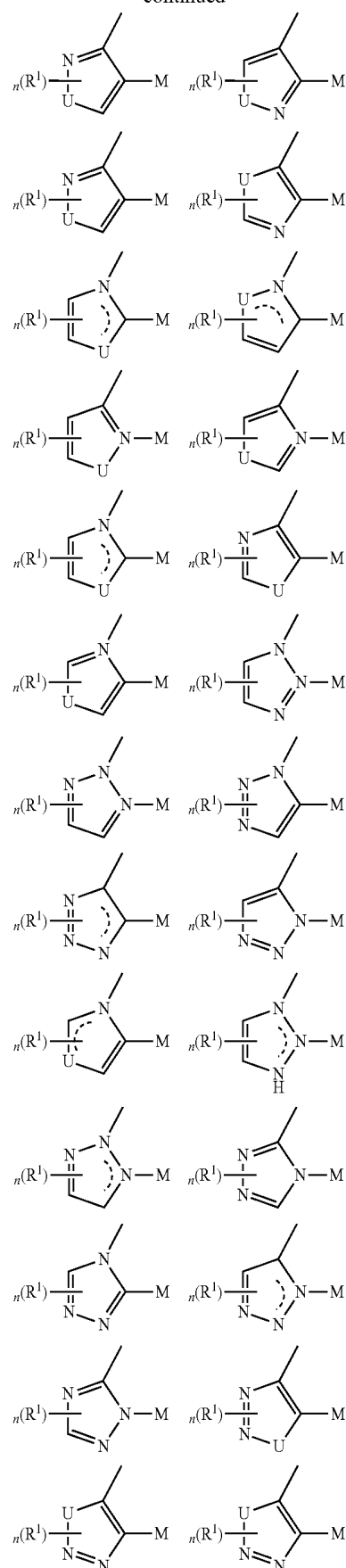
represents one of:



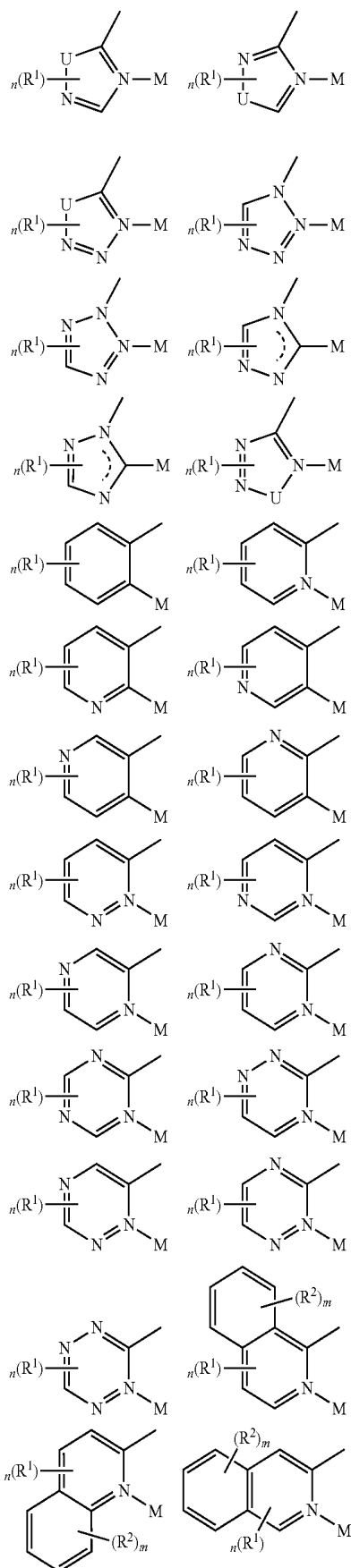
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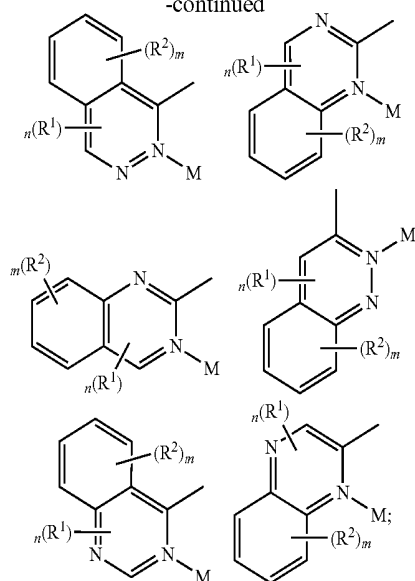
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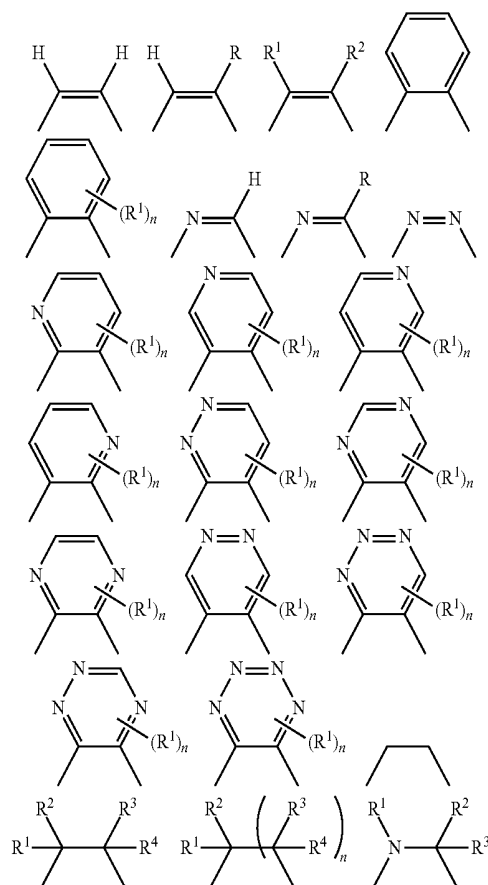


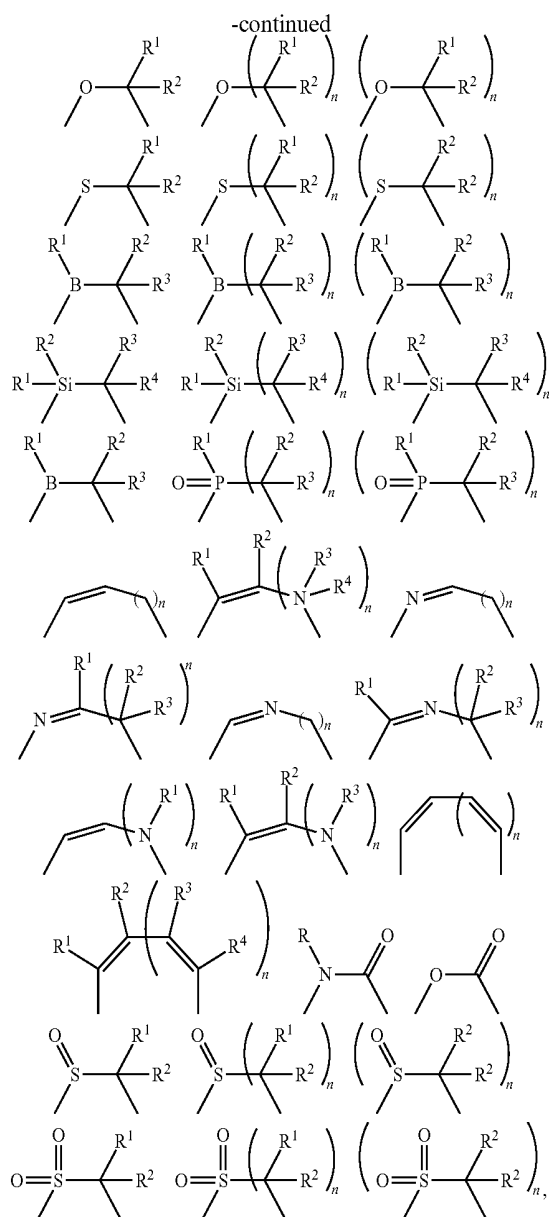
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each of V, V¹, and V², if present, independently represents N, C, CH, P, B, Si, or SiH, valency permitting;

each of A, A¹, X, X¹, Y, Y¹, Z, Z¹, and U, if present, independently represents O, S, S=O, SO₂, Se, N, NR³, P, PR³, R¹P=O, C, CR¹R², C=O, SiR¹R², GeR¹R², B, BH, P(O)H, PH, NH, CR¹H, CH, CH₂, Si, SiH, SiH₂, SiHR¹, BH, or BR³, valency permitting, or any one of

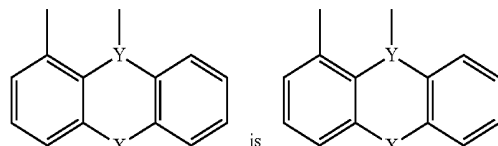




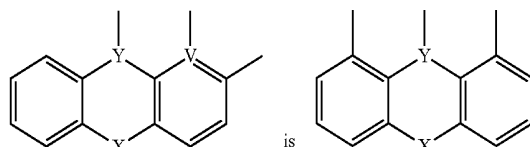
each of R , R^1 , R^2 , R^3 , R^4 , R^5 , and R^6 , if present, independently represents deuterium, halogen, hydroxyl, thiol, nitro, cyano, nitrile, isonitrile, sulfinyl, mercapto, sulfo, carboxyl, hydrazino; substituted or unsubstituted: aryl, cycloalkyl, cycloalkenyl, heterocyclyl, heteroaryl, alkyl, alkenyl, alkynyl, amino, monoalkylamino, dialkylamino, monoarylamino, diarylamino, alkoxy, aryloxy, haloalkyl, aralkyl, ester,

alkoxycarbonyl, acylamino, alkoxycarbonylamino, aryloxycarbonylamino, sulfonylamino, sulfamoyl, carbamoyl, alkylthio, ureido, phosphoramidate, silyl, polymeric; or any conjugate or combination thereof; and each of m , n , o , p , q , and r is independently is an integer of 0 to 4, valency permitting.

24. The compound of claim 23 wherein



25. The compound of claim 23, wherein



26. The compound of claim 23, wherein at least one of V , V^1 , and V^2 , if present, represents N.

27. The compound of claim 23, wherein the compound comprises at least one phenyl group and at least one pyridinyl group.

28. The compound of claim 23, wherein M represents Pt or Pd.

29. The compound of claim 28, wherein M represents Pt.

30. A device comprising the compound of claim 23.

31. The device of claim 30, wherein the device is a full color display.

32. The device of claim 30, wherein the device is an organic light emitting diode.

33. The device of claim 32, wherein the organic light emitting diode is a phosphorescent organic light emitting diode.

34. The device of claim 32, wherein the organic light emitting diode is a fluorescent organic light emitting diode.

35. A photovoltaic device comprising the compound of claim 23.

36. A luminescent display device comprising the compound of claim 23.

37. A light emitting device comprising the compound of claim 23.

* * * * *

专利名称(译)	金属化合物，方法和用途		
公开(公告)号	US20180226592A1	公开(公告)日	2018-08-09
申请号	US15/882358	申请日	2018-01-29
申请(专利权)人(译)	加州大学董事会代表亚利桑那州立大学亚利桑那板		
当前申请(专利权)人(译)	加州大学董事会代表亚利桑那州立大学亚利桑那板		
[标]发明人	LI JIAN LI GUIJIE		
发明人	LI, JIAN LI, GUIJIE		
IPC分类号	H01L51/00 C09K11/06 C07F15/00 H01L51/42 H01L51/50		
CPC分类号	H01L51/5036 H01L51/0087 C09K2211/1007 C09K2211/1029 C09K2211/1033 C09K2211/1037 C09K2211/1044 C09K2211/1059 H01L51/42 H01L51/5016 H01L51/5012 Y02E10/549 C07F15/0086 C09K2211/185 C09K11/06		
优先权	14/430454 2015-03-23 US PCT/US2013/061353 2013-09-24 WO 61/704880 2012-09-24 US		
其他公开文献	US10622571		
外部链接	Espacenet USPTO		

摘要(译)

本文公开了可用于器件的金属化合物，例如OLED。

